

Supplementary Information for

Rational Design of a Fluorescent Sensor to Simultaneously Determine Both the Enantiomeric Composition and Concentration of Chiral Functional Amines

Kaili Wen, Shanshan Yu,* Zeng Huang, Liming Chen, Meng Xiao, Xiaoqi Yu* and Lin Pu*

I. Preparation of Samples for Fluorescence Measurement.

For the interactions with the diamine: A 2 mM stock solution of (*R*)-**4** in CH₂Cl₂ and 2 mM Zn(OAc)₂·2H₂O in CH₃OH were freshly prepared for each measurement. For the fluorescence enhancement study, a sensor solution of (*R*)-**4**+Zn²⁺ (2 equiv) (2 mM) was mixed with various equiv of the diamine solution (15 mM in CH₃OH) in a 10 mL test tube. The resulting solution was allowed to stand at room temperature for 3 h, then was diluted to the desired concentration (0.02 mM) for 0.5 h before measurements. All the fluorescence spectra were taken within 2 h.

For the interactions with the amino alcohols: A 2 mM stock solution of (*R*)-**4** in CH₂Cl₂ and 2 mM Zn(OAc)₂·2H₂O in CH₃OH were freshly prepared for each measurement. For the fluorescence enhancement study, a sensor solution of (*R*)-**4**+Zn²⁺ (2 equiv) (2 mM) was mixed with various equiv of an amino alcohol solution (15 mM in CH₃OH) in a 10 mL test tube. The resulting solution was allowed to stand at room temperature for 3.5 h and be diluted to the desired concentration (0.02 mM) before measurements. All the fluorescence spectra were taken within 2 h.

For the interactions with the amino acids: A 2 mM stock solution of (*R*)-**4** in CH₂Cl₂, 2 mM Zn(OAc)₂·2H₂O in CH₃OH and a solution of an amino acid (15 mM) mixed with n-Bu₄NOH (16.5 mM) in CH₃OH were freshly prepared for each measurement. For the fluorescence enhancement study, a sensor solution of (*R*)-**4**+Zn²⁺ (2 equiv) (2 mM) was mixed with various equiv of the n-Bu₄NOH-amino acid mixed solution in a 10 mL test tube. The resulting solution was allowed to stand at room temperature for 3.5 h and be diluted to the desired concentration (0.02 mM) before measurements. All the fluorescence spectra were taken within 2 h.

II. Supplementary Fluorescence Spectra, UV/vis Absorption Spectra, TOF Mass Spectra and NMR Titration Plots

Time Dependence Fluorescence Responses

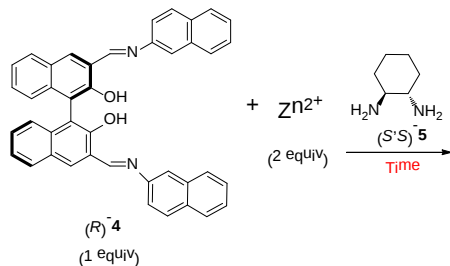


Figure S1. Fluorescence kinetics spectra of (R)-4 (2.0×10^{-3} M) with 25 equiv (S,S)-5 in the presence of 2 equiv Zn(OAc)₂·2H₂O. (Solvent: CH₃OH/2% CH₂Cl₂. λ_{exc} = 370 nm, slit = 5/5 nm.).

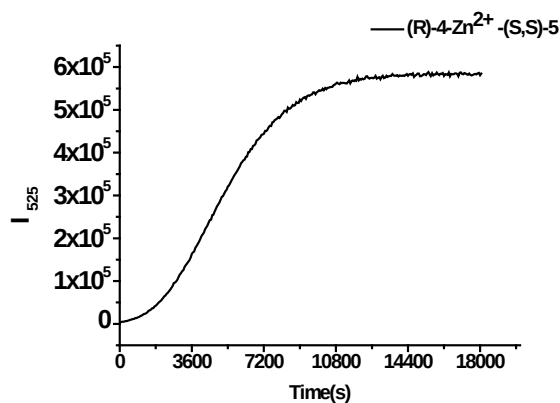


Figure S2. Fluorescence spectra of 2-naphthylamine (4.0×10^{-5} M) without/with the Zn(OAc)₂·2H₂O (4.0×10^{-5} M). (Solvent: CH₃OH/2% CH₂Cl₂. λ_{exc} = 370 nm, slit = 5/5 nm.).

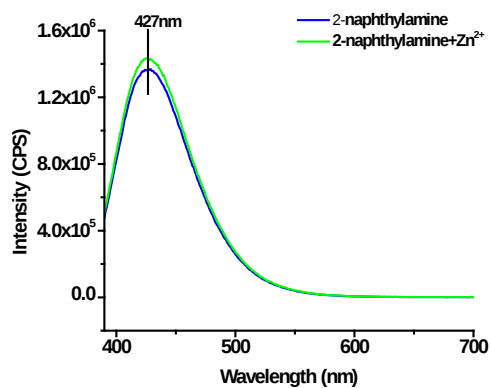


Figure S3. Fluorescent response of (S)-**4** (2.0×10^{-5} M in $\text{CH}_3\text{OH}/2\% \text{CH}_2\text{Cl}_2$) + Zn^{II} (2 equiv) to 5 – 20 eq (*R,R*)-**5** (a) and (*S,S*)-**5** (b). Plots of I_{427} (c), I_{525} (d), I_{525}/I_{500} (e) for (S)-**4** (2.0×10^{-5} M in $\text{CH}_3\text{OH}/2\% \text{CH}_2\text{Cl}_2$) + Zn^{II} (2 equiv) in the presence of varying concentrations of (*S,S*)-**5** and (*R,R*)-**5**. ($\lambda_{\text{exc}} = 370$ nm, slit: 5/5nm)

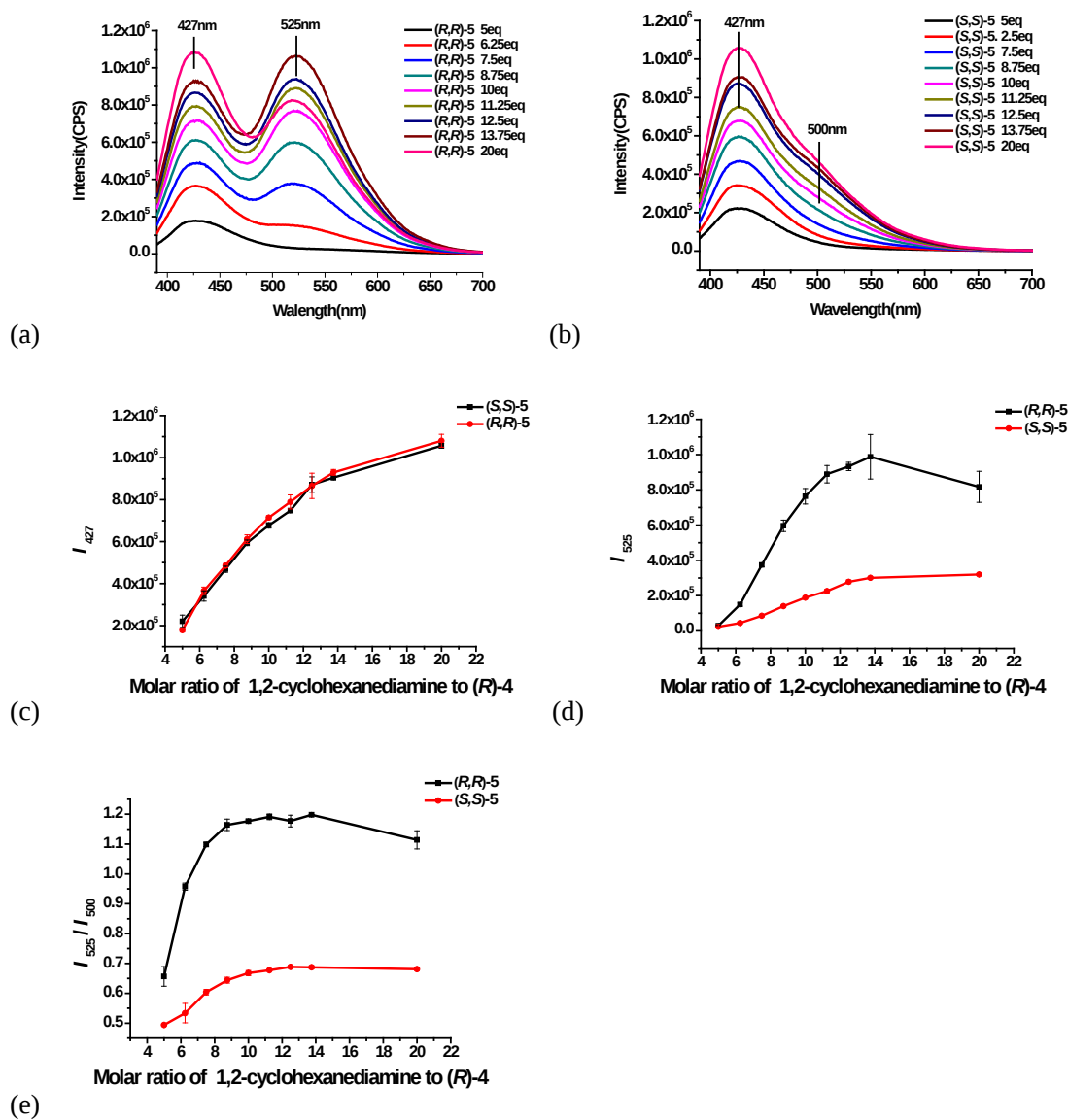


Figure S4. Excitation spectra of (*R*)-**4** (2.0×10^{-5} M CH₃OH/2% CH₂Cl₂) + Zn^{II} (2 equiv) toward 40 equiv *L*-**7** when the emission wavelengths are at 527 and 427 nm respectively.

