Supporting Information

Visible Light Photocatalytic Aerobic Oxygenation of Indoles and pH as Chemoselective Switch

Chenhao Zhang,^a Sanliang Li,^a Filip Bureš,^b Richmond Lee,^c Xinyi Ye,^d and Zhiyong Jiang^{*a}

E-mail: chmjzy@henu.edu.cn

^{a.}Key Laboratory of Natural Medicine and Immuno-Engineering of Henan Province, Henan University, Kaifeng, Henan

^{b.} Institute of Organic Chemistry and Technology, University of Pardubice, Faculty of Chemical Technology, Studentská 573, Pardubice, 53210, Czech Republic

^c ARC Centre of Excellence for Electromaterials Science and Research School of Chemistry, Australian National University, Canberra ACT 2601, Australia

^d Division of Chemistry and Biological Chemistry, Nanyang Technological University, 21 Nanyang Link, 637371, Singapore

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1. General information

Column chromatography was carried out with *silica gel* 60 (particle size 0.040–0.063 mm, 230–400 mesh) or neutral alumina (200–300 mesh) and commercially available solvents. Thin-layer chromatography (TLC) was conducted on aluminum sheets coated with *silica gel* 60 F254 with visualization by a UV lamp (254 or 360 nm). Melting points (m.p.) were measured in open capillaries and were uncorrected.

¹H and ¹³C NMR spectra were recorded at 300 and 75 MHz at 25 °C with a 300 MHz instrument. Chemical shifts are reported in parts per million (ppm), using the residual solvent signal as an internal standard: CDCl₃ (¹H NMR: δ 7.26, singlet; ¹³C NMR: δ 77.0, triplet). Multiplicities were given as: s (singlet), d (doublet), t (triplet), q (quartet), quintet, m (multiplets), dd (doublet of doublets), dt (doublet of triplets), and br (broad). Coupling constants (J) were recorded in Hertz (Hz). The number of proton atoms (n) for a given resonance was indicated by *n*H. The number of carbon atoms (n) for a given resonance was indicated by nC. HRMS was reported in units of mass of charge ratio (m/z). Mass samples were dissolved in DCM and MeOH (HPLC Grade) unless otherwise stated. Electrochemical measurements were carried out by cyclic voltammetry (CV). The cyclic voltammetry was performed with an Autolab Potentiostat under nitrogen atmosphere in a one-compartment electrolysis cell consisting of a platinum wire working electrode, a platinum wire counter electrode, and a quasi Ag/AgCl reference electrode. Cyclic voltammograms were monitored at scan rates of either 100 mV•s⁻¹ or 50 mV•s⁻¹ and recorded in distilled acetonitrile. The concentration of the complex was maintained at 0.5 mM or less and each solution contained 0.1 M of tetrabutylammonium hexafluorophosphate (TBAP) as the electrolyte.

2. General experimental procedure for the synthesis of 2



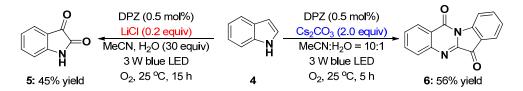
Conditions I: 28 µL (0.0005 mmol, 0.005 eq.) of **DPZ** solution (1.0 mg of **DPZ** in 160 µL of CH₃CN) was added into a 10.0 mL sample vial, and CH₃CN was then removed *in vacuo*. Subsequently, **1** (0.1 mmol, 1.0 eq.), LiBr (0.02 mmol, 0.2 eq.) and solvent (2.0 mL CH₃CN and 30 eq. of H₂O) were sequentially added. The reaction mixture was stirred under the irradiation by a 3 W blue LED ($\lambda = 450-455$ nm) at 25 °C under an oxygen atmosphere (The sample vial was fitted with an oxygen balloon and the temperature was maintained in an incubator). The reaction was monitored by TLC. Upon complete consumption of **1**, the solvent was removed *in vacuo*. The reaction mixture was then loaded onto a short *silica gel* column, followed by gradient elution with petroleum ether/ethyl acetate (10/1–3/1 ratio). Removing the solvent *in vacuo*, afforded products **2a-k**.

