# Synthesis of $\alpha$-Hydroxy Acids by Silylene Transfer to $\alpha$-Keto Esters 

Brett E. Howard and K. A. Woerpel ${ }^{*}$<br>Department of Chemistry, University of California, Irvine, California 92697-2025

## Supporting Information

## Contents:

I. Synthesis of Substituted $\alpha$-Keto Esters ..... $S$-2
II. Synthesis of $\alpha$-Hydroxy Acids ..... $S-5$
III. Chiral $\alpha$-Keto Ester Synthetic Intermediates and Products ..... $S-8$
IV. $\alpha$-Imino Ester Synthesis and Silylene Transfer ..... $S-12$
V. References ..... $S$-13
VI. X-ray Crystallographic Data ..... $S$-13
VII. Selected Spectra ..... S-28

## Experimental Section

General. Melting points were obtained using a Büchi 510 melting point apparatus and are reported uncorrected. Chiral gas chromatography was performed using a Hewlett-Packard 5890 series II gas chromatograph with a G-TA $\gamma$-cyclodextrin, trifluoroacetyl column ( $20 \mathrm{~m} \times 0.25 \mathrm{~mm}, 100: 1$ split ratio), with helium as the carrier gas. Chiral HPLC was accomplished using a Varian Prostar 320 UV-Vis detector in combination with 2 Varian Prostar 210 solvent delivery modules and a Daicel Chemical Industries LTD. Chiralcel OD-H column ( $25 \mathrm{~cm} \times 0.46 \mathrm{~cm}$ ). Method of chromatography is as follows: 199:1 hexanes:IPA, flow rate $0.6 \mathrm{~mL} / \mathrm{min}$. Analytical thin later chromatography was performed on EMD Silical Gel $60 \mathrm{~F}_{254}$ precoated plates. Liquid chromatography utilized force flow (flash chromatography) of the indicated solvent system on Silacycle Sila-P silica gel $\left(\mathrm{SiO}_{2}\right) 60 \AA$ pore size, $40-63 \mu \mathrm{~m}$ mesh. Infrared spectroscopy was performed on an Applied Systems React IR 1000. High-resolution mass spectra were acquired on a Walters LCT Premier and were obtained by peak matching. Microanalyses were performed by Atlantic Microlabs, Atlanta, GA. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were recorded at $25^{\circ} \mathrm{C}$ at 400 and 100 , and 500 and 125 MHz respectively, using Bruker DRX 400 or DRX 500 spectrometers as indicated. These data are reported as follows: chemical shift in ppm from internal tetramethylsilane on the $\delta$ scale, multiciplity ( $\mathrm{br}=\mathrm{broad}, \mathrm{s}=$ singlet, $\mathrm{d}=$ doublet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, $\mathrm{m}=$ multiplet $)$, coupling constants $(\mathrm{Hz})$, and integration. Silacyclopropanes were stored and manipulated in an Innovative Technologies nitrogen-atmosphere dry box. All reactions were performed under an atmosphere of nitrogen or argon in glassware that had been flame-dried under vacuum prior to use. Solvents were distilled or filtered before use. Cyclohexenesilacyclopropane (6) was constructed by known methods. ${ }^{1}$


5

1,1-di-tert-butyl-2,2-dimethylsilirane (5). To a cooled ( $-78^{\circ} \mathrm{C}$ ) solution of lithium granules ( $3.9 \mathrm{~g}, 470$ mmol ) in THF ( 23 mL ), isobutylene ( $14.5 \mathrm{~g}, 259 \mathrm{mmol}$ ) and di-tert-butyldichlorosilane ( $10 \mathrm{~g}, 47 \mathrm{mmol}$ ) were added. After warming to room temperature and stirring for 72 h , the reaction mixture was concentrated in vacuo and filtered through a Celite pad using Schlenk technique. The Celite was rinsed with hexanes ( $4 \times 50 \mathrm{~mL}$ ) and the filtrate was concentrated. The reaction mixture was then purified by bulb to bulb distillation to give 5 ( $4.7 \mathrm{~g}, 50 \%$ yield) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.54(\mathrm{~s}, 6 \mathrm{H}), 1.24(\mathrm{~s}, 18 \mathrm{H}), 0.66(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 32.0,29.2,20.5,17.9,15.3$; IR (thin film) $2936,2857,1475 \mathrm{~cm}^{-1}$.

## I. Synthesis of Substituted $\alpha$-Keto Esters

General Coupling Procedure for the Formation of $\boldsymbol{\alpha}$-Keto Esters: To a solution of the $\alpha$-hydroxy acid (1.5-2.0 equiv) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 0.1 M ) was added the allylic alcohol ( 1.0 equiv) and DMAP ( 0.1 equiv). The solution was cooled to $0^{\circ} \mathrm{C}$ and stirred for 15 min prior to the addition of DCC (1.5-2.0 equiv). After stirring for $3-6 \mathrm{~h}$, the reaction mixture was filtered, diluted with 20 mL of saturated aqueous $\mathrm{NaHCO}_{3}$, and extracted three times with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 30 mL per mmol of alcohol). The organic layers were combined, dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated in vacuo. The resulting oil was purified as indicated.


1

Allyl 2-oxo-2-phenylacetate (1). The general coupling procedure was performed on benzoyl formic acid (3.30 $\mathrm{g}, 22.0 \mathrm{mmol})$ and allyl alcohol ( $0.748 \mathrm{~mL}, 11.0 \mathrm{mmol}$ ). Purification of the concentrated reaction mixture by flash chromatography ( $10: 1$ hexanes:EtOAc) gave $1(1.01 \mathrm{~g}, 48 \%)$ as a yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.00-7.97(\mathrm{~m}, 2 \mathrm{H}), 7.63-7.60(\mathrm{~m}, 1 \mathrm{H}), 7.49-7.46(\mathrm{~m}, 2 \mathrm{H}), 5.99(\mathrm{ddt}, J=17.0,10.4,5.9,1 \mathrm{H}), 5.42(\mathrm{~m}, 1 \mathrm{H})$, $5.31(\mathrm{~m}, 1 \mathrm{H}), 4.85(\mathrm{dt}, J=5.9,1.3,2 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 186.3,163.7,135.2,132.6,131.0$, 130.2, 129.2, 120.1, 66.7; IR (thin film) 3070, 1737, 1686, 1598, 1451, 1194, $1175 \mathrm{~cm}^{-1}$; HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$213.0528, found 213.0529. Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{O}_{3}: \mathrm{C}, 69.46 ; \mathrm{H}, 5.30$. Found: C, 69.10; H, 5.26.


Cinnamyl 2-oxopropanoate (6a). The general coupling procedure was performed on pyruvic acid ( 0.557 mL , 8.00 mmol ) and cinnamyl alcohol ( $0.537 \mathrm{~g}, 4.00 \mathrm{mmol}$ ). Purification of the concentrated reaction mixture by flash chromatography ( $10: 1$ hexanes:EtOAc) gave $\mathbf{6 a}\left(0.434 \mathrm{~g}, 53 \%\right.$ ) as a yellow oil: ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.42-7.27(\mathrm{~m}, 5 \mathrm{H}), 6.74(\mathrm{~d}, J=15.9,1 \mathrm{H}), 6.33(\mathrm{dt}, J=15.6,6.7,1 \mathrm{H}), 4.91(\mathrm{dd}, J=6.7,1.1,2 \mathrm{H}), 2.50$ (s, 3H); ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 192.0,160.7,136.3,136.0,128.9,128.7,127.0,121.6,67.2,27.0$; IR (thin film) 3369 (br), 2939, 1727, 1677, $1135 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$ 227.0684, found 227.0687. Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{O}_{3}$ : C, 70.57; H, 5.92. Found: C, 70.56; H, 6.02.


6b
Cinnamyl 2-oxobutanoate (6b). The general coupling procedure was performed on 2-ketobutyric acid ( 0.817 $\mathrm{g}, 8.00 \mathrm{mmol})$ and cinnamyl alcohol ( $0.537 \mathrm{~g}, 4.00 \mathrm{mmol}$ ). Purification of the concentrated reaction mixture by column chromatography ( $10: 1$ hexanes:EtOAc) gave $\mathbf{6 b}\left(0.667 \mathrm{~g}, 76 \%\right.$ ) as a pale yellow oil: ${ }^{1} \mathrm{H}$ NMR ( 500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.41-7.27(\mathrm{~m}, 5 \mathrm{H}), 6.72(\mathrm{~d}, J=15.9,1 \mathrm{H}), 6.32(\mathrm{dt}, J=15.9,6.6,1 \mathrm{H}), 4.9(\mathrm{dd}, J=6.6,1.1,2 \mathrm{H})$, $2.88(\mathrm{q}, J=7.2,2 \mathrm{H}), 1.13(\mathrm{t}, J=7.2,3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 194.3,160.5,135.5,135.2,128.3$, 128.0, 126.4, 121.3, 66.2, 32.4, 6.5; IR (thin film) 3451, 2940, $1731,1269 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$241.0841, found 241.0845.


Cinnamyl 3-methyl-2-oxobutanoate ( $\mathbf{6 c}$ ). The general coupling procedure was performed on 2-ketovaline ( $0.318 \mathrm{~g}, 2.75 \mathrm{mmol}$ ) and cinnamyl alcohol ( $0.184 \mathrm{~g}, 1.37 \mathrm{mmol}$ ). Purification of the concentrated reaction mixture by column chromatography ( $10: 1$ hexanes:EtOAc) gave $\mathbf{6 c}\left(0.254 \mathrm{~g}, 80 \%\right.$ ) as a light yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.42-7.27(\mathrm{~m}, 5 \mathrm{H}), 6.74(\mathrm{~d}, J=15.9,1 \mathrm{H}), 6.34(\mathrm{~m}, 1 \mathrm{H}), 4.92(\mathrm{~d}, J=6.7,2 \mathrm{H}), 3.30$ (septet, $J=7.0,1 \mathrm{H}), 1.19(\mathrm{dd}, J=7.0,0.7,6 \mathrm{H}){ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 198.0,161.5,135.85,135.79$, 128.6, 128.4, 126.7, 121.5, 66.6, 37.1, 17.1; IR (thin film) 2977, 2937, 1729, 1449, 1264, $1027 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+} 255.0997$, found 255.0996. Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{O}_{3}$ : C, 72.39; H , 6.94. Found: C, 72.32; H, 7.08.


Cinnamyl 3,3-dimethyl-2-oxobutanoate (6d). The general coupling procedure was performed on 3,3-dimethyl-2-oxobutanoic acid ( $0.484 \mathrm{~g}, 3.73 \mathrm{mmol}$ ) and cinnamyl alcohol ( $0.250 \mathrm{~g}, 1.86 \mathrm{mmol}$ ). Purification of the concentrated reaction mixture by column chromatography ( $10: 1$ hexanes:EtOAc) gave $\mathbf{6 d}(0.408 \mathrm{~g}, 90 \%)$ as a white solid: mp $45{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.43-7.30(\mathrm{~m}, 5 \mathrm{H}), 6.74(\mathrm{~d}, J=15.9,1 \mathrm{H}), 6.33(\mathrm{dt}, J=$ $15.9,6.7,1 \mathrm{H}), 4.91(\mathrm{dd}, J=6.7,1.2,2 \mathrm{H}), 1.29(\mathrm{~s}, 9 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 202.1,163.8,136.1$, 128.9, 128.7, 127.0, 121.8, 66.4, 43.0, 26.0; IR (thin film) 2973, 1733, 1717, 1480, 1449, 1287, $1233 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$269.1154, found 269.1157. Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{O}_{3}$ : C, 73.15; H, 7.37. Found: C, 73.44; H, 7.40.


Cinnamyl 2-0x0-2-phenylacetate (6e). The general coupling procedure was performed on benzoylformic acid $(0.60 \mathrm{~g}, 4.0 \mathrm{mmol})$ and cinnamyl alcohol $(0.268 \mathrm{~g}, 2.00 \mathrm{mmol})$. Purification of the concentrated reaction
mixture by column chromatography ( $10: 1$ hexanes:EtOAc) gave $\mathbf{6 e}\left(0.481 \mathrm{~g}, 90 \%\right.$ ) as a yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.07-8.04(\mathrm{~m}, 2 \mathrm{H}), 7.70-7.65(\mathrm{~m}, 1 \mathrm{H}), 7.55-7.50(\mathrm{~m}, 2 \mathrm{H}), 7.47-7.41(\mathrm{~m}, 2 \mathrm{H}), 7.38-7.28$ $(\mathrm{m}, 3 \mathrm{H}), 6.80(\mathrm{~d}, J=15.9,1 \mathrm{H}), 6.40(\mathrm{dt}, J=15.9,6.6,1 \mathrm{H}), 5.06(\mathrm{dd}, J=6.6,1.2,2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $(125 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 186.4,163.8,136.2,136.1,135.2,132.7,130.3,129.2,128.9,128.7,127.0,121.8,66.9$; IR (thin film) 3060, 3028, 1735, 1686, 1596, 1449, $1195 \mathrm{~cm}^{-1}$; HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$289.0841, found 289.0837. Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{O}_{3}$ : C, 76.68; H, 5.30. Found: C, 76.25; H, 5.27.

