Supporting information

χ-Shaped bis(areno)-1,4-dihydropyrrolo[3,2-*b*]pyrroles generated by oxidative aromatic coupling

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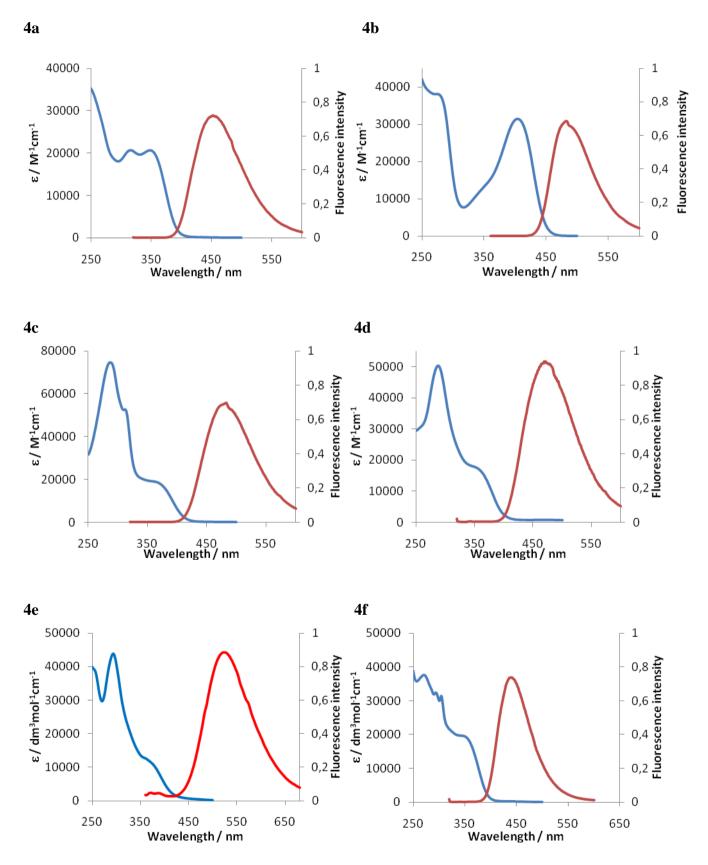
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1. General remarks.

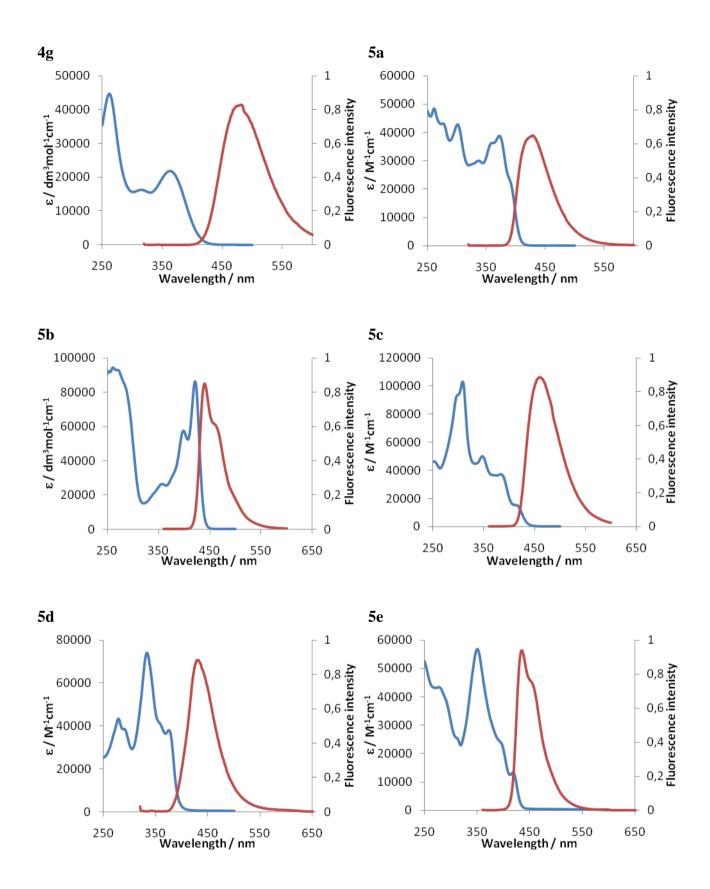
All reagents and solvents were purchased from commercial sources and were used as received unless otherwise noted. Reagent grade solvents (CH₂Cl₂, hexanes) were distilled prior to use. DMF was dried over magnesium sulfate, then distilled and stored under argon. Transformations with moisture and oxygen sensitive compounds were performed under a stream of argon. The reaction progress was monitored by means of thin layer chromatography (TLC), which was performed on aluminum foil plates, covered with Silica gel 60 F₂₅₄ or Aluminum oxide 60 F₂₅₄ (neutral). Products purification was done by means of column chromatography with Kieselgel 60 or Aluminum oxide. Occasionally, dry column vacuum chromatography (DCVC) for purification of products obtained was performed using Silica gel Type D 5F. The identity and purity of prepared compounds were proved by ¹H NMR and ¹³C NMR spectrometry as well as by MS-spectrometry (*via* EI-MS or ESI-MS). For HRMS measurements both quadruple and TOF mass analyzer types were used. NMR spectra were measured on 500 MHz, 600 MHz or 400 MHz instruments with TMS as internal standard. All chemical shifts are given in ppm. All melting points for crystalline products were measured with automated melting point apparatus and were given without correction. The absorbance and fluorescence spectra were measured in dichloromethane.

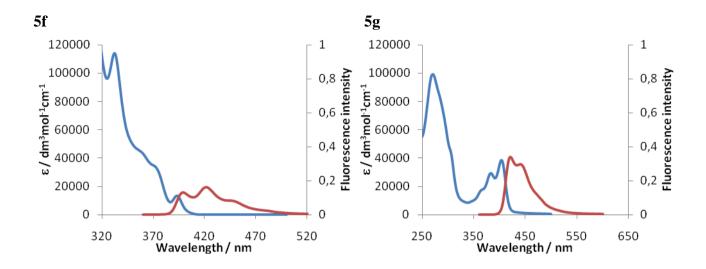
Linear optical measurements

Steady-state fluorescence measurements were performed with dilute solutions (10⁻⁶M, optical density<0.1) contained in standard 1 cm quartz cuvettes at room temperature. Compounds were dissolved in dichloromethane unless otherwise noted. Emission spectra were obtained, for each compound, under excitation at λ =350 nm. Fluorescence quantum yields were measured by using quinine hemisulphate monohydrate in 0.5m sulfuric acid as a standard.

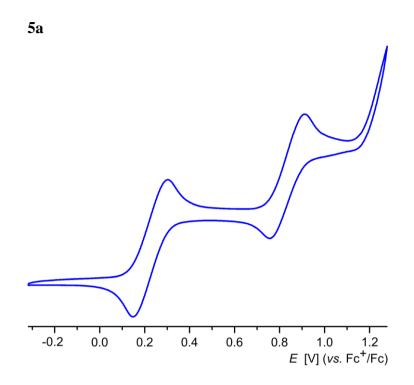


2. Normalized absorption and emission spectra for synthesized compounds.

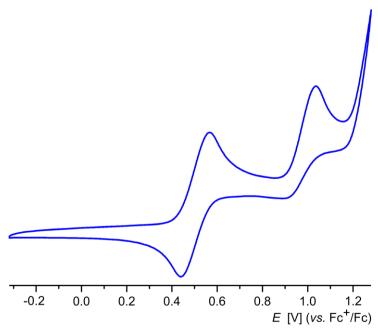


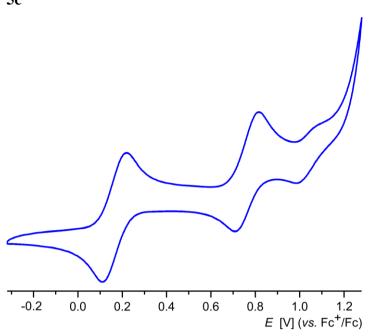


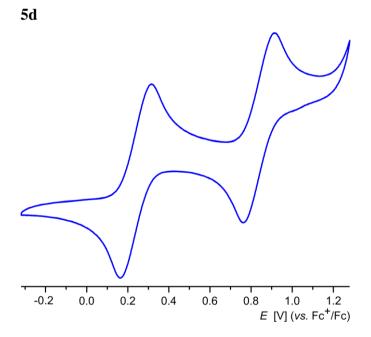
3. Cyclic voltammetry for selected compounds.