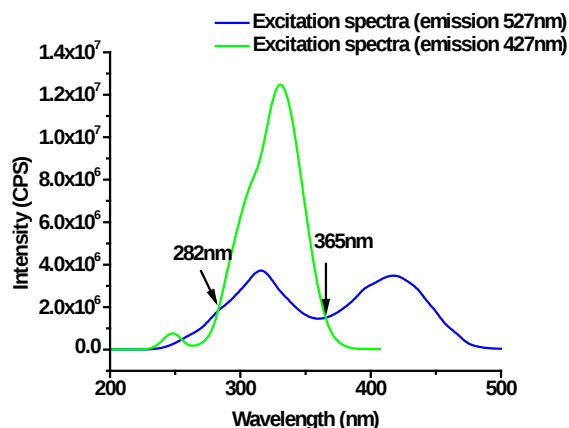
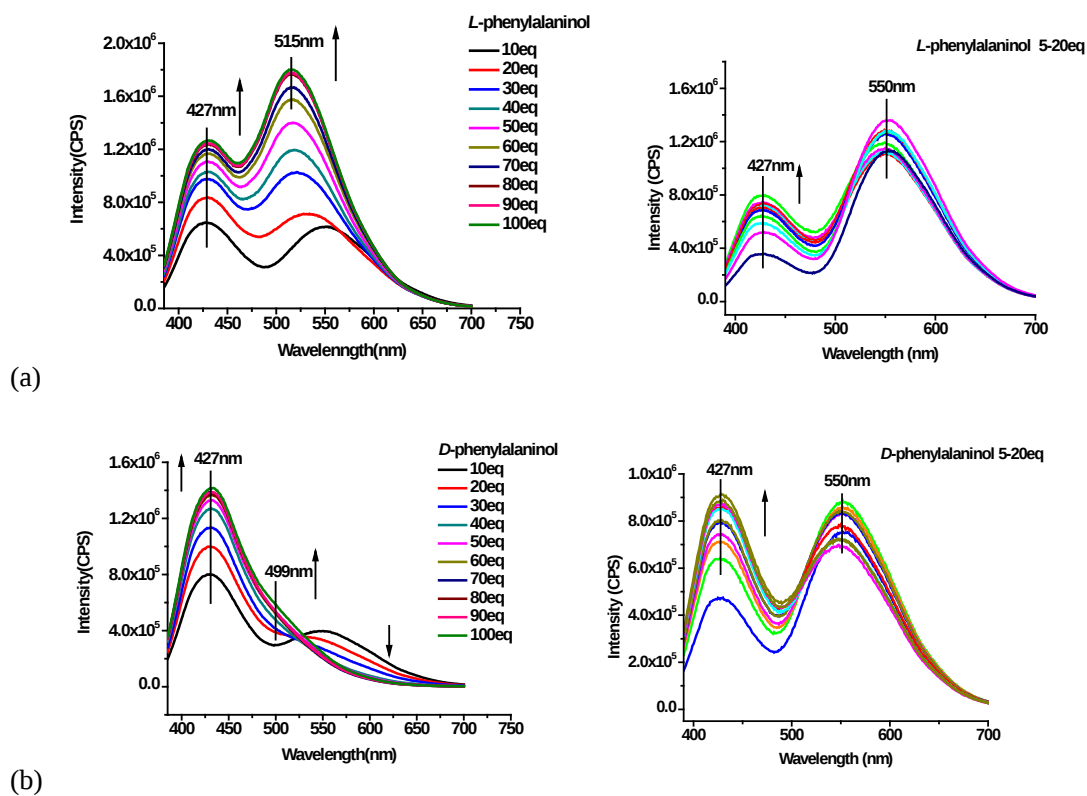


Figure S5. Fluorescence spectra of (*R*)-**4** (2.0×10^{-5} M) + Zn^{II} (2 equiv) to 10, 20, 30, 40, 50, 60, 70, 80, 90 and 100 equiv *L*-**8** (a) and *D*-**8** (b). Plots of I_{427} (c), I_{515} (d), I_{515}/I_{499} (e) for (*R*)-**4** (2.0×10^{-5} M) + Zn^{II} (2 equiv) in the presence of varying concentrations of *L*-**8** and *D*-**8**. (Solvent: CH₃OH/2% CH₂Cl₂. $\lambda_{exc} = 365$ nm, slit = 4/4 nm.).



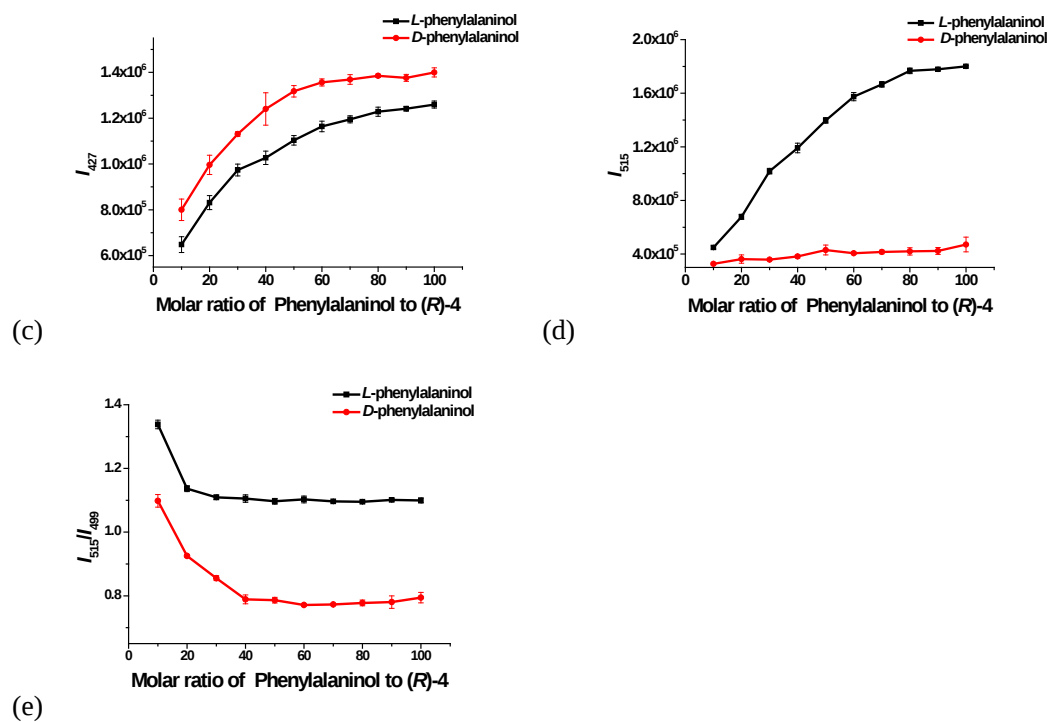


Figure S6. Excitation spectra of (R)-4 (2.0×10^{-5} M $\text{CH}_3\text{OH}/2\% \text{CH}_2\text{Cl}_2$) + Zn^{II} (2 equiv) toward 40 equiv L-9 when the emission wavelengths are at 521 and 427 nm respectively.