$6 f$

Crotyl 2-oxo-2-phenylacetate (6f). The general coupling procedure was performed on benzoyl formic acid $(3.36 \mathrm{~g}, 22.4 \mathrm{mmol})$ and crotyl alcohol $(0.941 \mathrm{~mL}, 11.2 \mathrm{mmol})$. Purification of the concentrated reaction mixture by column chromatography ( $10: 1$ hexanes:EtOAc) gave $\mathbf{6 f}\left(1.11 \mathrm{~g}, 49 \%\right.$ ) as a yellow oil: ${ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.96-7.94(\mathrm{~m}, 2 \mathrm{H}), 7.61-7.56(\mathrm{~m}, 1 \mathrm{H}), 7.46-7.42(\mathrm{~m}, 2 \mathrm{H}), 5.91-5.82(\mathrm{~m}, 1 \mathrm{H}), 5.64(\mathrm{dtq}, J=$ $15.1,6.7,1.6,1 \mathrm{H}), 4.76(\mathrm{dt}, J=6.8,1.2,2 \mathrm{H}), 1.70-1.67(\mathrm{~m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 186.5,163.9$, $135.1,133.7,132.7,130.2,129.1,124.0,67.0,18.0$; IR (thin film) $3031,1735,1686,1598,1451,1194,1173$ $\mathrm{cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$227.0684, found 227.0685. Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{O}_{3}$ : C, 70.57; H, 5.92. Found: C, 70.59; H, 6.00.

( $\boldsymbol{E}$ )-Hept-2-enyl 2-oxo-2-phenylacetate ( $\mathbf{6 g}$ ). The general coupling procedure was performed on benzoyl formic acid ( $0.500 \mathrm{~g}, 3.33 \mathrm{mmol}$ ) and ( $E$ )-hept-2-en-1-ol ( $0.253 \mathrm{~g}, 2.22 \mathrm{mmol}$ ). Purification of the concentrated reaction mixture by column chromatography ( $20: 1$ hexanes:EtOAc) gave $\mathbf{6 g}(0.497 \mathrm{~g}, 91 \%$ ) as a light yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.03-8.00(\mathrm{~m}, 2 \mathrm{H}), 7.69-7.64(\mathrm{~m}, 1 \mathrm{H}), 7.55-7.49(\mathrm{~m}, 2 \mathrm{H}), 5.96-5.88(\mathrm{~m}$, $1 \mathrm{H}), 5.75-5.64(\mathrm{~m}, 1 \mathrm{H}), 4.83(\mathrm{~d}, J=6.7,2 \mathrm{H}), 2.10(\mathrm{q}, J=6.9,2 \mathrm{H}), 1.45-1.28(\mathrm{~m}, 4 \mathrm{H}), 0.91(\mathrm{t}, J=7.1,3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 186.6,163.9,139.0,135.1,132.7,130.3,129.1,122.7,67.2,32.2,31.1,22.4$, 14.1; IR (thin film) 2958, 2931, 1735, 1688, 1598, 1451, $1194 \mathrm{~cm}^{-1}$; HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{NaO}_{3}$ $(\mathrm{M}+\mathrm{Na})^{+}$269.1154, found 269.1154. Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{O}_{3}: \mathrm{C}, 73.15 ; \mathrm{H}, 7.37$. Found: C, 73.39; H, 7.42.


6h
(E)-4-(tert-butyldimethylsilyloxy)but-2-enyl 2-oxo-2-phenylacetate ( 6 h ). The general coupling procedure was performed on benzoyl formic acid $(0.500 \mathrm{~g}, 3.33 \mathrm{mmol})$ and ( $E$ )-4-(tert-butyldimethylsilyloxy)but-2-en-1ol ( $0.449 \mathrm{~g}, 2.22 \mathrm{mmol}$ ). Purification of the concentrated reaction mixture by column chromatography ( $20: 1$ hexanes:EtOAc) gave $\mathbf{6 h}(0.676 \mathrm{~g}, 91 \%)$ as a light yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.04-8.00(\mathrm{~m}$, $2 \mathrm{H}), 7.70-7.64(\mathrm{~m}, 1 \mathrm{H}), 7.55-7.48(\mathrm{~m}, 2 \mathrm{H}), 6.04-5.87(\mathrm{~m}, 2 \mathrm{H}), 4.90(\mathrm{~d}, J=5.6,2 \mathrm{H}), 4.23(\mathrm{dd}, J=3.7,1.3$, 2H), $0.92(\mathrm{~s}, 9 \mathrm{H}), 0.09(\mathrm{~s}, 6 \mathrm{H}){ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 186.4,163.8,136.3,135.1,132.7,130.3,129.1$,

$6 i$
(E)-5-(benzyloxy)pent-2-enyl 2-oxobutanoate (6i). The general coupling procedure was performed on 2ketobutyric acid ( $0.153 \mathrm{~g}, 1.50 \mathrm{mmol}$ ) and ( $E$ )-5-(benzyloxy)pent-2-en-1-ol ( $0.192 \mathrm{~g}, 1.00 \mathrm{mmol}$ ). Purification of the concentrated reaction mixture by column chromatography ( $20: 1$ hexanes:EtOAc) gave $\mathbf{6 i}(0.186 \mathrm{~g}, 67 \%)$ as a light yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.36-7.24(\mathrm{~m}, 5 \mathrm{H}), 5.94-5.85(\mathrm{~m}, 1 \mathrm{H}), 5.74-5.64(\mathrm{~m}, 1 \mathrm{H})$, $4.68(\mathrm{~d}, J=6.6,2 \mathrm{H}), 4.50(\mathrm{~s}, 2 \mathrm{H}), 3.52(\mathrm{t}, J=6.6,2 \mathrm{H}), 2.84(\mathrm{q}, J=7.2,2 \mathrm{H}), 2.39(\mathrm{q}, J=6.5,2 \mathrm{H}), 1.11(\mathrm{t}, J=$ $7.2,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 195.1,161.1,138.6,134.7,128.6,127.9,127.8,124.7,73.1,69.3$, $66.9,33.1,32.9,7.2$; IR (thin film) $3029,2860,1727,1455,1218,1100 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{NO}_{4}\left(\mathrm{M}+\mathrm{NH}_{4}\right)^{+}$294.1705, found 294.1700.

## II. Synthesis of $\boldsymbol{\alpha}$-Hydroxy Acids

Procedure for the Formation of $\boldsymbol{\alpha}$-Hydroxy Acids: Silacyclopropane $\mathbf{5}$ (1.25-1.65 equiv) was added to a solution of the $\alpha$-keto ester ( 1.0 equiv) in toluene ( 0.1 M ) inside an inert atmosphere glove box. Upon cooling to $-24^{\circ} \mathrm{C}$, silver tosylate ( $10 \mathrm{~mol} \%$ ) was added, and the reaction mixture was kept at $-24^{\circ} \mathrm{C}$. After 3 h , the reaction mixture was removed from the glove box and warmed to ambient temperature under nitrogen atmosphere over 1 h , then $\mathrm{HF} \cdot \operatorname{Pyr}(70 \% \mathrm{HF}$ solution in pyridine, 4.0 equiv of HF$)$ was added. After an additional 1 h , the reaction mixture was diluted with 20 mL of saturated aqueous $\mathrm{NaHCO}_{3}$ and washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 25 \mathrm{~mL})$. The aqueous layer was then acidified to pH 1 with 1 M aqueous HCl and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \times 20 \mathrm{~mL})$. The organic layers were combined, dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated in vacuo to provide pure products.


3
2-hydroxy-2-phenylpent-4-enoic acid (3). The general transfer procedure was performed on $\mathbf{1}(0.190 \mathrm{~g}, 1.00$ $\mathrm{mmol})$, silacyclopropane $5(0.277 \mathrm{~g}, 1.40 \mathrm{mmol})$, and AgOTs ( $0.028 \mathrm{~g}, 0.10 \mathrm{mmol}$ ) to give $3(0.092 \mathrm{~g}, 48 \%)$ as a white solid: mp $100^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.65-7.63(\mathrm{~m}, 2 \mathrm{H}), 7.40-7.32(\mathrm{~m}, 3 \mathrm{H}), 5.81$ (ddt, $J=$ $17.2,10.1,7.2,1 H), 5.23(\mathrm{dd}, J=17.2,1.3,1 \mathrm{H}), 5.19(\mathrm{~d}, J=10.2,1 \mathrm{H}), 3.03(\mathrm{dd}, J=14.0,7.4,1 \mathrm{H}), 2.81(\mathrm{dd}, J$ $=14.1,6.9,1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 179.0,140.2,131.7,128.4,128.2,125.5,120.4,77.8,44.1$; IR (thin film) 3435 (br), 3062, 2981, 1713, 1449, 1229, $1189 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{NaO}_{3}(\mathrm{M}+$ $\mathrm{Na})^{+}$215.0684, found 215.0683. Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{O}_{3}: \mathrm{C}, 68.74 ; \mathrm{H}, 6.29$. Found: C, 68.48; H, 6.21.


7a

2-Hydroxy-2-methyl-3-phenylpent-4-enoic acid (7a). The general transfer procedure was performed on 6a $(0.204 \mathrm{~g}, 1.00 \mathrm{mmol})$, silacyclopropane $5(0.287 \mathrm{~g}, 1.45 \mathrm{mmol})$, and AgOTs ( $0.028 \mathrm{~g}, 0.10 \mathrm{mmol}$ ) to give 7 a
$(0.126 \mathrm{~g}, 62 \%)$ as a light yellow oil, single diastereomer (as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy): ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.33-7.29(\mathrm{~m}, 5 \mathrm{H}), 6.32(\mathrm{dt}, J=17.0,10.0,1 \mathrm{H}), 5.33(\mathrm{dd}, J=10.2,1.7,1 \mathrm{H}), 5.27(\mathrm{dd}, J=$ $17.0,1.1,1 \mathrm{H}), 3.64(\mathrm{~d}, J=9.8,1 \mathrm{H}), 1.55(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C} \operatorname{NMR}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 181.2,139.6,135.6,128.8$, $128.7,127.5,119.1,77.5,57.9,24.7$; IR (thin film) 3450 (br), 3064, 2983, 1713, 1455, $1148 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$229.0841, found 229.0844. Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{3}: \mathrm{C}, 69.88 ; \mathrm{H}, 6.84$. Found: C, 69.70; H, 6.80.


7b
2-Ethyl-2-hydroxy-3-phenylpent-4-enoic acid (7b). The general transfer procedure was performed on $\mathbf{6 b}$ $(0.050 \mathrm{~g}, 0.23 \mathrm{mmol})$, silacyclopropane $5(0.072 \mathrm{~g}, 1.6 \mathrm{mmol})$, and AgOTs ( $0.006 \mathrm{~g}, 0.02 \mathrm{mmol}$ ) to give 7b $(0.038 \mathrm{~g}, 84 \%)$ as a white solid, single diastereomer (as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy): mp $80{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.27-7.22(\mathrm{~m}, 5 \mathrm{H}), 6.26(\mathrm{dt}, J=17.0,10.0,1 \mathrm{H}), 5.23(\mathrm{dd}, J=10.2,1.7,1 \mathrm{H}), 5.19$ $(\mathrm{dd}, J=17.1,1.1,1 \mathrm{H}), 3.57(\mathrm{~d}, J=9.8,1 \mathrm{H}), 1.95-1.74(\mathrm{~m}, 2 \mathrm{H}), 0.88(\mathrm{t}, J=7.4,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $(125 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) $\delta 180.8,139.8,135.9,128.8,128.6,127.4,118.7,80.7,57.9,30.7,8.3$; IR (thin film) 3460 (br), 2973, 1710, 1231, $1135 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$243.0997, found 243.0995. Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{O}_{3}$ : C, 70.89; H, 7.32. Found: C, 70.80; H, 7.39.


7c

2-Hydroxy-2-isopropyl-3-phenylpent-4-enoic acid (7c). The general transfer procedure was performed on $\mathbf{6 c}$ $(0.232 \mathrm{~g}, 1.00 \mathrm{mmol})$, silacyclopropane $5(0.287 \mathrm{~g}, 1.45 \mathrm{mmol})$, and AgOTs $(0.028 \mathrm{~g}, 0.10 \mathrm{mmol})$ to give 7c $(0.127 \mathrm{~g}, 54 \%)$ as a white solid, single diastereomer (as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy): mp $125{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.37-7.20(\mathrm{~m}, 5 \mathrm{H}), 6.31(\mathrm{dt}, J=16.9,9.8,1 \mathrm{H}), 5.22(\mathrm{~m}, 2 \mathrm{H}), 3.86(\mathrm{~d}, J=9.6,1 \mathrm{H})$, 2.12 (septet, $J=6.9,1 \mathrm{H}), 1.14(\mathrm{~d}, J=6.9,3 \mathrm{H}), 0.96(\mathrm{~d}, J=6.9,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 179.3$, $140.0,135.8,129.0,128.3,127.0,118.0,82.4,54.0,34.6,17.2,16.9$; IR (thin film) 3489 (br), 2973, 1706, 1231, 1140, $1073 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$257.1154, found 257.1159. Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{O}_{3}$ : C, 71.77; H, 7.74. Found: C, 71.47; H, 7.70.


