Appendix A Descriptors

This Appendix is based on an overview of the QSAR descriptors in MOE (Molecular Operating Environment) by the Chemical Computing Group. Please refer to http://www.chemcomp.com/journal/descr.htm for the original source. The following list shows the sub-groups of the 2D descriptors:

• 2D Molecular Descriptors

- Physical Properties $(2D_{phys})$
- Subdivided Surface Areas $(2D_{sa})$
- Atom Counts and Bond Counts (2D_{counts})
- Kier & Hall Connectivity and Kappa Shape Indices $(2D_{kh})$
- Adjacency and Distance Matrix Descriptors $(2D_{adj})$
- Pharmacophore Feature Descriptors $(2D_{pharm})$
- Partial Charge Descriptors $(2D_{charge})$

Table A.1: List of QSAR descriptors in the Molecular Operating Environment (MOE). The descriptions are based on the Chemical Computing Group's webpage [1]

| Code | Type | Used | Description |
|-------------------|-----------------------------|------|--|
| apol | $2\mathrm{D}_{phys}$ | | Sum of the atomic polarizabilities (including implicit hy- drogens) with polarizabilities taken from [2]. |
| bpol | $2\mathrm{D}_{phys}$ | | Sum of the absolute value of the difference between atomic polarizabilities of all bonded atoms in the molecule (in- cluding implicit hydrogens) with polarizabilities taken from [2]. |
| density | $2\mathrm{D}_{phys}$ | | Molecular mass density: Weight divided by vdw_vol. |
| FCharge | $2 \operatorname{D}_{phys}$ | V | Total charge of the molecule (sum of formal charges). |
| mr | 2 D _{phys} | | Molecular refractivity (including implicit hydrogens). This property is calculated from an 11 descriptor linear model [3] with R^2 =0.997, RMSE=0.168 on 1,947 small molecules. |
| SMR | $2\mathrm{D}_{phys}$ | | Molecular refractivity (including implicit hydrogens). This property is an atomic contribution model [4] that as- sumes the correct protonation state (washed structures). The model was trained on 7000 structures and results |
| | | | may vary from the mr descriptor. |
| $AM1_dipole$ | $2\mathrm{D}_{phys}$ | R | The dipole moment calculated using the AM1 Hamiltonian $$ |
| AM1_E | $2\mathrm{D}_{phys}$ | R | [5]. The total energy calculated using the AM1 Hamiltonian [5]. |
| $AM1_Eele$ | $2\mathrm{D}_{phys}$ | R | The electronic energy calculated using the AM1 Hamilto- |
| AM1_HF | $2\mathrm{D}_{phys}$ | R | nian [5]. The heat of formation calculated using the AM1 Hamilto- nian [5]. |
| AM1_IP | $2\mathrm{D}_{phys}$ | R | The ionization potatnial calculated using the AM1 Hamil- tonian [5]. |
| AM1_HOMO | $2 \operatorname{D}_{phys}$ | R | The energy of the Highest Unoccupied Molecular Orbital calculated using the AM1 Hamiltonian [5]. |
| AM1_LUMO | $2 \mathrm{D}_{phys}$ | R | The energy of the Lowest Unoccupied Molecular Orbital calculated using the AM1 Hamiltonian [5]. |
| MNDO_dipole | $2 \operatorname{D}_{phys}$ | R | The dipole moment calculated using MNDO Hamiltonian [5]. |
| MNDO_E | $2 \mathrm{D}_{phys}$ | R | The total energy calculated using the MNDO Hamiltonian [5]. |
| $MNDO_Eele$ | $2 \mathrm{D}_{phys}$ | R | The electronic energy calculated using the MNDO Hamil- tonian [5]. |
