

Supporting Information For:

Water-Soluble Pd Nanoparticles Synthesized from ω -Carboxyl-*S*-alkanethiosulfate Ligand Precursors as Unimolecular Micelle Catalysts

Diego J. Gavia, May S. Maung, and Young-Seok Shon*

Department of Chemistry and Biochemistry, California State University, Long Beach, 1250 Bellflower Blvd., Long Beach, CA 90840, United States

**For correspondence: Email: ys.shon@csulb.edu. Fax: (562) 985-8557.*

I.	Characterization of ω -Carboxyl- <i>S</i> -alkanethiosulfate Ligand	2
	Sodium ω -Carboxyl- <i>S</i> -undecanethiosulfate	2
	Sodium ω -Carboxyl- <i>S</i> -hexanethiosulfate	3
II.	^1H NMR Study of Thiosulfate Hydrolysis in Aqueous Solution	4
III.	Characterization of ws-MUA-PdNP	5
IV.	Characterization of ws-MHA-PdNP	8
V.	Post-Catalysis Characterization of ws-MHA-PdNP	10

I. Characterization of ω -Carboxyl-*S*-alkanethiosulfate Ligands

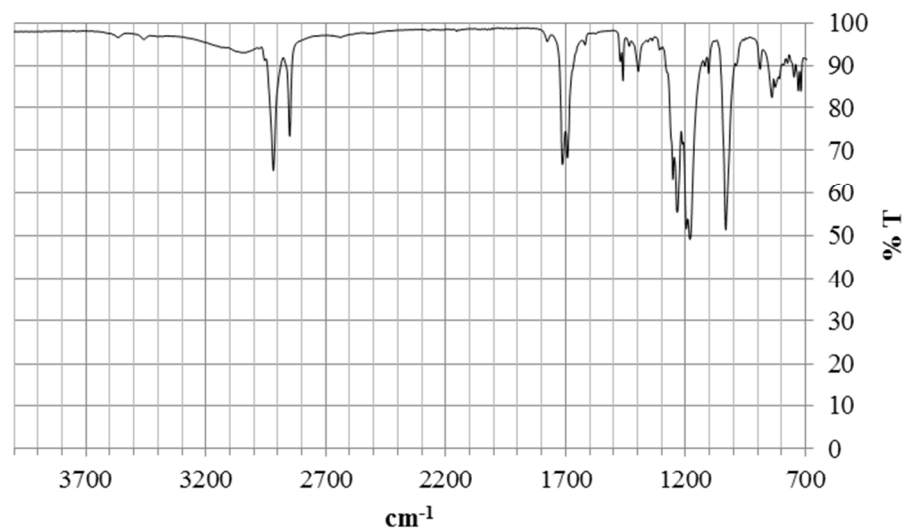


Figure S1. Infrared spectrum of sodium ω -carboxyl-*S*-undecanethiosulfate. The peaks at >3400 cm⁻¹ is due to C=O overtone.

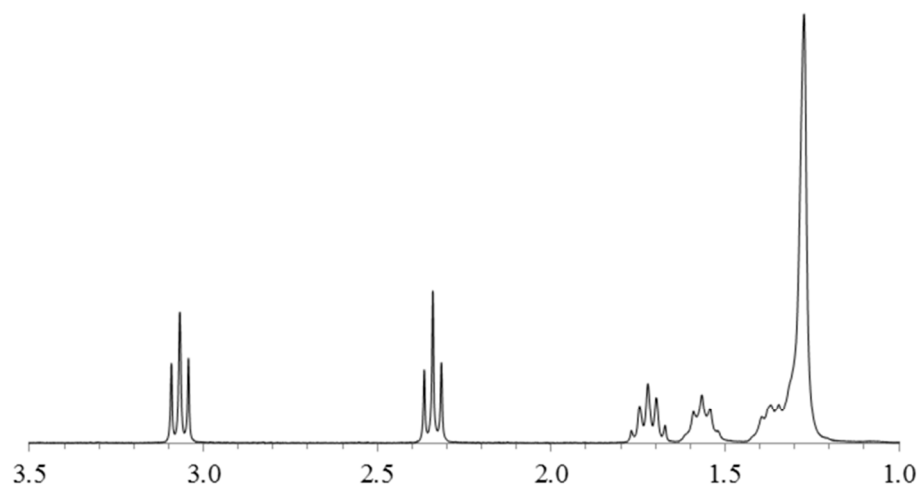


Figure S2. ¹H NMR spectrum of sodium ω -carboxyl-*S*-undecanethiosulfate in D₂O.