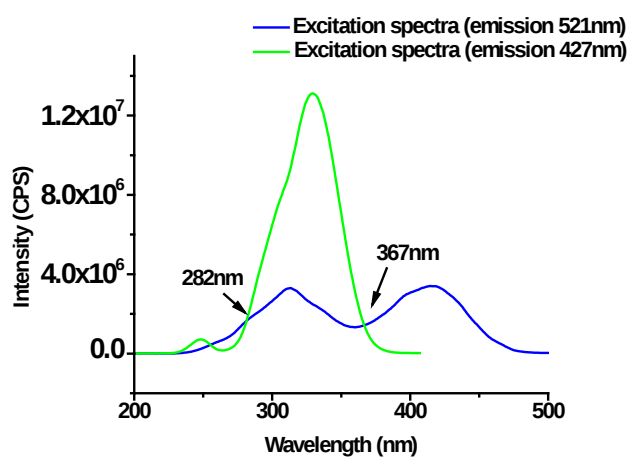


Figure S7. Fluorescence spectra of (*R*)-**4** (2.0×10^{-5} M) + Zn^{II} (2 equiv) toward 10, 20, 30, 40, 50, 60, 70, 80, 90 and 100 equiv *L*-**9** (a) and *D*-**9** (b). Plots of I_{427} (c), I_{521} (d), I_{521}/I_{501} (e) for (*R*)-**4** (2.0×10^{-5} M) + Zn^{II} (2 equiv) in the presence of varying concentrations of *L*-**9** and *D*-**9**. (Solvent: $\text{CH}_3\text{OH}/2\% \text{CH}_2\text{Cl}_2$. $\lambda_{\text{exc}} = 367$ nm, slit = 4/4 nm.).

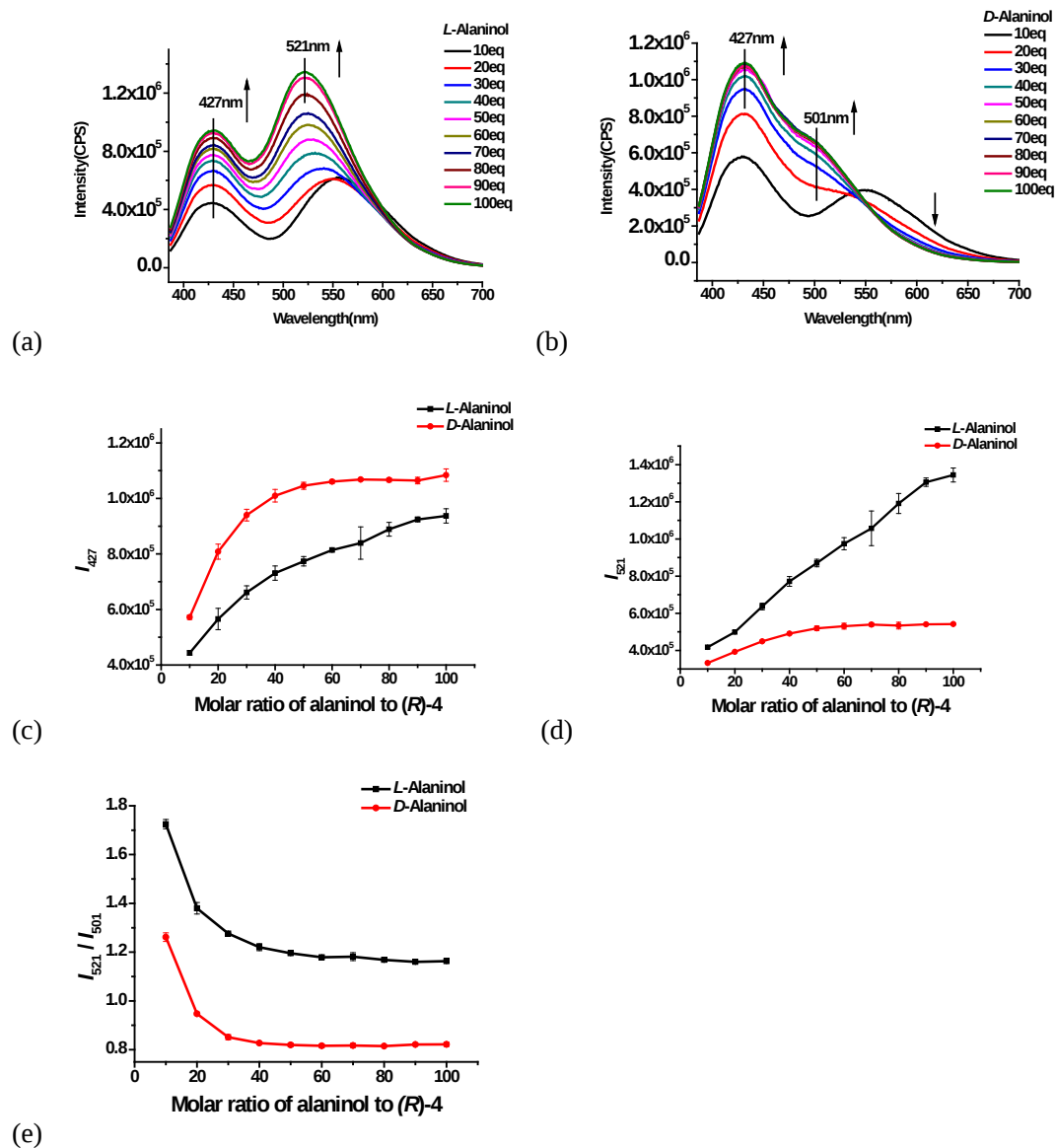


Figure S8. Fluorescence spectra of (*R*)-**4** (2.0×10^{-5} M) + Zn^{II} (2 equiv) toward 16, 20, 24, 28, 32, 40, 48, 56, 64 and 72 μL of *L*-**14** (15.0×10^{-3} M)+*n*-Bu₄NOH (16.5×10^{-3} M) (a) and of *D*-**14** (15.0×10^{-3} M)+*n*-Bu₄NOH (16.5×10^{-3} M) (b). Plots of I_{427} (c), I_{518}/I_{507} (d) for (*R*)-**4** (2.0×10^{-5} M) + Zn^{II} (2 equiv) in the presence of varying concentrations of *L*-**14**+*n*-Bu₄NOH and *D*-**14**+*n*-Bu₄NOH. (Solvent: CH₃OH/2% CH₂Cl₂. λ_{exc} = 368 nm, slits = 5/5 nm.).