| MNDO_HF | $2\mathrm{D}_{phys}$ | R | The heat of formation calculated using the MNDO Hamil- tonian [5]. |
| MNDO_IP | $2 \mathrm{D}_{phys}$ | R | The ionization potatnial calculated using the MNDO Hamiltonian [5]. |
| MNDO_HOMO | $2 \mathrm{D}_{phys}$ | R | The energy of the Highest Unoccupied Molecular Orbital calculated using the MNDO Hamiltonian [5]. |
| MNDO_LUMO | $2 \operatorname{D}_{phys}$ | R | The energy of the Lowest Unoccupied Molecular Orbital calculated using the MNDO Hamiltonian [5]. |
| PM3_dipole | $2 D_{phys}$ | U | The dipole moment calculated using PM3 Hamiltonian [5]. |
| РМ3_Е | $2\mathrm{D}_{phys}$ | | The total energy calculated using the PM3 Hamiltonian [5]. |
| $PM3_Eele$ | $2\mathrm{D}_{phys}$ | | The electronic energy calculated using the PM3 Hamilto- nian [5]. |
| PM3_HF | $2 \operatorname{D}_{phys}$ | | The heat of formation calculated using the PM3 Hamilto- nian [5]. |
| PM3_IP | $2\mathrm{D}_{phys}$ | | The ionization potatnial calculated using the PM3 Hamil- tonian [5]. |
| РМ3 _ НОМО | $2\mathrm{D}_{phys}$ | | The energy of the Highest Unoccupied Molecular Orbital calculated using the PM3 Hamiltonian [5]. |

| PM3_LUMO | $2\mathrm{D}_{phys}$ | | The energy of the Lowest Unoccupied Molecular Orbital |
|-----------------|----------------------------|---|--|
| TPSA | $2\mathrm{D}_{phys}$ | | calculated using the PM3 Hamiltonian [5]. Total Polar Surface Area calculated using group contribu- tions to approximate the polar surface area from connec- |
| Weight | $2 \mathrm{D}_{phys}$ | | tion table information only [6]. Molecular weight (including implicit hydrogens) with |
| LogS | $2\mathrm{D}_{phys}$ | R | atomic weights taken from [2]. Log of the aqueous solubility calculated from an atom con- |
| $\rm logP(o/w)$ | $2\mathrm{D}_{phys}$ | | tributiontilear atom type model [7]. Log of the octanol/water partition coefficient (including implicit hydrogens). This property is calculated from a linear atom type model [8] with R^2 =0.931, RMSE=0.393 |
| SlogP | 2 D _{phys} | | on 1,847 molecules. Log of the octanol/water partition coefficient (including implicit hydrogens). This property is an atomic contribu- tion model [4] that calculates logP from the given struc- ture; i.e., the correct protonation state (washed struc- tures). Results may vary from the logP(o/w) descriptor. The training set for SlogP was 7000 structures. |
| vdw_vol | $2\mathrm{D}_{phys}$ | | van der Waals volume calculated using a connection table approximation. |
| vdw_area | $2\mathrm{D}_{phys}$ | | Area of van der Waals surface calculated using a connec- tion table approximation. |
| SlogP_VSA0 | $2 D_{sa}$ | | Sum of v_i such that $L_i := -0.4$. |
| SlogP_VSA1 | $2D_{sa}$ | | Sum of v_i such that L_i is in [-0.4,-0.2]. |
| SlogP_VSA2 | $2D_{sa}$ | U | Sum of v_i such that L_i is in [-0.2,0]. |
| SlogP_VSA3 | $^{2}\mathrm{D}_{sa}^{sa}$ | 0 | Sum of v_i such that L_i is in [0,0.1]. |
| - | | | |
| SlogP_VSA4 | $2D_{sa}$ | | Sum of v_i such that L_i is in [0.1,0.15]. |
| SlogP_vsa5 | $2D_{sa}$ | | Sum of v_i such that L_i is in $[0.15, 0.20]$. |
| SlogP_VSA6 | $2 D_{sa}$ | | Sum of v_i such that L_i is in [0.20,0.25]. |