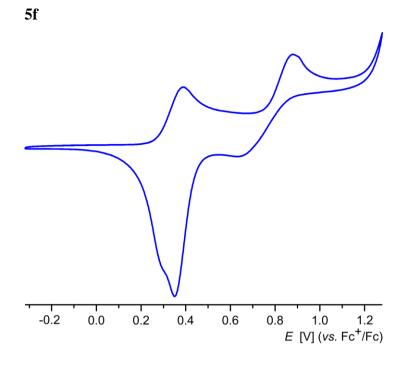
5b



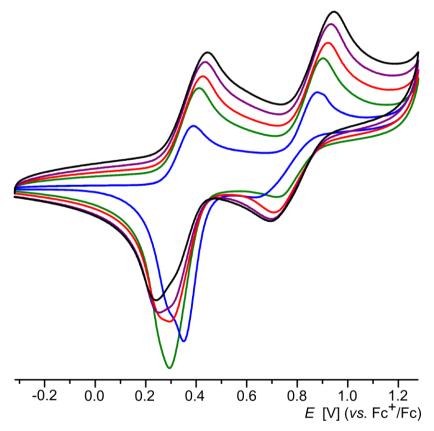


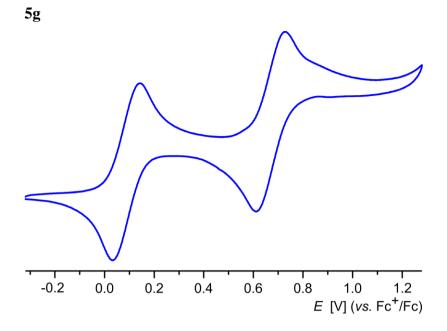


5c

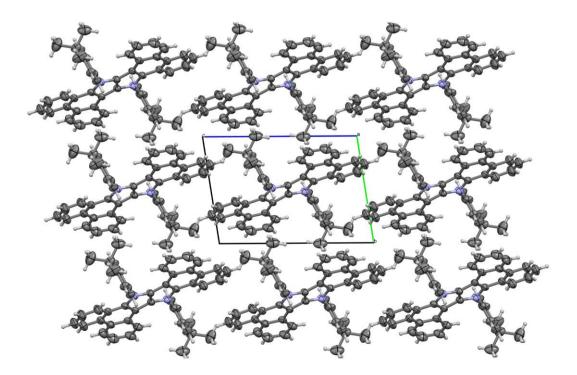


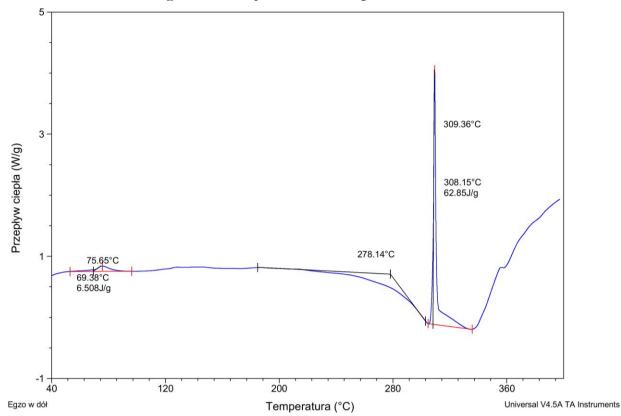
5f at v = 50, 250, 500, 750 and 1000 mVs⁻¹





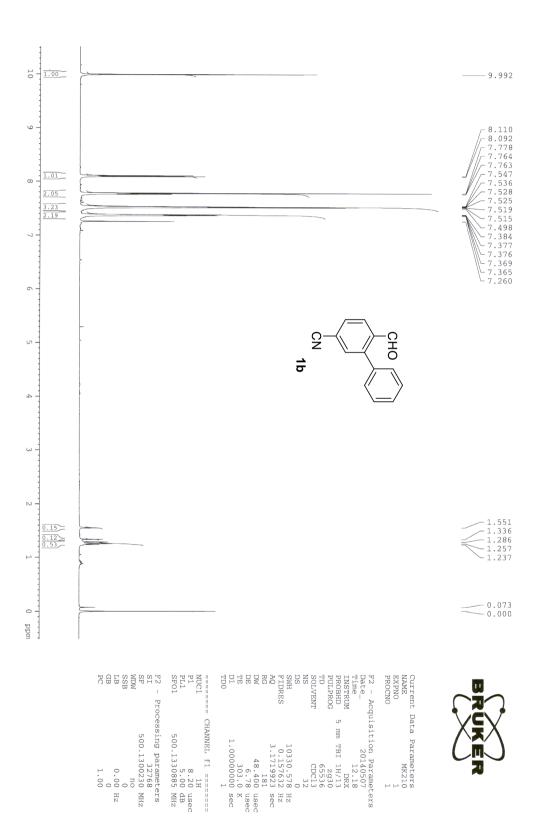
4. Crystal packing for compound 5a.

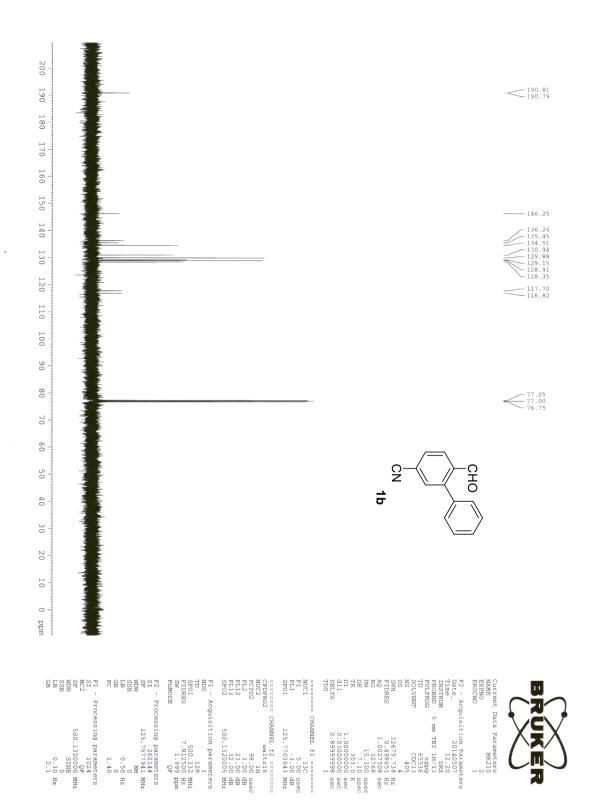


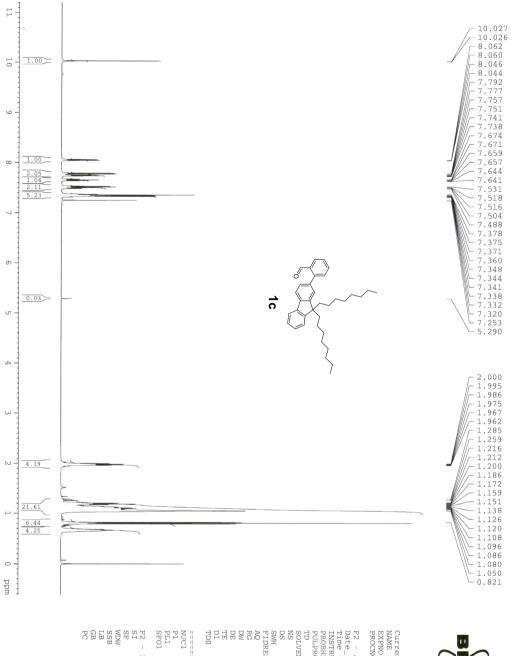


5. Differential Scanning Calorimetry curve for compound 5b.

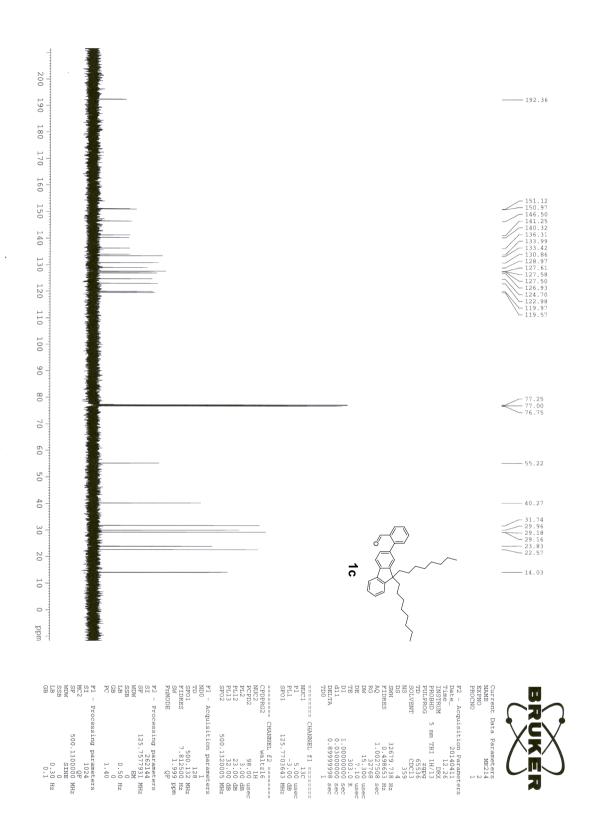
6. ¹H and ¹³C NMR spectra of compounds 1b-g, 4a-g and 5a-g.