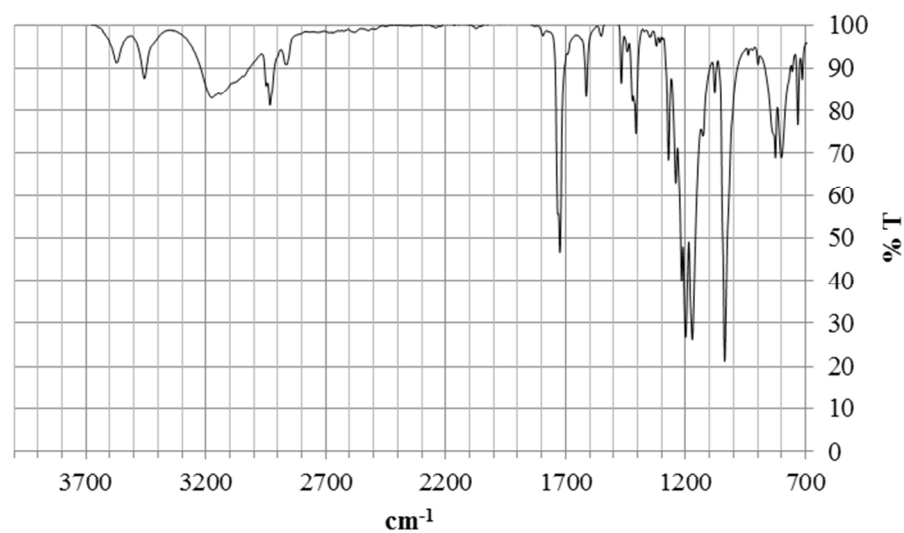


Figure S3. Infrared spectrum of sodium ω -carboxyl-*S*-hexanethiosulfate. The peaks at >3400 cm^{-1} is due to C=O overtone.

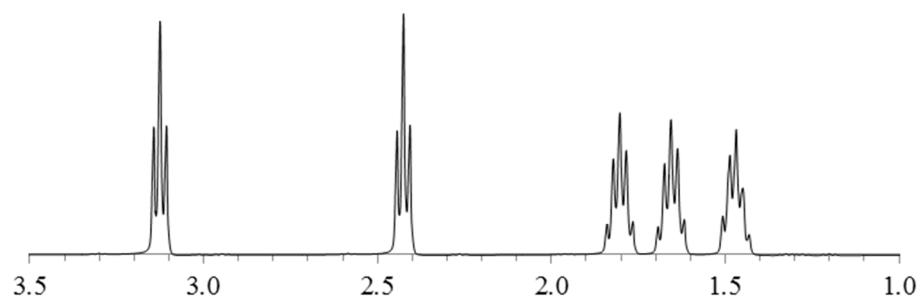


Figure S4. ^1H NMR spectrum of sodium ω -carboxyl-*S*-hexanethiosulfate in D_2O .

II. ^1H NMR Study of Thiosulfate Hydrolysis in Aqueous Solution

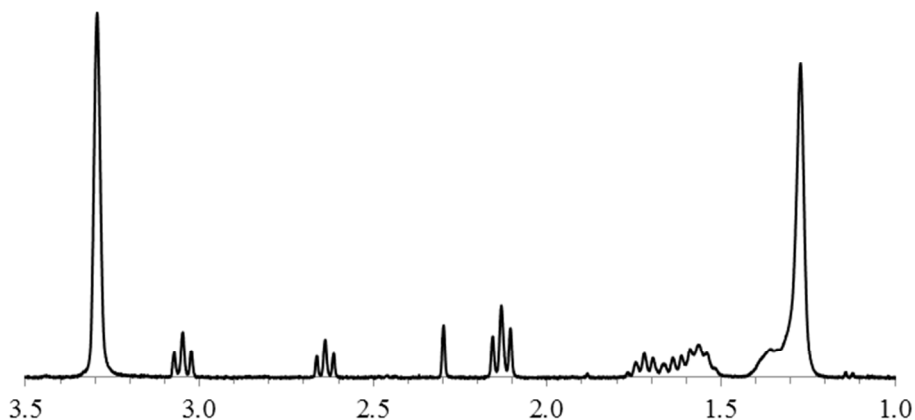


Figure S5. ^1H NMR spectra of sodium ω -carboxyl-*S*-undecanethiosulfate hydrolyzed by aqueous sodium borohydride solution in a one phase system of $\text{D}_2\text{O}/\text{MeOD}$. The resonances at 2.64 ppm indicate the formation of disulfide after the hydrolysis of thiosulfate.