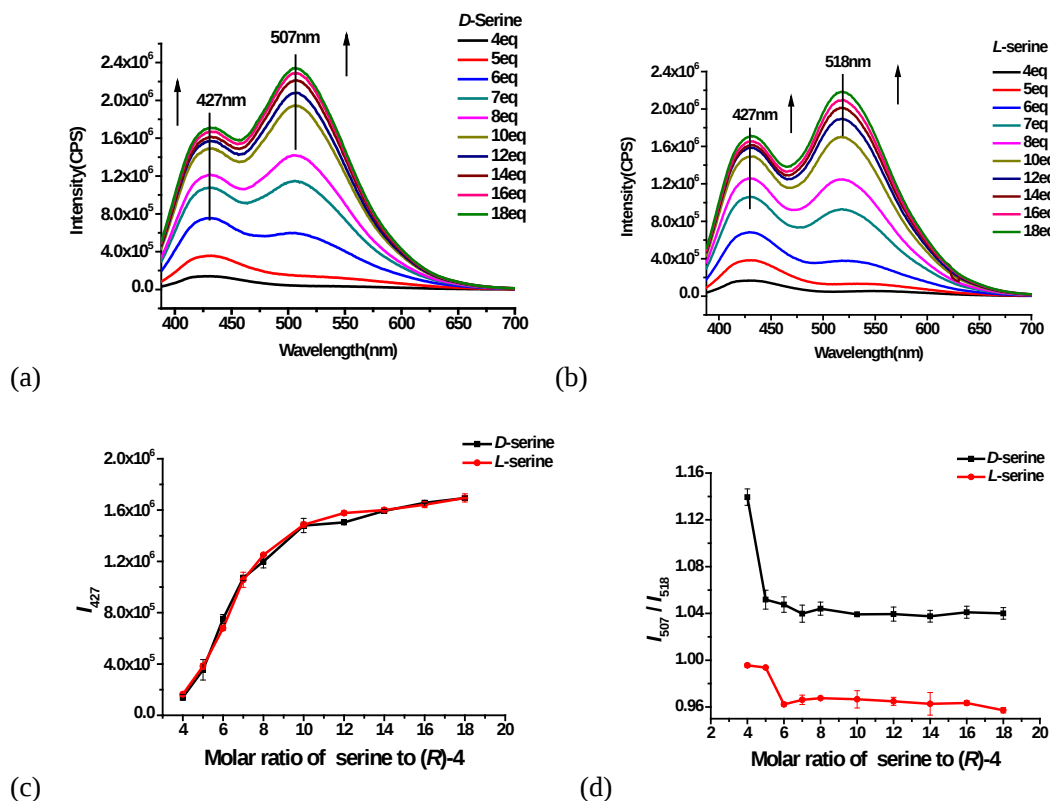
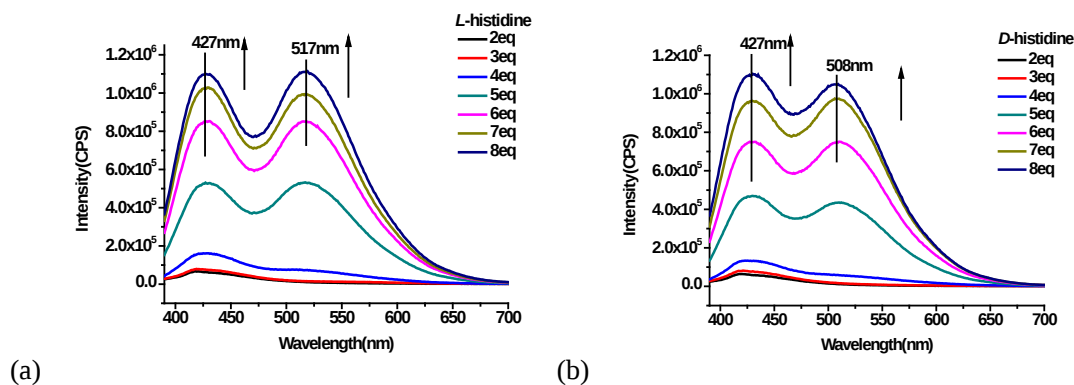
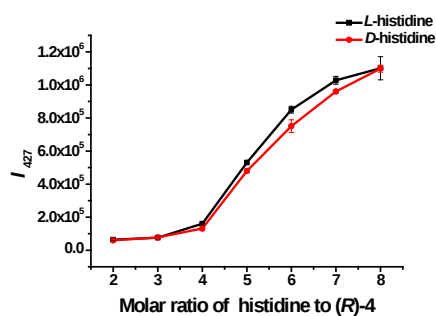
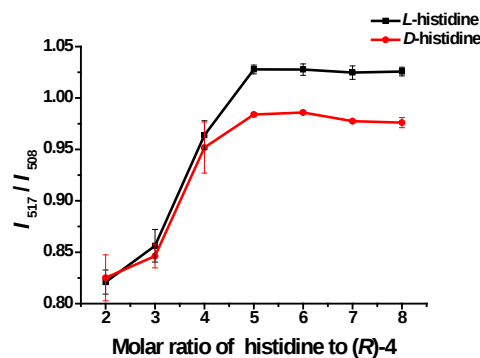


Figure S9. Fluorescence spectra of (*R*)-**4** (2.0×10^{-5} M) + Zn^{II} (2 equiv) toward 8, 12, 16, 20, 24, 28 and 32 μL of *L*-**13** (15×10^{-3} M)+*n*-Bu₄NOH (16.5×10^{-3} M) (a) and of *D*-**13** (15×10^{-3} M)+*n*-Bu₄NOH (16.5×10^{-3} M) (b). Plots of I_{427} (c), I_{517}/I_{508} (d) for (*R*)-**4** (2.0×10^{-5} M) + Zn^{II} (2 equiv) in the presence of varying concentrations of *L*-**13**+*n*-Bu₄NOH and *D*-**13**+*n*-Bu₄NOH. (Solvent: CH₃OH/2% CH₂Cl₂. λ_{exc} = 370 nm, slit = 5/5 nm.).



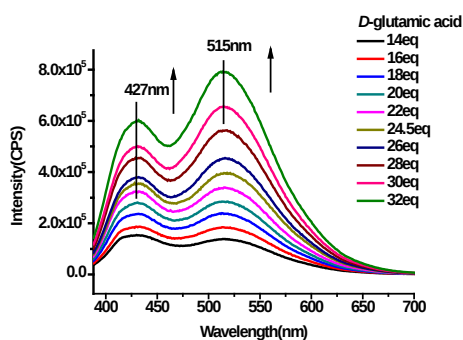


(c)

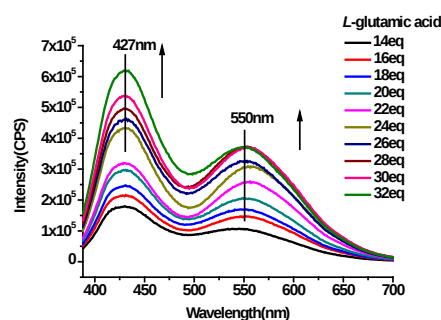


(d)

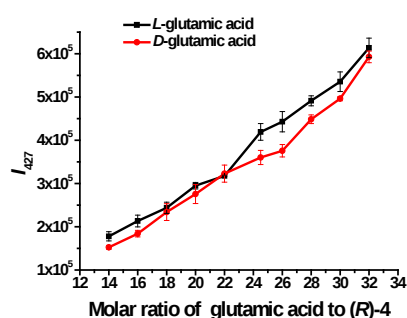
Figure S10. Fluorescence spectra of (R)-4 (2.0×10^{-5} M) + Zn^{II} (2 equiv) toward 56, 64, 72, 80, 88, 98, 104, 112, 120 and 128 μL of D-12 (15×10^{-3} M) + n-Bu₄NOH (16.5×10^{-3} M) (a) and of L-12 (15×10^{-3} M) + n-Bu₄NOH (16.5×10^{-3} M) (b). Plots of I_{427} (c), I_{515} (d), I_{515}/I_{550} (e) for (R)-4 (2.0×10^{-5} M) + Zn^{II} (2 equiv) in the presence of varying concentrations of D-12 (15×10^{-3} M) + n-Bu₄NOH (16.5×10^{-3} M) and L-12 (15×10^{-3} M) + n-Bu₄NOH (16.5×10^{-3} M). (Solvent: $\text{CH}_3\text{OH}/2\% \text{CH}_2\text{Cl}_2$, $\lambda_{\text{exc}} = 368$ nm, slit = 5/5 nm.).



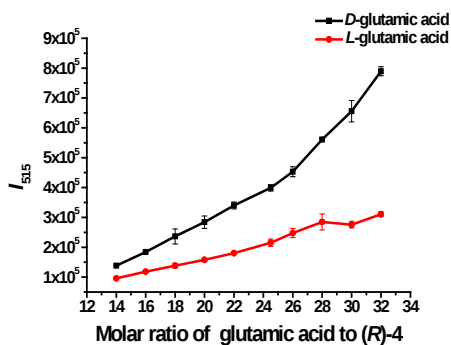
(a)



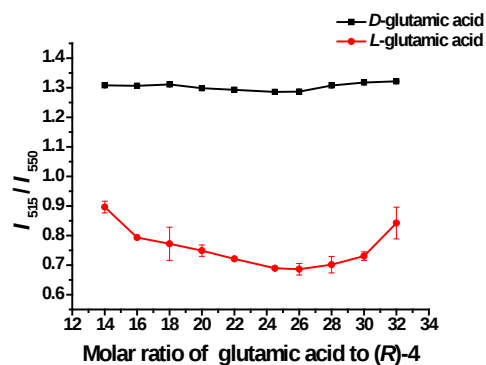
(b)



(c)

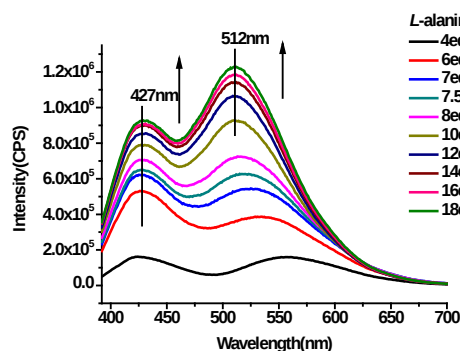


(d)

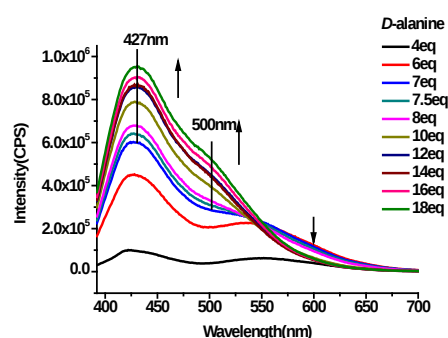


(d)