| SlogP_VSA7 | $2 D_{sa}$ | | Sum of v_i such that L_i is in [0.25,0.30]. |
| $SlogP_VSA8$ | $2 D_{sa}$ | | Sum of v_i such that L_i is in [0.30,0.40]. |
| SlogP_VSA9 | $2 D_{sa}$ | | Sum of v_i such that $L_i \gtrsim 0.40$. |
| SMR_VSA0 | $2D_{sa}$ | | Sum of v_i such that R_i is in [0,0.11]. |
| SMR_VSA1 | $2D_{sa}$ | | Sum of v_i such that R_i is in [0.11,0.26]. |
| SMR_VSA2 | $2D_{sa}$ | | Sum of v_i such that R_i is in [0.26,0.35]. |
| SMR_VSA3 | $^{2}\mathrm{D}_{sa}$ | | Sum of v_i such that R_i is in [0.35,0.39]. |
| SMR_VSA4 | $2D_{sa}^{a}$ | | |
| | | | Sum of v_i such that R_i is in [0.39,0.44]. |
| SMR_VSA5 | $2D_{sa}$ | | Sum of v_i such that R_i is in [0.44,0.485]. |
| SMR_VSA6 | $2 D_{sa}$ | | Sum of v_i such that R_i is in [0.485,0.56]. |
| SMR_VSA7 | $2 D_{sa}$ | U | Sum of v_i such that $R_i \downarrow 0.56$. |
| nmol | $2D_{count}$ | V | Number of molecules. |
| a_aro | $2D_{count}$ | | Number of aromatic atoms. |
| a_count | $2D_{count}$ | | Number of atoms (including implicit hydrogens). |
| a_heavy | $2D_{count}$ | | Number of heavy $atoms(Z_i > 1)$. |
| a_ICM | $2\mathrm{D}_{count}$ | | Atom information content (mean). |
| a_IC | $2D_{count}$ | | Atom information content (total). |
| a_nH | $2D_{count}$ | | Number of hydrogen atoms (including implicit hydrogens). |
| a_nB | $2D_{count}$ | V | Number of boron atoms. |
| a_nC | $2D_{count}$ | • | Number of carbon atoms. |
| a⊒n0 a_nN | | | Number of nitrogen atoms. |
| a_nO | $2D_{count}$ | | 0 |
| | $2D_{count}$ | | Number of oxygen atoms. |
| a_nF | $2D_{count}$ | | Number of fluorine atoms. |
| a_nP | $2D_{count}$ | V | Number of phosphorus atoms. |
| a_nS | $2D_{count}$ | | Number of sulfur atoms. |
| a_nCl | $2 D_{count}$ | | Number of chlorine atoms. |
| a_nBr | $2D_{count}$ | V | Number of bromine atoms. |
| a_nI | $2D_{count}$ | | Number of iodine atoms. |
| b_1rotN | $2D_{count}$ | U | Number of rotatable single bonds. |
| b_1rotR | $2D_{count}$ | | Fraction of rotatable single bonds. |
| b_ar | $2D_{count}$ | | Number of aromatic bonds. |
| b_count | $2D_{count}$ | | Number of bonds (including implicit hydrogens). |
| | | | |
| | | | |

| b_double | $2\mathrm{D}_{count}$ | U | Number of double bonds. Aromatic bonds are not consid- |
|-------------------|--------------------------------|---|--|
| 1.1 | аD | | ered to be double bonds. |
| b_heavy b_notN | $2D_{count}$ | | Number of bonds between heavy atoms. Number of rotatable bonds |
| b_rotN | $2D_{count}$ | | Number of rotatable bonds. |
| b_rotR | $2D_{count}$ | | Fraction of rotatable bonds. |
| b_single | $2D_{count}$ | | Number of single bonds (including implicit hydrogens). Aromatic bonds are not considered to be single bonds. |
| b_triple | $2D_{count}$ | | Number of triple bonds. |
| VAdjMa | $2D_{count}$ | | Vertex adjacency information (magnitude): $1 + \log_2 m$ |
| | | | where m is the number of heavy-heavy bonds. If m is |
| | | | zero, then zero is returned. |
| VAdjEq | $2D_{count}$ | | Vertex adjacency information (equality): $-(1-f)log_2(1-f)$ |
| | | | f) – $f \log_2 f$ where $f = (n^2 - m)/n^2$, n is the number of |
