Supporting Information

A Dual-sensing DNA Nanostructure with an Ultra-broad Detection Range

Byunghwa Kang¹, Soyeon V. Park¹, Hyongsok Tom Soh^{2*}, Seung Soo Oh^{1,3**}

 ¹Department of Materials Science and Engineering, Pohang University of Science and Technology (POSTECH), 77 Cheongam-Ro, Nam-Gu, Pohang, Gyeongbuk 37673, South Korea
 ²Department of Electrical Engineering and Department of Radiology, Canary Center at Stanford University, 3155 Porter Drive, Stanford, CA 94305 (USA)
 ³School of Interdisciplinary Bioscience and Bioengineering, Pohang University of Science and Technology (POSTECH), 77 Cheongam-Ro, Nam-Gu, Pohang, Gyeongbuk 37673, South Korea
 * E-mail address: tsoh@stanford.edu
 ** E-mail address: seungsoo@postech.ac.kr

Table of Contents

- 1. Materials and Methods
 - 1.1 Reagents and Materials
 - 1.2 Detection range measurement of the DMRD
 - 1.3 Tuning target detection range of the DDM
 - 1.4 Tuning target detection range of the EDM

2. Supplementary Figures and Tables

Table S1: DNA sequences

Table S2: ATP detection range of the DDM

Table S3: hybridization energies of antisense DNAs and EC₅₀ value of the DDM

Figure S1: Arrhenius plot

Figure S2: Fitting graph for obtaining K_{d} value of the ATP aptamer

- Figure S3: Antisense-free target specificity of the DDM
- Table S4: ATP detection range of the EDM
- Figure S4: Tris-free target specificity of the EDM
- Table S5: Detection of ATP targets across five orders of magnitude concentration range

1. Materials and Methods

1.1 Reagents and Materials

DMRD and antisense DNAs AD11–18 were synthesized and purified by either Integrated DNA Technologies (Coralville, IA) or BIONEER (Daejeon, Korea). Adenosine triphosphate (ATP), Tris(hydroxymethyl)aminomethane (Tris), Tris hydrochloride, sodium chloride (NaCl), magnesium chloride (MgCl₂), potassium chloride (KCl), Triton X-100, HEPES, and hydrogen peroxide (H₂O₂) were purchased from GA Biochem (Chuncheon, Korea). Adenosine diphosphate (ADP), adenosine monophosphate (AMP), adenosine, dimethyl sulfoxide (DMSO), sodium hydroxide (NaOH) and hemin were purchased from Sigma-Aldrich (St. Louis, MO). Guanosine triphosphate (GTP), cytosine triphosphate (CTP) and uracil triphosphate (UTP) were purchased from Tech and Innovation (Chuncheon, Korea). 2,2'-Azino-bis(3ethylbenzothiazoline-6-sulphonic acid) (ABTS) was purchased from Roche (Basel, Switzerland).

1.2 Detection range measurement of the DMRD

We measured the affinity of the DDM for ATP via microscale thermophoresis (2bind GmbH; Regensburg, Germany). A titration series from 21.4 nM to 350 μ M ATP was prepared in 1X ATP Binding Buffer (ATPBB; 20 mM Tris-HCl, 300 mM NaCl, and 5 mM MgCl₂, pH 7.5), with the same concentration of DMRD (2 nM) added to each. The binding behavior of each sample was analyzed in the Monolith NT.115 pico device (NanoTemper Technologies GmbH; Munich, Germany). This experiment was performed in duplicate. In parallel, we also conducted FAM fluorescence visualization in 1.5 ml tubes containing 100 μ l 1X ATPBB with 5 μ M DMRD, in which we compared ATP-induced fluorescence decrease with 1 mM ATP or a negative control without ATP. Fluorescence signals were obtained using the Azure c600 (Azure Biosystems; Dublin, CA) at room temperature.

To confirm the ATP concentration range that can be detected by the EDM, we performed an ATP-assisted peroxidation test. Solutions containing 100 nM DMRD and 5 mM HEPES (pH 7.5) were prepared in 1X ATP Reaction Buffer (ATPRB; 10 mM

KCl, 200 mM NaCl, and 0.002% (v/v) Triton X-100). To this solution, we added 5 mM H_2O_2 and 500 µM ABTS, and then incubated at 25 °C for 10 min. These reaction mixtures were challenged with varying concentrations of ATP (0–5 mM) and then transferred to a flat-bottomed transparent 96-well microplate (Greiner; Kremsmünster, Austria), after which 200 nM hemin was added to each well (final reaction volume: 50 µl). The reactions were incubated at 25 °C for 1 h. Absorbance increases at 418 nm were measured with the Spark 10M microplate reader, with experiments performed in triplicate. In parallel, we performed naked eye visualization of ABTS⁺-associated color change (green) in 1.5 ml tubes, in which we compared ATP-assisted peroxidation in a 100 µl sample with 6 mM ATP versus a negative control without ATP, with results recorded by the digital camera of a Samsung Galaxy S9+ smartphone.