7d
2-tert-Butyl-2-hydroxy-3-phenylpent-4-enoic acid (7d). The general transfer procedure was performed on $\mathbf{6 d}$ ( $0.246 \mathrm{~g}, 1.00 \mathrm{mmol}$ ), silacyclopropane $5(0.287 \mathrm{~g}, 1.45 \mathrm{mmol})$, and AgOTs ( $0.028 \mathrm{~g}, 0.10 \mathrm{mmol}$ ) to give 7d ( $0.115 \mathrm{~g}, 47 \%$ ) as a white solid, single diastereomer (as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy): $\mathrm{mp} 123{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR (400 MHz, acetone- $d_{6}$ ) $\delta 7.43-7.30(\mathrm{~m}, 5 \mathrm{H}), 6.47(\mathrm{~d}, J=15.9,1 \mathrm{H}), 6.25(\mathrm{ddd}, J=15.8,8.1,6.3,1 \mathrm{H})$, $2.76(\mathrm{dd}, J=13.6,8.2,1 \mathrm{H}), 2.56(\mathrm{ddd}, J=13.6,6.2,0.9,1 \mathrm{H}), 1.05(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}\right.$, acetone- $\left.d_{6}\right) \delta$ 176.2, 137.9, 133.3, 128.6, 127.2, 126.2, 125.7, 81.8, 37.3, 36.9, 25.3; IR (thin film) 3435 (br), 2960, 1702, 1368, 1212, $1106 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$271.1310, found 271.1317. Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{O}_{3}$ : C, $72.55 ; \mathrm{H}, 8.12$. Found: C, $72.33 ; \mathrm{H}, 8.08$.

$7 e$

2-Hydroxy-2,3-diphenylpent-4-enoic acid (7e). The general transfer procedure was performed on $\mathbf{6 e}(0.266 \mathrm{~g}$, $1.00 \mathrm{mmol})$, silacyclopropane $5(0.277 \mathrm{~g}, 1.40 \mathrm{mmol})$, and AgOTs $(0.028 \mathrm{~g}, 0.10 \mathrm{mmol})$ to give $7 \mathrm{e}(0.189 \mathrm{~g}$, $71 \%$ ) as a white solid, single diastereomer (as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy): mp $183{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 500 MHz , acetone $-d_{6}$ ) $\delta 7.89-7.86(\mathrm{~m}, 2 \mathrm{H}), 7.57-7.56(\mathrm{~m}, 2 \mathrm{H}), 7.40-7.36(\mathrm{~m}, 2 \mathrm{H}), 7.30-7.26(\mathrm{~m}, 3 \mathrm{H})$, $7.23-7.19(\mathrm{~m}, 1 \mathrm{H}), 6.03(\mathrm{ddd}, J=17.2,10.4,8.6,1 \mathrm{H}), 4.84-4.78(\mathrm{~m}, 2 \mathrm{H}), 4.38(\mathrm{~d}, J=8.6,1 \mathrm{H}){ }^{13}{ }^{3} \mathrm{C}$ NMR $(125$ MHz , acetone- $d_{6}$ ) $\delta 174.8,141.7,140.8,137.4,130.1,128.5,128.01,127.98,127.5,126.9,126.8,116.8,80.7$, 57.6; IR (thin film) 3423, 3064 (br), 3029, 1721, 1694, 1449, $1100 \mathrm{~cm}^{-1}$; HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$291.0997, found 291.1000. Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{O}_{3}$ : C, 76.10; H, 6.01. Found: C, 75.82; H, 6.19.


7f

2-Hydroxy-3-methyl-2-phenylpent-4-enoic acid (7f). The general transfer procedure was performed on $\mathbf{6 f}$ $(0.204 \mathrm{~g}, 1.00 \mathrm{mmol})$, silacyclopropane $5(0.287 \mathrm{~g}, 1.45 \mathrm{mmol})$, and AgOTs ( $0.028 \mathrm{~g}, 0.10 \mathrm{mmol}$ ) to give 7 f $(0.126 \mathrm{~g}, 62 \%)$ as a white solid, single diastereomer (as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy): mp $119{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.66-7.64(\mathrm{~m}, 2 \mathrm{H}), 7.38-7.28(\mathrm{~m}, 3 \mathrm{H}), 5.58(\mathrm{ddd}, J=17.5,10.6,7.2,1 \mathrm{H}), 4.98$ (d, $J=8.4,1 \mathrm{H}), 4.95(\mathrm{~s}, 1 \mathrm{H}), 3.26$ (quintet, $J=6.7,1 \mathrm{H}), 1.19(\mathrm{~d}, J=6.7,3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $179.0,139.8,137.2,128.3,128.0,126.0,117.1,80.5,45.1,14.5$; IR (thin film) 3481 (br), 3072, 2977, 1706, 1449, 1241, 1192, $1135 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$229.0841, found 229.0841. Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{3}$ : C, 69.88; H, 6.84. Found: C, 69.49; H, 6.93.


79
2-Hydroxy-2-phenyl-3-vinylheptanoic acid (7g). The general transfer procedure was performed on $\mathbf{6 g}$ ( 0.150 $\mathrm{g}, 0.609 \mathrm{mmol})$, silacyclopropane $5(0.174 \mathrm{~g}, 1.45 \mathrm{mmol})$, and AgOTs $(0.017 \mathrm{~g}, 0.061 \mathrm{mmol})$ to give $7 \mathrm{~g}(0.110$ $\mathrm{g}, 72 \%$ ) as a white solid with $98: 2$ diastereoselectivity (as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy): mp $110{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.64-7.59(\mathrm{~m}, 2 \mathrm{H}), 7.37-7.25(\mathrm{~m}, 3 \mathrm{H}), 5.45(\mathrm{dt}, J=17.1,10.1,1 \mathrm{H}), 4.94(\mathrm{dd}, J$ $=10.3,1.8,1 \mathrm{H}), 4.82(\mathrm{dd}, J=17.3,1.6,1 \mathrm{H}), 2.96(\mathrm{t}, J=9.6,1 \mathrm{H}), 1.68-1.55(\mathrm{~m}, 1 \mathrm{H}), 1.43-1.18(\mathrm{~m}, 5 \mathrm{H}), 0.89$ $(\mathrm{t}, J=6.6,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 180.3,140.2,136.0,128.3,128.0,126.3,119.0,81.3,52.1$, 29.8, 29.3, 22.7, 14.2; IR (thin film) 3450 (br), $3074,2958,2933,1706,1449,1131 \mathrm{~cm}^{-1}$; HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{15} \mathrm{H}_{19} \mathrm{O}_{3}(\mathrm{M}-\mathrm{H})^{-} 247.1334$, found 247.1333. Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{O}_{3}$ : C, 72.55; H, 8.12. Found: C, 72.53; H, 8.14.


7h
3-((tert-Butyldimethylsilyloxy)methyl)-2-hydroxy-2-phenylpent-4-enoic acid (7h). The general transfer procedure was performed on $\mathbf{6 h}(0.100 \mathrm{~g}, 0.299 \mathrm{mmol})$, silacyclopropane $5(0.071 \mathrm{~g}, 0.36 \mathrm{mmol})$, and AgOTs $(0.008 \mathrm{~g}, 0.03 \mathrm{mmol})$. The reaction mixture was concentrated in vacuo and purified by column chromatography (80:18:2 hexanes:EtOAc:AcOH) to give $7 \mathrm{~h}(0.071 \mathrm{~g}, 71 \%)$ as a white solid, single diastereomer (as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy): mp $91-93{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.62-7.58(\mathrm{~m}, 2 \mathrm{H}), 7.37-7.25(\mathrm{~m}$, 3 H ), 5.72 (ddd, $J=17.4,10.5,8.5,1 \mathrm{H}), 5.05-4.95(\mathrm{~m}, 2 \mathrm{H}), 4.09(\mathrm{dd}, J=10.3,3.1,1 \mathrm{H}), 3.92(\mathrm{dd}, J=10.3,3.8$, $1 \mathrm{H}), 3.24(\mathrm{~m}, 1 \mathrm{H}), 0.90(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 176.3,139.4,133.5,128.4,128.1$, $125.8,119.3,81.6,66.1,50.7,25.9,18.3,-5.6,-5.7$; IR (thin film) $3529,2950,2931,2858,1698,1259,1098$ $\mathrm{cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{O}_{4} \mathrm{Si}(\mathrm{M}-\mathrm{H})^{-}$335.1679, found 335.1672.


7i
3-(2-(benzyloxy)ethyl)-2-Ethyl-2-hydroxypent-4-enoic acid (7i). The general transfer procedure was performed on $6 \mathbf{i}(0.050 \mathrm{~g}, 0.18 \mathrm{mmol})$, silacyclopropane $5(0.061 \mathrm{~g}, 0.31 \mathrm{mmol})$, and $\mathrm{AgOTs}(0.005 \mathrm{~g}, 0.02$ mmol ) to give $7 \mathbf{i}(0.038 \mathrm{~g}, 75 \%)$ as a white solid, single diastereomer (as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy): mp $81-82{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.38-7.26(\mathrm{~m}, 5 \mathrm{H}), 5.62(\mathrm{dt}, J=17.2,10.1,1 \mathrm{H})$, $5.18(\mathrm{dd}, J=10.2,1.8,1 \mathrm{H}), 5.06(\mathrm{dd}, J=17.2,1.7,1 \mathrm{H}), 4.57(\mathrm{~d}, J=11.8,1 \mathrm{H}), 4.50(\mathrm{~d}, J=11.8,1 \mathrm{H})$, $3.63-3.58(\mathrm{~m}, 1 \mathrm{H}), 3.50-3.45(\mathrm{~m}, 1 \mathrm{H}), 2.63(\mathrm{td}, J=10.3,3.2,1 \mathrm{H}), 1.78-1.54(\mathrm{~m}, 4 \mathrm{H}), 0.81(\mathrm{t}, J=7.4,3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 178.8,137.3,136.0,128.7,128.4,128.2,119.5,79.8,73.4,68.0,48.3,31.0$, 29.4, 8.0; IR (thin film) 3438 (br), 2971, 2937, 2879, 1719, $1237 \mathrm{~cm}^{-1}$; HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{O}_{4}$ $(\mathrm{M}+\mathrm{H})^{+}$279.1596, found 279.1588. Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{O}_{4}: \mathrm{C}, 69.04 ; \mathrm{H}, 7.97$. Found: C, 69.07; H, 7.98.

## III. Chiral $\alpha$-Keto Ester Synthetic Intermediates and Products



S-1
Lactate S-1. (S)-2-(tert-Butyldiphenylsilyloxy)propanoic acid ${ }^{2}$ ( $0.20 \mathrm{~g}, 0.61 \mathrm{mmol}$ ) was added to 10 mL of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ followed by oxalyl chloride ( $0.058 \mathrm{~mL}, 0.67 \mathrm{mmol}$ ) and DMF ( 1 drop). After stirring for 11 h , pyridine $(0.072 \mathrm{~mL}, 0.91 \mathrm{mmol})$ and ( $S, E$ )-oct-3-en-2-ol${ }^{3}(0.097 \mathrm{~g}, 0.76 \mathrm{mmol}$, ee $>97 \%$ as determined by chiral GC) were then added. After 3 h , the reaction mixture was diluted with 15 mL of aqueous saturated ammonium chloride and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The organic layers were combined, dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated in vacuo. The resultant yellow oil was purified by column chromatography ( $20: 1$ hexanes:EtOAc) to give $\mathbf{S - 1}(0.106 \mathrm{~g}, 40 \%)$ as a colorless oil: $[\alpha]^{23}{ }_{\mathrm{D}}-55.3\left(c 5.29, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.77-$ $7.69(\mathrm{~m}, 4 \mathrm{H}), 7.47-7.36(\mathrm{~m}, 6 \mathrm{H}), 5.71-5.65(\mathrm{~m}, 1 \mathrm{H}), 5.37(\mathrm{ddt}, J=15.3,6.9,1.3,1 \mathrm{H}), 5.27$ (pentet, $J=6.5$,
$1 \mathrm{H}), 4.28(\mathrm{q}, J=6.7,1 \mathrm{H}), 2.02(\mathrm{q}, J=6.6,2 \mathrm{H}), 1.39(\mathrm{~d}, J=6.7,3 \mathrm{H}), 1.34-1.29(\mathrm{~m}, 4 \mathrm{H}), 1.19(\mathrm{~d}, J=6.4,3 \mathrm{H})$, $1.13(\mathrm{~s}, 9 \mathrm{H}), 0.91(\mathrm{t}, J=7.2,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 173.2,136.2,136.0,134.0,133.8,133.5$, 129.96, 129.95, 129.5, 127.9, 127.8, 71.8, 69.3, 32.1, 31.3, 27.1, 22.4, 21.5, 20.5, 19.5, 14.1; IR (thin film) 3074, 2960, 2933, 2860, 1750, $1428 \mathrm{~cm}^{-1}$; HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{27} \mathrm{H}_{42} \mathrm{NO}_{3} \mathrm{Si}\left(\mathrm{M}+\mathrm{NH}_{4}\right)^{+} 456.2934$, found 456.2936. Anal. Calcd for $\mathrm{C}_{27} \mathrm{H}_{38} \mathrm{O}_{3} \mathrm{Si}$ : C, 73.92; H, 8.73. Found: C, 73.97; H, 8.89.