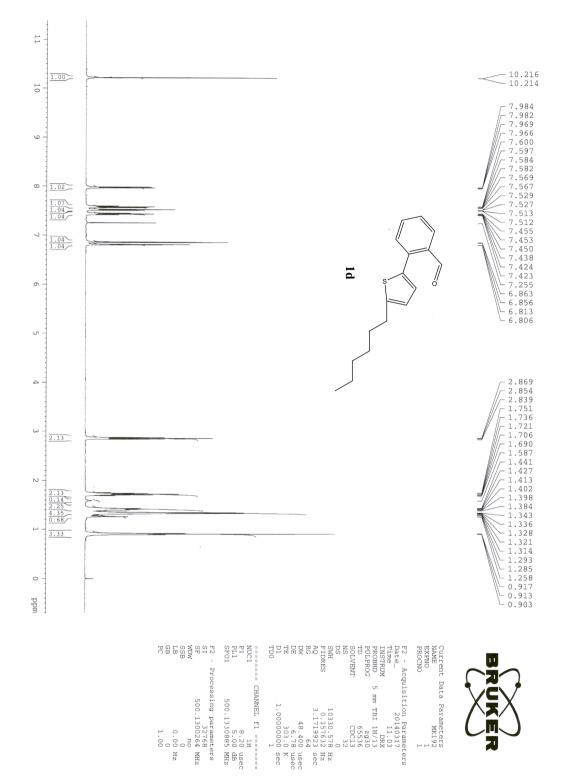


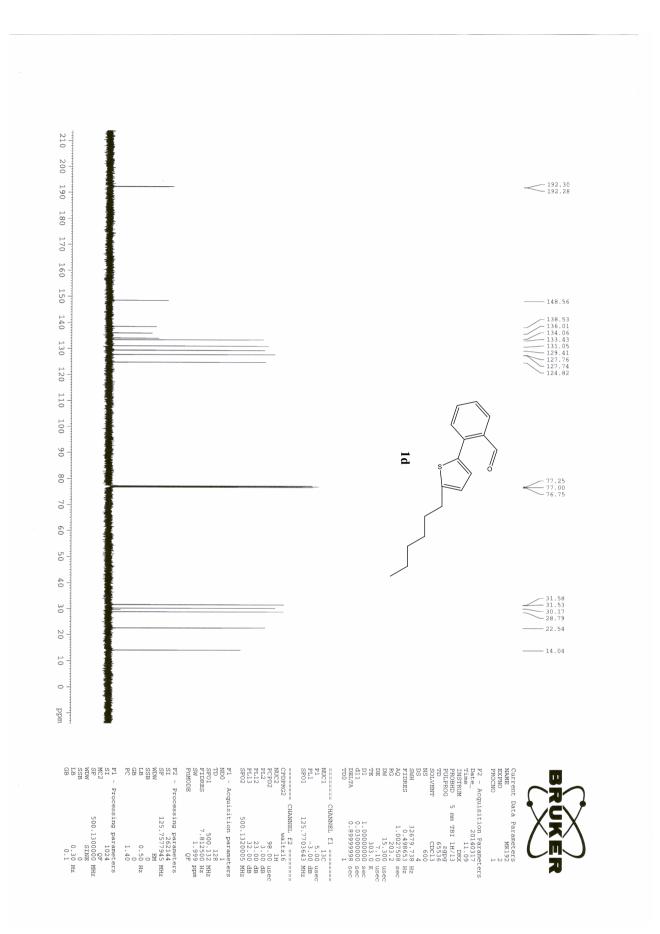


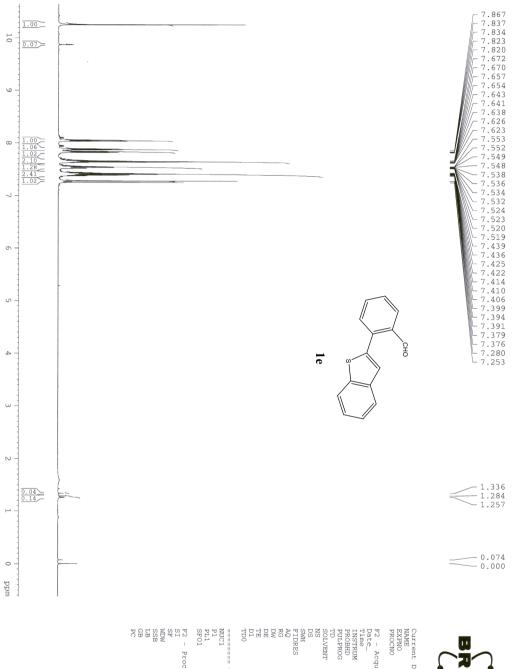


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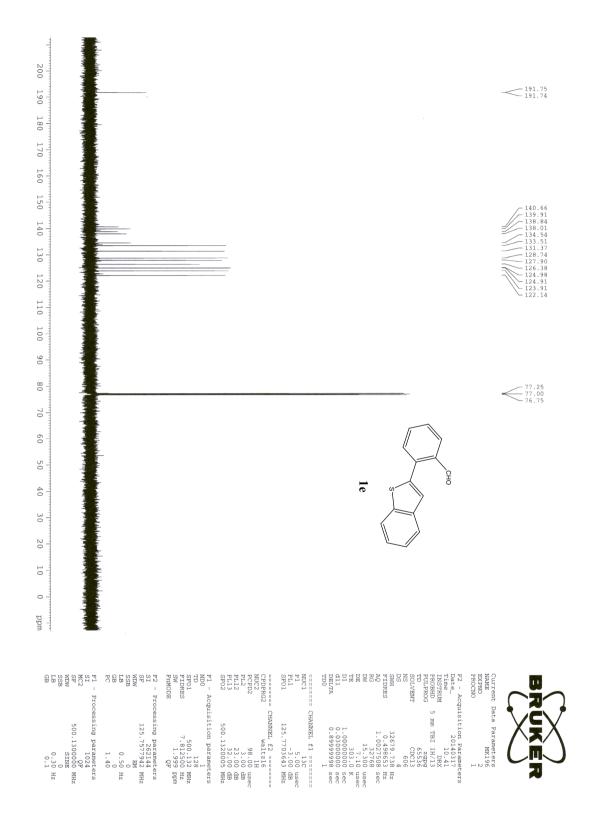