1.3 Tuning target detection range of the DDM

We prepared eight samples of 100 nM DMRD in 1X ATPBB, and then added 150 nM of an antisense DNA (AD11–18) to each reaction (final volume: 100 μ I). Samples were then heated to 95 °C for 10 min followed by ice-quenching for 10 min, and finally 10 min at room temperature. All samples were challenged with varying ATP concentrations (0–10,000 μ M) and incubated for 30 min at 25 °C, and then transferred to a flat-bottomed black 96-well microplate (Greiner; Kremsmünster, Austria). Fluorescence signals were monitored by Spark 10M microplate reader with the following settings: excitation = 485 nm, emission = 535 nm, integration time = 40 μ s. This experiment was performed in triplicate.

We tested DDM selectivity in the presence of AD11, with the reaction conditions and heat-treatment described above. The concentration of ATP, ADP, AMP, adenosine, GTP, CTP, or UTP was 1 mM. Fluorescence signals were monitored by Spark 10M microplate reader, with experiments performed in quadruplicate.

1.4 Tuning target detection range of the EDM

We prepared multiple sets of eight experimental samples (volume: 50 μ l) containing 100 nM DMRD in 1X ATPRB with varying Tris concentrations (20 mM–2.1 M). After

incubation at 25 °C for 10 min, each set of samples was challenged with varying ATP concentrations (1–720 mM), after which 200 nM hemin was added. The reaction was incubated at room temperature for 1 h. The increase of absorbance at 418 nm was measured by Spark 10M microplate reader, with experiments performed in triplicate.

We tested EDM selectivity in 20 mM Tris reaction conditions as described above, where each ATP analogue was present at 3 mM. Absorbance was monitored at 418 nm by Spark 10M microplate reader, with experiments performed in triplicate.

2. Supplementary Figures and Tables

 Table S1. Sequences of DNAs used in this work. Blue sequences show components of the DDM;

 magenta sequences show the EDM.

Name	Sequence $(5' \rightarrow 3')$
	FAM-AAT TCT GGG GGA GCC TTT TGT GGG TAG
DMRD	GGC GGG TTG GTT TTG CCC CGG AGG AGG AAT
	T-BHQ1
AD11	AAC CCG CCC TA
AD12	AAC CCG CCC TAC
AD13	CAA CCC GCC CTA C
AD14	CAA CCC GCC CTA CC
AD15	CAA CCC GCC CTA CCC
AD16	CAA CCC GCC CTA CCC A
AD17	CAA CCC GCC CTA CCC AC
AD18	CCA ACC CGC CCT ACC CAC

Antisense strand (# reflects length in nt)	Detection range (µM)
AD11	33.9 ~ 1,782
AD12	29.7 ~ 1,514
AD13	57.2 ~ 2,998
AD14	132.8 ~ 5,189
AD15	160.9 ~ 5,678
AD16	180.3 ~ 5,968
AD17	389.3 ~ 7,663
AD18	346.8 ~ 7,440

Table S2. ATP detection range of the antisense-tuned DDM.

Table S3. The hybridization energies between antisense DNAs (AD11 to AD18) and the DMRD, and EC_{50} values in terms of ATP-induced structure-switching. The hybridization energies (ΔG , kcal/mole) between the antisense DNAs and the DMRD were calculated with the OligoAnalyzer Tool (Version 3.1) from Integrated DNA Technologies (IDT; Coralville, IA). Temperature was fixed at 298.15 K. EC_{50} values of our DDM increased depending on the length of antisense DNAs.

Name	Hybridization Energy (∆G _{hybridization}) (kcal/mole)	EC ₅₀ (μΜ)
AD11	-24.87	238.23
AD12	-26.22	205.79
AD13	-28.17	494.17
AD14	-31.24	1,346.40
AD15	-34.31	1,694.10
AD16	-36.26	1,984.21
AD17	-37.6	4,827.45
AD18	-40.67	4,768.02

