

Supporting Information for

Hydrothermal Fabrication of $\text{MnCO}_3@\text{rGO}$ Composite as an Anode Material for High Performance Lithium Ion Batteries

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Table S1. Atom ratio of C, O and Mn elements in MGC based on XPS data

Element	C1s	O1s	Mn2p
Atom ratio percent (at.%)	43.9	45.11	10.99

$$\text{Total mass} = 43.9 \times 12 + 45.11 \times 16 + 10.99 \times 54.94 = 526.8 + 721.76 + 603.79 = 1852.35$$

$$\text{MnCO}_3 \text{ wt.}\% = 10.99 \times 114.95 / 1852.35 \times 100\% = 68.20\%$$

$$\text{rGO wt.}\% = 100\% - 68.20\% = 31.80\%$$

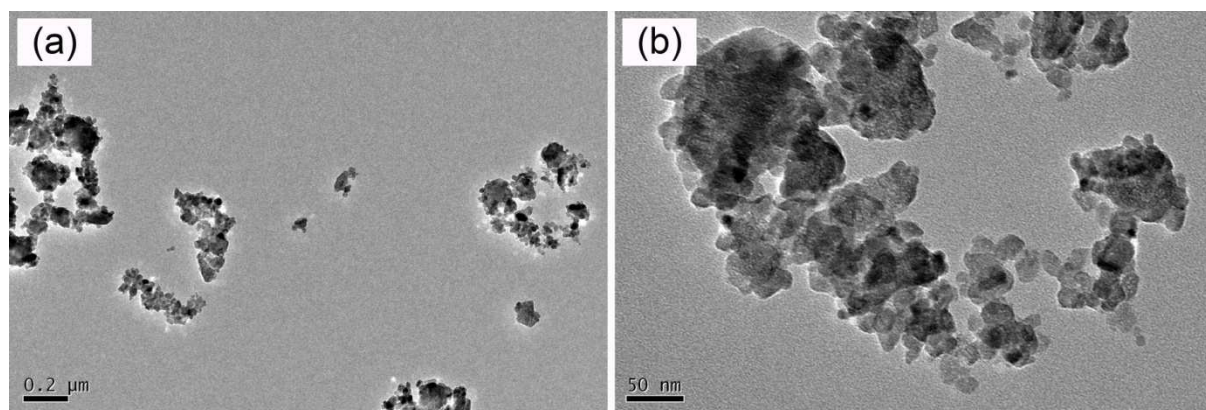


Figure S1. (a) and (b) TEM images of the pristine MnCO_3 with different magnification powers.

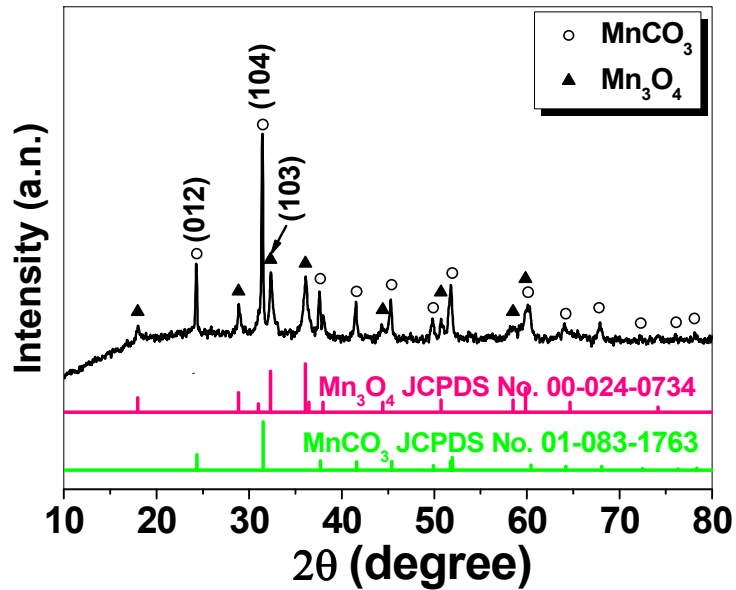


Figure S2. XRD pattern of the as-synthesized MnCO_3 with Mn_3O_4 purity by a similar synthesis process to the MGC just in the absence of GO.