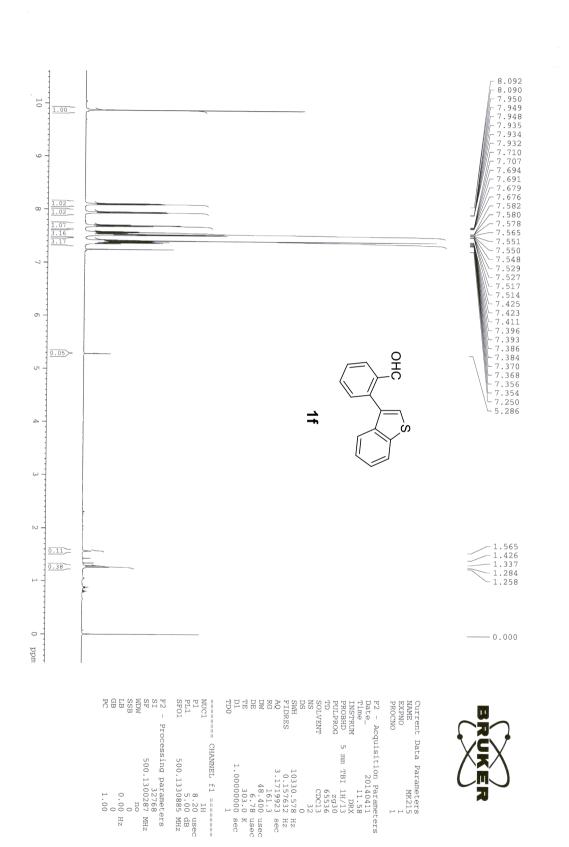


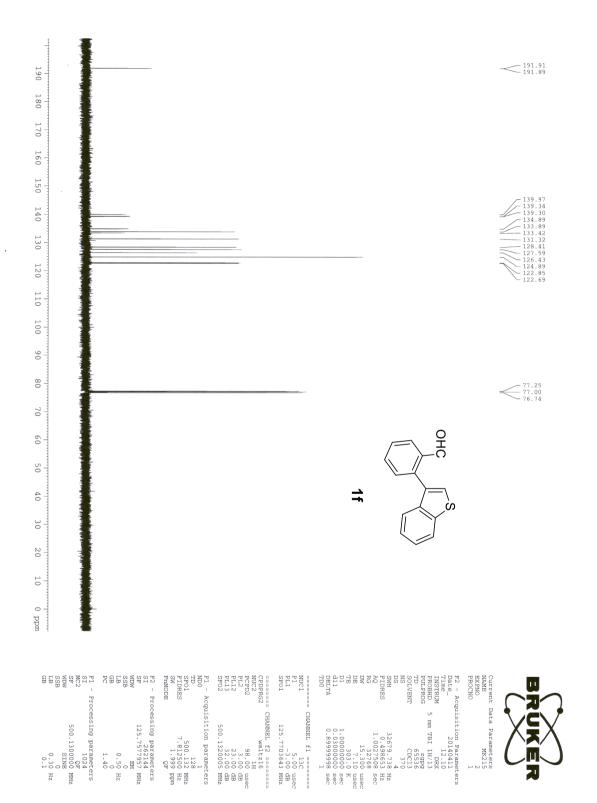


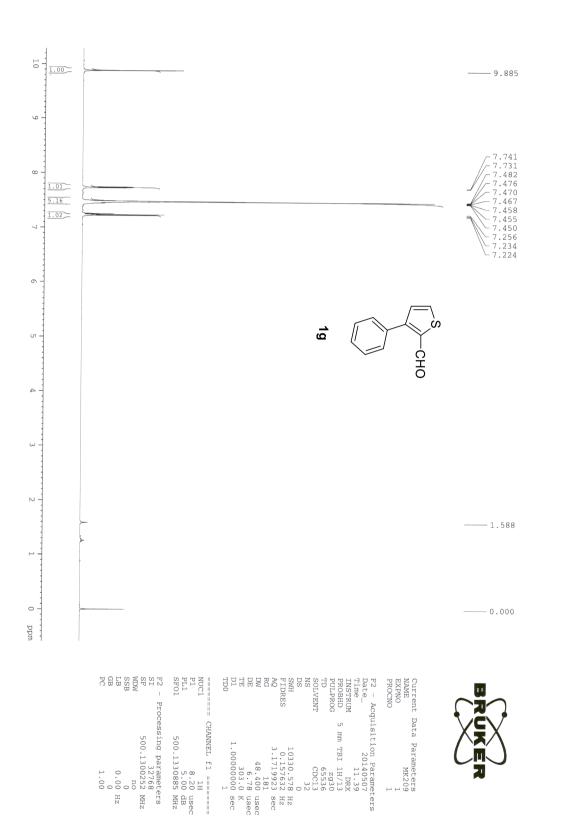


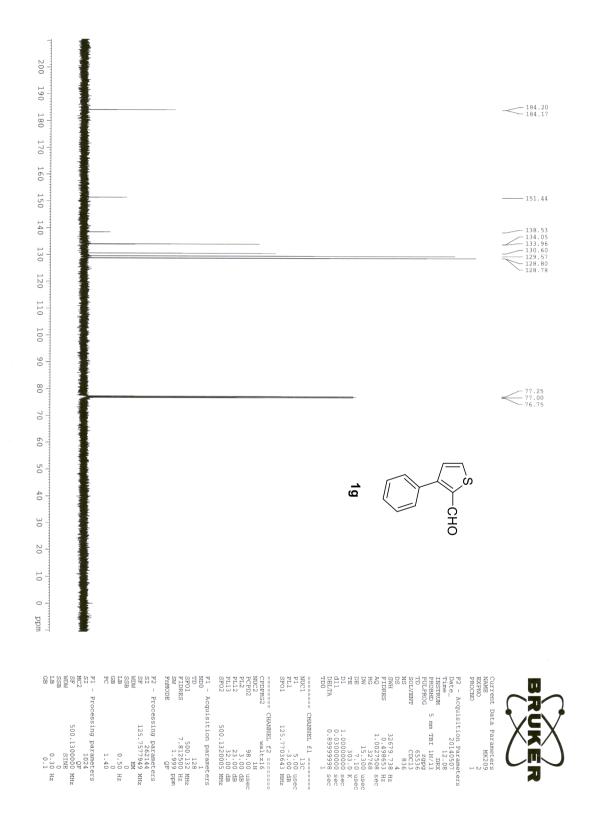
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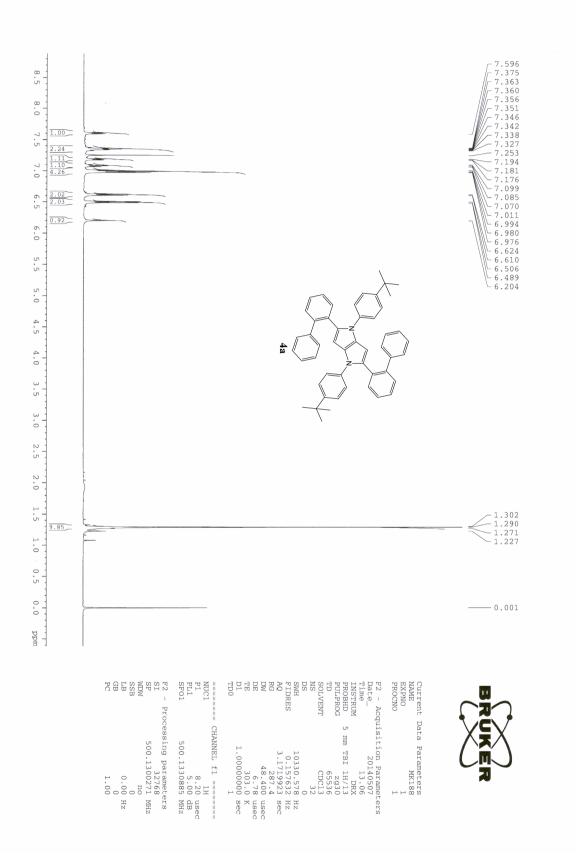