| | | | heavy atoms and m is the number of heavy-heavy bonds. |
| | | | If f is not in the open interval $(0,1)$, then 0 is returned. |
| lip_acc | $2\mathrm{D}_{counts}$ | | Number of O and N atoms. |
| lip_don | $2 D_{counts}$ | | Number of OH and NH atoms. |
| lip_druglike | $2 D_{counts}$ | R | 1 if $lip_violation < 2$ otherwise 0. |
| lip_violation | $2 D_{counts}$ | R | Number of violations of Lipinski's Rule of Five [9]. |
| opr_brigid | $2 D_{counts}$ | | Number of rigid bonds [10]. |
| opr_nring | $2D_{counts}$ | | Number of ring bonds [10]. |
| opr_nrot | $2D_{counts}$ | | Number of rotatable bonds [10]. |
| opr_leadlike | $2D_{counts}$ | R | 1 if $opr_violation < 2$ otherwise 0. |
| opr_violation | $2D_{counts}$ | R | Number of violations of Oprea's lead-like test [10]. |
| chi0 | $2D_{kh}$ | | Atomic connectivity index (order 0) [11, 12]. |
| chi0_C | $2D_{kh}^{\kappa n}$ | | Carbon connectivity index (order 0). |
| chi1 | $2D_{kh}^{\kappa n}$ | | Atomic connectivity index (order 1) [11, 12]. |
| chi1_C | $2D_{kh}$ | | Carbon connectivity index (order 1). |
| chi0v | $2D_{kh}$ | | Atomic valence connectivity index (order 0) [11, 12]. |
| chi0v_C | $2D_{kh}$ | | Carbon valence connectivity index (order 0). |
| chi1v | $2D_{kh}$ | | Atomic valence connectivity index (order 1) [11, 12]. |
| chi1v_C | $2D_{kh}$ | | Carbon valence connectivity index (order 1). |
| Kier1 | $2D_{kh}^{kh}$ | | First kappa shape index [11]. |
| Kier2 | $2D_{kh}^{\kappa n}$ | | Second kappa shape index [11]. |
| Kier3 | $2D_{kh}^{kh}$ | | Third kappa shape index [11]. |
| KierA1 | $2D_{kh}$ | | First alpha modified shape index [11]. |
| KierA2 | $2D_{kh}$ | | Second alpha modified shape index [11]. |
| KierA3 | $2D_{kh}$ | | Third alpha modified shape index [11]. |
| KierFlex | $2D_{kh}$ | | Kier molecular flexibility index [11]. |
| zagreb | $2D_{kh}$ | | Zagreb index. |
| balabanJ | $2D_{adj}$ | | Balaban's connectivity topological index [13]. |
| diameter | $^{2}\mathrm{D}_{adj}$ | | Largest value in the distance matrix [14]. |
| petitjean | $2D_{adj}$ | | Value of (diameter-radius) / diameter as defined in [14]. |
| radius | $2D_{adj}$ | | Radius is defined as the smallest of the r_i [14]. |
| BCUT_PEOE_0 | $2 D_{a d j}$ $2 D_{a d j}$ | | The BCUT descriptors are calculated from the eigenvalues |
| _ = = 1 | - a aj | | of a modified adjancy matrix [15]. Smallest eigenvalue. |
| BCUT_PEOE_1 | $2\mathrm{D}_{adj}$ | | The BCUT descriptors are calculated from the eigenvalues |
| | <i>a ay</i> | | of a modified adjancy matrix [15]. 1/3-ile eigenvalue. |
| BCUT_PEOE_2 | $2 D_{adj}$ | | The BCUT descriptors are calculated from the eigenvalues |
| Deelin Lollin | 2D a aj | | of a modified adjancy matrix [15]. 2/3-ile eigenvalue. |
| BCUT_PEOE_3 | $2D_{adj}$ | | The BCUT descriptors are calculated from the eigenvalues |
| Deel_I Holl_5 | 2 D a dj | | of a modified adjancy matrix [15]. Largest eigenvalue. |
| BCUT_SLOGP_0 | $2D_{adi}$ | | The BCUT descriptors using LogP contribution instead of |
| DOOT_DDOOT_0 | $2D_{adj}$ | | partial charge. Smallest eigenvalue. |
| BCUT_SLOGP_1 | $2 D_{adj}$ | | The BCUT descriptors using LogP contribution instead of |