III. Characterization of 11-Mercaptoundecanoic acid-capped Palladium Nanoparticles (ws-MUA-PdNP)

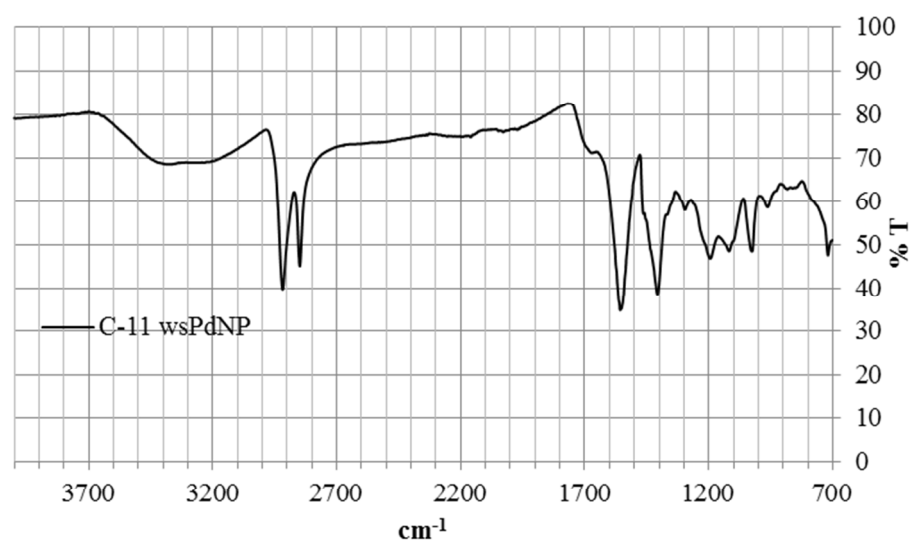


Figure S6. Infrared spectrum of ws-MUA-PdNP.

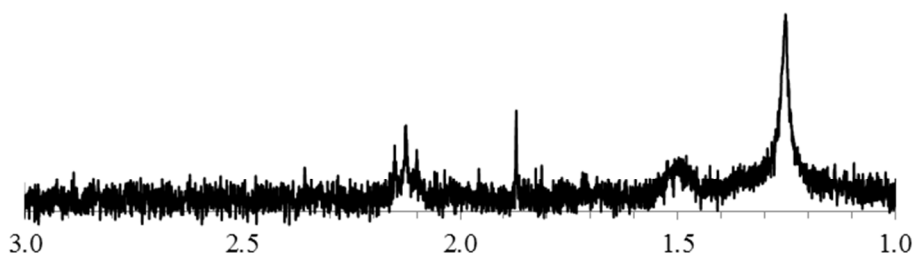


Figure S7. ¹H NMR spectrum of ws-MUA-PdNP.