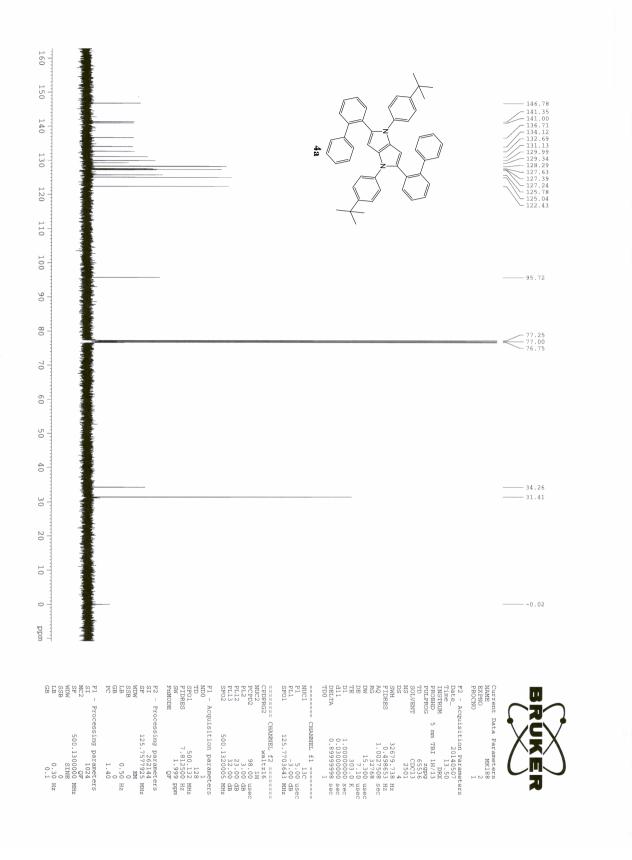


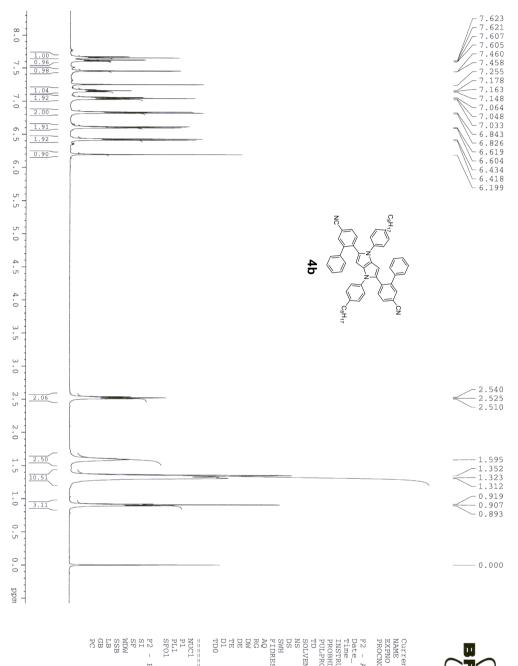






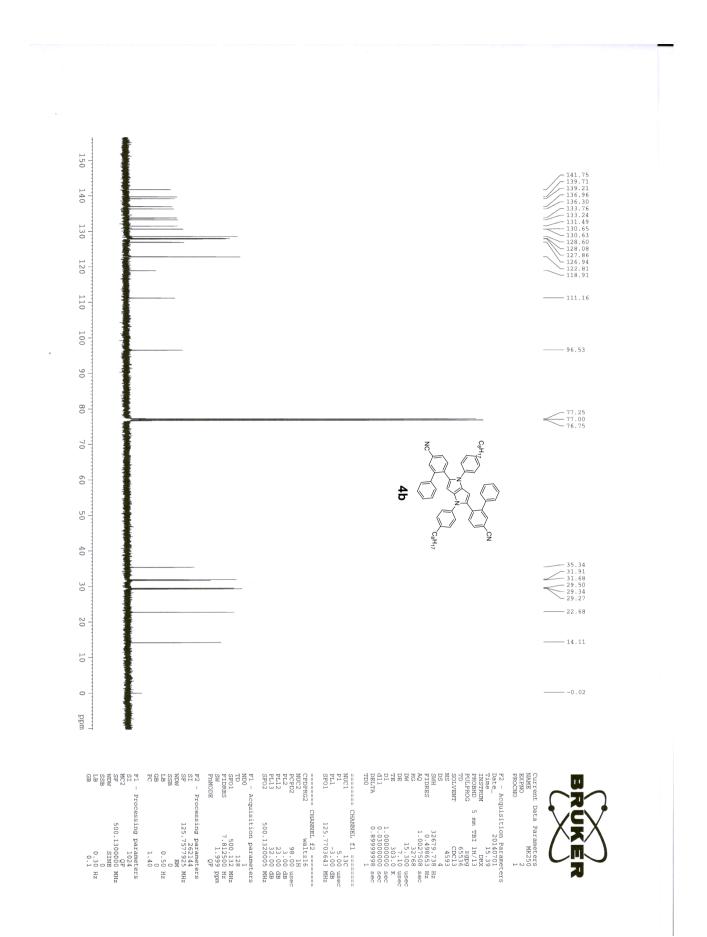


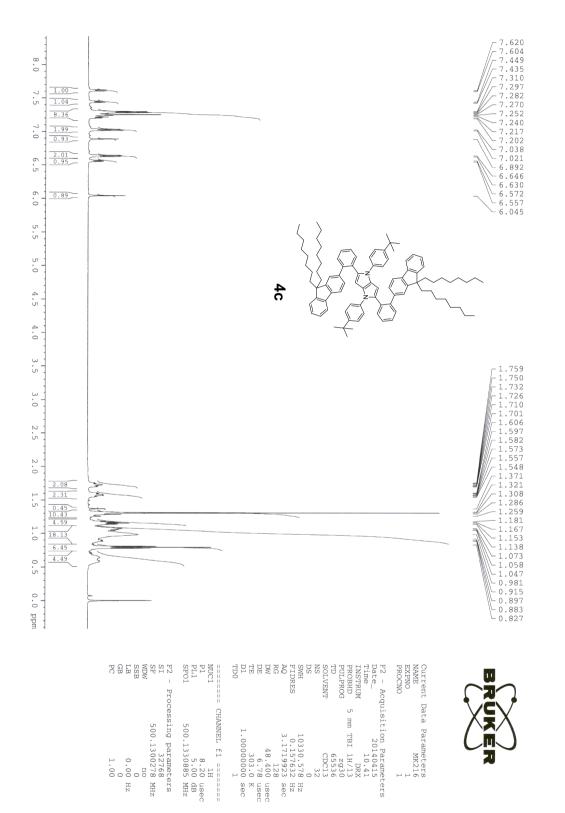


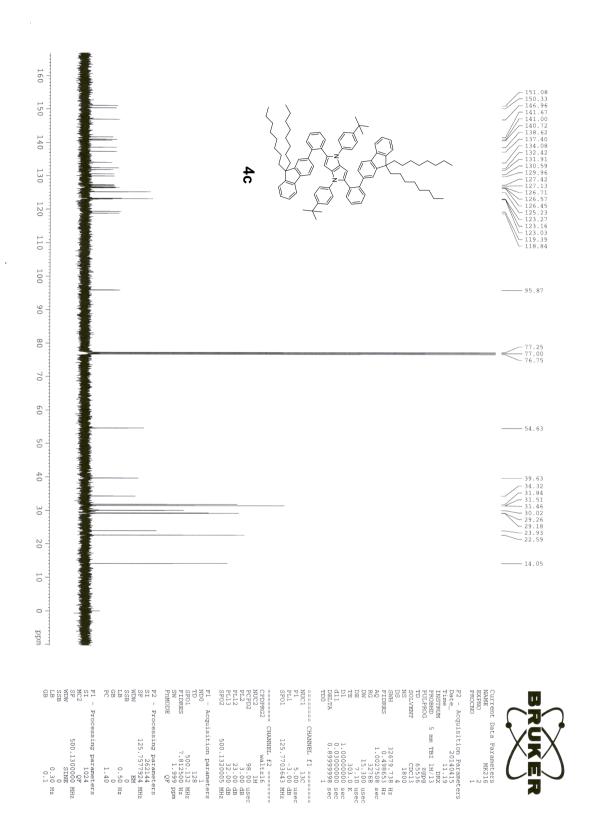


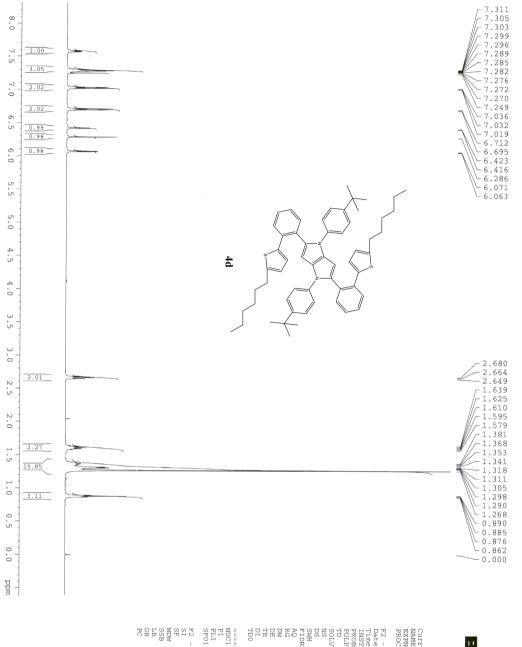
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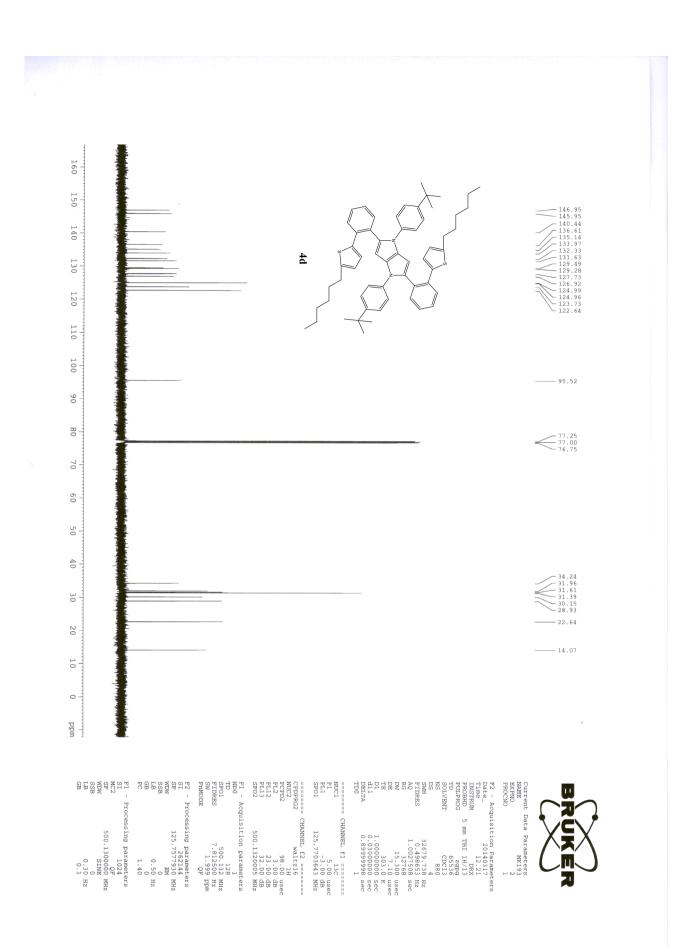
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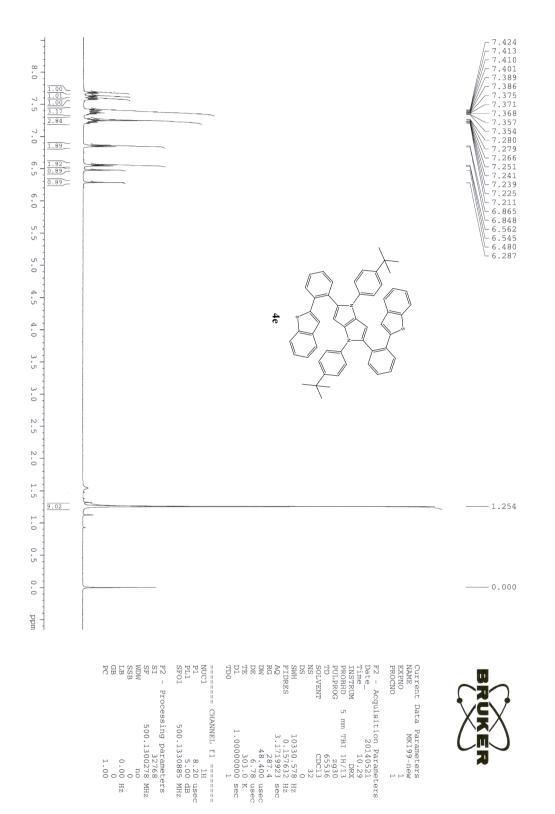