| DCUI_SLOGI_I | $2D_{adj}$ | | |
| DOUT SLOOD A | аD | | partial charge. 1/3-ile eigenvalue. |
| BCUT_SLOGP_2 | $2 D_{a dj}$ | | The BCUT descriptors using LogP contribution instead of |
| BOUT SLOOP : | эD | | partial charge. 2/3-ile eigenvalue. |
| BCUT_SLOGP_3 | $2 D_{a dj}$ | | The BCUT descriptors using LogP contribution instead of |
| DOUT GMD A | эD | | partial charge. Largest eigenvalue. |
| BCUT_SMR_0 | $2 D_{adj}$ | | The BCUT descriptors using atomic contribution to molar refractivity instead of partial charge. Smallest eigenvalue |
| | | | refractivity instead of partial charge. Smallest eigenvalue. |
| | | | |

| BCUT_SMR_1 | $2 \mathbf{D}_{adj}$ | | The BCUT descriptors using atomic contribution to molar |
|--------------|------------------------|---|--|
| BCUT_SMR_2 | $2\mathrm{D}_{adj}$ | | refractivity instead of partial charge. 1/3-ile eigenvalue. The BCUT descriptors using atomic contribution to molar |
| BCUT_SMR_3 | $2 \mathbf{D}_{adj}$ | | refractivity instead of partial charge. 2/3-ile eigenvalue. The BCUT descriptors using atomic contribution to molar refractivity instead of partial charge. Largest eigenvalue |
| GCUT_PEOE_0 | $2\mathbf{D}_{adj}$ | | refractivity instead of partial charge. Largest eigenvalue. The GCUT descriptors are calculated from the eigenvalues of a modified graph distance adjancy matrix. Smallest |
| GCUT_PEOE_1 | $2 D_{adj}$ | | eigenvalue. The GCUT descriptors are calculated from the eigenval- ues of a modified graph distance adjancy matrix. 1/3-ile |
| GCUT_PEOE_2 | $2 D_{adj}$ | U | eigenvalue. The GCUT descriptors are calculated from the eigenval- ues of a modified graph distance adjancy matrix. 2/3-ile |
| GCUT_PEOE_3 | $2 \mathrm{D}_{adj}$ | | eigenvalue. The GCUT descriptors are calculated from the eigenval- ues of a modified graph distance adjancy matrix. Largest |
| GCUT_SLOGP_0 | $2 \mathbf{D}_{adj}$ | | eigenvalue. The GCUT descriptors using LogP contribution instead of partial charge. Smallest eigenvalue. |
| GCUT_SLOGP_1 | $2 \mathbf{D}_{adj}$ | | The GCUT descriptors using LogP contribution instead of partial charge. 1/3-ile eigenvalue. |
| GCUT_SLOGP_2 | $2\mathrm{D}_{adj}$ | | The GCUT descriptors using LogP contribution instead of partial charge. 2/3-ile eigenvalue. |
| GCUT_SLOGP_3 | $2\mathrm{D}_{adj}$ | | The GCUT descriptors using LogP contribution instead of partial charge. Largest eigenvalue. |
| GCUT_SMR_0 | $2\mathrm{D}_{adj}$ | | The GCUT descriptors using atomic contribution to molar refractivity instead of partial charge. Smallest eigenvalue. |
| GCUT_SMR_1 | $2\mathrm{D}_{adj}$ | | The GCUT descriptors using atomic contribution to molar refractivity instead of partial charge. 1/3-ile eigenvalue. |
| GCUT_SMR_2 | $2\mathrm{D}_{adj}$ | | The GCUT descriptors using atomic contribution to molar refractivity instead of partial charge. 2/3-ile eigenvalue. |
| GCUT_SMR_3 | $2\mathrm{D}_{adj}$ | | The GCUT descriptors using atomic contribution to molar refractivity instead of partial charge. Largest eigenvalue. |