3. General experimental procedure for the synthesis of 3



Conditions II: 28 µL (0.0005 mmol, 0.005 eq.) of **DPZ** solution (1.0 mg of **DPZ** in 160 µL of MeCN) was added into a 10.0 mL sample vial, and CH₃CN was then removed *in vacuo*. Next, **1** (0.1 mmol, 1.0 eq.), K₃PO₄ (0.3 mmol, 3.0 eq.) and solvent (2.0 mL, CH₃CN:H₂O = 10:1, v/v) were added into reaction system successively. The reaction mixture was stirred under the irradiation by a 3 W blue LED (λ = 450–455 nm) at 25 °C under an oxygen atmosphere (The sample vial was fitted with an oxygen balloon and the temperature was maintained in an incubator). The reaction was monitored by TLC. Upon complete consumption of **1**, the solvent was removed *in vacuo*. The reaction mixture was then loaded onto a short *silica gel* column, followed by gradient elution with petroleum ether/ethyl acetate (10/1–2/1 ratio). Removing the solvent *in vacuo*, afforded products **3a-k**. The yields could be improved when 10 mol% TEMPO as an additive was added.

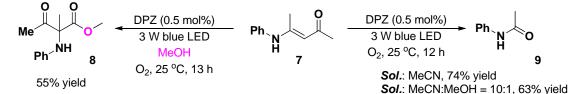
4. Experimental procedure for the synthesis of 5 and 6



The synthesis of 5: 28 μ L (0.0005 mmol, 0.005 eq) of DPZ solution (1.0 mg of DPZ in 160 μ L of CH₃CN) was added into a 10.0 mL sample vial, and CH₃CN was then removed *in vacuo*. Subsequently, **4** (0.1 mmol, 1.0 eq.), LiCl (0.2 mmol, 2.0 eq.) and solvent (2.0 mL of CH₃CN and 30.0 eq. of H₂O) were added sequentially. The reaction mixture was stirred under the irradiation by a 3 W blue LED (λ = 450–455 nm) at 25 °C under an oxygen atmosphere (The sample vial was fitted with an oxygen balloon and the temperature was maintained in an incubator). After 15 hours, the solvent was removed *in vacuo*. The reaction mixture was then loaded onto a short *silica gel* column, followed by gradient elution with petroleum ether/ethyl acetate (10/1–3/1 ratio). Removing the solvent *in vacuo*, afforded product **5**.

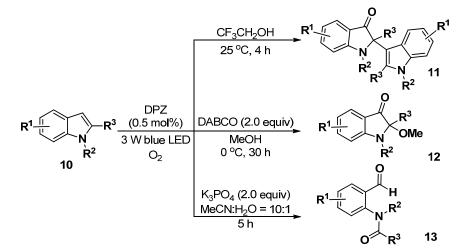
The synthesis of 6: 28 μ L (0.0005 mmol, 0.005 eq.) of DPZ solution (1.0 mg of DPZ in 160 μ L of MeCN) was added into a 10.0 mL sample vial, and CH₃CN was then removed *in vacuo*. Next, **4** (0.1 mmol, 1.0 eq.), Cs₂CO₃ (0.2 mmol, 2.0 eq.) and solvent (2.0 mL, CH₃CN/H₂O = 10:1, v/v) were added into the reaction system successively. The reaction mixture was stirred under the irradiation by a 3 W blue LED (λ = 450–455 nm) at 25 °C under an oxygen atmosphere (The sample vial was fitted with an oxygen balloon and the temperature was maintained in an incubator). After 5 hours, the solvent was removed *in vacuo*. The reaction with petroleum ether/ ethyl acetate (10/1–5/1 ratio). Removing the solvent *in vacuo*, afforded product **6**.