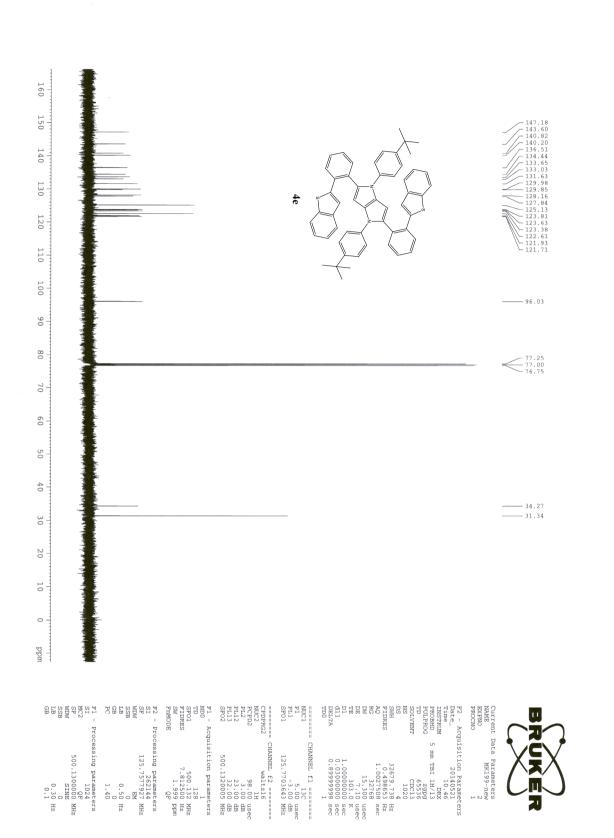


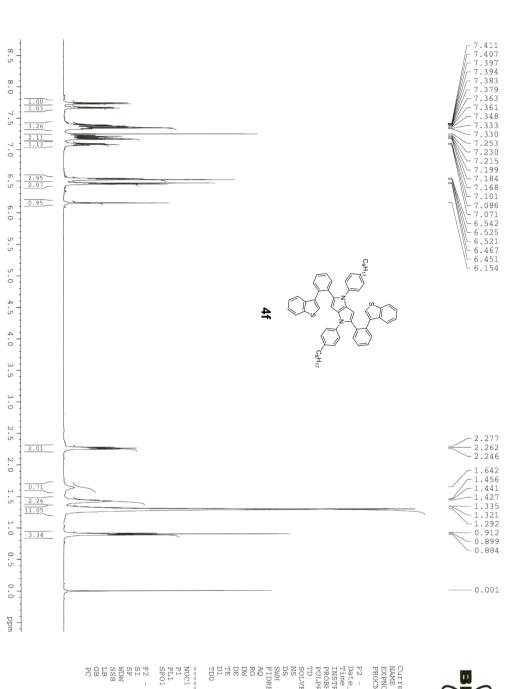






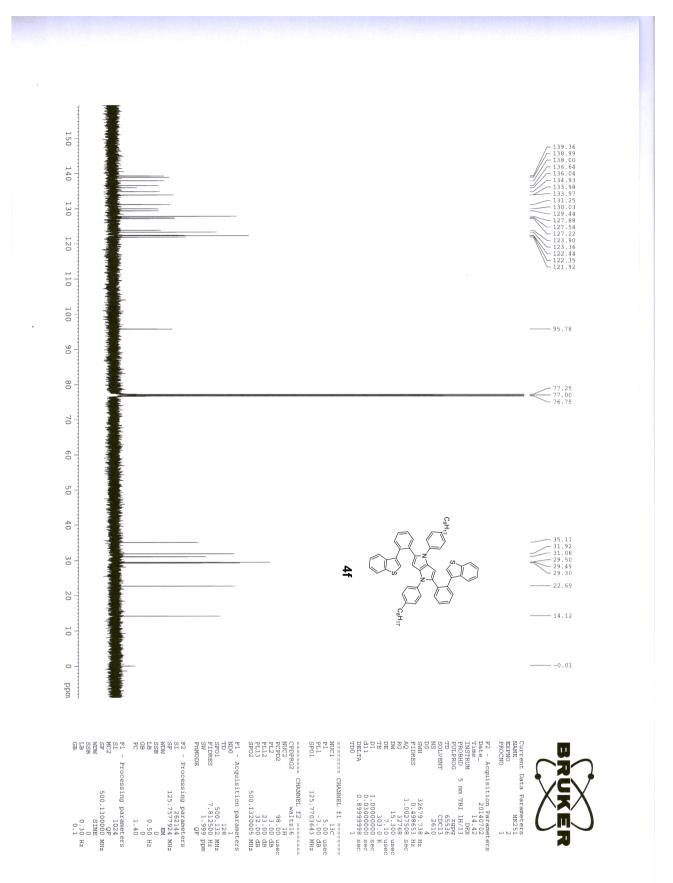


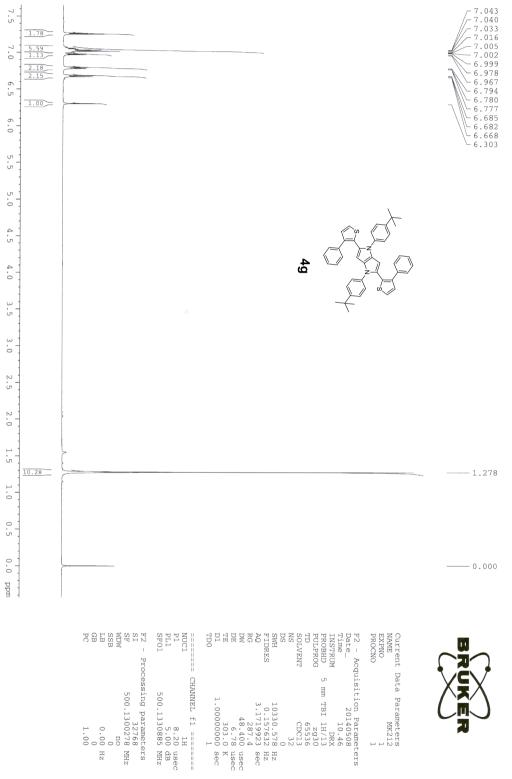




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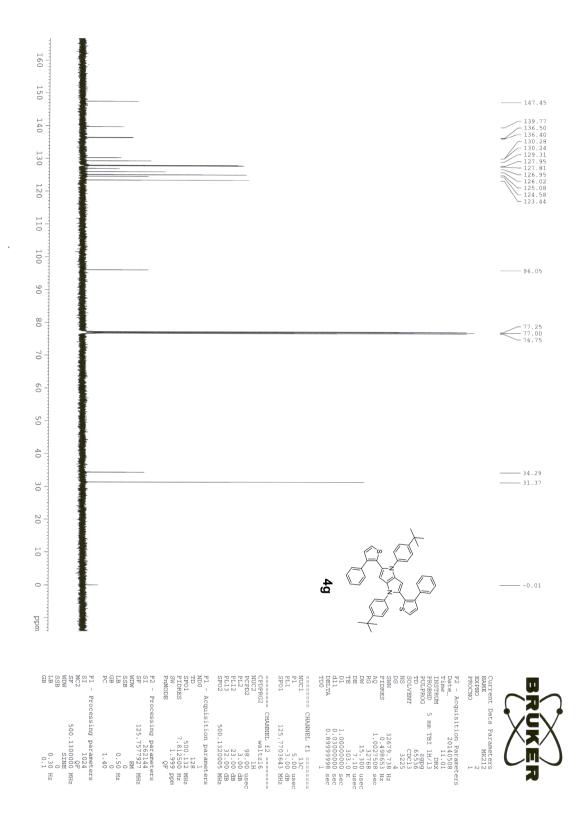