| VDistEq | $2 \mathbf{D}_{adj}$ | | VdistEq is defined as the sum of $log_2 m - log_2 p_i/m$ where p_i is the number of distance matrix entries equal to i. |
| VDistMa | $2\mathrm{D}_{adj}$ | | VDistMa is defined as the sum of $log_2 m - D_{ij} log_2 D_{ij}/m$ over all i and j. |
| weinerPath | $2D_{adj}$ | | Wiener path number [16, 17]. |
| weinerPol | $2 D_{a dj}$ | | Wiener polarity number [16]. |
| a_acc | $2 D_{pharm}$ | | Number of hydrogen bond acceptor atoms (not counting |
| | | | acidic atoms but counting atoms that are both hydrogen |
| | | | bond donors and acceptors such as -OH). |
| a_acid | $2 D_{pharm}$ | V | Number of acidic atoms. |
| a_base | $2 D_{pharm}$ | V | Number of basic atoms. |
| a_don | $2 D_{pharm}$ | | Number of hydrogen bond donor atoms (not counting ba- |
| | | | sic atoms but counting atoms that are both hydrogen bond |
| | | | donors and acceptors such as -OH). |
| a_hyd | $2D_{pharm}$ | | Number of hydrophobic atoms. |
| vsa_acc | $2 D_{pharm}$ | | Approximation to the sum of VDW surface areas of pure |
| | | | hydrogen bond acceptors (not counting acidic atoms and atoms that are both hydrogen bond donors and acceptors such as -OH). |
| vsa_acid | $2 \mathrm{D}_{pharm}$ | V | Approximation to the sum of VDW surface areas of acidic atoms. |
| vsa_base | $2 \mathrm{D}_{pharm}$ | V | Approximation to the sum of VDW surface areas of basic atoms. |
| vsa_don | $2 \mathrm{D}_{pharm}$ | | Approximation to the sum of VDW surface areas of pure hydrogen bond donors (not counting basic atoms and atoms that are both hydrogen bond donors and acceptors such as -OH). |

| vsa_hyd | $2\mathrm{D}_{pharm}$ | Approximation to the sum of VDW surface areas of hy- drophobic atoms. |
|----------------|-------------------------|--|
| vsa_other | $2 D_{pharm}$ | Approximation to the sum of VDW surface areas of atoms |
| vsa_pol | $2 D_{pharm}$ | typed as "other". Approximation to the sum of VDW surface areas of polar (both hydrogen bond donors and acceptors) atoms (such as OH) |
| PC+ | $2\mathrm{D}_{charge}$ | as -OH). Total positive partial charge (Suffix: Q for external charges, PEOE for calculated charges). |
| PC- | $2 D_{charge}$ U | Total negative partial charge (Suffix: Q for external charges, PEOE for calculated charges). |
| RPC+ | $2\mathrm{D}_{charge}$ | Relative positive partial charge (Suffix: Q for external charges, PEOE for calculated charges). |
| PRC- | $2 D_{charge}$ | Relative negative partial charge (Suffix: Q for external charges, PEOE for calculated charges). |
| VSA_POS | $2 D_{charge}$ U | Total positive van der Waals surface area (Suffix: Q for external charges, PEOE for calculated charges). |
| VSA_NEG | $2D_{charge}$ | Total negative van der Waals surface area (Suffix: Q for external charges, PEOE for calculated charges). |
| VSA_PPOS | $2 D_{charge}$ | Total positive polar van der Waals surface area (Suffix: Q for external charges, PEOE for calculated charges). |
| VSA_PNEG | $2 \mathrm{D}_{charge}$ | Total negative polar van der Waals surface area (Suffix: Q for external charges, PEOE for calculated charges). |
| VSA_HYD | $2 \mathrm{D}_{charge}$ | Total hydrophobic van der Waals surface area (Suffix: Q for external charges, PEOE for calculated charges). |
