

Supporting Information for:

Layered Double Hydroxide-Supported Carbon Dots as an Efficient
Heterogeneous Fenton-Like Catalyst for Generation of Hydroxyl Radicals

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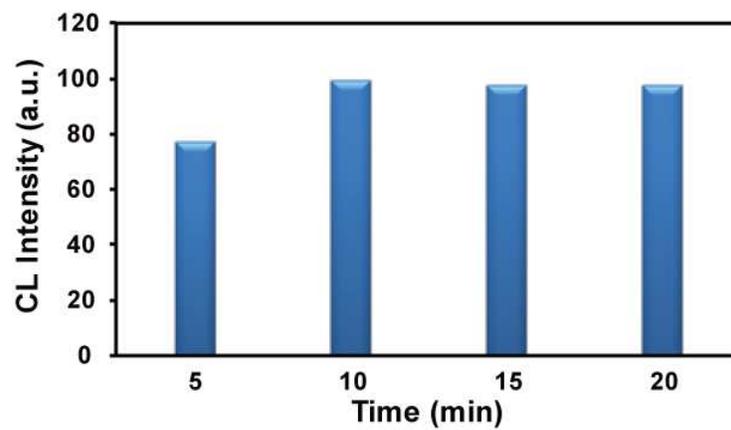


Figure S1 Effect of stirring time for the CL intensity of the carbon dot-DBS-LDH-acidified H_2O_2 system.

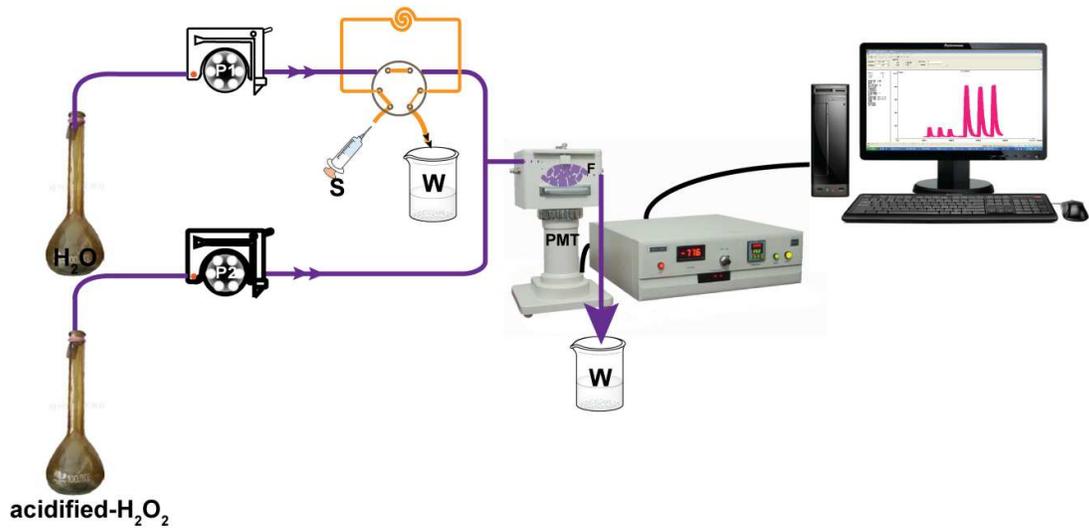


Figure S2 Schematic diagram of the CL flow-through device. P₁ and P₂, peristaltic pumps; S, sample injector (carbon dot-DBS-LDHs); F, flow cell; W, waste; PMT, photomultiplier tube (-1100 V); H₂O at 1.0 mL/min; acidified H₂O₂ at 2.0 mL/min.

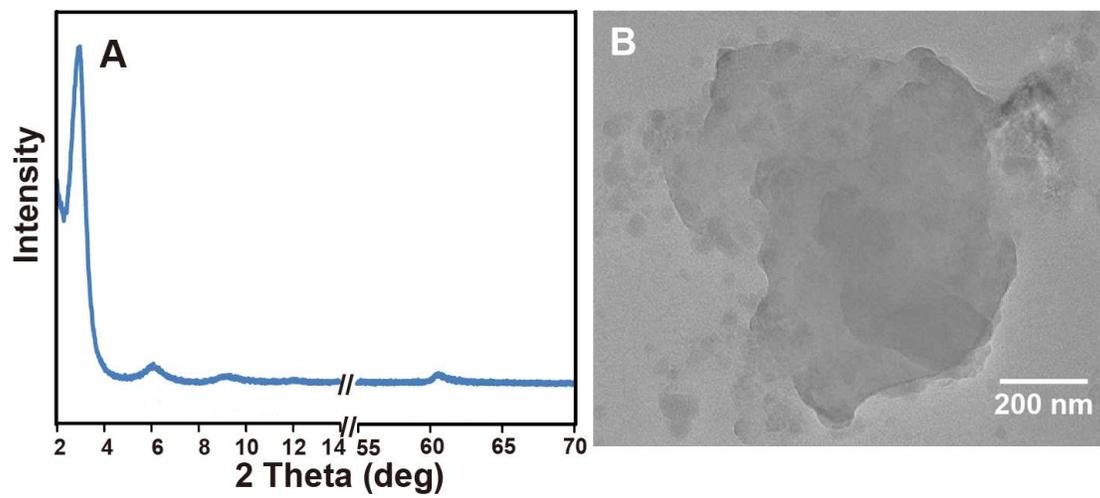


Figure S3 (A) Powder XRD patterns, (B) TEM image of DBS-LDHs.