S-2
(S)-((S,E)-oct-3-en-2-yl) 2-Hydroxypropanoate (S-2). To a cooled solution ( $-20^{\circ} \mathrm{C}$ ) of S-1 (1.77 g, 4.03 mmol ) in 40 mL THF was added $n-\mathrm{Bu}_{4} \mathrm{NF}(6.05 \mathrm{~mL}$ of a 1 M solution in THF, 6.05 mmol ) dropwise. After 48 $\mathrm{h}, 10 \mathrm{~mL}$ of aqueous saturated $\mathrm{NH}_{4} \mathrm{Cl}$ was added and the reaction mixture was warmed to room temperature. ${ }^{4}$ The mixture was then extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$. The organic layers were combined, dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated in vacuo. The resultant oil was purified by column chromatography (5:1 hexanes:EtOAc) affording $\mathbf{S - 2}(0.597 \mathrm{~g}, 74 \%)$ as a colorless oil: $[\alpha]^{23}{ }_{\mathrm{D}}-67.2\left(c 3.44, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.73-5.68(\mathrm{~m}, 1 \mathrm{H}), 5.45-5.32(\mathrm{~m}, 2 \mathrm{H}), 4.25-4.18(\mathrm{~m}, 1 \mathrm{H}), 3.00(\mathrm{~d}, J=4.4,1 \mathrm{H}), 2.01(\mathrm{q}, J=$ $7.0,2 \mathrm{H}), 1.38(\mathrm{~d}, J=6.9,3 \mathrm{H}), 1.36-1.25(\mathrm{~m}, 7 \mathrm{H}), 0.88(\mathrm{t}, J=7.0,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 175.1$, $134.2,128.8,77.8,66.8,31.8,31.0,22.1,20.39,20.35,13.9$; IR (thin film) 3464 (br), 2958, 2929, 2858, 1735, 1214, $1129 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{11} \mathrm{H}_{20} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+} 223.1310$, found 223.1302. Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{20} \mathrm{O}_{3}$ : C, 65.97; H, 10.07. Found: C, 65.77; H, 10.21.


13
(S,E)-Oct-3-en-2-yl 2-oxopropanoate (13). To S-2 ( $0.50 \mathrm{~g}, 2.5 \mathrm{mmol}$ ) in 25 mL of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was added pyridine ( $1.0 \mathrm{~mL}, 12 \mathrm{mmol}$ ) followed by Dess-Martin periodinane $(1.6 \mathrm{~g}, 3.7 \mathrm{mmol}) .{ }^{5}$ After 3 h , the reaction mixture was diluted with 15 mL of saturated aqueous $\mathrm{Na}_{2} \mathrm{SO}_{3}$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The organic layers were combined, dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated in vacuo. The resultant oil was purified by column chromatography ( $5: 1$ hexanes: EtOAc ) to give $13\left(0.23 \mathrm{~g}, 84 \%\right.$ ) as a pale yellow oil: $[\alpha]^{23}{ }_{\mathrm{D}}-32.9(c$ $\left.1.70, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 5.79(\mathrm{td}, J=6.7,15.0,1 \mathrm{H}), 5.50(\mathrm{ddt}, J=15.1,7.3,1.3,1 \mathrm{H}), 5.42$ (quintet, $J=6.5,1 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}), 2.03(\mathrm{q}, J=6.8,2 \mathrm{H}), 1.42-1.23(\mathrm{~m}, 7 \mathrm{H}), 0.89(\mathrm{t}, J=7.1,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 192.6,160.4,135.6,128.3,74.5,32.0,31.1,26.9,22.4,20.5,14.1$; IR (thin film) 2960 , 2933, 2861, 1725, 1430, 1295, $1144 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+} 221.1154$, found 221.1148. Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}_{3}$ : C, 66.64; H, 9.15. Found: C, 66.92; H, 9.08.


14
$\boldsymbol{\alpha}$-Hydroxy acid 14. The general transfer procedure was performed on $\mathbf{1 3}(0.498 \mathrm{~g}, 2.51 \mathrm{mmol})$,
silacyclopropane $5(0.871 \mathrm{~g}, 4.40 \mathrm{mmol})$, and $\mathrm{AgOTs}(0.07 \mathrm{~g}, 0.3 \mathrm{mmol})$ to give $14(0.389 \mathrm{~g}, 77 \%, 97 \%$ ee based on HPLC data on the derived benzyl ester, absolute and relative stereochemistry determined from phenylethylamine salt vida infra) as a white solid: mp $105{ }^{\circ} \mathrm{C} ;[\alpha]^{23}{ }_{\mathrm{D}}+10.6\left(c 0.38, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.60-5.49(\mathrm{~m}, 1 \mathrm{H}), 5.21(\mathrm{ddd}, J=15.3,9.7,1.5,1 \mathrm{H}), 2.23(\mathrm{t}, J=10.2,1 \mathrm{H}), 1.73(\mathrm{dd}, J=6.4$, $1.4,3 \mathrm{H}), 1.43-1.09(\mathrm{~m}, 9 \mathrm{H}), 0.87(\mathrm{t}, J=6.9,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 182.2,130.3,129.2,51.4$, 29.9, 29.3, 24.6, 22.7, 18.3, 14.2; IR (thin film) 3440 (br), 2958, 2933, 2860, 1717, 1453, $1239 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{11} \mathrm{H}_{20} \mathrm{NaO}_{3}(\mathrm{M}+\mathrm{Na})^{+}$223.1310, found 223.1301. Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{20} \mathrm{O}_{3}$ : C, 65.97; H , 10.07. Found: C, 65.91 ; H, 10.09 .


S-3
(2S,3R)-3-(Benzyloxy)-2-(tert-Butyldimethylsilyloxy)butanoic acid (S-3). (2S,3R)-3-(benzyloxy)-2hydroxybutanoic acid $^{6}(0.61 \mathrm{~g}, 2.9 \mathrm{mmol})$ was combined with imidazole $(0.40 \mathrm{~g}, 5.8 \mathrm{mmol})$ and tertbutyldimethylsilyl chloride ( $1.7 \mathrm{~g}, 11 \mathrm{mmol}$ ) in 2 mL of DMF. ${ }^{7}$ The reaction mixture was stirred overnight then diluted with 50 mL of $1: 1 \mathrm{Et}_{2} \mathrm{O}: \mathrm{EtOAc}$. This solution was washed with 15 mL of $10 \%$ aqueous citric acid, 15 mL of $\mathrm{H}_{2} \mathrm{O}$, and 15 mL of saturated aqueous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The mixture was then concentrated in vacuo, and dilute with in 20 mL of $\mathrm{MeOH} . \mathrm{K}_{2} \mathrm{CO}_{3}(0.69 \mathrm{~g}, 5.0 \mathrm{mmol})$ in 6 mL of $\mathrm{H}_{2} \mathrm{O}$ was then added dropwise. After stirring for 16 h , the solution was concentrated, diluted with 20 mL of $\mathrm{H}_{2} \mathrm{O}$, and acidified with $10 \%$ aqueous citric acid (about 15 mL ) to pH 2 . The mixture was then extracted with EtOAc ( $3 \times 25 \mathrm{~mL}$ ), the organic layers were combined, dried with $\mathrm{MgSO}_{4}$, and concentrated in vacuo to give a viscous oil which was purified by column chromatography (80:18:2 hexanes:EtOAc:AcOH) to give S-3 ( $0.83 \mathrm{~g}, 88 \%$ ) as a colorless oil. S-3 decomposed rapidly and was carried on to the next synthetic step immediately after isolation: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.36-7.25(\mathrm{~m}, 5 \mathrm{H}), 4.64(\mathrm{~d}, J=11.6,1 \mathrm{H}), 4.49(\mathrm{~d}, J=11.6,1 \mathrm{H}), 4.10(\mathrm{~d}, J=3.0,1 \mathrm{H}), 4.06-3.97(\mathrm{~m}, 1 \mathrm{H}), 1.31$ $(\mathrm{d}, J=6.4,3 \mathrm{H}), 0.92(\mathrm{~s}, 9 \mathrm{H}), 0.11(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 175.7,137.5,128.7,128.3,128.1$, $75.1,73.6,71.6,25.8,18.3,15.6,-3.4$; IR (thin film) 3355 (br), 2971, $1380,1129 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{O}_{4} \mathrm{Si}(\mathrm{M}-\mathrm{H})^{-} 323.1679$, found 323.1688 .


S-4
(2S,3R)-Cinnamyl 3-(benzyloxy)-2-(tert-butyldimethylsilyloxy)butanoate (S-4). To a solution of S-3 (1.89 $\mathrm{g}, 5.83 \mathrm{mmol}$ ) in 58 mL of benzene, were added cinnamyl alcohol ( $0.601 \mathrm{~g}, 4.49 \mathrm{mmol}$ ), triethylamine ( 1.68 $\mathrm{mL}, 12.1 \mathrm{mmol})$ and DMAP $(0.680 \mathrm{~g}, 5.57 \mathrm{mmol})^{7}$ After stirring for $30 \mathrm{~min}, 2,4,6$-trichlorobenzoyl chloride $(1.43 \mathrm{~mL}, 9.15 \mathrm{mmol})$ was added dropwise. After stirring for 16 h , the solution was diluted with 20 mL of $10 \%$ aqueous citric acid and extracted with EtOAc $(3 \times 25 \mathrm{~mL})$. The organic layers were combined, dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated in vacuo to give an oil which was purified by column chromatography (20:1 hexanes:EtOAc) to give $\mathbf{S}-4(1.08 \mathrm{~g}, 54 \%)$ as a colorless oil: $[\alpha]_{\mathrm{D}}^{23}-12.6\left(c 0.55, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.38-7.24(\mathrm{~m}, 10 \mathrm{H}), 6.67(\mathrm{~d}, J=15.9,1 \mathrm{H}), 6.26(\mathrm{dt}, J=15.9,6.5,1 \mathrm{H}), 4.78(\mathrm{dt}, J=6.5,1.5,2 \mathrm{H})$, $4.65(\mathrm{~d}, J=12.0,1 \mathrm{H}), 4.58(\mathrm{~d}, J=12.0,1 \mathrm{H}), 4.27(\mathrm{~d}, J=4.6,1 \mathrm{H}), 3.88(\mathrm{dq}, J=6.4,4.7,1 \mathrm{H}), 1.24(\mathrm{~d}, J=6.4$, $3 \mathrm{H}), 0.93(\mathrm{~s}, 9 \mathrm{H}), 0.12(\mathrm{~s}, 3 \mathrm{H}), 0.08(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 171.9,138.8,136.4,134.8,128.5$, $128.0,126.8,123.1,76.5,76.3,71.5,65.6,26.0,18.7,15.9,-4.7,-5.0$; IR (thin film) 2954, 2858, 1752, 1254,

1162, $1123 \mathrm{~cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{40} \mathrm{NO}_{4} \mathrm{Si}\left(\mathrm{M}+\mathrm{NH}_{4}\right)^{+} 458.2727$, found 458.2735. Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{36} \mathrm{O}_{4} \mathrm{Si}$ : C, 70.87; H, 8.23. Found: C, 71.20 ; H, 8.27.