| VSA_POL | $2\mathrm{D}_{charge}$ | Total polar van der Waals surface area (Suffix: Q for ex- ternal charges, PEOE for calculated charges). |
| VSA_FPOS | $2\mathrm{D}_{charge}$ | Fractional positive van der Waals surface area (Suffix: Q for external charges, PEOE for calculated charges). |
| VSA_FNEG | $2 D_{charge}$ | Fractional negative van der Waals surface area (Suffix: Q for external charges, PEOE for calculated charges). |
| VSA_FPPOS | $2 D_{charge}$ | Fractional positive polar van der Waals surface area (Suf- fix: Q for external charges, PEOE for calculated charges). |
| VSA_FPNEG | $2 D_{charge}$ | Fractional negative polar van der Waals surface area (Suf- fix: Q for external charges, PEOE for calculated charges). |
| VSA_FHYD | $2 D_{charge}$ | Fractional hydrophobic van der Waals surface area (Suffix: Q for external charges, PEOE for calculated charges). |
| VSA_FPOL | $2 D_{charge}$ | Fractional polar van der Waals surface area (Suffix: Q for external charges, PEOE for calculated charges). |
| PEOE_VSA+6 | $2 D_{charge}$ | Sum of v_i where q_i is greater than 0.3. |
| PEOE_VSA+5 | $2D_{charge}$ | Sum of v_i where q_i is in the range [0.25,0.30]. |
| PEOE_VSA+4 | $2 D_{charge}$ | Sum of v_i where q_i is in the range [0.20,0.25]. |
| PEOE_VSA+3 | $2 D_{charge}$ | Sum of v_i where q_i is in the range [0.15,0.20]. |
| $PEOE_VSA+2$ | $2 D_{charge}$ | Sum of v_i where q_i is in the range [0.10,0.15]. |
| PEOE_VSA+1 | $2 D_{charge}$ | Sum of v_i where q_i is in the range $[0.05, 0.10]$. |
| $PEOE_VSA+0$ | $2 D_{charge}$ | Sum of v_i where q_i is in the range [0.00,0.05]. |
| PEOE_VSA-0 | $2 D_{charge}$ | Sum of v_i where q_i is in the range [-0.05,0.00]. |
| PEOE_VSA-1 | $2 D_{charge}$ | Sum of v_i where q_i is in the range [-0.10,-0.05]. |
| PEOE_VSA-2 | $2 D_{charge}$ | Sum of v_i where q_i is in the range [-0.15,-0.10]. |
| PEOE_VSA-3 | $2 D_{charge}$ | Sum of v_i where q_i is in the range [-0.20,-0.15]. |
| $PEOE_VSA-4$ | $2 D_{charge}$ | Sum of v_i where q_i is in the range [-0.25,-0.20]. |
| $PEOE_{VSA-5}$ | $2 D_{charge}$ | Sum of v_i where q_i is in the range [-0.30,-0.25]. |
| PEOE_VSA-6 | $2D_{charge}$ | Sum of v_i where q_i is less than -0.30. |
| E | $3D_{Epot}$ | Value of the potential energy. |
| E_ang | $3D_{Epot}$ | Angle bend potential energy. |
| E_ele | $3D_{Epot}$ | Electrostatic component of the potential energy. |
| E_nb | $3D_{Epot}$ | Value of the potential energy with all non-bonded terms disabled. |
| E_oop | $3D_{Epot}$ | Out-of-plane potential energy. |
| E_sol | $3D_{Epot}$ U | Solvation energy. |
| E_stb | $3D_{Epot}$ | Bond stretch-bend cross-term potential energy. |
| E_{str} | $3D_{Epot}$ | Bond stretch potential energy. |
| | | |

| E_strain | $3D_{Epot}$ | Local strain energy: the current energy minus the value of the energy at a near local minimum. |
|-----------------|------------------------------|--|
| E_tor | $3D_{Epot}$ | Torsion potential energy. |
| E_vdw | | van der Waals component of the potential energy. |
| E_rele | $3D_{Epot}$ $3D_{Epot}$ V | Electrostatic interaction energy (x3d). |