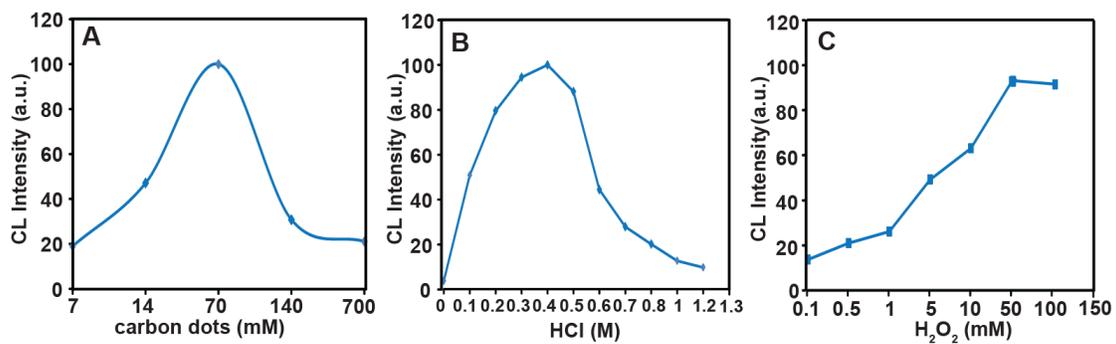


Figure S4 Effects of various conditions on the CL intensity, (A) concentration of carbon dots, (B) concentration of HCl, (C) concentration of H₂O₂.

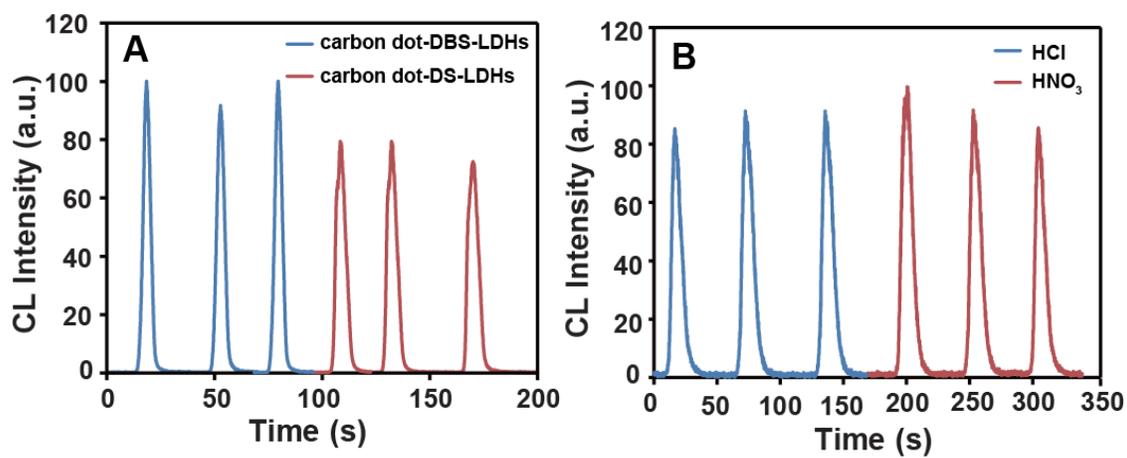


Figure S5 CL intensity of (A) acidified H₂O₂ mixed with carbon dot-DBS-LDHs or carbon dot-DS-LDHs, (B) carbon dot-DBS-LDH-H₂O₂ mixed with HCl or HNO₃.

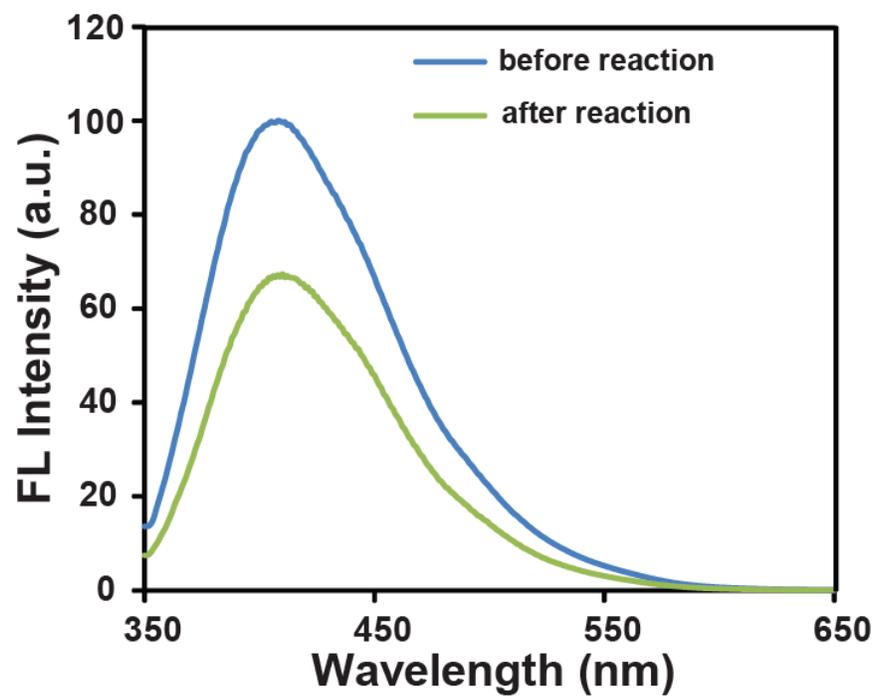


Figure S6 Fluorescence spectra of carbon dot-DBS-LDHs before and after reaction.

Table S1 Effects of radical scavengers on carbon dot-DBS-LDH-acidified H₂O₂ system^a.

Radical scavengers	Concentration (mM)	CL intensity^b
H ₂ O		100
ascorbic acid	10	28
thiourea	50	33.9
histidine	30	99.8
DABCO	30	97.1
NaN ₃	30	99.7
NBT	30	98.4

^a The experiments were carried out with flow injection system.

^b All of the results were the mean of three determinations.