Supporting Information

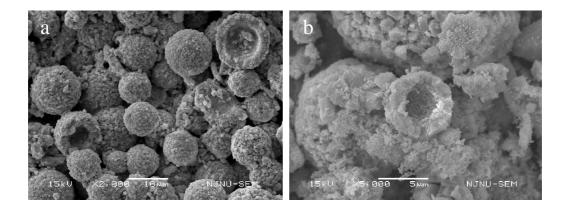
A Facile Preparation Method of Rare Earth Phosphate Hollow Spheres

and Their Photoluminescent Properties

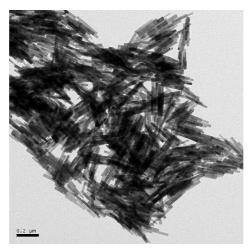
Mingyun Guan,^{†,‡} Feifei Tao,[†] Jianhua Sun,[†] and Zheng Xu[†]*

State Key Laboratory of Coordination Chemistry, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing 210093, P. R. China. School of Chemistry and Chemical Engineering, Jiangsu Teachers University of Technology, Changzhou, Jiangsu 213001, P. R. China.

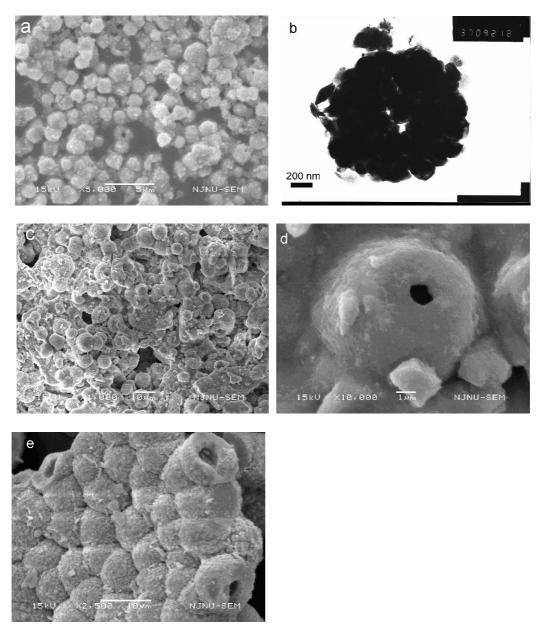
SI-1. SEM images of Tb^{3+} or Dy^{3+} doped GdPO₄•H₂O hollow spheres



- (a) SEM image of Tb^{3+} doped GdPO₄•H₂O hollow spheres. (b) SEM image of Dy^{3+} doped GdPO₄•H₂O hollow spheres.
- SI-2. TEM image of EuPO₄•H₂O nanorod prepared at the same condition as experimental section, except using H_3PO_4 instead of $H_6P_4O_{13}$.

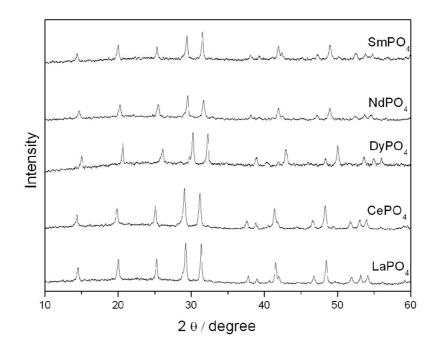


SI-3. SEM images of DyPO₄•1.5H₂O, LaPO₄•nH₂O, NdPO₄ hollow spheres



(a) SEM image of DyPO₄•1.5H₂O hollow microspheres; (b) TEM image of DyPO₄•1.5H₂O hollow microspheres; (c), (d) SEM images of LaPO₄•nH₂O hollow microspheres and an enlargement image of individual hollow sphere; (e) SEM image of NdPO₄ hollow microspheres.

SI-4. XRD pattern of the different samples



XRD patterns of SmPO₄ (JCPDS No.34-0537), NdPO₄ (JCPDS No. 04-0644), DyPO₄•1.5H₂O (JCPDS No. 21-0316), CePO₄ (JCPDS No. 34-1380), LaPO₄•nH₂O (JCPDS No. 46-1439) hollow microspheres.