5. Experimental procedure for the synthesis of 8 and 9



The synthesis of 8: 28 μ L (0.0005 mmol, 0.005 eq.) of DPZ solution (1.0 mg of DPZ in 160 μ L of MeCN) was added into a 10.0 mL sample vial, and CH₃CN was then removed *in vacuo*. Next, 7 (0.1 mmol, 1.0 eq.) and solvent (2.0 mL MeOH) were added into the reaction system successively. The reaction mixture was stirred under the irradiation by a 3 W blue LED (λ = 450–455 nm) at 25 °C from a 5 cm distance. The reaction was monitored by TLC. Upon complete consumption of 7, the solvent was removed *in vacuo*. The reaction mixture was then loaded onto a short *silica gel* column, followed by gradient elution with petroleum ether/ethyl acetate. Removing the solvent *in vacuo*, afforded product 8.

The synthesis of 9: 28 μ L (0.0005 mmol, 0.005 eq.) of DPZ solution (1.0 mg of DPZ in 160 μ L of MeCN) was added into a 10.0 mL sample vial, and CH₃CN was then removed *in vacuo*. Next, **7** (0.1 mmol, 1.0 eq.) and solvent (2.0 mL MeCN or 2.0 mL MeCN/MeOH =10:1) were added into the reaction system successively. The reaction mixture was stirred under the irradiation by a 3 W blue LED (λ = 450–455 nm) at 25 °C from a 5 cm distance. The reaction was monitored by TLC. Upon complete consumption of **7**, the solvent was removed *in vacuo*. The reaction mixture was then loaded onto a short *silica gel* column, followed by gradient elution with petroleum ether/ethyl acetate. Removing the solvent *in vacuo*, afforded product **9**.

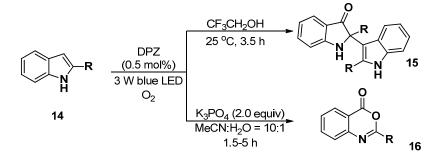


6. General experimental procedure for the synthesis of 11, 12 and 13

The synthesis of 11: 28 μ L (0.0005 mmol, 0.005 eq.) of DPZ solution (1.0 mg of DPZ in 160 μ L of MeCN) was added into a 10.0 mL sample vial, and CH₃CN was then removed *in vacuo*. Next, **10** (0.1 mmol, 1.0 eq.) and solvent (1.0 mL, CF₃CH₂OH) were added into the reaction system successively. The reaction mixture was stirred under the irradiation by a 3 W blue LED ($\lambda = 450-455$ nm) at 25°C (The sample vial was fitted with an oxygen balloon and the temperature was maintained in an incubator). The reaction was monitored by TLC. Upon complete consumption of **10**, the solvent was removed *in vacuo*. The reaction mixture was then loaded onto a short neutral alumina column, followed by gradient elution with petroleum ether/ethyl acetate (10/1–5/1 ratio). Removing the solvent *in vacuo*, afforded products **11a-e**.

The synthesis of 12: 28 μ L (0.0005 mmol, 0.005 eq.) of DPZ solution (1.0 mg of DPZ in 160 μ L of MeCN) was added into a 10.0 mL sample vial, and then solvent was removed *in vacuo*. Next, **10** (0.1 mmol, 1.0 eq.), DABCO (0.2 mmol, 2.0 equiv) and solvent (2.0 mL, MeOH) were added into the reaction system successively. The reaction mixture was stirred under irradiation by a 3 W blue LED ($\lambda = 450-455$ nm) at 0 °C (The sample vial was fitted with an oxygen balloon and the temperature was maintained in an incubator). The reaction was monitored by TLC. Upon complete consumption of **10**, the solvent was removed *in vacuo*. The reaction mixture was then loaded onto a short neutral alumina column, followed by gradient elution with petroleum ether/ethyl acetate (20/1–5/1 ratio). Removing the solvent *in vacuo*, afforded products **12a-e**.