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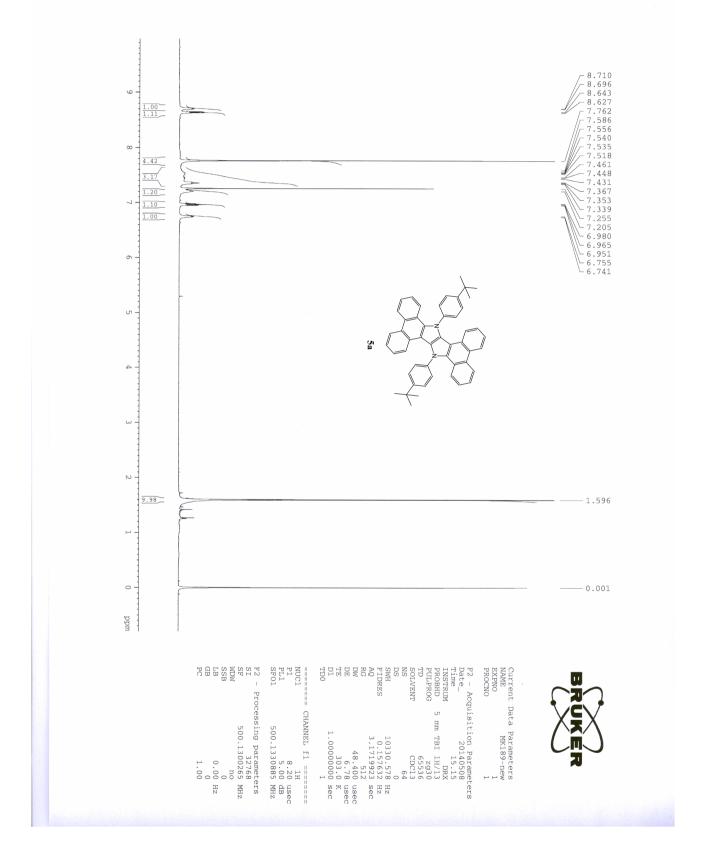


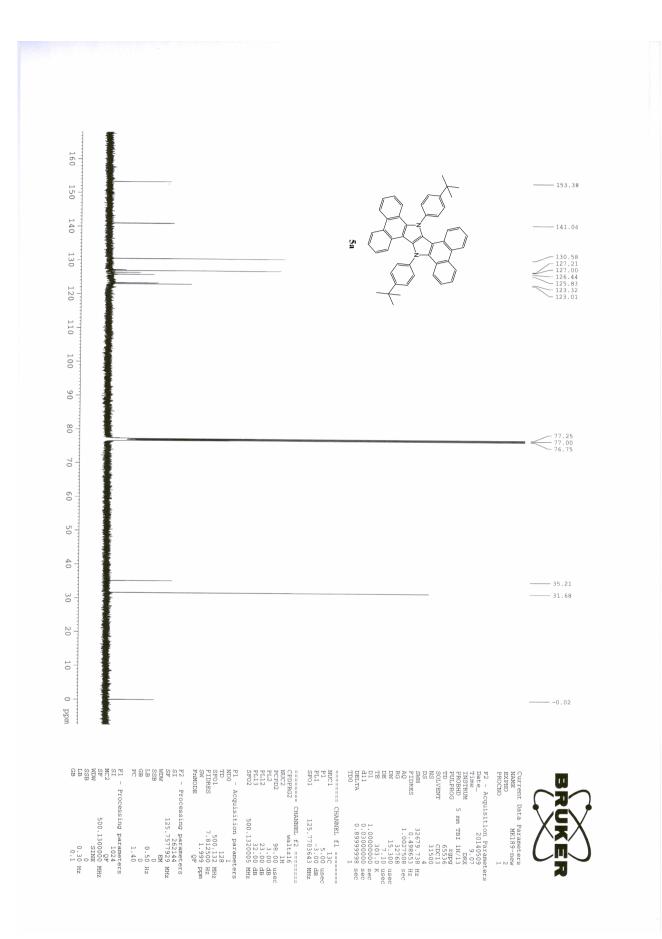


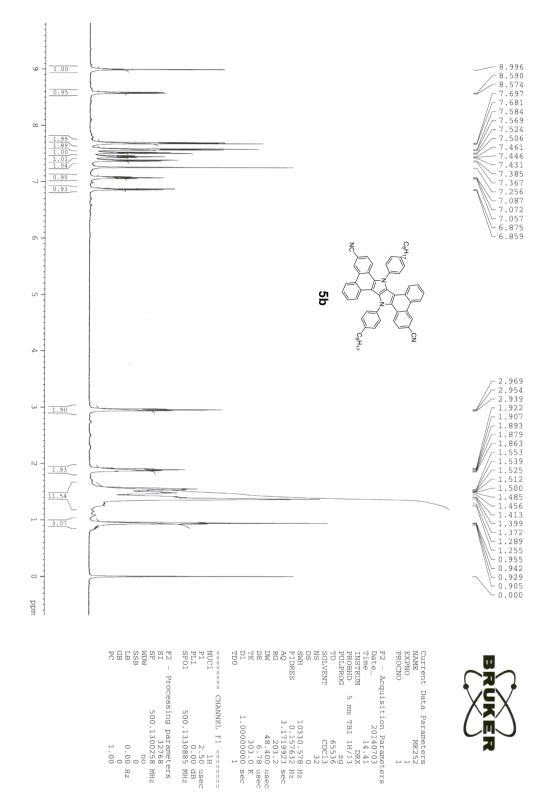
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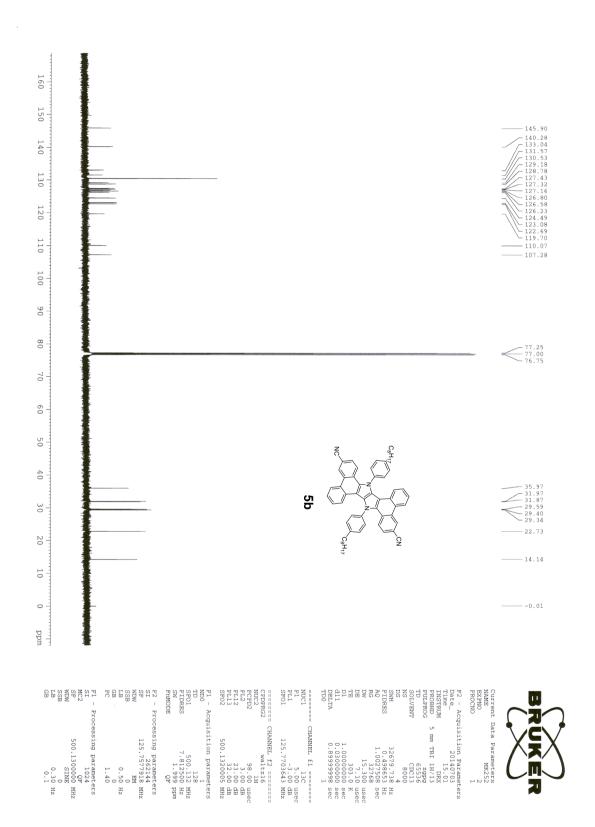