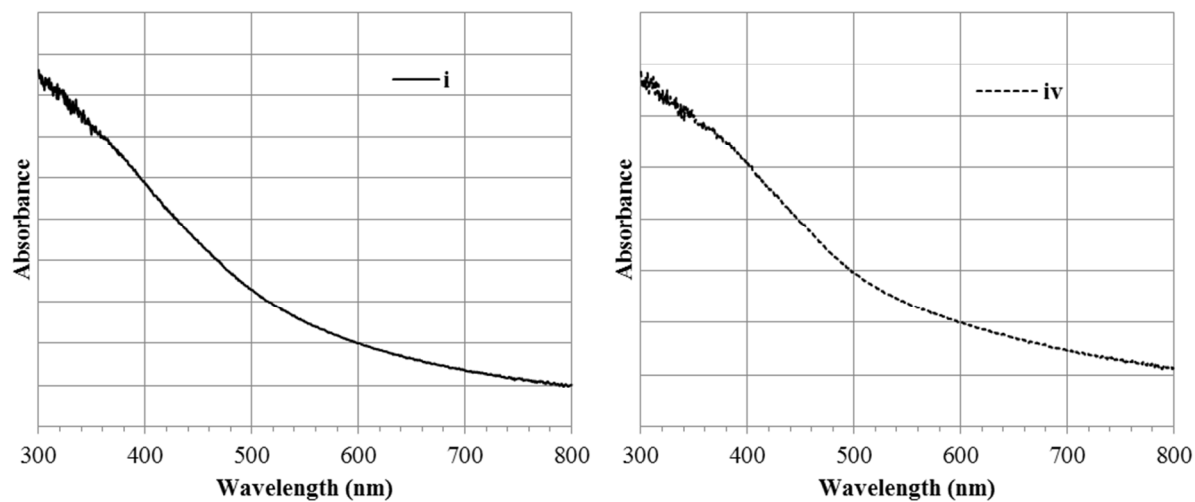


Figure S8. UV-Vis Spectra of ws-MUA-PdNPs (i) (left) and (iv) (right).

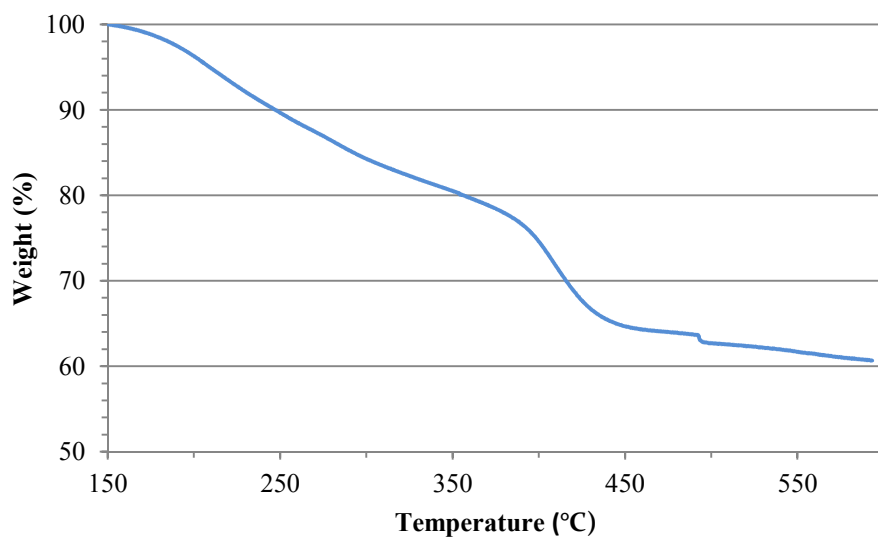


Figure S9. Thermogravimetric profile of ws-MUA-PdNP (i).

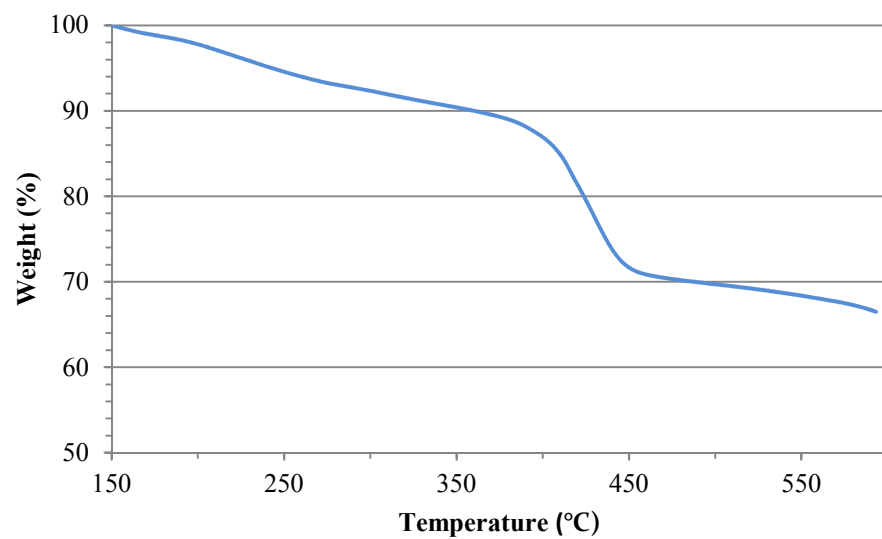


Figure S10. Thermogravimetric profile of ws-MUA-PdNP (iv).