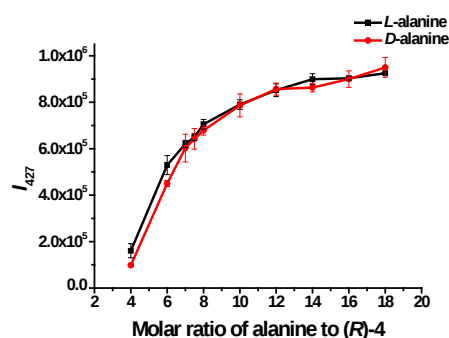
Figure S11. Fluorescence spectra of *(R)*-4 (2.0×10^{-5} M) + Zn^{II} (2 equiv) toward 16, 24, 28, 30, 32, 40, 48, 56, 64 and 72 μL of *D*-11 (15×10^{-3} M) + *n*-Bu₄NOH (16.5×10^{-3} M) (a) and of *L*-11 (15×10^{-3} M) + *n*-Bu₄NOH (16.5×10^{-3} M) (b). Plots of I_{427} (c), I_{512} (d), I_{512}/I_{550} (e) for *(R)*-4 (2×10^{-5} M) + Zn^{II} (2 equiv) in the presence of varying concentrations of *D*-11 (15×10^{-3} M) + *n*-Bu₄NOH (16.5×10^{-3} M) and *L*-11 (15×10^{-3} M) + *n*-Bu₄NOH (16.5×10^{-3} M). (Solvent: $\text{CH}_3\text{OH}/2\% \text{CH}_2\text{Cl}_2$. $\lambda_{\text{exc}} = 372$ nm, slit = 5/5 nm.).



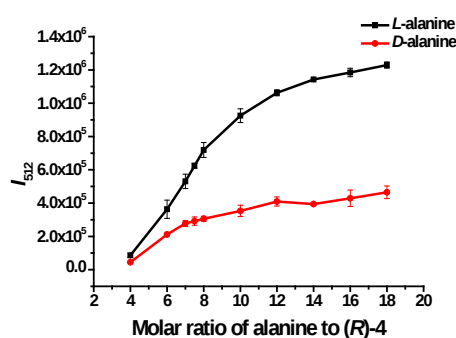
(a)



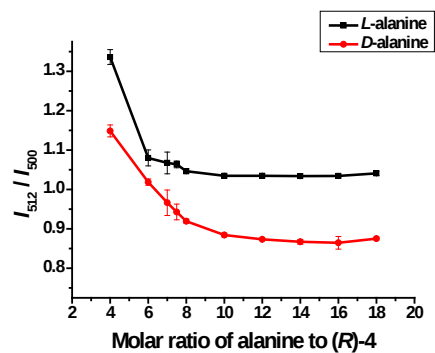
(b)



(c)

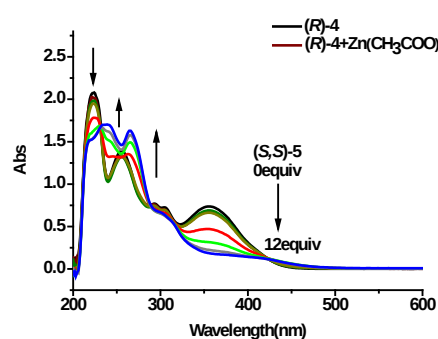


(d)

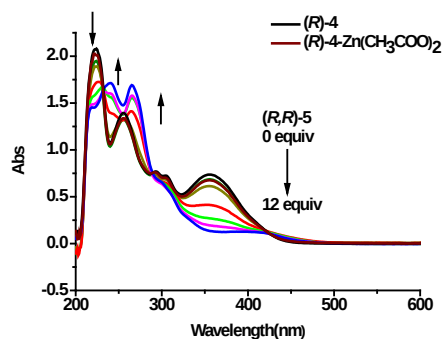


(e)

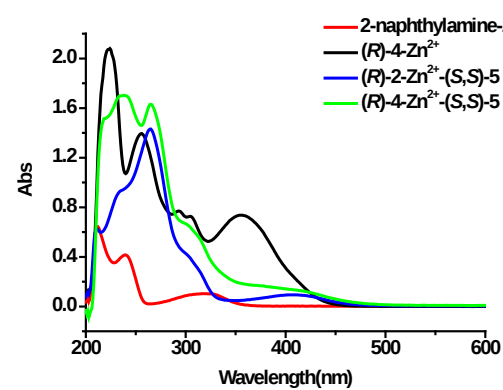
Figure S12. UV/ vis absorption titration of (R)-4 with 2, 4, 6, 8, 10 and 12 equiv (S,S)-5 (a) and (R,R)-5 (b) in the presence of Zn^{II} . (c) UV/vis absorption spectra of (R)-2 and (R)-4 with (S,S)-5 in the presence of Zn^{II} .



(a)



(b)



(c)

Figure S13. TOF mass spectra of (*R*)-**4** + Zn(OAc)₂·2H₂O (2 equiv) + (*S,S*)-**5** (7 equiv).

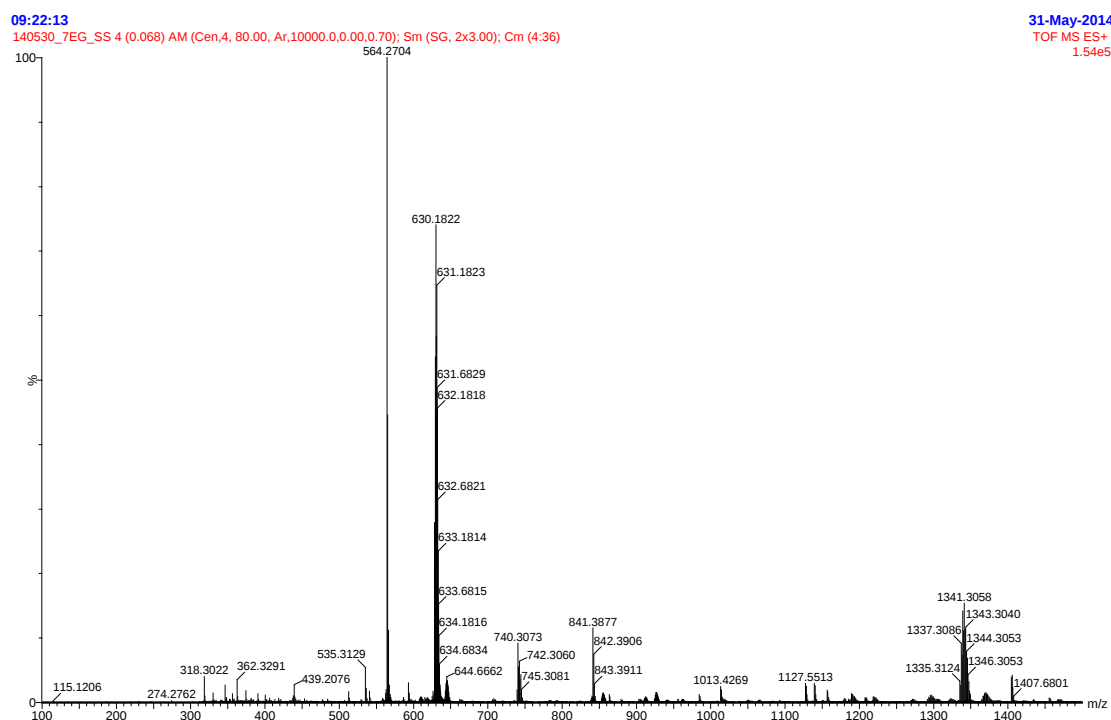


Figure S14. TOF mass spectra of (*R*)-**4** + Zn(OAc)₂·2H₂O (2 equiv) + (*R,R*)-**5** (7 equiv).