(2S,3R)-Cinnamyl 3-(benzyloxy)-2-hydroxybutanoate (S-5). S-4 ( $0.624 \mathrm{~g}, 1.42 \mathrm{mmol}$ ) was added to 15 mL of THF and cooled to $-20^{\circ} \mathrm{C}$. After stirring for $30 \mathrm{~min}, n-\mathrm{Bu} \mathrm{u}_{4} \mathrm{NF}(2.12 \mathrm{~mL}$ of a 1 M solution in THF, 2.12 mmol ) was added dropwise. After $48 \mathrm{~h}, 5 \mathrm{~mL}$ of aqueous saturated $\mathrm{NH}_{4} \mathrm{Cl}$ was added and the reaction mixture was warmed to room temperature. ${ }^{4}$ The mixture was then extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The organic layers were combined, dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated in vacuo. The resultant oil was purified by column chromatography ( $5: 1$ hexanes:EtOAc) to give $\mathbf{S - 5}(0.303 \mathrm{~g}, 66 \%)$ as a colorless oil: $[\alpha]^{23}{ }_{\mathrm{D}}-62.9(c 0.781$, $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.37-7.24(\mathrm{~m}, 10 \mathrm{H}), 6.65(\mathrm{~d}, J=15.9,1 \mathrm{H}), 6.22(\mathrm{dt}, J=15.9,6.6,1 \mathrm{H})$, 4.83 (ddd, $J=12.6,6.7,1.1,1 \mathrm{H}), 4.75$ (ddd, $J=12.6,6.6,1.2,1 \mathrm{H}), 4.62(\mathrm{~d}, J=11.9,1 \mathrm{H}), 4.41(\mathrm{~d}, J=11.9$, $1 \mathrm{H}), 4.12(\mathrm{dd}, J=8.4,2.3,1 \mathrm{H}), 3.98(\mathrm{qd}, J=6.4,2.3,1 \mathrm{H}), 2.95(\mathrm{~d}, J=8.4,1 \mathrm{H}), 1.35(\mathrm{~d}, J=6.4,3 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 171.0,138.2,136.2,135.4,128.9,128.6,128.5,127.9,126.9,122.6,75.3,74.7,71.1$, 66.4, 15.8; IR (thin film) 3489 (br), 3029, 2935, 2875, 1742, 1266, $1198 \mathrm{~cm}^{-1}$; HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{NaO}_{4}(\mathrm{M}+\mathrm{Na})^{+} 349.1416$, found 349.1429 .


15
( $\boldsymbol{R}$ )-Cinnamyl 3-(benzyloxy)-2-oxobutanoate (17). To $\mathbf{S - 5}(0.280 \mathrm{~g}, 0.858 \mathrm{mmol})$ in 10 mL of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was added pyridine $(0.345 \mathrm{~mL}, 4.28 \mathrm{mmol})$ followed by Dess-Martin periodinane $(0.545 \mathrm{~g}, 1.28 \mathrm{mmol}) .{ }^{5}$ After 3 h , the reaction mixture was diluted with 8 mL of saturated aqueous $\mathrm{Na}_{2} \mathrm{SO}_{3}$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20$ mL ). The organic layers were combined, dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated in vacuo. The resultant oil was purified by column chromatography ( $5: 1$ hexanes:EtOAc) to give $15(0.233 \mathrm{~g}, 84 \%)$ as a yellow oil: $[\alpha]^{23}{ }_{\mathrm{D}}$ $+27.3\left(c 1.15, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.42-7.26(\mathrm{~m}, 10 \mathrm{H}), 6.74(\mathrm{~d}, J=15.9,1 \mathrm{H}), 6.31(\mathrm{dt}, J=$ $15.9,6.7,1 \mathrm{H}), 4.94(\mathrm{ddd}, J=6.7,2.5,1.2,2 \mathrm{H}), 4.70(\mathrm{~d}, J=11.5,1 \mathrm{H}), 4.62-4.54(\mathrm{~m}, 2 \mathrm{H}), 1.49(\mathrm{~d}, J=6.9$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 195.5,162.6,137.4,136.2,136.0,128.9,128.7,128.6,128.3,128.2,127.0$, 121.6, 77.7, 72.8, 67.0, 16.8; IR (thin film) 3477 (br), 3031, 2985, 2939, 2873, 1729, $1272 \mathrm{~cm}^{-1}$; HRMS (ESI) $m$ $/ z$ calcd for $\mathrm{C}_{21} \mathrm{H}_{28} \mathrm{NO}_{5}\left(\mathrm{M}+\mathrm{NH}_{4}+\mathrm{CH}_{3} \mathrm{OH}\right)^{+} 374.1967$, found 374.1957. HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{21} \mathrm{D}_{3} \mathrm{H}_{21} \mathrm{NaO}_{5}\left(\mathrm{M}+\mathrm{Na}+\mathrm{CD}_{3} \mathrm{OH}\right)^{+} 382.1710$, found 382.1704 .


16

2-((R)-1-(Benzyloxy)ethyl)-2-Hydroxy-3-phenylpent-4-enoic acid (16). The general transfer procedure was performed on $15(0.050 \mathrm{~g}, 0.15 \mathrm{mmol})$, silacyclopropane $5(0.050 \mathrm{~g}, 1.65 \mathrm{mmol})$, and $\mathrm{AgOTs}(0.004 \mathrm{~g}, 0.02$ $\mathrm{mmol})$ to give $16\left(0.032 \mathrm{~g}, 63 \%, 80 \%\right.$ diastereoselectivity as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy. The
product is predominantly one diastereomer, but $20 \%$ of other compounds can be observed by ${ }^{1} \mathrm{H}$ NMR spectroscopy. These materials are likely to be isomers because the compound exhibits satisfactory elementary analysis.) as a viscous light yellow oil. The relative stereochemistry of the product was assigned based upon analogies to similar systems, as described in the text: ${ }^{1} \mathrm{H}$ NMR major diastereomer $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.47-$ $7.24(\mathrm{~m}, 10 \mathrm{H}), 6.36(\mathrm{td}, J=10.0,17.1,1 \mathrm{H}), 5.27(\mathrm{dd}, J=10.1,1.7,1 \mathrm{H}), 5.21(\mathrm{dd}, J=17.1,1.2,1 \mathrm{H}), 4.81(\mathrm{~d}, J$ $=10.9,1 \mathrm{H}), 4.55(\mathrm{~d}, J=10.9,1 \mathrm{H}), 4.06(\mathrm{q}, J=6.3,1 \mathrm{H}), 3.98(\mathrm{~d}, J=9.7,1 \mathrm{H}), 1.34(\mathrm{~d}, J=6.3,3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 174.9,138.9,136.7,136.1,129.1,128.9,128.80,128.76,128.7,128.4,127.7,118.8,81.2$, $76.4,72.3,54.0,14.0$; IR (thin film) 3460 (br), 3006, 2917, 1710, $1420,1360 \mathrm{~cm}^{-1}$; HRMS (ESI) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{NaO}_{4}(\mathrm{M}+\mathrm{Na})^{+} 349.1416$, found 349.1414. Anal. Calcd for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{O}_{4}: \mathrm{C}, 73.60 ; \mathrm{H}, 6.79$. Found: C, 73.24; H, 6.85.

## IV. $\alpha$-Imino Ester Synthesis and Silylene Transfer



17
Cinnamyl 2-(4-methoxyphenylimino)acetate (17). To a solution of anisidine ( $0.073 \mathrm{~g}, 0.60 \mathrm{mmol}$ ) in 3 mL of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was added cinnamyl 2-oxoacetate (obtained as a mixture of aldehyde, hydrate, and hydrate oligomers ${ }^{8}$, $0.136 \mathrm{~g}, 0.654 \mathrm{mmol}$ ) and $\mathrm{MgSO}_{4}$ (about 0.2 g ). ${ }^{9}$ After stirring for 2 h , the reaction mixture was filtered and concentrated in vacuo to give $17(0.172 \mathrm{~g}, 98 \%)$ as a yellow-green solid: mp $76-81^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.98(\mathrm{~s}, 1 \mathrm{H}), 7.43-7.25(\mathrm{~m}, 7 \mathrm{H}), 6.97-6.92(\mathrm{~m}, 2 \mathrm{H}), 6.76(\mathrm{~d}, J=15.9,1 \mathrm{H}), 6.41(\mathrm{dt}, J=15.9,6.6$, $1 \mathrm{H}), 5.01(\mathrm{dd}, J=6.6,1.0,2 \mathrm{H}), 3.84(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 163.6,160.9,147.9,141.5,136.3$, 135.7, 128.9, 128.6, 127.0, 124.0, 122.6, 114.8, 66.6, 55.8; IR (thin film) $3054,2919,2840,1740,1717,1590$ $\mathrm{cm}^{-1}$; HRMS (ESI) $m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{NNaO}_{3}(\mathrm{M}+\mathrm{Na})^{+} 318.1106$, found 318.1096.


Azasilalactone (18). The general transfer procedure was performed on 17 ( $0.082 \mathrm{~g}, 0.28 \mathrm{mmol}$ ), silacyclopropane $5(0.079 \mathrm{~g}, 0.40 \mathrm{mmol})$, and AgOTs $(0.008 \mathrm{~g}, 0.03 \mathrm{mmol})$ to give $\mathbf{1 8}(0.059 \mathrm{~g}, 48 \%)$ as a tan viscous oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.45-7.21(\mathrm{~m}, 5 \mathrm{H}), 6.88(\mathrm{~s}, 4 \mathrm{H}), 6.16(\mathrm{dt}, J=16.9,9.7,1 \mathrm{H}), 5.10$ (dd, $J=10.0,1.6,1 \mathrm{H}), 5.01(\mathrm{~d}, J=16.9,1 \mathrm{H}), 4.79(\mathrm{~d}, J=1.8,1 \mathrm{H}), 3.83-3.80(\mathrm{~m}, 4 \mathrm{H}), 1.21(\mathrm{~s}, 9 \mathrm{H}), 0.90(\mathrm{~s}$, $9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 171.8,154.9,141.8,137.3,128.9,128.7,128.6,127.0,121.5,118.8,115.2$, $67.2,55.8,52.0,28.5,27.3,24.1,21.5$; IR (thin film) 2937, 2861, 1766, 1509, 1474, $1243 \mathrm{~cm}^{-1}$; HRMS (ESI) $m$ $/ z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{35} \mathrm{NNaO}_{3} \mathrm{Si}(\mathrm{M}+\mathrm{Na})^{+} 460.2284$, found 460.2275 .


Silyl ketene acetal (12). Ethyl pyruvate ( $0.022 \mathrm{~g}, 0.19 \mathrm{mmol}$ ) was added to a solution of silirane $5(0.045 \mathrm{~g}$, $0.23 \mathrm{mmol})$ in benzene- $d_{6}(0.65 \mathrm{~mL})$. The solution was left under an inert atmosphere for several minutes before the addition of AgOTs $(0.005 \mathrm{~g}, 0.02 \mathrm{mmol})$. This reaction was performed to determine the viability of 12 as an intermediate; no product was collected: ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , benzene- $d_{6}$ ) $\delta 3.89(\mathrm{q}, J=5.7,2 \mathrm{H}), 1.98$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 1.16-1.14 (m, 21H); ${ }^{29} \mathrm{Si}$ NMR ( 100 MHz , benzene- $d_{6}$ ) $\delta 14.2$.

## V. References

(1) Driver, T. G.; Franz, A. K.; Woerpel, K. A. J. Am. Chem. Soc. 2002, 124, 6524-6525.
(2) Harris, J. M.; O'Doherty, G. A. Tetrahedron Lett. 2002, 43, 8195-8199.
(3) Gao, Y.; Hanson, R. M.; Klunder, J. M.; Ko, S. Y.; Masamune, H.; Sharpless, K. B. J. Am. Chem. Soc. 1987, 109, 5765-5780.
(4) Nicewicz, D. A.; Johnson, J. S. J. Am. Chem. Soc. 2005, 127, 6170-6171.
(5) Dess, D. B.; Martin, J. C. J. Org. Chem. 1983, 48, 4155-4156.
(6) Deechongkit, S.; You, S. L.; Kelly, J. W. Org. Lett. 2004, 6, 497-500.
(7) Díez, E.; Dixon, D. J.; Ley, S. V.; Polara, A.; Rodríguez, F. Helv. Chim. Acta. 2003, 86, 3717-
3729.
(8) Spaller, M. R.; Thielemann, W. T.; Brennan, P. E.; Bartlett, P. A. J. Comb. Chem. 2002, 4, 516522.
(9) Cozzi, F.; Annunziata, R.; Cinquini, M.; Poletti, L.; Perboni, A.; Tamburini, B. Chirality 1998, 10, 91-94.

## VI. X-Ray Crystallographic Data



## SI-6

Phenylethylamine Salt (SI-6): To a solution of $\mathbf{1 4}(0.050 \mathrm{mg}, 0.25 \mathrm{mmol})$ in benzene ( 2 mL ) was added Sphenylethylamine ( $0.030 \mathrm{mg}, 0.25 \mathrm{mmol}$ ). The mixture was left at $4^{\circ} \mathrm{C}$ for 21 d , at which point crystals had formed and were submitted for X-ray crystallographic analysis.

X-ray Data Collection, Structure Solution and Refinement for SI-6.
A colorless crystal of approximate dimensions $0.14 \times 0.19 \times 0.35 \mathrm{~mm}$ was mounted on a glass fiber and transferred to a Bruker CCD platform diffractometer. The SMART ${ }^{1}$ program package was used to determine the unit-cell parameters and for data collection ( $40 \mathrm{sec} /$ frame scan time for a hemisphere of diffraction data). The raw frame data was processed using SAINT ${ }^{2}$ and $\mathrm{SADABS}^{3}$ to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL ${ }^{4}$ program. The diffraction symmetry was $2 / m$ and the systematic absences were consistent with the monoclinic space groups $P 2_{1}$ and $P 2_{1} / m$. It was later determined that the noncentrosymmetric space group $P 2_{1}$ was correct.