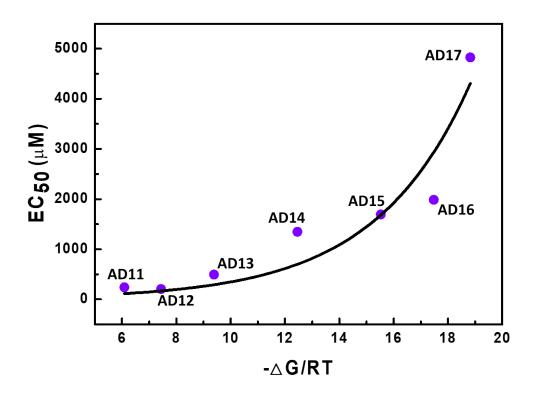


Figure S1. The increase of the EC₅₀ value for antisense strands of increasing length relative to the Gibbs free energy changes ($-\Delta G/RT = (\Delta G_{hybridization} - \Delta G_{mfold})/RT$) at room temperature (298.15 K). Regression, *y*: 19.98*exponential(0.285**x*); r² = 0.873. $\Delta G_{hybridization}$ is the hybridization energy of antisense DNAs with the DMRD, and ΔG_{mfold} is the intrinsic folding energy of our DMRD (obtained by mfold software¹). From extrapolation of this Arrhenius plot, we calculated a K_d value of 19.98 µM for the ATP-binding aptamer, which is quite similar to the experimentally measured value of 23.14 µM.

(1) Juker, M., Mfold web server for nucleic acid folding and hybridization prediction. *Nucleic Acids Res.* **2003**, *31*, 3406-3415.

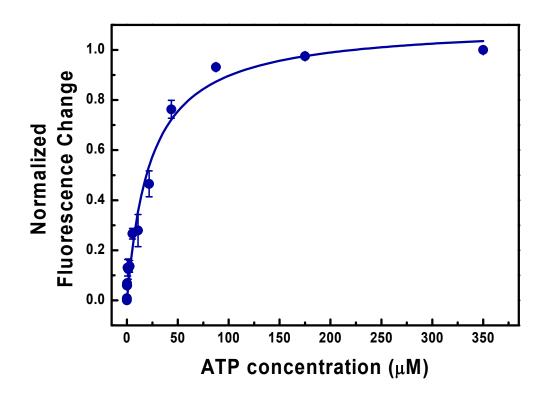


Figure S2. The fitting graph for obtaining the K_d of the ATP-binding aptamer based on microscale thermophoresis. Regression, y: $1.10^*x/(23.1 + x)$; r² = 0.977.

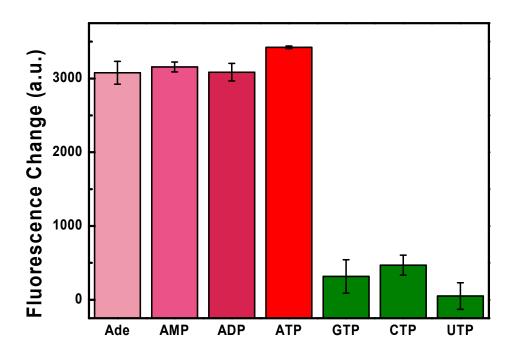


Figure S3. Target specificity of the antisense-free DDM. All ATP analogues, including adenosine (Ade), were present at 1 mM.

Tris concentration (mM)	Detection range (mM)
20	1.36 ~ 4.19
50	3.91 ~ 15.1
100	15.0 ~ 32.9
250	32.6 ~ 66.1
400	51.5 ~ 114
700	88.6 ~ 198
1000	126 ~ 273
1500	213 ~ 443
2100	282 ~ 506

Table S4. ATP detection range of the EDM at varying Tris concentrations.

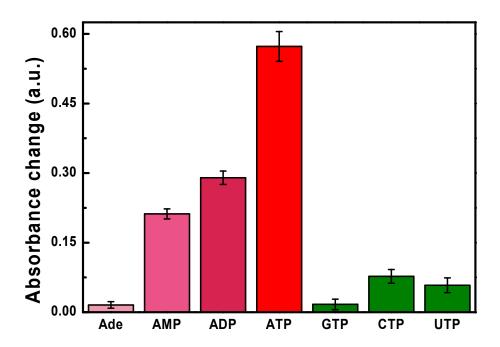


Figure S4. EDM specificity in the absence of Tris. All ATP analogues were present at 3 mM, and the peroxidation reaction occurred for 1 h.

Table S5. Detection of ATP targets across a concentration range spanning five orders of magnitude. Two DMRD assay conditions were sufficient to successfully identify ATP concentrations ranging from 1 μ M to 100 mM. Experimental signals (fluorescence and absorbance) were well-matched to the expected signals obtained from calibration curves of **Figures 2B** and **3B**. The numbers in parentheses are the differences in our experimental data relative to expected values.

[Set 1 – 20 mM Tris without any antisense strands]

	[ATP] (µM)			
_	1	10	100	1,000
Fluorescence signal	0.02	0.18	0.79 (-0.03)	-
Absorbance signal	< 0.01		0.02	0.07 (+0.01)

[Set 2 – AD11 and 400 mM Tris]

[ATP] (µM)

	100	1,000	10,000	100,000
Fluorescence signal	0.28 (-0.01)	0.76 (-0.04)	0.97	-
Absorbance signal		< 0.01		0.72 (+0.01)