

Supporting Information

Infrared laser spectroscopy of the CH_3OO radical formed from the reaction of CH_3 and O_2 within a helium nanodroplet

Alexander M. Morrison,^b Jay Agarwal,^a H. F. Schaefer III^a and Gary E. Douberly^{b,*}

^a Center for Computational Chemistry, University of Georgia, Athens, GA 30602

^b Department of Chemistry, University of Georgia, Athens, GA 30602

- Corresponding Author: douberly@uga.edu

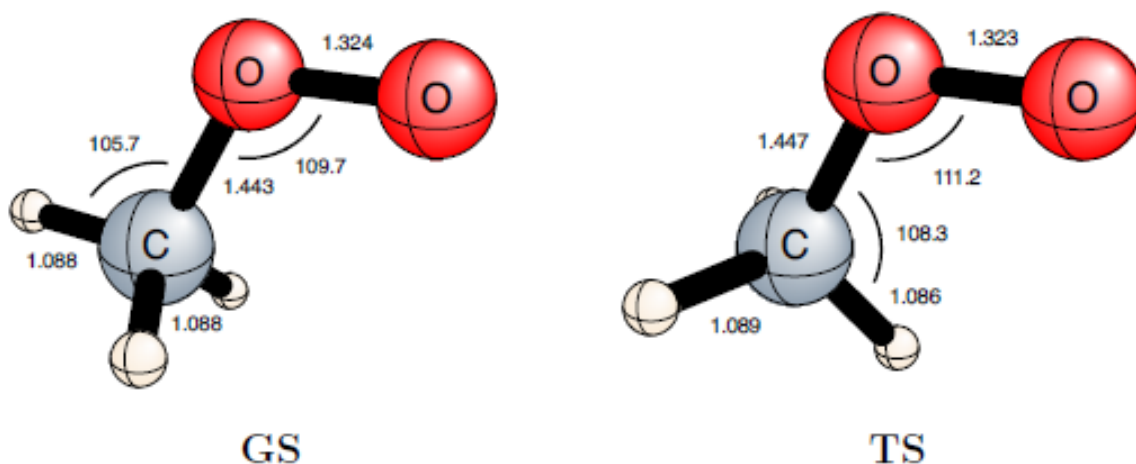


Figure S1: Optimized geometry for the ground-state and transition-state at the CCSD(T)/cc-pVTZ level of theory. Bond lengths are shown in Angstroms and bond angles in degrees.

Supporting Information

Table S1: Focal Point analysis of the CH₃OO torsional barrier in the (X)²A'' electronic state.^a

Basis set	ΔE_e HF	+ δ MP2	+ δ CCSD	+ δ CCSD(T)	NET
cc-pVTZ	+1.36	-0.66	+0.33	-0.08	[+0.95]
cc-pVQZ	+1.38	-0.69	+0.33	-0.08	[+0.94]
cc-pV5Z	+1.38	-0.69	[+0.33]	[-0.08]	[+0.94]
CBS LIMIT	[+1.38]	[-0.70]	[+0.33]	[-0.08]	[+0.93]
$\Delta E_t(\text{final}) = \Delta E_t[\text{CBS CCSD(T)}] + \Delta_{\text{ZPVE}}[\text{CCSD(T)/cc-pVTZ}]$					
$\Delta E_t(\text{final}) = 0.93 - 0.21 = \mathbf{0.72 \text{ kcal mol}^{-1}}$					

^aEnergies shown in kcal mol⁻¹. Delta (δ) denotes the change in relative energy (ΔE_e) with respect to the preceding level of theory.

Note: The barrier for torsional rotation (0.72 kcal mol⁻¹), has a suggested accuracy of ± 0.1 kcal mol⁻¹. This can be deduced from looking at the focal-point table, where we see oscillatory convergence to within 0.08 kcal mol⁻¹.

Ground-State: (Cartesian Coordinates)

H -1.97621996614	0.487796731042	0.0
C -1.1661393889	-0.238632305755	0.0
O 0.0526172527545	0.534467925974	0.0
O 1.09753254814	-0.278193009495	0.0
H -1.19623505374	-0.856855813713	0.895207315689
H -1.19623505374	-0.856855813713	-0.895207320981

Ground-State Frequencies [CCSD(T)/cc-pVTZ]:

0.0000i
0.0000i
0.0000i
0.0000i
0.0000
0.0000
149.7274
493.5895
949.3199
1144.0846
1160.2828
1212.0275
1449.0832
1484.1431
1497.6945
3061.9382
3159.3911
3168.8276

ZPVE: 27.0619 kcal/mol

Supporting Information

Transition State: (Cartesian Coordinates)