IV. Characterization of 6-Mercaptohexanoic acid-capped Palladium Nanoparticles (ws-MHA-PdNP)

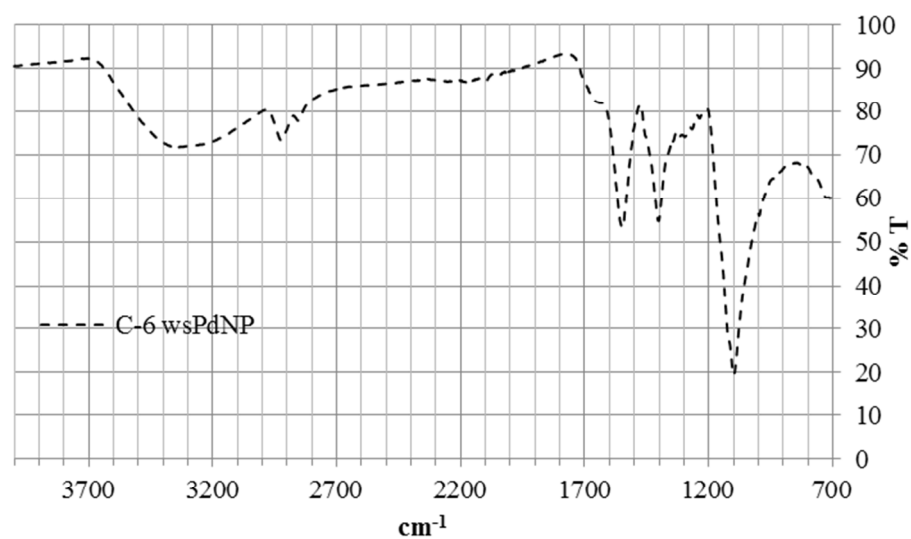


Figure S11. Infrared spectrum of ws-MHA-PdNP.

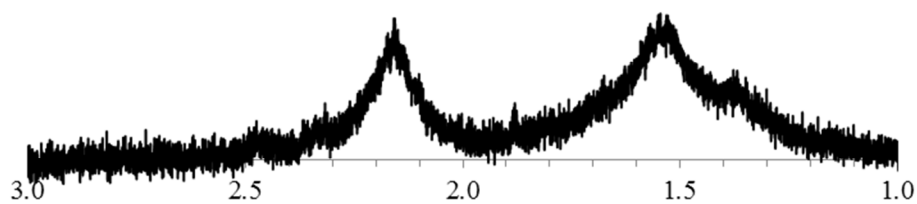


Figure S12. ¹H NMR spectrum of ws-MHA-PdNP in D₂O.

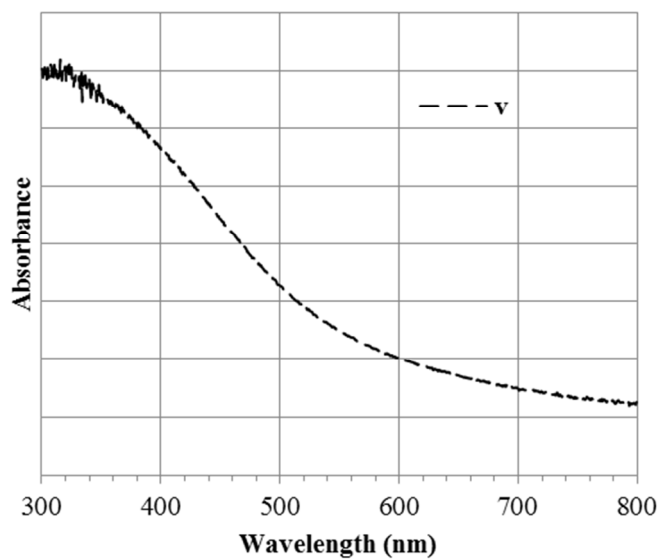


Figure S13. UV-Vis Spectra of ws-MHA-PdNP (v).