The synthesis of 13: 28 μ L (0.0005 mmol, 0.005 eq.) of DPZ solution (1.0 mg of DPZ in 160 μ L of MeCN) was added into a 10.0 mL sample vial, and CH₃CN was then removed *in vacuo*. Next, **10** (0.1 mmol, 1.0 eq.), K₃PO₄ (0.2 mmol, 2.0 eq.) and solvent (2.0 mL, MeCN/H₂O = 10:1, v/v) were added into the reaction system successively. The reaction mixture was stirred under the irradiation by a 3 W blue LED (λ = 450–455 nm) at 25 °C (The sample vial was fitted with an oxygen balloon and the temperature was maintained in an incubator). The reaction was monitored by TLC. Upon complete consumption of **10**, the solvent was removed *in vacuo*. The reaction mixture was then loaded onto a short *silica gel* column, followed by gradient elution with petroleum ether/ethyl acetate (10/1–2/1 ratio). Removing the solvent *in vacuo*, afforded products **13a-e**. The yields could be improved when extra 10 mol% TEMPO was added.

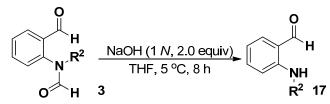


7. General experimental procedure for the synthesis of 15 and 16

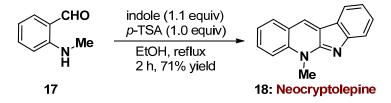
The synthesis of 15: 28 μ L (0.0005 mmol, 0.005 eq.) of DPZ solution (1.0 mg of DPZ in 160 μ L of MeCN) was added into a 10.0 mL sample vial, and CH₃CN was then removed *in vacuo*. Next, **14** (0.1 mmol, 1.0 eq.) and solvent (1.0 mL CF₃CH₂OH) were added into the reaction system successively. The reaction mixture was stirred under the irradiation by a 3 W blue LED ($\lambda = 450-455$ nm) at 25 °C (The sample vial was fitted with an oxygen balloon and the temperature was maintained in an incubator). The reaction was monitored by TLC. Upon complete consumption of **14**, the solvent was removed *in vacuo*. The reaction mixture was then loaded onto a short neutral alumina column, followed by gradient elution with petroleum ether/ethyl acetate (10/1–5/1 ratio). Removing the solvent *in vacuo*, afforded products **15a-b**

The synthesis of 16: 28 μ L (0.0005 mmol, 0.005 eq.) of DPZ solution (1.0 mg of DPZ in 160 μ L of MeCN) was added into a 10.0 mL sample vial, and CH₃CN was then removed *in vacuo*. Next, **14** (0.1 mmol, 1.0 eq.), K₃PO₄ (0.2 mmol, 2.0 eq.), AcOH (0.01mmol, 0.1 equiv) and solvent (2.0 mL, MeCN/H₂O = 10:1, v/v) were added into the reaction system successively. The reaction mixture was stirred under the irradiation by a 3 W blue LED (λ = 450–455 nm) at 25 °C from a 5 cm distance. The reaction was monitored by TLC. Upon complete consumption of **14**, the solvent was removed *in vacuo*. The reaction mixture was then loaded onto a short *silica gel* column, followed by gradient elution with petroleum ether/ethyl acetate (10/1–5/1 ratio). Removing the solvent *in vacuo*, afforded products **16a-b**.

8. Experimental procedure for the synthesis of 18



An ice-cooled solution of compounds **3** (0.5 mmol, 1.0 eq.) in THF (3.0 mL) was treated with 1 *N* aqueous sodium hydroxide (1.0 mmol, 2.0 eq.) keeping the temperature below 8 °C. After the starting material was consumed, the mixture was diluted with brine and extracted twice with ethyl ether. The organic solution was washed with water until neutral and then with brine, dried over sodium sulfate and concentrated *in vacuo* to afford crude product. The crude product was further purified by SiO₂ column chromatography, and the corresponding aldehyde **17** was obtained.



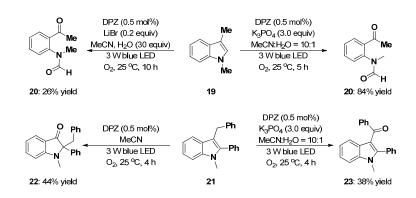
A mixture of compound **17** (0.5 mmol, 1.0 eq.), indole (0.5 mmol, 1.0 eq.) and *p*-TsOH (95 mg, 0.5 mmol, 1.0 eq.) in ethanol (5.0 mL) was stirred open to air in a 25 mL round bottom flask in reflux. After 2 hours, the mixture was cooled to room temperature and washed using 1 *N* NaOH (30 mL). The aqueous layer was extracted with CH_2Cl_2 (3 x 40 mL). The combined organic layers were then dried with anhydrous sodium sulfate. The organic layer was concentrated under reduced pressure. The crude product was subsequently purified by flash chromatography on alumina, affording neocryptolepine **18** in 71% yield.