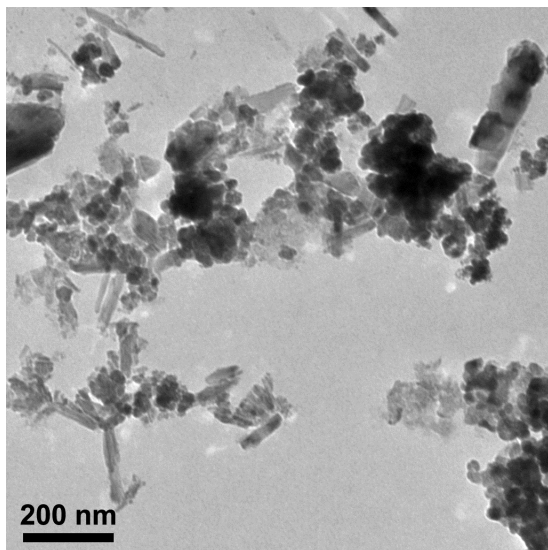


Figure S3. TEM image of the as-synthesized MnCO_3 with Mn_3O_4 purity by a similar synthesis process to the MGC just in the absence of GO.

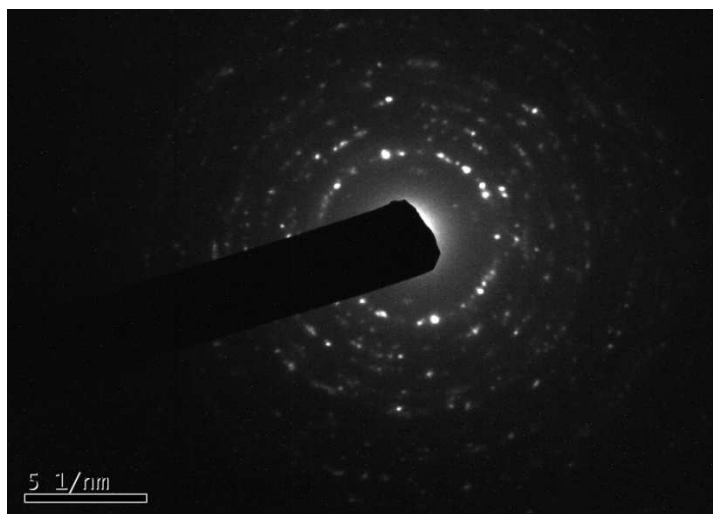


Figure S4. Selected area electron diffraction pattern of MGC.

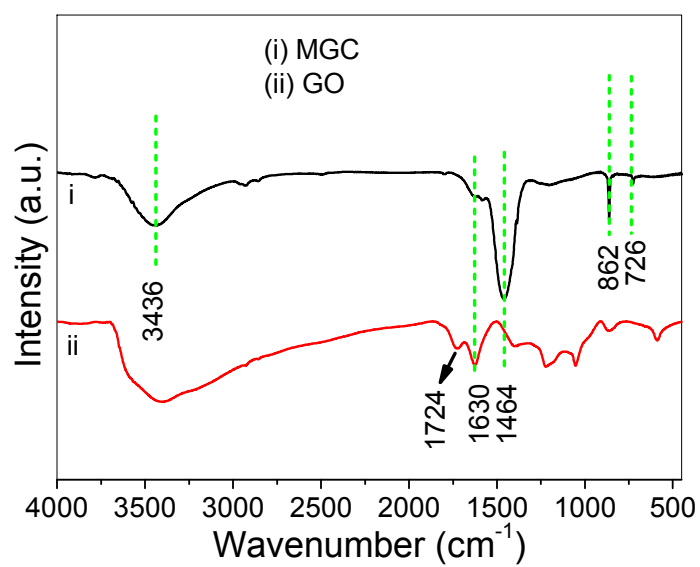


Figure S5. FT-IR spectra of MGC and GO.

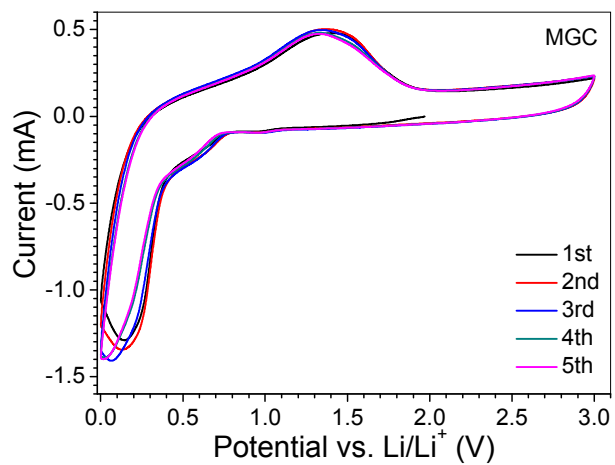


Figure S6. CV plots of MnCO₃@rGO composite (MGC) after cycled for 3 times in Li ion battery (LIBs).