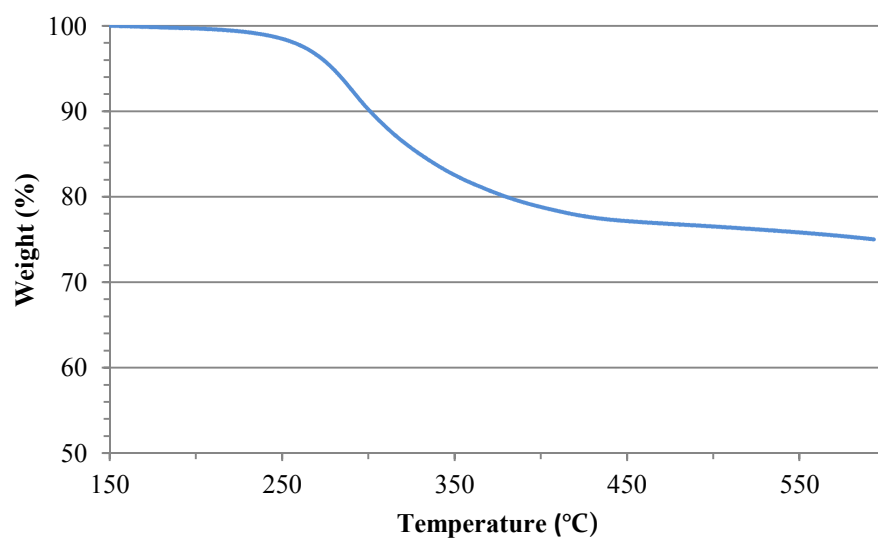


Figure S14. Thermogravimetric profile of ws-MHA-PdNP (v).

V. Post-Catalysis Characterization of 6-Mercaptohexanoic acid-capped Palladium Nanoparticles (ws-MHA-PdNP)

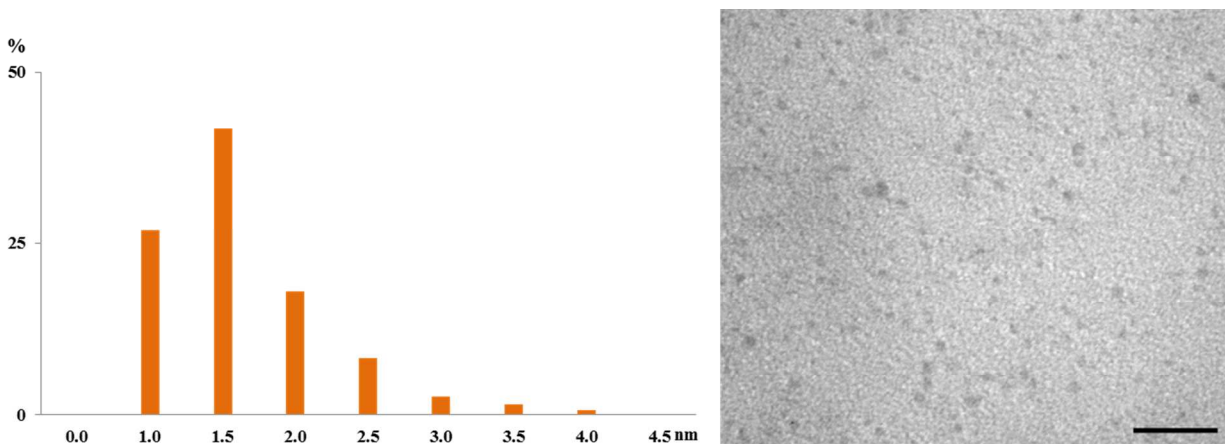


Figure S15. TEM image and size histogram of ws-MHA-PdNP (v) after 4 hour reaction. Histogram documents at least 400 counts (1.60 ± 0.56 nm). Scale bar in TEM image is 20 nm.

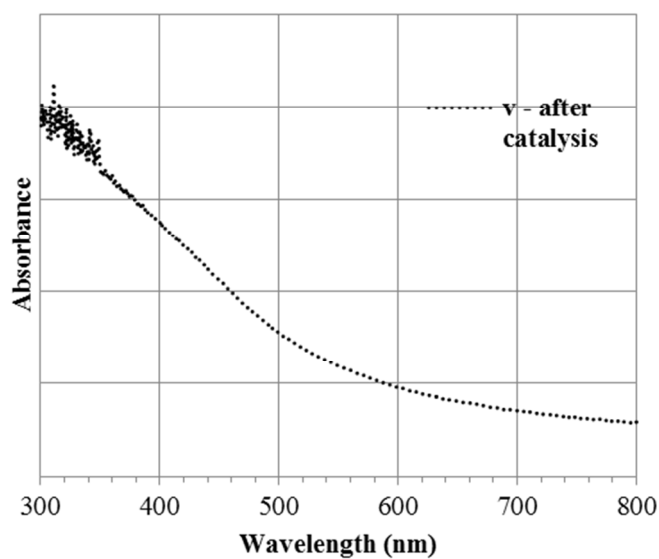


Figure S16. UV-vis spectrum of ws-MHA-PdNP (v) after 4 hour reaction.