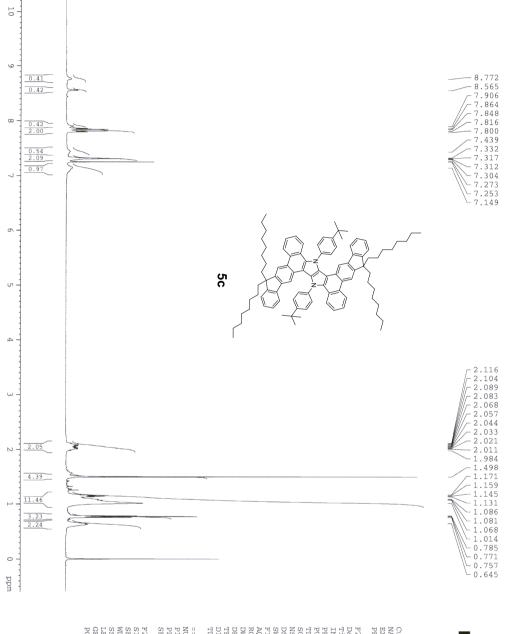










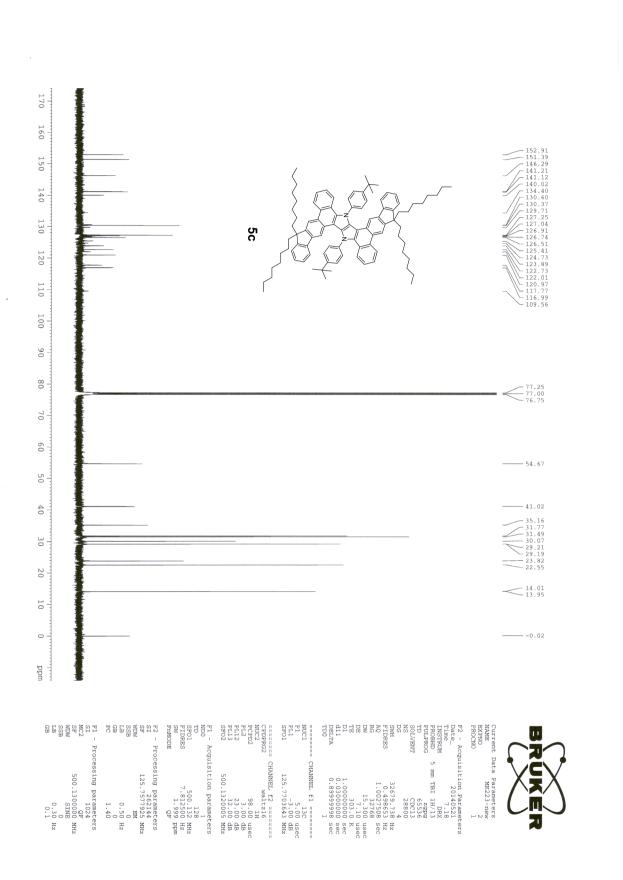


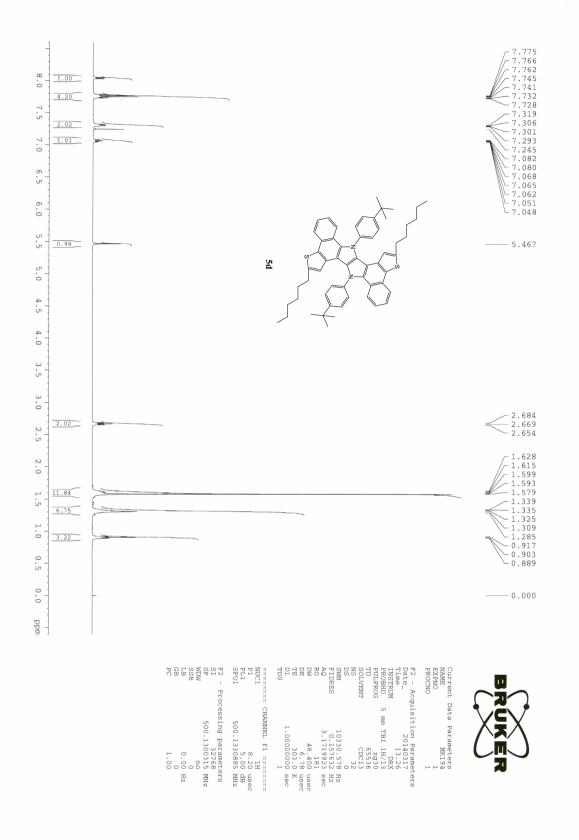
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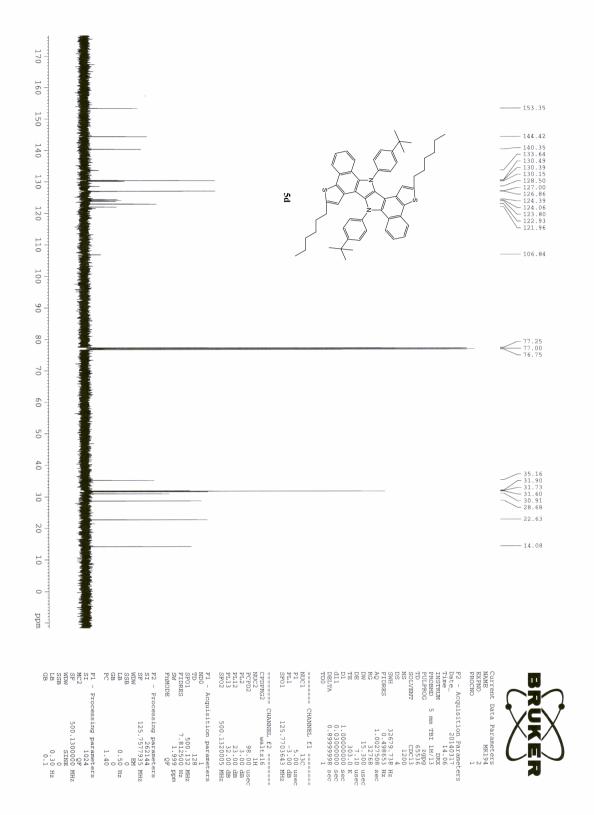
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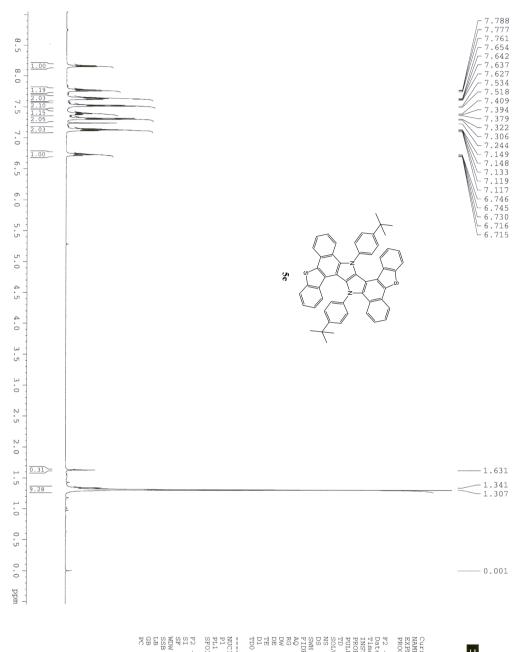
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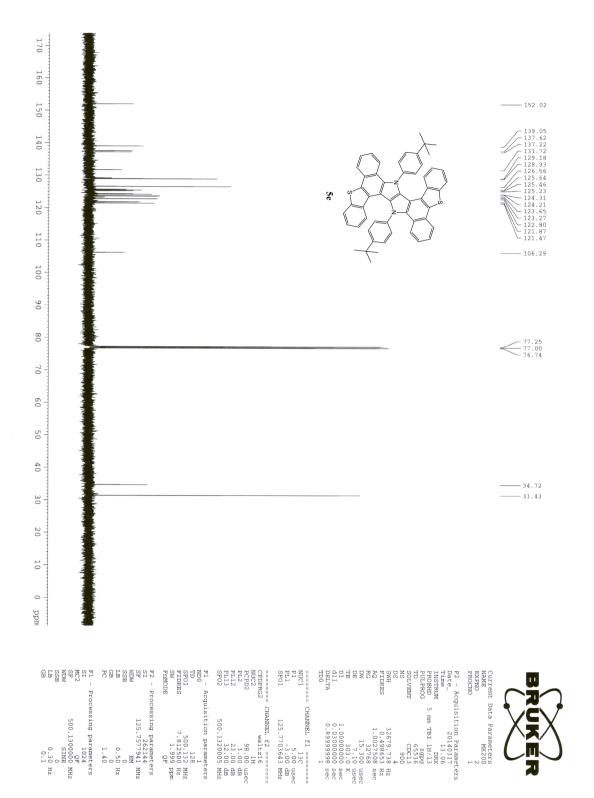


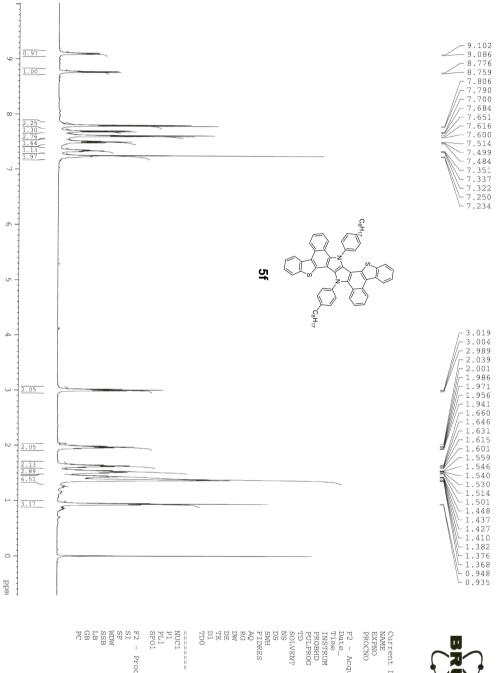






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