(Supporting Information)

A New Osmium-Based Reagent for the Dihydroxylation of Alkenes

Timothy J. Donohoe, ** Robert M. Harris, * Sam Butterworth, * Jeremy N. Burrows * Andew Cowley, * and Jeremy S. Parker*

timothy.donohoe@chem.ox.ac.uk

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^a Department of Chemistry, University of Oxford, Chemistry Research Laboratory, Mansfield Road, Oxford, OX1 3TA, UK

AstraZeneca Pharmaceuticals, Mereside, Alderley Park, Macclesfield, Cheshire, SK10 4TG, UK
AstraZeneca, Process R&D Avlon/Charnwood, Avlon Works, Severn Road, Hallen, Bristol,
BS10 7ZE, UK

General Details.

Proton nuclear magnetic resonance spectra (NMR) were recorded at 400 MHz. ¹³C NMR spectra were recorded at 100 MHz. Infrared spectra (IR) were recorded as evaporated films. Chemical ionisation (CI) was performed using NH₃. Electrospray ionisation (ESI) was performed using tetraoctylammoniun bromide as the lock mass. All solvents and reagents requiring purification were purified using standard laboratory techniques according to methods published by Perrin, ¹ apart from CH₂Cl₂, THF and Et₂O which were dried by filtration through an activated alumina purification column. Petrol refers to petroleum ether in the boiling range 40-60 °C.

General Experimental Procedures

General Procedure A. Osmium(VI) intermediate formation with *t*-BuOOH:

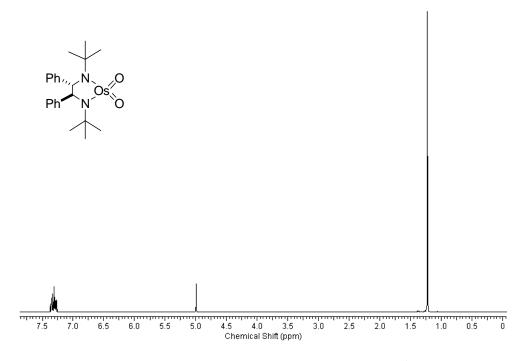
A 70% aqueous solution of *tert*-butyl hydroperoxide (12 eq.) was added to a solution of osmium(VI) reagent **7** (1 eq.) and alkene (2 eq.) in CHCl₃ (50 ml/mmol). The reaction was stirred for 5 min at room temperature, dried (MgSO₄) and evaporated directly onto silica. The product was isolated by flash column chromatography (SiO₂, petrol \rightarrow petrol-Et₂O₃ 80:20).

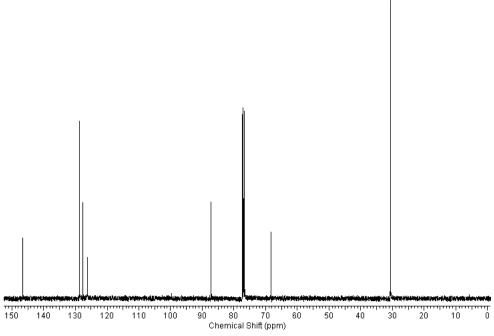
General Procedure B. Osmium(VI) intermediate formation from 1,2-diamines:

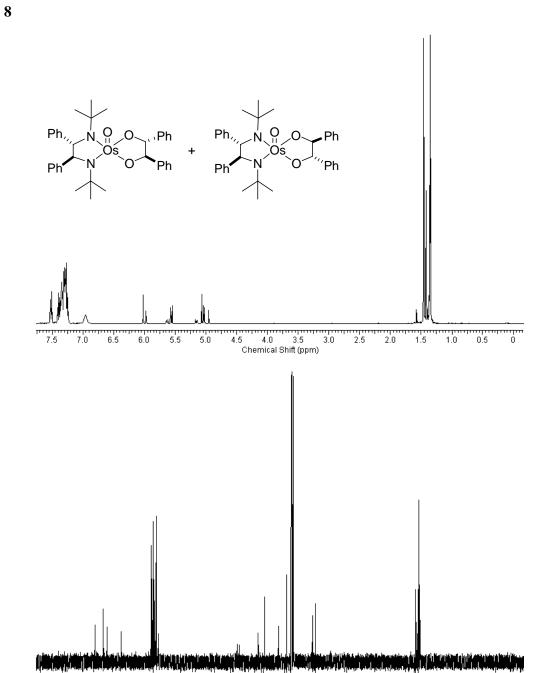
A solution of diamine (1 eq.) and alkene (1.5 eq.) in toluene (50 ml/mmol) was cooled to -78°C and stirred for 15 min. A solution of osmium tetroxide (1.1 eq.) in toluene (10 ml/mmol) was added dropwise, the reaction stirred for 10 min at -78°C and allowed to warm to room temperature. The reaction mixture was evaporated under reduced pressure and the product isolated directly by flash column chromatography (SiO₂, petrol \rightarrow petrol-Et₂O, 80:20).