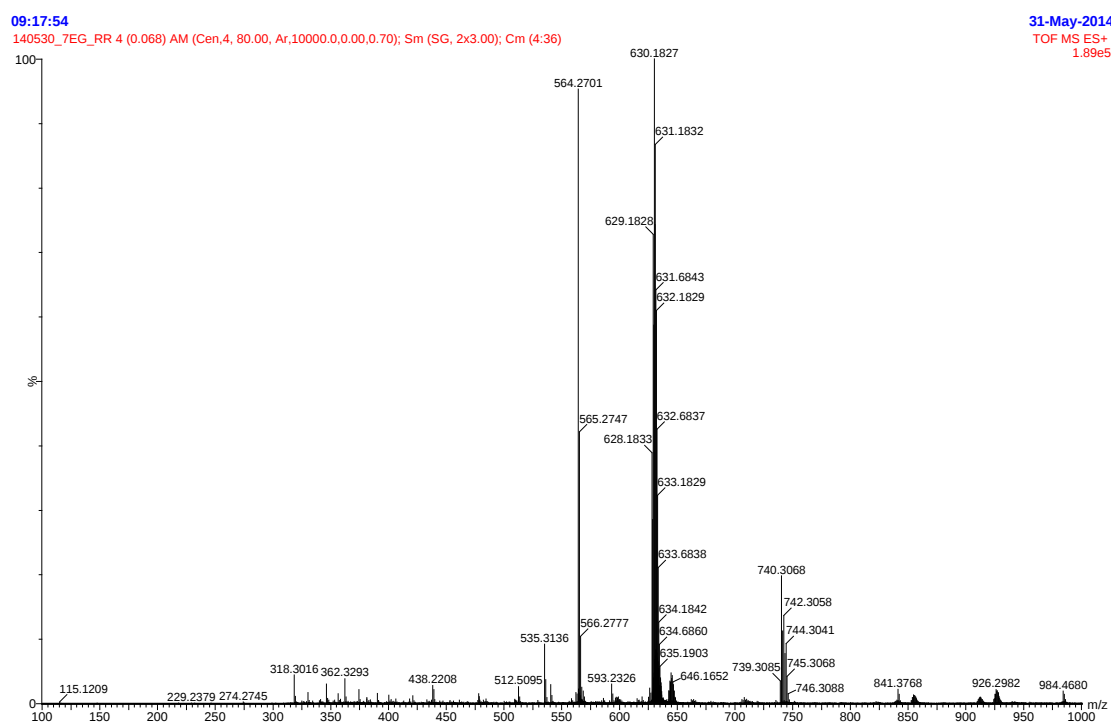


Figure S15. ^1H NMR titration of (*R*)-**4** (5mM) + ZnBr_2 (2 equiv) with (*S,S*)-**5** in CDCl_3 : CD_3OD (2: 1) in comparison with the naphthalene-2-amine + ZnBr_2 (10 mM). (The ^1H NMR spectra were taken after the solution was allowed to stand at room temperature for 3 h).

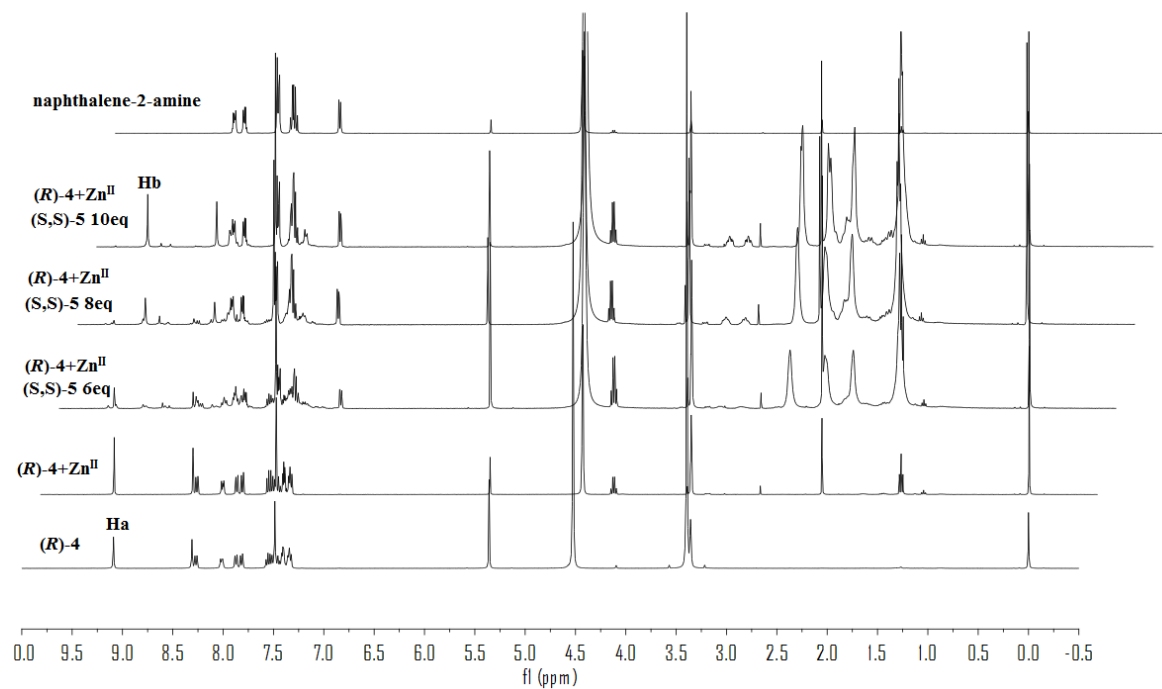


Figure S16. ^1H NMR titration of (*R*)-**4** (5 mM) + ZnBr_2 (2 equiv) with (*R,R*)-**5** in CDCl_3 : CD_3OD (2: 1) in comparison with the 2-naphthylamine + ZnBr_2 (10 mM). (The ^1H NMR spectra were taken after the solution was allowed to stand at room temperature for 3 h).

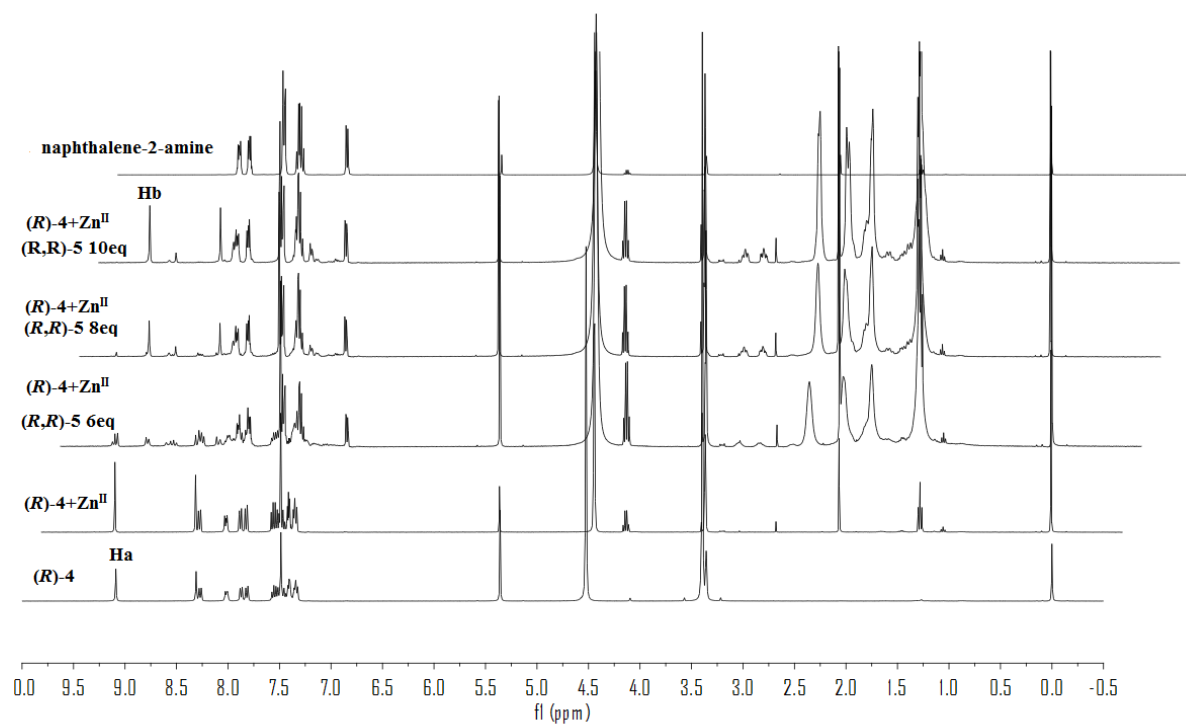


Figure S17. ^1H NMR spectrum of (*R*)-**4** (5mM) with ZnBr_2 (2 equiv) toward (*S,S*)-**5** and (*R,R*)-**5** (10 equiv) in $\text{CDCl}_3 : \text{CD}_3\text{OD}$ (2: 1) in comparison with the (*R*)-**4** (5mM) without ZnBr_2 (1 equiv) toward (*S,S*)-**5** and (*R,R*)-**5**. (The ^1H NMR spectra were taken after the solution was allowed to stand at room temperature for 3 h).