The structure was solved by direct methods and refined on $F^{2}$ by full-matrix least-squares techniques. The analytical scattering factors ${ }^{5}$ for neutral atoms were used throughout the analysis. Hydrogen atoms were either located from a difference-Fourier map and refined ( $x, y, z$ and $U_{i s o}$ ) or were included using a riding model. There were two molecules of the formula unit present $(Z=4)$.

Least-squares analysis yielded wR2 $=0.1698$ and GOF $=1.050$ for 434 variables refined against 5903 data $(0.85 \AA), \mathrm{R} 1=0.0597$ for those 4209 data with $\mathrm{I}>2.0 \sigma(\mathrm{I})$. The absolute structure could not be assigned by inversion of the model or by refinement of the Flack parameter. ${ }^{6}$ Although the quality of the structure was not high, the data establishes the relative stereochemistry between the acid and the amine.

## References.

1. SMART Software Users Guide, Version 5.1, Bruker Analytical X-Ray Systems, Inc.; Madison, WI 1999.
2. SAINT Software Users Guide, Version 6.0, Bruker Analytical X-Ray Systems, Inc.; Madison, WI 1999.
3. Sheldrick, G. M. SADABS Version 2.10, Bruker Analytical X-Ray Systems, Inc.; Madison, WI 2002.
4. Sheldrick, G. M. SHELXTL Version 6.12, Bruker Analytical X-Ray Systems, Inc.; Madison, WI 2001.
5. International Tables for X-Ray Crystallography 1992, Vol. C., Dordrecht: Kluwer Academic Publishers.
6. Flack, H. D. Acta. Cryst. 1983, A39, 876-881.

Definitions:
$\mathrm{wR} 2=\left[\Sigma\left[\mathrm{w}\left(\mathrm{F}_{\mathrm{o}}{ }^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2}\right] / \Sigma\left[\mathrm{w}\left(\mathrm{F}_{\mathrm{o}}{ }^{2}\right)^{2}\right]\right]^{1 / 2}$
$\mathrm{R} 1=\Sigma| | \mathrm{F}_{\mathrm{o}}\left|-\left|\mathrm{F}_{\mathrm{c}}\right|\right| / \Sigma\left|\mathrm{F}_{\mathrm{o}}\right|$
Goof $=\mathrm{S}=\left[\Sigma\left[\mathrm{w}\left(\mathrm{F}_{\mathrm{o}}{ }^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2}\right] /(\mathrm{n}-\mathrm{p})\right]^{1 / 2}$ where n is the number of reflections and p is the total number of parameters refined.

The thermal ellipsoid plot is shown at the $50 \%$ probability level.


Table 1. Crystal data and structure refinement for SI-6.

Identification code
Empirical formula
Formula weight
Temperature
Wavelength
Crystal system
Space group
Unit cell dimensions
kaw 109 (Brett Howard)
$\mathrm{C}_{19} \mathrm{H}_{31} \mathrm{~N} \mathrm{O}_{3}$
321.45

153(2) K
$0.71073 \AA$
Monoclinic
$P 2_{1}$
$\mathrm{a}=17.558(3) \AA \quad \alpha=90^{\circ}$.
$\mathrm{b}=5.4049(9) \AA \quad \beta=92.475(3)^{\circ}$.
$\mathrm{c}=20.711(3) \AA \quad \gamma=90^{\circ}$.

## Volume <br> 1963.6(6) $\AA^{3}$

Z
Density (calculated)
Absorption coefficient
F(000)
Crystal color
Crystal size
Theta range for data collection
Index ranges
4
$1.087 \mathrm{Mg} / \mathrm{m}^{3}$
$0.072 \mathrm{~mm}^{-1}$
704
colorless
$0.35 \times 0.19 \times 0.14 \mathrm{~mm}^{3}$

Reflections collected
1.49 to $24.79^{\circ}$

Independent reflections
Completeness to theta $=24.79^{\circ}$
$-20 \leq h \leq 20,-6 \leq k \leq 6,-24 \leq l \leq 23$

Absorption correction
Max. and min. transmission
Refinement method
Data / restraints / parameters
12061

Goodness-of-fit on $\mathrm{F}^{2}$
Final R indices [I>2sigma(I) $=4209$ data $]$
R indices (all data; 0.85A)
Absolute structure parameter
$5903[\mathrm{R}(\mathrm{int})=0.0366]$
99.4 \%

Semi-empirical from equivalents
0.9899 and 0.9751

Full-matrix least-squares on $\mathrm{F}^{2}$

Extinction coefficient
5903 / 1 / 434
1.050

Largest diff. peak and hole
$\mathrm{R} 1=0.0597, \mathrm{wR} 2=0.1453$
$R 1=0.0945, w R 2=0.1698$
-0.8(15)
$0.005(2)$
0.392 and -0.255 e. $\AA^{-3}$

Table 2. Atomic coordinates ( $\mathrm{x} 10^{4}$ ) and equivalent isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for SI-6. U(eq) is defined as one third of the trace of the orthogonalized $U^{i j}$ tensor.

|  | x | y | z | $\mathrm{U}(\mathrm{eq})$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O}(1)$ | 1343(1) | 9490(5) | 526(1) | 38(1) |
| $\mathrm{O}(2)$ | 887(1) | 5696(5) | 425(1) | 35(1) |
| $\mathrm{O}(3)$ | 2276(2) | 3699(6) | 360(2) | 74(1) |
| $\mathrm{N}(1)$ | -3(1) | 11643(5) | 688(1) | 26(1) |
| C(1) | 1430(2) | 7213(7) | 505(2) | 26(1) |
| C(2) | 2236(2) | 6164(7) | 607(2) | 39(1) |
| C(3) | 2806(2) | 7673(7) | 219(2) | 32(1) |
| C(4) | 2426(2) | 6257(13) | 1331(2) | 71(2) |
| C(5) | 2589(2) | 7671(8) | -505(2) | 35(1) |
| C(6) | 3015(2) | 9559(8) | -892(2) | 36(1) |
| C(7) | 2780(2) | 9619(9) | -1605(2) | 45(1) |
| C(8) | 3174(3) | 11607(10) | -1983(2) | 62(1) |
| C(9) | 3608(2) | 6731(8) | 345(2) | 35(1) |
| C(10) | 4182(2) | 8080(8) | 581(2) | 39(1) |
| $\mathrm{C}(11)$ | 4976(2) | 7147(10) | 722(2) | 52(1) |
| C(12) | -440(2) | 11092(7) | 1270(2) | 29(1) |
| C(13) | 104(2) | 11229(7) | 1865(2) | 35(1) |
| C(14) | 105(3) | 13202(9) | 2278(2) | 62(1) |
| C(15) | 626(3) | 13328(11) | 2815(2) | 73(2) |
| C(16) | 1137(3) | 11548(10) | 2921(2) | 64(1) |
| C(17) | 1175(3) | 9617(11) | 2514(2) | 67(1) |
| C(18) | 657(2) | 9416(10) | 1980(2) | 57(1) |
| C(19) | -826(2) | 8580(7) | 1179(2) | 36(1) |
| $\mathrm{O}(4)$ | 3692(1) | 8360(5) | 4564(1) | 31(1) |
| $\mathrm{O}(5)$ | 4018(1) | 4387(4) | 4630(1) | 29(1) |
| $\mathrm{O}(6)$ | 2657(2) | 2772(5) | 4218(2) | 49(1) |
| N(2) | 5021(1) | 10580(5) | 4332(1) | 25(1) |
| C(20) | 3547(2) | 6104(7) | 4506(2) | 27(1) |
| C(21) | 2736(2) | 5369(6) | 4261(2) | 30(1) |
| C(22) | 2166(2) | 6429(7) | 4735(2) | 34(1) |
| C(23) | 2594(2) | 6460(8) | 3588(2) | 36(1) |
| C(24) | 2323(2) | 5433(11) | 5426(2) | 59(1) |
| C(25) | 1901(2) | 6730(17) | 5936(2) | 93(2) |


| $\mathrm{C}(26)$ | $2064(3)$ | $5730(30)$ | $6619(3)$ | $176(6)$ |
| :--- | ---: | :---: | ---: | ---: |
| $\mathrm{C}(27)$ | $1623(4)$ | $5980(30)$ | $7053(3)$ | $207(8)$ |
| $\mathrm{C}(28)$ | $1352(2)$ | $5913(8)$ | $4525(2)$ | $38(1)$ |
| $\mathrm{C}(29)$ | $821(2)$ | $7588(9)$ | $4418(2)$ | $50(1)$ |
| $\mathrm{C}(30)$ | $-2(2)$ | $7078(13)$ | $4227(2)$ | $73(2)$ |
| $\mathrm{C}(31)$ | $5431(2)$ | $9915(7)$ | $3742(2)$ | $31(1)$ |
| $\mathrm{C}(32)$ | $5012(2)$ | $10820(7)$ | $3136(2)$ | $33(1)$ |
| $\mathrm{C}(33)$ | $5429(3)$ | $11399(8)$ | $2608(2)$ | $48(1)$ |
| $\mathrm{C}(34)$ | $5075(3)$ | $12157(10)$ | $2035(2)$ | $67(1)$ |
| $\mathrm{C}(35)$ | $4289(3)$ | $12369(11)$ | $1995(2)$ | $70(2)$ |
| $\mathrm{C}(36)$ | $3866(3)$ | $11814(11)$ | $2511(2)$ | $64(1)$ |
| $\mathrm{C}(37)$ | $4229(2)$ | $11021(9)$ | $3089(2)$ | $46(1)$ |
| $\mathrm{C}(38)$ | $5579(2)$ | $7152(8)$ | $3740(2)$ | $42(1)$ |

Table 3. Bond lengths $\left[\AA\right.$ ] and angles $\left[{ }^{\circ}\right]$ for SI-6.

| $\mathrm{O}(1)-\mathrm{C}(1)$ | $1.241(4)$ |
| :--- | :--- |
| $\mathrm{O}(2)-\mathrm{C}(1)$ | $1.262(4)$ |
| $\mathrm{O}(3)-\mathrm{C}(2)$ | $1.430(5)$ |
| $\mathrm{N}(1)-\mathrm{C}(12)$ | $1.487(4)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.532(5)$ |
| $\mathrm{C}(2)-\mathrm{C}(4)$ | $1.523(6)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.544(5)$ |
| $\mathrm{C}(3)-\mathrm{C}(9)$ | $1.508(4)$ |
| $\mathrm{C}(3)-\mathrm{C}(5)$ | $1.531(5)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | $1.516(5)$ |
| $\mathrm{C}(6)-\mathrm{C}(7)$ | $1.515(5)$ |
| $\mathrm{C}(7)-\mathrm{C}(8)$ | $1.515(6)$ |
| $\mathrm{C}(9)-\mathrm{C}(10)$ | $1.321(5)$ |
| $\mathrm{C}(10)-\mathrm{C}(11)$ | $1.500(5)$ |
| $\mathrm{C}(12)-\mathrm{C}(19)$ | $1.526(5)$ |
| $\mathrm{C}(12)-\mathrm{C}(13)$ | $1.527(5)$ |
| $\mathrm{C}(13)-\mathrm{C}(14)$ | $1.368(6)$ |
| $\mathrm{C}(13)-\mathrm{C}(18)$ | $1.392(6)$ |
| $\mathrm{C}(14)-\mathrm{C}(15)$ | $1.411(6)$ |
| $\mathrm{C}(15)-\mathrm{C}(16)$ | $1.326(7)$ |
| $\mathrm{C}(16)-\mathrm{C}(17)$ | $1.308(6)$ |
| $\mathrm{C}(17)-\mathrm{C}(18)$ | $1.345(7)$ |
| $\mathrm{O}(4)-\mathrm{C}(20)$ | $1.405(6)$ |
| $\mathrm{O}(5)-\mathrm{C}(20)$ | $1.251(4)$ |
| $\mathrm{O}(6)-\mathrm{C}(21)$ | $1.262(4)$ |
| $\mathrm{N}(2)-\mathrm{C}(31)$ | $1.413(5)$ |
| $\mathrm{C}(20)-\mathrm{C}(21)$ | $1.488(4)$ |
| $\mathrm{C}(21)-\mathrm{C}(23)$ | $1.543(4)$ |
| $\mathrm{C}(21)-\mathrm{C}(22)$ | $1.523(5)$ |
| $\mathrm{C}(22)-\mathrm{C}(28)$ | $1.502(5)$ |
| $\mathrm{C}(22)-\mathrm{C}(24)$ | C |
| $\mathrm{C}(24)-\mathrm{C}(25)$ | $\mathrm{C}(25)-\mathrm{C}(26)$ |
| $\mathrm{C}(26)-\mathrm{C}(27)$ | $\mathrm{C}(29)-\mathrm{C}(30)$ |
| $\mathrm{C}(29)$ | $1.59)$ |
| C |  |