9. Optimization of the reaction conditions

Table S1: Optimization of the reaction conditions^a

	U /L /	Z (0.5 mol%) W blue LED	+	O H N	
	1	, 25 ℃, 11 h 2a	3a	о∕∽н	
entry	solvent	additive	t (h)	$\operatorname{conv}(\%)^{\mathrm{b}}$	$2a/3a^b$
1	CH ₂ Cl ₂		9	7	N.D
2	Toluene		9	6	N.D
3	CH ₃ CN		9	>99	5:1
4	CH ₃ CN	NaH ₂ PO ₄ (0.1 eq.)	9	>99	9:1
5	CH ₃ CN	LiHPO ₄ (0.1 eq.)	9	>99	10:1
6	CH ₃ CN	LiCl (1.0 eq.)	9	>99	14:1
7	CH ₃ CN	LiBr (0.1 eq.)	9	>99 ^c	18:1
8	CH ₃ CN, H ₂ O (30 eq.)	LiCl (1.0 eq.), Py (1.0 eq.)	9	>99 ^d	20:1
9	CH ₃ CN, H ₂ O (30 eq.)	LiBr (0.1 eq.)	9	>99 ^e	20:1
9	CH ₃ CN, H ₂ O (30 eq.)	LiBr (0.2 eq.)	9	>99 ^f	20:1
10	CH ₃ CN	K ₂ CO ₃ (3 eq.)	9	62% ^g	1:4
11	CH ₃ CN	Cs ₂ CO ₃ (3 eq.)	9	58% ^g	1:6
12	CH ₃ CN	K ₃ PO ₄ (3 eq.)	9	66% ^g	1:7
13	CH ₃ CN/H ₂ O (10:1)	K ₃ PO ₄ (3 eq.)	9	74% ^g	1:20

^aThe reaction was performed on a 0.05 mmol scale, a 3 W blue LED (450–455 nm), 25 °C, ambient atmosphere. ^bDetermined by crude ¹H NMR. ^cYield (isol.) of 2a = 41%. ^dYield (*isol.*) of 2a = 53%. ^eYield (*isol.*) of 2a = 57%. ^fYield (*isol.*) of 2a = 68%. ^gIsolated yield.



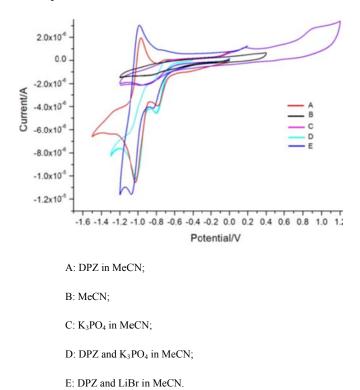
10. The preliminary study of photooxygenation of 1,3-disubstituted and 1,2,3-trisubstituted indoles.

In the presence of the reaction conditions A, the oxidation of 1,3-disubstituted indole **19** was messy, and product **20** was obtained in 26% yield. It was found that **20** could be achieved in 84% yield when under the reaction conditions B. While the products were same under two different reaction conditions, it could be deduced that the two reactions would go through SET and ET paths individually.

The reaction of 1,2,3-trisubstituted indole **21** was found to undergo an oxidation/semipinacol rearrangement process,¹ affording to product **22** in 44% yield. On the contrary, an oxygenation product **23** was obtained in 38% yield when in the presence of reaction conditions B. Interestingly, the double bond of **21** could not be oxidized, probably due to the more easily oxidized benzylic sp³ C–H.

