Fast Switching Water Processable Electrochromic Polymers

Pengjie Shi,[†] Chad M. Amb,[†] Aubrey L. Dyer,^{†‡} and John R. Reynolds^{*†‡}

\$School of Chemistry and Biochemistry, and School of Materials Science and Engineering, Center for Organic Photonics and Electronics, Georgia Institute of

Technology, Atlanta, GA 30332 (USA)

[†]The George and Josephine Butler Polymer Research Laboratory, Department of

Chemistry, Center for Macromolecular Science and Engineering, University of Florida,

Gainesville, Florida 32611 (USA)

*E-mail: reynolds@chemistry.gatech.edu

Table S1. GPC estimated Molecular Weights in THF, Elemental Analysis of the Polymers

	Monomer 5 ratio	Monomer 7 ratio (x)	Monomer 8 ratio (y)	M _n (g/mol)	M _w (g/mol)	PDI	EA (Calcd/Found)		
							С	Н	Ν
ECP-Blue-A	1	/	1	25,800	38,500	1.5	67.23/66.94	7.52/7.64	2.49/2.42
ECP-Blue-R	1	0.7	0.3	13,900	23,300	1.7	67.46/67.11	7.79/7.89	0.73/0.68

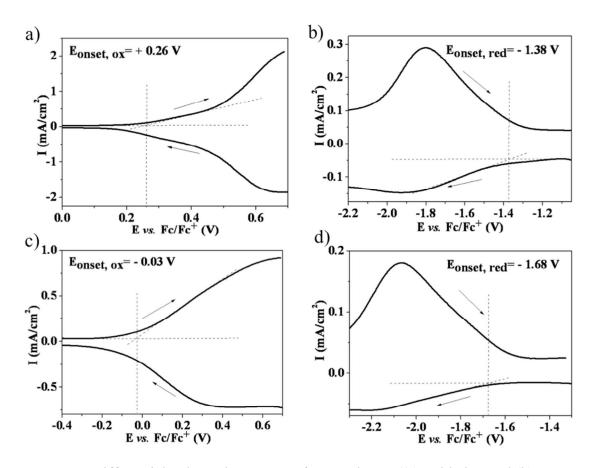


Figure S1. Differential pulse voltammetry of ECP-Blue-A ((a) oxidation and (b) reduction) and ECP-Blue-R ((c) oxidation and (d) reduction) drop-cast from toluene solution onto platinum button electrode (A= 0.02 cm^2). Measurements were performed in 0.2 M LiBTI /propylene carbonate (PC) with a Pt foil counter electrode and a Ag/Ag⁺ reference electrode calibrated vs Fc/Fc⁺ (E_{Fc/Fc+} = $0.085 \text{ V vs Ag/Ag}^+$).

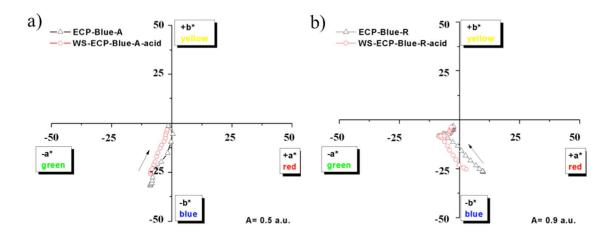


Figure S2.Plot of CIE 1976 a^*b^* color coordinates showing calculated values of (a) ECP-Blue-A and WS-ECP-Blue-A-acid, (b) ECP-Blue-R and WS-ECP-Blue-R-acid as a function of applied potential.