| E_rele E_rnb | 11 p 0 0 | 0. () |
| | Dpor | Non-bonded interaction energy $(x3d)$. |
| E_rsol | $3D_{Epot}$ V | Solvation free energy difference (x3d). |
| E_rvdw | $3D_{Epot}$ V | van der Waals interaction energy (x3d). |
| ASA | $3D_{shape}$ | Water accessible surface area calculated using a radius of |
| | _ | 1.4 A for the water molecule. |
| dens | $3D_{shape}$ | Mass density: molecular weight divided by van der Waals |
| | | volume. |
| glob | $3D_{shape}$ | Globularity, or inverse condition number of the covariance |
| | | matrix of atomic coordinates. |
| pmi | $3D_{shape}$ | Principal moment of inertia. |
| pmiX | $3D_{shape}$ | x component of the principal moment of inertia (x3d). |
| pmiY | $3D_{shape}$ | y component of the principal moment of inertia (x3d). |
| pmiZ | $3D_{shape}$ | z component of the principal moment of inertia (x3d). |
| rgyr | $3D_{shape}$ | Radius of gyration. |
| std_dim1 | 3D _{shape} | Standard dimension 1: the square root of the largest eigen- |
| | enapo | value of the covariance matrix of the atomic coordinates. |
| std_dim2 | $3D_{shape}$ | Standard dimension 2: the square root of the second |
| | snape | largest eigenvalue of the covariance matrix of the atomic |
| | | coordinates. |
| std_dim3 | $3D_{shape}$ | Standard dimension 3: the square root of the third largest |
| | 32 shape | eigenvalue of the covariance matrix of the atomic coordi- |
| | | nates. |
| vol | $3D_{shape}$ | van der Waals volume calculated using a grid approxima- |
| VOI | 5D _{shape} | tion (spacing 0.75 A). |
| VSA | 3D . | van der Waals surface area. |
| ASA+ | 3D _{shape} | Water accessible surface area of all atoms with positive |
| ASAT | $3D_{conf}$ | partial charge. |
| ASA- | ъD | . 0 |
| ASA- | $3D_{conf}$ | Water accessible surface area of all atoms with negative |
| ACA TI | аD | partial charge. |
| ASA_H | $3D_{conf}$ | Water accessible surface area of all hydrophobic atoms. |
| ASA_P | $3D_{conf}$ | Water accessible surface area of all polar atoms. |
| DASA | $3D_{conf}$ | Absolute value of the difference between ASA+ and ASA |
| CASA+ | $3D_{conf}$ | Positive charge weighted surface area [18]. |
| CASA- | $3D_{conf}$ | Negative charge weighted surface area [18]. |
| DCASA | $3D_{conf}$ | Absolute value of the difference between CASA+ and |
| | _ | CASA- [18]. |
| dipole | $3D_{conf}$ | Dipole moment calculated from the partial charges of the |
| | | molecule. |
| dipoleX | $3D_{conf}$ | The x component of the dipole moment (x3D). |
| dipoleY | $3D_{conf}$ | The y component of the dipole moment (x3D). |
| dipoleZ | $3D_{conf}$ | The z component of the dipole moment (x3D). |
| FASA+ | $3D_{conf}$ | Fractional ASA $+$ calculated as ASA $+$ / ASA. |
| FASA- | $3D_{conf}$ | Fractional ASA- calculated as ASA- / ASA. |
| FCASA+ | $3D_{conf}$ | Fractional CASA $+$ calculated as CASA $+$ / ASA. |
| FCASA- | $3D_{conf}$ | Fractional CASA- calculated as CASA- / ASA. |
| FASA_H | 3D _{conf} | Fractional ASA_H calculated as ASA_H / ASA. |
| FASA_P | 3D _{conf} | Fractional ASA_P calculated as ASA_P / ASA. |
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