

Figure S1: Optimized structures and relative energies of VO₃, VO₂CH₂, VO₂CHCH₃, and VO₂C₃H₄ in the doublet (M=2) and quartet (M=4) spin multiplicities. The energies (in eV) are given as ($\Delta H_{0 \text{ K}} / \Delta G_{298 \text{ K}}$) below each geometry. The bond lengths in 0.1 nm are given.

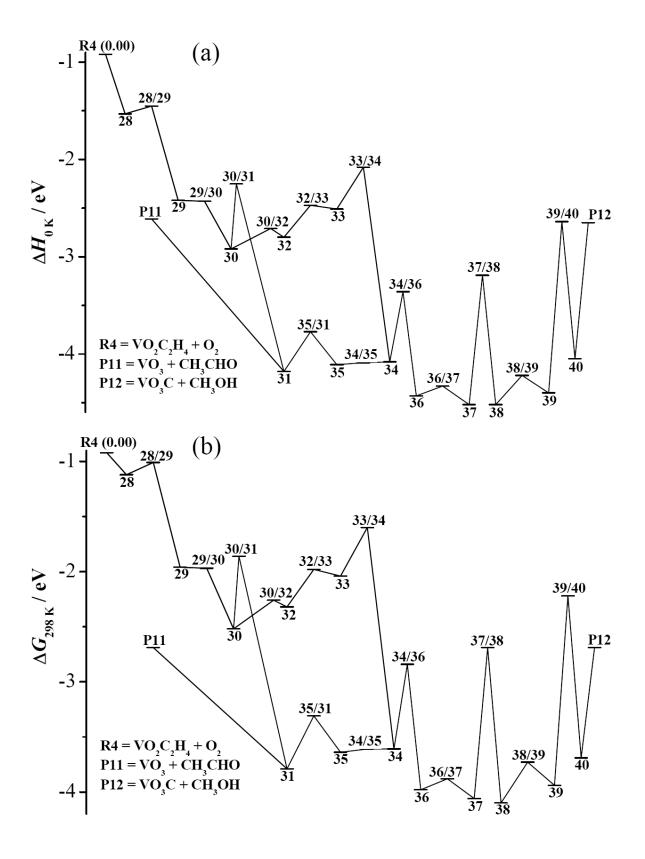


Figure S2: Potential energy profiles for the reaction of VO₂CHCH₃ with O₂. The profiles are plotted for zero-point vibration corrected energies (a) and Gibbs free energies at 298 K (b) relative to the separated reactants VO₂CHCH₃ + O₂. The reaction intermediates and transition states are denoted as n and n_1/n_2 , respectively.

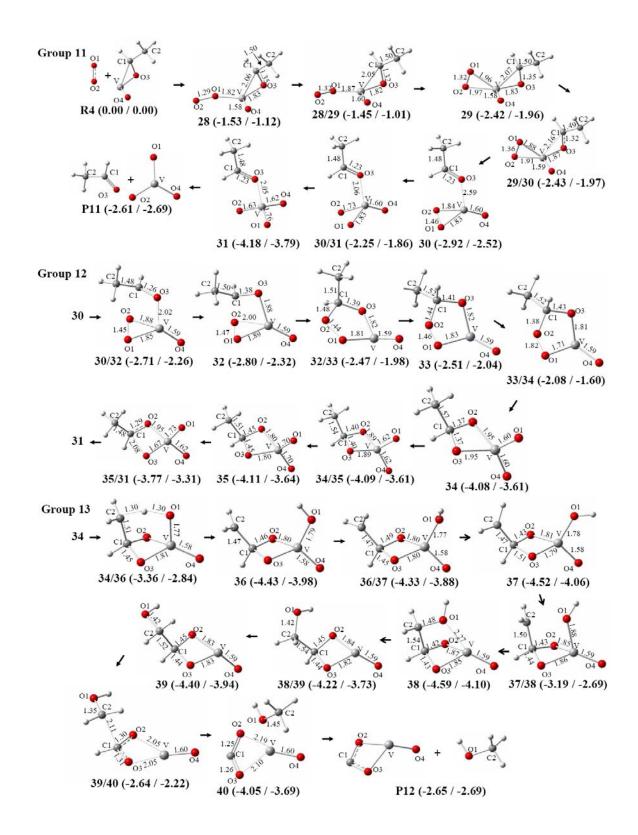


Figure S3: Optimized structures and energies of the species in Figure S2. The energies (in eV) are given as $(\Delta H_{0 \text{ K}} / \Delta G_{298 \text{ K}})$ below each geometry. The bond lengths in 0.1 nm and some bond angles in degree are given.