References

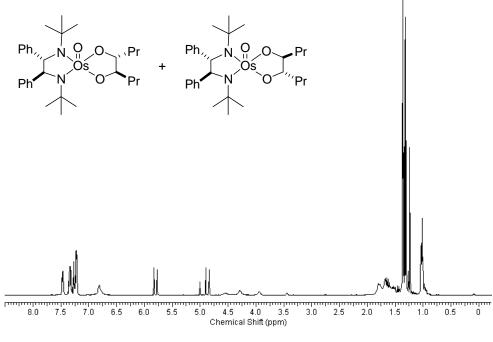
1. Armarego, W. L. F.; Perrin, D. O. *Purification of Laboratory Chemicals*, Butterworth and Heinemann, Oxford, 1996 (4th Edition).

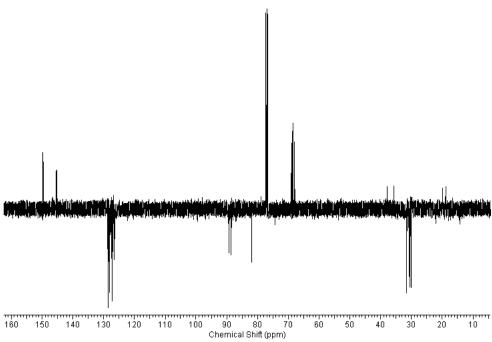


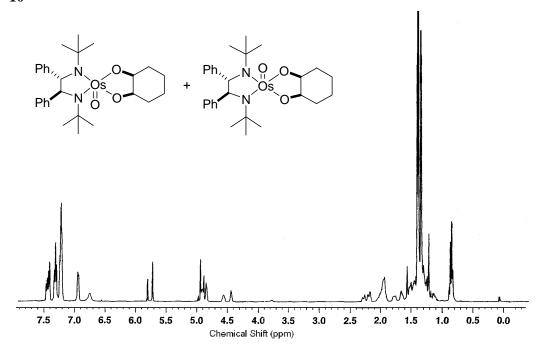


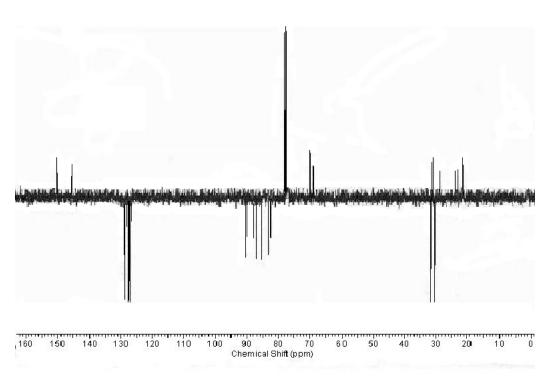


160 150 140 130 120 110 100 90 80 70 Chemical Shift (ppm)

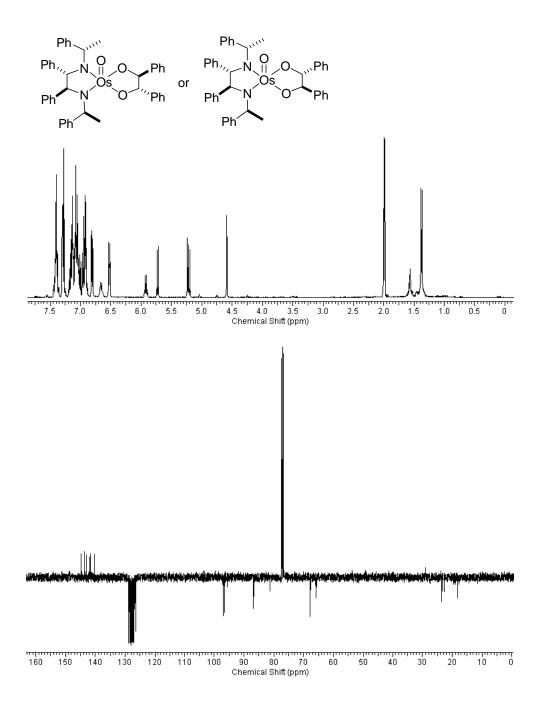








13 Steroisomer α :



Stereoisomer β :