| $\mathrm{C}(31)-\mathrm{C}(32)$ | $1.509(5)$ |
| :--- | :--- |
| $\mathrm{C}(31)-\mathrm{C}(38)$ | $1.516(5)$ |
| $\mathrm{C}(32)-\mathrm{C}(37)$ | $1.377(5)$ |
| $\mathrm{C}(32)-\mathrm{C}(33)$ | $1.379(5)$ |
| $\mathrm{C}(33)-\mathrm{C}(34)$ | $1.379(6)$ |
| $\mathrm{C}(34)-\mathrm{C}(35)$ | $1.384(7)$ |
| $\mathrm{C}(35)-\mathrm{C}(36)$ | $1.361(7)$ |
| $\mathrm{C}(36)-\mathrm{C}(37)$ | $1.399(5)$ |


| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{O}(2)$ | $123.8(3)$ |
| :--- | :--- |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(2)$ | $118.4(3)$ |
| $\mathrm{O}(2)-\mathrm{C}(1)-\mathrm{C}(2)$ | $117.8(3)$ |
| $\mathrm{O}(3)-\mathrm{C}(2)-\mathrm{C}(4)$ | $111.8(4)$ |
| $\mathrm{O}(3)-\mathrm{C}(2)-\mathrm{C}(1)$ | $110.7(3)$ |
| $\mathrm{C}(4)-\mathrm{C}(2)-\mathrm{C}(1)$ | $106.6(3)$ |
| $\mathrm{O}(3)-\mathrm{C}(2)-\mathrm{C}(3)$ | $105.3(3)$ |
| $\mathrm{C}(4)-\mathrm{C}(2)-\mathrm{C}(3)$ | $112.1(3)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $110.4(3)$ |
| $\mathrm{C}(9)-\mathrm{C}(3)-\mathrm{C}(5)$ | $111.1(3)$ |
| $\mathrm{C}(9)-\mathrm{C}(3)-\mathrm{C}(2)$ | $110.6(3)$ |
| $\mathrm{C}(5)-\mathrm{C}(3)-\mathrm{C}(2)$ | $111.7(3)$ |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(3)$ | $114.2(3)$ |
| $\mathrm{C}(7)-\mathrm{C}(6)-\mathrm{C}(5)$ | $114.2(3)$ |
| $\mathrm{C}(8)-\mathrm{C}(7)-\mathrm{C}(6)$ | $114.1(3)$ |
| $\mathrm{C}(10)-\mathrm{C}(9)-\mathrm{C}(3)$ | $124.7(4)$ |
| $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | $125.1(4)$ |
| $\mathrm{N}(1)-\mathrm{C}(12)-\mathrm{C}(19)$ | $108.7(3)$ |
| $\mathrm{N}(1)-\mathrm{C}(12)-\mathrm{C}(13)$ | $108.5(3)$ |
| $\mathrm{C}(19)-\mathrm{C}(12)-\mathrm{C}(13)$ | $113.7(3)$ |
| $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(18)$ | $117.3(4)$ |
| $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(12)$ | $121.7(4)$ |
| $\mathrm{C}(18)-\mathrm{C}(13)-\mathrm{C}(12)$ | $120.8(4)$ |
| $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)$ | $120.9(4)$ |
| $\mathrm{C}(16)-\mathrm{C}(15)-\mathrm{C}(14)$ | $120.4(5)$ |
| $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)$ | $120.7(4)$ |
| $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)$ | $120.3(5)$ |
| $\mathrm{C}(13)-\mathrm{C}(18)-\mathrm{C}(17)$ | $120.2(5)$ |
| $\mathrm{O}(4)-\mathrm{C}(20)-\mathrm{O}(5)$ | $124.6(3)$ |
|  |  |


| $\mathrm{O}(4)-\mathrm{C}(20)-\mathrm{C}(21)$ | $117.6(3)$ |
| :--- | :--- |
| $\mathrm{O}(5)-\mathrm{C}(20)-\mathrm{C}(21)$ | $117.7(3)$ |
| $\mathrm{O}(6)-\mathrm{C}(21)-\mathrm{C}(23)$ | $108.4(3)$ |
| $\mathrm{O}(6)-\mathrm{C}(21)-\mathrm{C}(20)$ | $111.3(3)$ |
| $\mathrm{C}(23)-\mathrm{C}(21)-\mathrm{C}(20)$ | $108.3(3)$ |
| $\mathrm{O}(6)-\mathrm{C}(21)-\mathrm{C}(22)$ | $110.1(3)$ |
| $\mathrm{C}(23)-\mathrm{C}(21)-\mathrm{C}(22)$ | $110.7(3)$ |
| $\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(22)$ | $108.0(3)$ |
| $\mathrm{C}(28)-\mathrm{C}(22)-\mathrm{C}(24)$ | $109.5(3)$ |
| $\mathrm{C}(28)-\mathrm{C}(22)-\mathrm{C}(21)$ | $112.4(3)$ |
| $\mathrm{C}(24)-\mathrm{C}(22)-\mathrm{C}(21)$ | $111.5(3)$ |
| $\mathrm{C}(25)-\mathrm{C}(24)-\mathrm{C}(22)$ | $114.8(4)$ |
| $\mathrm{C}(24)-\mathrm{C}(25)-\mathrm{C}(26)$ | $114.2(6)$ |
| $\mathrm{C}(27)-\mathrm{C}(26)-\mathrm{C}(25)$ | $122.9(7)$ |
| $\mathrm{C}(29)-\mathrm{C}(28)-\mathrm{C}(22)$ | $125.5(4)$ |
| $\mathrm{C}(28)-\mathrm{C}(29)-\mathrm{C}(30)$ | $125.8(5)$ |
| $\mathrm{N}(2)-\mathrm{C}(31)-\mathrm{C}(32)$ | $111.6(3)$ |
| $\mathrm{N}(2)-\mathrm{C}(31)-\mathrm{C}(38)$ | $109.2(3)$ |
| $\mathrm{C}(32)-\mathrm{C}(31)-\mathrm{C}(38)$ | $113.2(3)$ |
| $\mathrm{C}(37)-\mathrm{C}(32)-\mathrm{C}(33)$ | $119.2(3)$ |
| $\mathrm{C}(37)-\mathrm{C}(32)-\mathrm{C}(31)$ | $122.3(3)$ |
| $\mathrm{C}(33)-\mathrm{C}(32)-\mathrm{C}(31)$ | $118.4(3)$ |
| $\mathrm{C}(34)-\mathrm{C}(33)-\mathrm{C}(32)$ | $121.0(4)$ |
| $\mathrm{C}(33)-\mathrm{C}(34)-\mathrm{C}(35)$ | $119.1(4)$ |
| $\mathrm{C}(36)-\mathrm{C}(35)-\mathrm{C}(34)$ | $120.8(4)$ |
| $\mathrm{C}(35)-\mathrm{C}(36)-\mathrm{C}(37)$ | $119.7(4)$ |
| $\mathrm{C}(32)-\mathrm{C}(37)-\mathrm{C}(36)$ | $120.1(4)$ |
|  |  |

Supporting Information: Brett E. Howard and K. A. Woerpel
Table 4. Anisotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for SI-6. The anisotropic displacement factor exponent takes the form: $-2 \pi^{2}\left[h^{2} a^{* 2} U^{11}+\ldots+2 h k a^{*} b^{*} U^{12}\right]$

|  | $\mathrm{U}^{11}$ | $\mathrm{U}^{22}$ | $\mathrm{U}^{33}$ | $\mathrm{U}^{23}$ | $\mathrm{U}^{13}$ | $\mathrm{U}^{12}$ |
| :--- | :--- | :--- | :--- | :---: | :--- | :--- |
|  |  |  |  |  |  |  |
| $\mathrm{O}(1)$ | $20(1)$ | $23(2)$ | $71(2)$ | $-2(1)$ | $-3(1)$ | $-1(1)$ |
| $\mathrm{O}(2)$ | $32(1)$ | $35(2)$ | $37(1)$ | $2(1)$ | $-9(1)$ | $-9(1)$ |
| $\mathrm{O}(3)$ | $51(2)$ | $26(2)$ | $148(3)$ | $14(2)$ | $46(2)$ | $6(2)$ |
| $\mathrm{N}(1)$ | $20(1)$ | $24(2)$ | $33(2)$ | $-1(1)$ | $-7(1)$ | $1(1)$ |
| $\mathrm{C}(1)$ | $21(2)$ | $27(2)$ | $31(2)$ | $1(2)$ | $-2(1)$ | $-3(2)$ |
| $\mathrm{C}(2)$ | $28(2)$ | $29(2)$ | $60(2)$ | $10(2)$ | $8(2)$ | $12(2)$ |
| $\mathrm{C}(3)$ | $21(2)$ | $29(2)$ | $44(2)$ | $0(2)$ | $-3(2)$ | $3(2)$ |
| $\mathrm{C}(4)$ | $28(2)$ | $129(5)$ | $58(3)$ | $46(3)$ | $2(2)$ | $20(3)$ |
| $\mathrm{C}(5)$ | $19(2)$ | $44(2)$ | $43(2)$ | $-4(2)$ | $-5(2)$ | $-1(2)$ |
| $\mathrm{C}(6)$ | $30(2)$ | $40(2)$ | $38(2)$ | $-1(2)$ | $-4(2)$ | $-6(2)$ |
| $\mathrm{C}(7)$ | $33(2)$ | $55(3)$ | $48(2)$ | $-2(2)$ | $-5(2)$ | $5(2)$ |
| $\mathrm{C}(8)$ | $71(3)$ | $66(3)$ | $48(2)$ | $15(2)$ | $3(2)$ | $-1(3)$ |
| $\mathrm{C}(9)$ | $25(2)$ | $42(2)$ | $38(2)$ | $6(2)$ | $0(2)$ | $9(2)$ |
| $\mathrm{C}(10)$ | $21(2)$ | $49(3)$ | $45(2)$ | $6(2)$ | $-7(2)$ | $1(2)$ |
| $\mathrm{C}(11)$ | $24(2)$ | $76(3)$ | $56(3)$ | $20(2)$ | $-6(2)$ | $2(2)$ |
| $\mathrm{C}(12)$ | $23(2)$ | $30(2)$ | $35(2)$ | $0(2)$ | $0(1)$ | $4(2)$ |
| $\mathrm{C}(13)$ | $39(2)$ | $35(2)$ | $32(2)$ | $3(2)$ | $-4(2)$ | $-6(2)$ |
| $\mathrm{C}(14)$ | $81(3)$ | $47(3)$ | $57(3)$ | $-6(2)$ | $-25(2)$ | $16(3)$ |
| $\mathrm{C}(15)$ | $86(4)$ | $62(3)$ | $68(3)$ | $-26(3)$ | $-33(3)$ | $5(3)$ |
| $\mathrm{C}(16)$ | $79(3)$ | $54(3)$ | $57(3)$ | $-1(3)$ | $-29(2)$ | $2(3)$ |
| $\mathrm{C}(17)$ | $63(3)$ | $77(4)$ | $57(3)$ | $2(3)$ | $-25(2)$ | $5(3)$ |
| $\mathrm{C}(18)$ | $52(3)$ | $72(3)$ | $44(2)$ | $-10(2)$ | $-22(2)$ | $14(3)$ |
| $\mathrm{C}(19)$ | $28(2)$ | $44(2)$ | $36(2)$ | $5(2)$ | $-4(2)$ | $-9(2)$ |
| $\mathrm{O}(4)$ | $19(1)$ | $30(2)$ | $45(1)$ | $0(1)$ | $-7(1)$ | $-1(1)$ |
| $\mathrm{O}(5)$ | $21(1)$ | $29(1)$ | $36(1)$ | $-2(1)$ | $-10(1)$ | $1(1)$ |
| $\mathrm{O}(6)$ | $30(1)$ | $26(2)$ | $89(2)$ | $0(1)$ | $-26(1)$ | $-2(1)$ |
| $\mathrm{N}(2)$ | $20(1)$ | $26(2)$ | $28(2)$ | $1(1)$ | $-7(1)$ | $1(1)$ |
| $\mathrm{C}(20)$ | $21(2)$ | $27(2)$ | $31(2)$ | $-1(2)$ | $-3(1)$ | $2(2)$ |
| $\mathrm{C}(21)$ | $19(2)$ | $26(2)$ | $43(2)$ | $2(2)$ | $-10(2)$ | $-1(1)$ |
| $\mathrm{C}(22)$ | $20(2)$ | $35(2)$ | $46(2)$ | $6(2)$ | $-5(2)$ | $-7(2)$ |
| $\mathrm{C}(23)$ | $25(2)$ | $46(2)$ | $35(2)$ | $-4(2)$ | $-10(2)$ | $2(2)$ |
| $\mathrm{C}(24)$ | $30(2)$ | $98(4)$ | $46(2)$ | $24(3)$ | $-10(2)$ | $-6(2)$ |
| $\mathrm{C}(25)$ | $39(3)$ | $197(8)$ | $43(3)$ | $13(4)$ | $2(2)$ | $-12(4)$ |
|  |  |  |  |  |  |  |