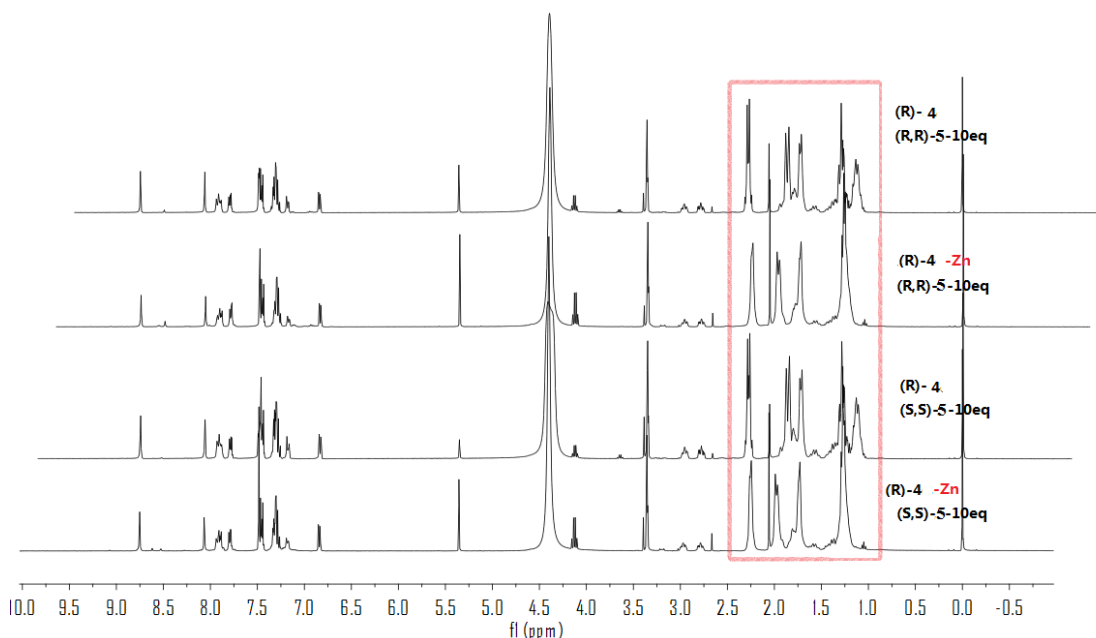
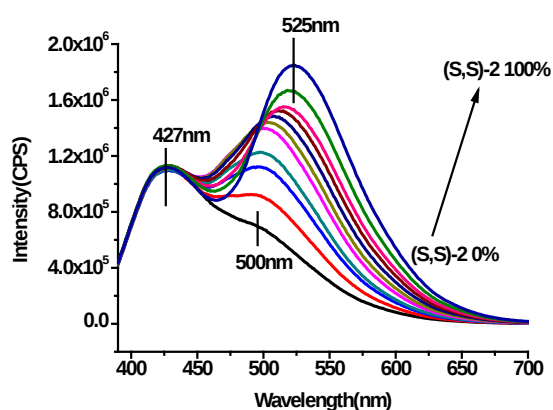
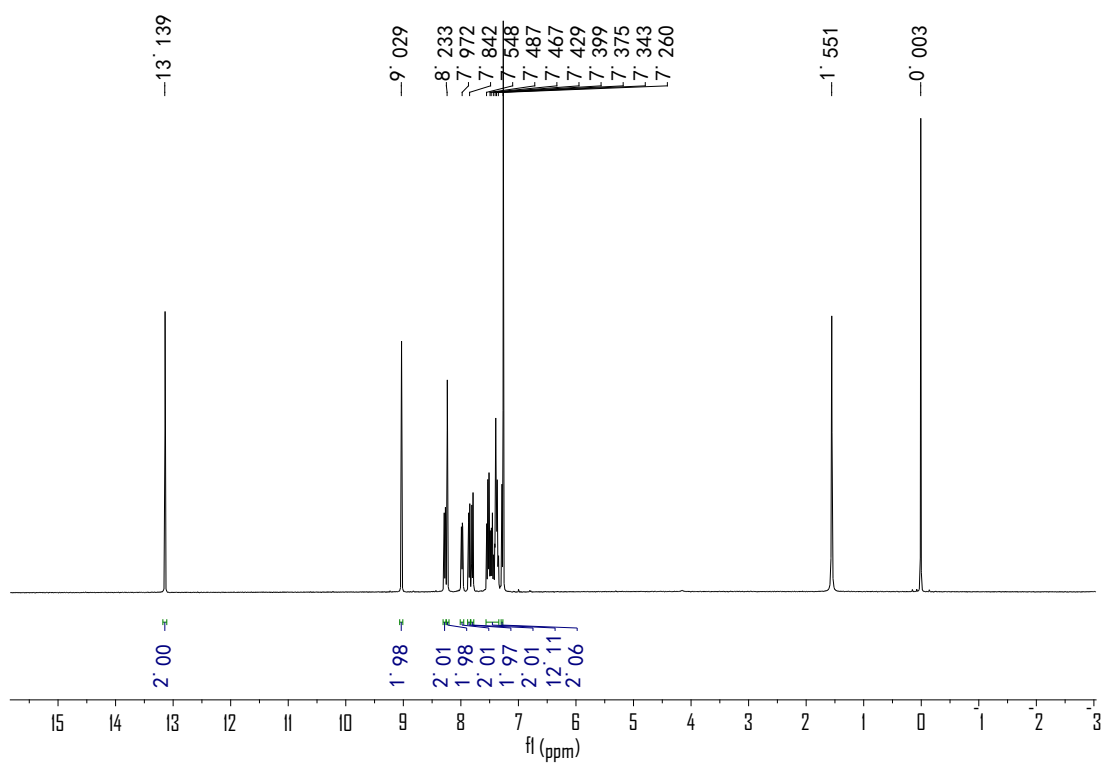


Figure S18. Fluorescence spectra of (*R*)-**4** (2.0×10^{-5} M) + Zn^{2+} (2 equiv) in the presence of the enantiomeric mixture of *trans*-1,2-cyclohexanediamine [from 0%, 10%, 20%, 30%, 40%, 50%, 60%, 70%, 80%, 90% to 100% (*S,S*)-**5**] at a total concentration of 22.5×10^{-5} M. (Solvent: CH_3OH with 2% CH_2Cl_2 . $\lambda_{\text{exc}} = 370$ nm, slit = 5/5 nm.).