H	-0.925965420482	1.29146315085	0.0
C	-1.17814575391	0.234897748448	0.0
O	0.053280041967	-0.525593962184	0.0
O	1.10782046825	0.272944428453	0.0
H	-1.73678128951	-0.0393094821325	-0.893941920592
H	-1.73678128951	-0.0393094821325	0.893941920592

Transition-State Frequencies [CCSD(T)/cc-pVTZ]:

142.6587i

0.0000i

0.0000i

0.0000i

0.0000

0.0000

0.0000

533.7710

941.3993

1126.2488

1138.5946

1229.6803

1437.7611

1486.6697

1496.9888

3059.1017

3148.8551

3182.0058

ZPVE: 26.8489 kcal/mol

Supporting Information

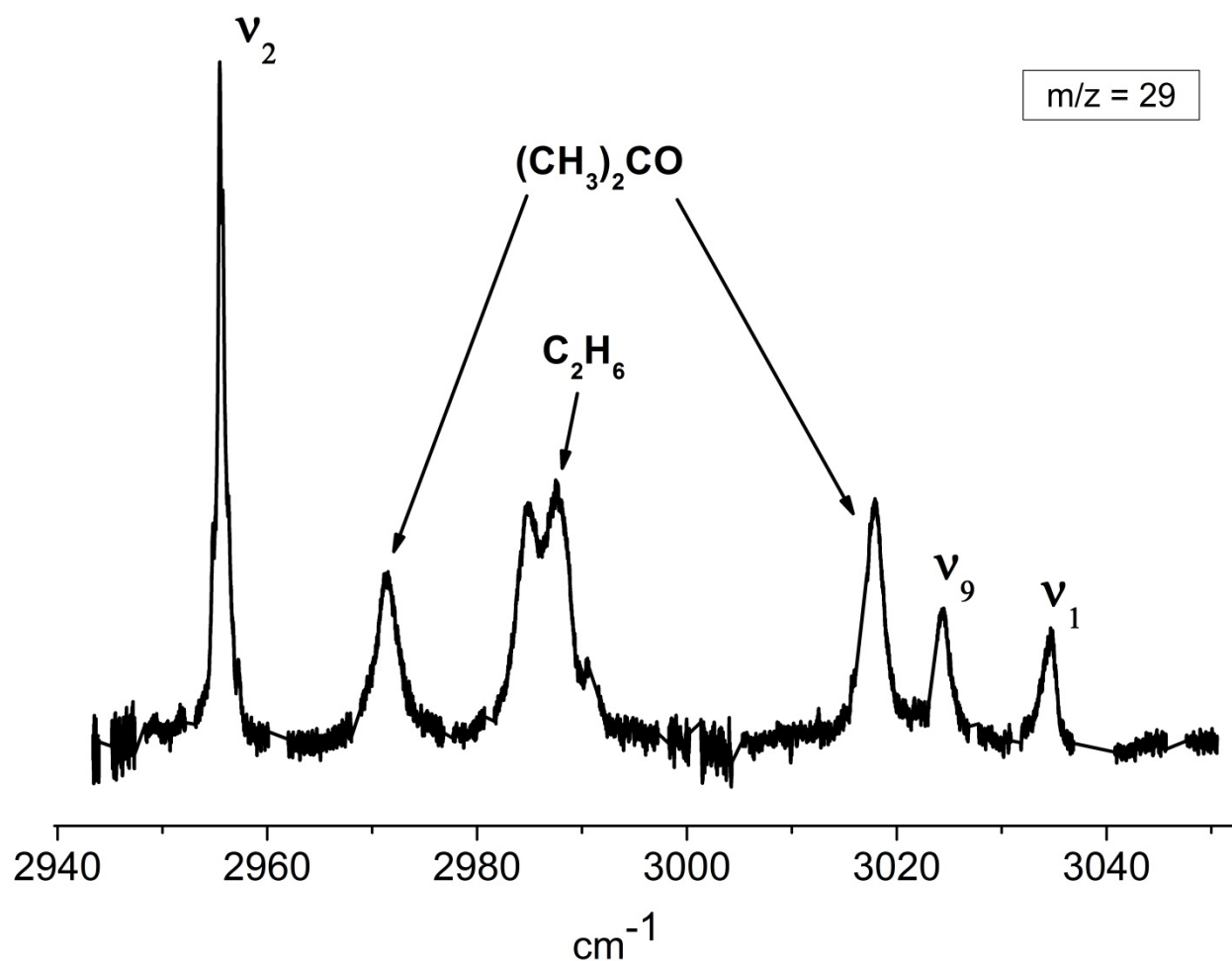


Figure S2: Survey Spectrum of the CH stretch region by measuring the depletion in mass channel $m/z = 29$. Two acetone bands as well as an ethane band are observed in addition to the three transitions reported in this manuscript for CH_3O_2 .