| $\mathrm{C}(26)$ | $43(3)$ | $432(19)$ | $53(3)$ | $44(7)$ | $0(3)$ | $14(6)$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{C}(27)$ | $78(5)$ | $460(20)$ | $83(5)$ | $96(9)$ | $-12(4)$ | $-59(9)$ |
| $\mathrm{C}(28)$ | $22(2)$ | $44(2)$ | $47(2)$ | $6(2)$ | $-6(2)$ | $-3(2)$ |
| $\mathrm{C}(29)$ | $28(2)$ | $65(3)$ | $56(3)$ | $3(2)$ | $-2(2)$ | $13(2)$ |
| $\mathrm{C}(30)$ | $25(2)$ | $129(5)$ | $65(3)$ | $0(3)$ | $-7(2)$ | $22(3)$ |
| $\mathrm{C}(31)$ | $27(2)$ | $32(2)$ | $34(2)$ | $3(2)$ | $3(2)$ | $1(2)$ |
| $\mathrm{C}(32)$ | $38(2)$ | $29(2)$ | $31(2)$ | $-4(2)$ | $-2(2)$ | $3(2)$ |
| $\mathrm{C}(33)$ | $64(3)$ | $47(3)$ | $33(2)$ | $5(2)$ | $7(2)$ | $6(2)$ |
| $\mathrm{C}(34)$ | $93(4)$ | $66(3)$ | $42(3)$ | $7(2)$ | $10(2)$ | $23(3)$ |
| $\mathrm{C}(35)$ | $99(4)$ | $81(4)$ | $29(2)$ | $2(2)$ | $-12(2)$ | $39(3)$ |
| $\mathrm{C}(36)$ | $63(3)$ | $86(4)$ | $41(3)$ | $-4(2)$ | $-16(2)$ | $26(3)$ |
| $\mathrm{C}(37)$ | $47(2)$ | $59(3)$ | $31(2)$ | $3(2)$ | $-6(2)$ | $4(2)$ |
| $\mathrm{C}(38)$ | $47(2)$ | $40(2)$ | $40(2)$ | $-1(2)$ | $3(2)$ | $11(2)$ |

Table 5. Hydrogen coordinates ( $\mathrm{x} 10^{4}$ ) and isotropic displacement parameters $\left(\AA^{2} \times 10^{3}\right)$ for SI-6.

|  | X | y | z | $\mathrm{U}(\mathrm{eq})$ |
| :---: | :---: | :---: | :---: | :---: |
| H(4) | 1970(30) | 2800(110) | 560(20) | 78(17) |
| H(1) | -307 | 11401 | 327 | 39 |
| H(2) | 408 | 10622 | 679 | 39 |
| H(3) | 157 | 13244 | 702 | 39 |
| H(3A) | 2789 | 9424 | 374 | 38 |
| H(4A) | 2052 | 5279 | 1559 | 107 |
| H(4B) | 2411 | 7978 | 1480 | 107 |
| H(4C) | 2937 | 5576 | 1421 | 107 |
| H(5A) | 2687 | 6004 | -681 | 42 |
| H(5B) | 2035 | 7997 | -563 | 42 |
| H(6A) | 2932 | 11221 | -706 | 44 |
| H(6B) | 3568 | 9197 | -847 | 44 |
| H(7A) | 2892 | 7988 | -1796 | 54 |
| H(7B) | 2222 | 9882 | -1649 | 54 |
| H(8A) | 3004 | 11505 | -2440 | 92 |
| H(8B) | 3727 | 11365 | -1943 | 92 |
| H(8C) | 3045 | 13238 | -1813 | 92 |
| H(9A) | 3706 | 5046 | 248 | 42 |
| H(10A) | 4084 | 9774 | 668 | 46 |
| H(11A) | 5338 | 8163 | 491 | 78 |
| H(11B) | 5013 | 5424 | 579 | 78 |
| H(11C) | 5097 | 7243 | 1188 | 78 |
| H(12A) | -843 | 12383 | 1311 | 35 |
| H(14A) | -250 | 14507 | 2203 | 75 |
| H(15A) | 612 | 14695 | 3102 | 87 |
| H(16A) | 1478 | 11637 | 3288 | 77 |
| H(17A) | 1555 | 8385 | 2587 | 80 |
| H(18A) | 683 | 8037 | 1698 | 68 |
| H(19A) | -1168 | 8622 | 793 | 54 |
| H(19B) | -1121 | 8203 | 1558 | 54 |
| H(19C) | -438 | 7300 | 1129 | 54 |
| H(8) | 3020(30) | 2100(130) | 4360(30) | 100(20) |

Supporting Information: Brett E. Howard and K. A. Woerpel

| H(5) | 5342 | 10419 | 4686 | 37 |
| :--- | ---: | ---: | ---: | ---: |
| H(6) | 4614 | 9556 | 4369 | 37 |
| H(7) | 4856 | 12173 | 4300 | 37 |
| H(22A) | 2235 | 8265 | 4749 | 40 |
| H(23A) | 2989 | 5883 | 3304 | 54 |
| H(23B) | 2608 | 8270 | 3614 | 54 |
| H(23C) | 2093 | 5928 | 3413 | 54 |
| H(24A) | 2187 | 3656 | 5433 | 70 |
| H(24B) | 2876 | 5565 | 5534 | 70 |
| H(25A) | 1347 | 6592 | 5831 | 112 |
| H(25B) | 2035 | 8508 | 5930 | 112 |
| H(26A) | 2152 | 3929 | 6575 | 211 |
| H(26B) | 2555 | 6461 | 6775 | 211 |
| H(27A) | 1857 | 5332 | 7457 | 311 |
| H(27B) | 1151 | 5081 | 6945 | 311 |
| H(27C) | 1509 | 7745 | 7107 | 311 |
| H(28A) | 1210 | 4230 | 4464 | 45 |
| H(29A) | 969 | 9271 | 4465 | 59 |
| H(30A) | -129 | 7852 | 3809 | 110 |
| H(30B) | -329 | 7764 | 4555 | 110 |
| H(30C) | -83 | 5288 | 4195 | 110 |
| H(31A) | 5937 | 10765 | 3771 | 37 |
| H(33A) | 5969 | 11274 | 2640 | 58 |
| H(34A) | 5368 | 12528 | 1671 | 80 |
| H(35A) | 4042 | 12909 | 1603 | 84 |
| H(36A) | 3327 | 11964 | 2479 | 77 |
| H(37A) | 3936 | 10621 | 3449 | 55 |
| H(38A) | 5894 | 6731 | 3377 | 64 |
| H(38B) | 5093 | 6265 | 3696 | 64 |
| H(38C) | 5846 | 6673 | 4146 | 64 |
|  |  |  |  | 54 |

Table 6. Torsion angles [ ${ }^{\circ}$ ] for SI-6.

| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{O}(3)$ | $-160.7(3)$ |
| :--- | :---: |
| $\mathrm{O}(2)-\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{O}(3)$ | $21.7(4)$ |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(4)$ | $77.5(4)$ |
| $\mathrm{O}(2)-\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(4)$ | $-100.1(4)$ |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $-44.5(5)$ |
| $\mathrm{O}(2)-\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $137.9(3)$ |
| $\mathrm{O}(3)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(9)$ | $-63.4(4)$ |
| $\mathrm{C}(4)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(9)$ | $58.4(4)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(9)$ | $177.1(3)$ |
| $\mathrm{O}(3)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(5)$ | $61.0(4)$ |
| $\mathrm{C}(4)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(5)$ | $-177.2(3)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(5)$ | $-58.5(4)$ |
| $\mathrm{C}(9)-\mathrm{C}(3)-\mathrm{C}(5)-\mathrm{C}(6)$ | $-68.8(4)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(5)-\mathrm{C}(6)$ | $167.1(3)$ |
| $\mathrm{C}(3)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | $-178.0(3)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | $176.6(4)$ |
| $\mathrm{C}(5)-\mathrm{C}(3)-\mathrm{C}(9)-\mathrm{C}(10)$ | $113.9(4)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(9)-\mathrm{C}(10)$ | $-121.4(4)$ |
| $\mathrm{C}(3)-\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | $178.5(3)$ |
| $\mathrm{N}(1)-\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(14)$ | $104.2(4)$ |
| $\mathrm{C}(19)-\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(14)$ | $-134.7(4)$ |
| $\mathrm{N}(1)-\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(18)$ | $-71.3(4)$ |
| $\mathrm{C}(19)-\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(18)$ | $-59.3(4)$ |
| $\mathrm{C}(18)-\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)$ | $49.8(5)$ |
| $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)$ | $-2.6(7)$ |
| $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16)$ | $-178.2(4)$ |
| $\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)$ | $1.3(8)$ |
| $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)$ | $1.2(9)$ |
| $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(18)-\mathrm{C}(17)$ | $-2.2(9)$ |
| $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(18)-\mathrm{C}(17)$ | $1.6(7)$ |
| $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(13)$ | $177.3(4)$ |
| $\mathrm{O}(4)-\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{O}(6)$ | $0.7(8)$ |
| $\mathrm{O}(5)-\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{O}(6)$ | $179.8(3)$ |
| $\mathrm{O}(4)-\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(23)$ | $-\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(23)$ |
| $\mathrm{O}(4)-\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(22)$ | $-4(4)$ |


| $\mathrm{O}(5)-\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(22)$ | $121.3(3)$ |
| :--- | :---: |
| $\mathrm{O}(6)-\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(28)$ | $-60.8(4)$ |
| $\mathrm{C}(23)-\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(28)$ | $59.0(4)$ |
| $\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(28)$ | $177.5(3)$ |
| $\mathrm{O}(6)-\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(24)$ | $62.6(4)$ |
| $\mathrm{C}(23)-\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(24)$ | $-177.5(3)$ |
| $\mathrm{C}(20)-\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(24)$ | $-59.1(4)$ |
| $\mathrm{C}(28)-\mathrm{C}(22)-\mathrm{C}(24)-\mathrm{C}(25)$ | $-66.4(5)$ |
| $\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(24)-\mathrm{C}(25)$ | $168.5(4)$ |
| $\mathrm{C}(22)-\mathrm{C}(24)-\mathrm{C}(25)-\mathrm{C}(26)$ | $-179.8(6)$ |
| $\mathrm{C}(24)-\mathrm{C}(25)-\mathrm{C}(26)-\mathrm{C}(27)$ | $-156.7(12)$ |
| $\mathrm{C}(24)-\mathrm{C}(22)-\mathrm{C}(28)-\mathrm{C}(29)$ | $113.1(5)$ |
| $\mathrm{C}(21)-\mathrm{C}(22)-\mathrm{C}(28)-\mathrm{C}(29)$ | $-122.4(4)$ |
| $\mathrm{C}(22)-\mathrm{C}(28)-\mathrm{C}(29)-\mathrm{C}(30)$ | $-177.9(4)$ |
| $\mathrm{N}(2)-\mathrm{C}(31)-\mathrm{C}(32)-\mathrm{C}(37)$ | $-31.0(5)$ |
| $\mathrm{C}(38)-\mathrm{C}(31)-\mathrm{C}(32)-\mathrm{C}(37)$ | $92.7(4)$ |
| $\mathrm{N}(2)-\mathrm{C}(31)-\mathrm{C}(32)-\mathrm{C}(33)$ | $150.8(3)$ |
| $\mathrm{C}(38)-\mathrm{C}(31)-\mathrm{C}(32)-\mathrm{C}(33)$ | $-85.5(4)$ |
| $\mathrm{C}(37)-\mathrm{C}(32)-\mathrm{C}(33)-\mathrm{C}(34)$ | $-0.4(6)$ |
| $\mathrm{C}(31)-\mathrm{C}(32)-\mathrm{C}(33)-\mathrm{C}(34)$ | $177.8(4)$ |
| $\mathrm{C}(32)-\mathrm{C}(33)-\mathrm{C}(34)-\mathrm{C}(35)$ | $0.9(7)$ |
| $\mathrm{C}(33)-\mathrm{C}(34)-\mathrm{C}(35)-\mathrm{C}(36)$ | $-0.7(8)$ |
| $\mathrm{C}(34)-\mathrm{C}(35)-\mathrm{C}(36)-\mathrm{C}(37)$ | $0.1(8)$ |
| $\mathrm{C}(33)-\mathrm{C}(32)-\mathrm{C}(37)-\mathrm{C}(36)$ | $-0.2(7)$ |
| $\mathrm{C}(31)-\mathrm{C}(32)-\mathrm{C}(37)-\mathrm{C}(36)$ | $-178.5(4)$ |
| $\mathrm{C}(35)-\mathrm{C}(36)-\mathrm{C}(37)-\mathrm{C}(32)$ | $0.4(8)$ |
|  |  |

## VII. Selected Spectra















## $=$





79








- $\quad$, out




S－5





