

Quantifying Intrinsic Ion-driven Conformational Changes in Diphenylacetylene Supramolecular Switches with Cryogenic Ion Vibrational Spectroscopy

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Supporting Information

Supplemental Table 1. Comparison of Experimental ($\pm 4 \text{ cm}^{-1}$) and Calculated (B3LYP/6-31+G(d,p)) Frequencies for NH and CO Fundamentals of the Molecular Switch Complexes Characterized in This Study (cm^{-1})

complex	NH _a	NH _b	NH _c	amide CO ^a	ester CO
1-TMA ⁺ (D ₂) ₂					
experimental	3395	3335	3467	1701	1731
calculated ^b	3394	3342	3464	1704, 1694	1731
1-Na ⁺ (D ₂) ₂					
experimental	3432	3443	3464	1705, 1695	1747
calculated ^b	3450	3461	3464	1715, 1707	1747
2-Cl ⁻ (D ₂) ₂					
experimental	2744	3194 ^c	3357	1725, 1711	1749
calculated ^b	2886	3244	3374	1735, 1717	1750
1-Cl ⁻ (D ₂) ₂					
experimental	2867 ^c	3196 ^c	3357	1709	1753
calculated ^b	2982	3260	3376	1726, 1714	1750
1-Br ⁻ (D ₂) ₂					
experimental	2945 ^c	3208 ^c	3357	1710	1752
calculated ^b	3080	3293	3377	1725, 1717	1751
1-I ⁻ (D ₂) ₂					
experimental	3034 ^c	3231 ^c	3357	1712	1748
calculated ^b	3139	3321	3376	1725, 1717	1751

^aWhen the two amide CO functionalities are observed as a doublet, both frequencies are given.

^bThe empirically scaled calculated frequencies are given. Energies above 2000 cm^{-1} are scaled by .957 (to match the free NH frequency to that observed in 1-Na⁺(D₂)₂) and below 2000 cm^{-1} by .987 (to match the free CO frequency to that observed in 2-Na⁺(D₂)₂).

^cOccurs in a congested spectral region, approximate assignment given.