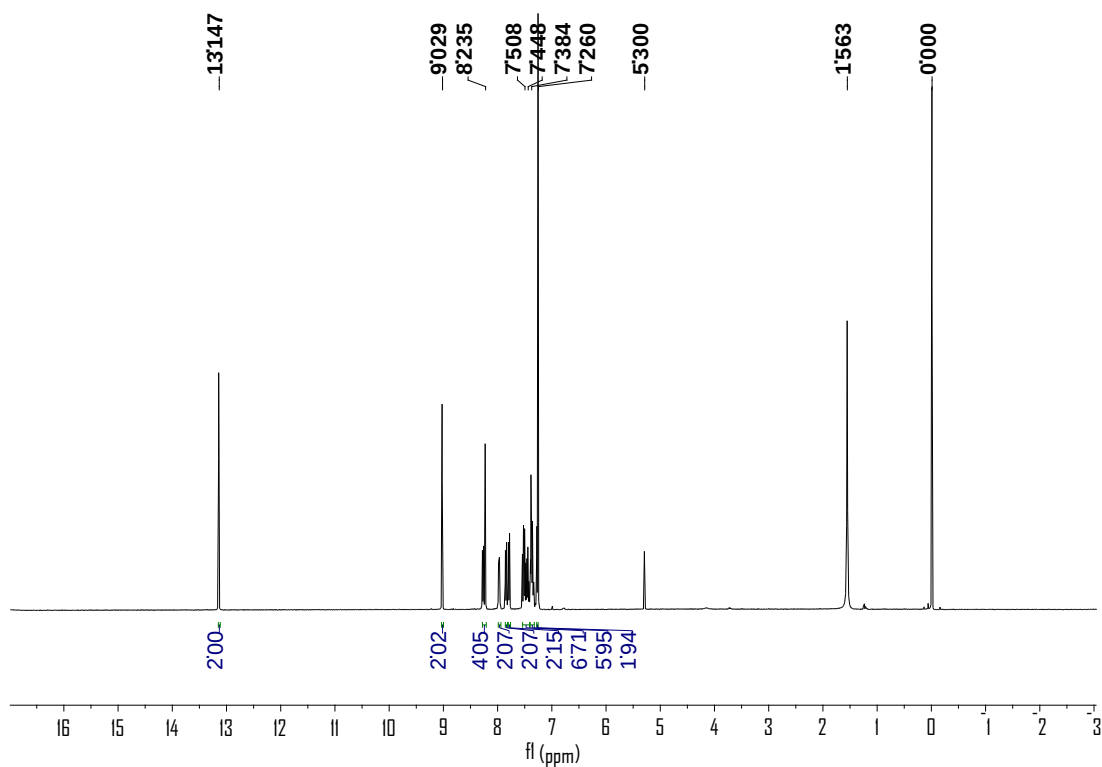


III. Preparation and Characterization of the (*R*)-4 and (*S*)-4

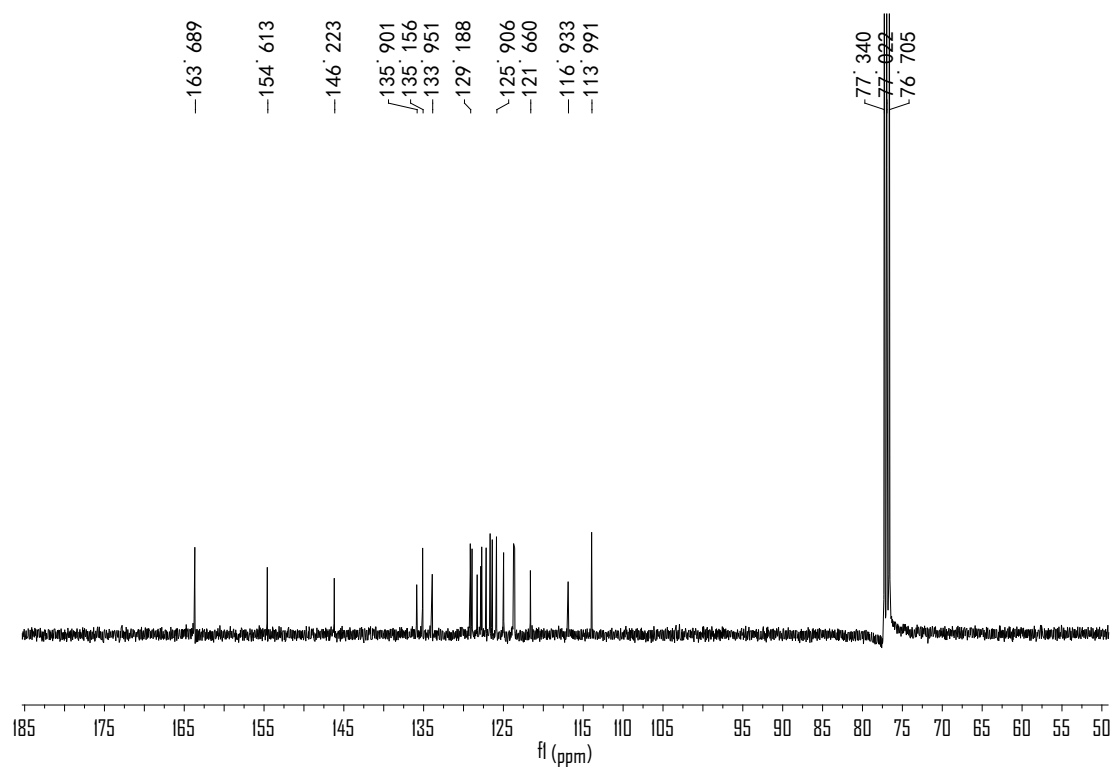
^1H -NMR of the (*R*)-4 (CDCl_3 , 400 MHz)



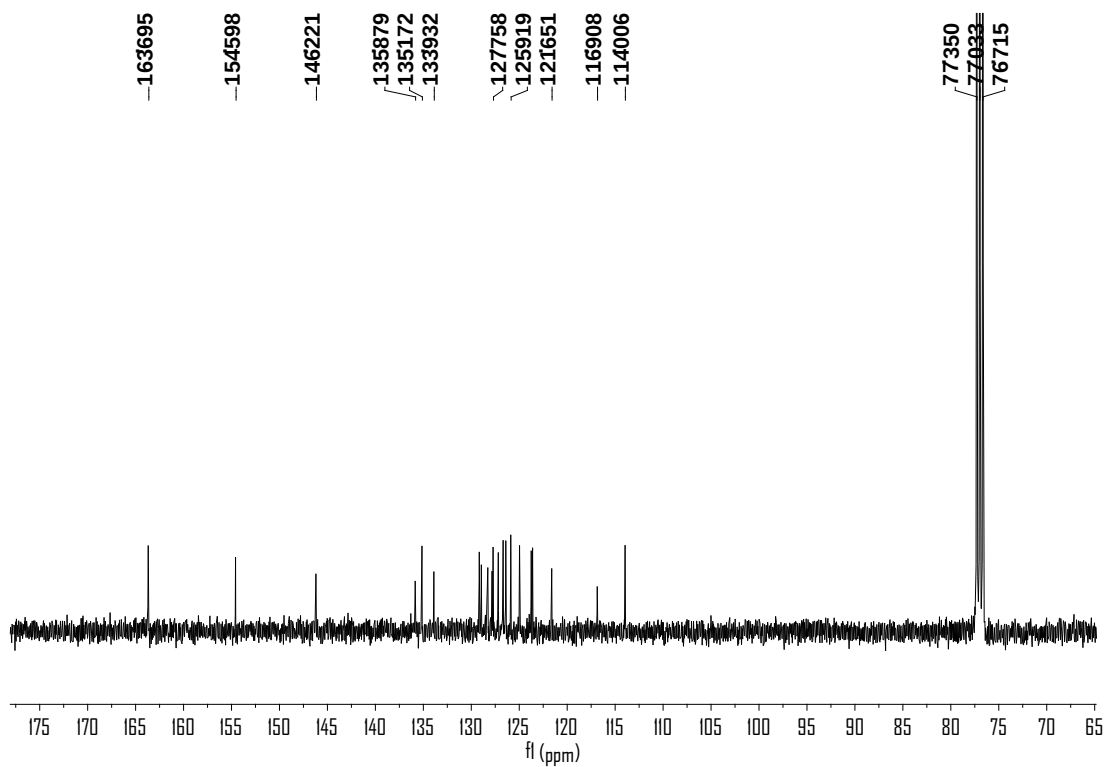
^1H -NMR of the (*S*)-4 (CDCl_3 , 400 MHz)



^{13}C -NMR of the (*R*)-4 (CDCl_3 , 100 MHz)



^{13}C -NMR of the (*S*)-4 (CDCl_3 , 100 MHz)



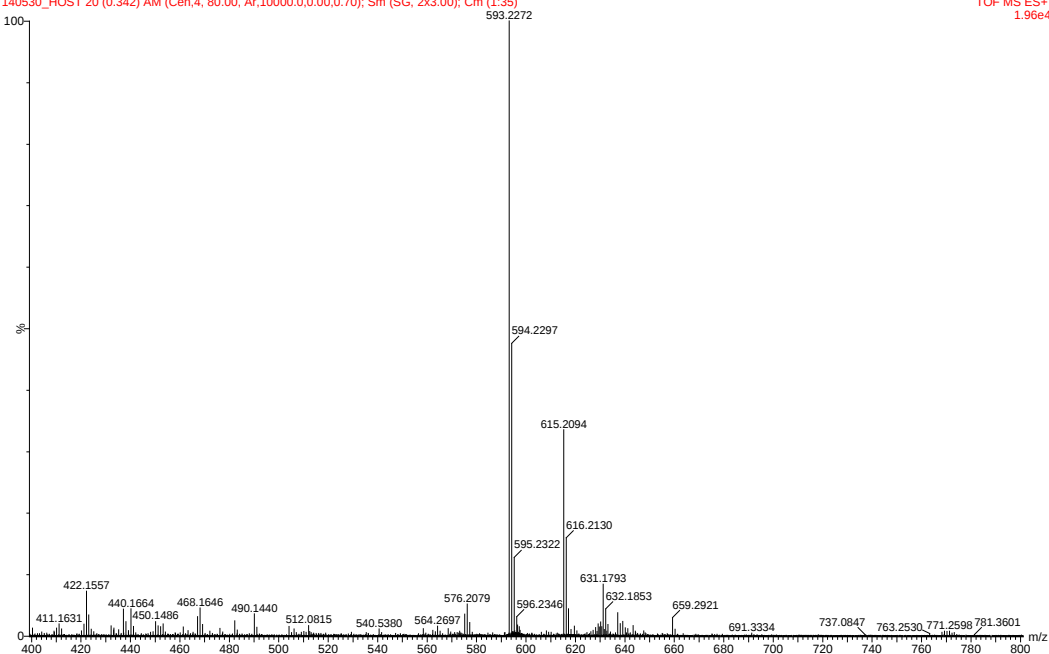
HRMS of the (R)-4

09:29:43

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31-May-2014

TOF MS ES+
1.96e4



HRMS of the (S)-4