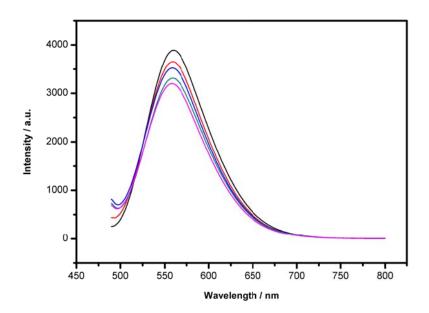
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 (a) Lerch, S.; Unkel, L.-N.; Brasholz, M. Angew. Chem., Int. Ed. 2014, 53, 6558–6562. (b) Ding, W.; Zhou, Q.-Q.; Xuan, J.; Li, T.-R.; Lu, L.-Q.; Xiao, W.-J. Tetrahedron Lett. 2014, 55, 4648–4652.)



11. . Cyclic voltammetry (CV) measurements

12. Phosphorescence quenching experiments



1.8 mg (0.005 mmol, 0.005 equiv) of **DPZ**, 637 mg of K_3PO_4 (3 mmol, 3.0 equiv) and 131.2 mg (1mmol, 1.0 equiv) of 1-methylindole **1a** were added in a 25 mL sample vial. 20.0 mL of CH₃CN (HPLC Grade) were sequentially added, and vigorously shaked for two minutes. Then 4.0 mL of mixture was used for the phosphorescence quenching experiment. The following four experiments were performed through increasing amounts of **1a** 65.6 mg (0.5 mmol, 0.5 equiv) successively. It was found that the phosphorescence intensity decreased in the presence of increasing concentration of **1a**.

13. Computational method

All density functional theory electronic structure calculations were carried out with Gaussian 09¹. Gas-phase geometry optimization of these structures were done with Truhlar's meta-hyrid functional M06-2X² with Pople's 6-31+G(d,p) basis set³. Subsequent single point calculations were completed with the same functional and larger triple zeta Wiegend and Aldrich basis set⁴ under SMD⁵ polarizable continuum model: def2TZVPP/SMD. SMD water parameters were used for fluorescein anion and acetonitrile for DPZ. Zero-point corrections (298K) at M06-2X/6-31+G(d,p) level of theory were used to correct the energies attained from single point M06-2X/def2TZVPP/SMD. Eigenvalues of the minimum structures' Hessian matrix were checked to ensure all are positive. Benchmarking calculations with fluorescein anion to ascertain that the described DFT method is accurate. The calculated singlet (ground state) and triplet energy gap, $E_{\rm T}$, for the fluorescein anion is 1.93 eV. This singlet-triplet energy gap is close to the experimental of 1.96 eV.⁶

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Cartesian Coordinates and Energies

All energies in hatrees.

 $DPZ(S_0)$

E = -1781.474323

ZPE = 0.2352600

- C 0.9520568085 0.1312772114 -0.7796283197
- C -0.0328855802 0.7749976808 -0.0719690765
- C -1.2862853225 0.1228366595 -0.2260598011
- C -1.2475724915 -0.9626948273 -1.0612167363
- H 0.124502481 1.6599998862 0.5291585117
- H-2.2005706317 0.4329015041 0.2665173479
- S 0.3717517026 -1.2467603686 -1.6544003732
- O 2.2572755637 0.4048355482 -0.9002392948
- C 2.7132181143 1.5369599539 -0.1718339135
- H 3.7800578235 1.6146294196 -0.3731816729
- H 2.5425105395 1.3947046318 0.9006878036
- H 2.199768348 2.441950405 -0.5142495157
- C -2.3508313269 -1.8935786199 -1.2965329747
- C -4.1476324959 -3.8800408496 -1.4945875541
- C -2.5624841839 -2.5926762932 -2.5259128128
- N -3.1710624531 -2.0945851326 -0.264824841
- C -4.0920209914 -3.0493608164 -0.3659044105
- N -3.4127286415 -3.6185974489 -2.572343173
- C -1.919108521 -2.2407769634 -3.7913640286
- C -1.1124955658 -1.1266390435 -5.8462374478
- C -1.1119934445 -2.4938535854 -5.9726822838
- C -1.5979844221 -3.1193189737 -4.7925570328
- H -0.7892005184 -3.0264061337 -6.8567338931
- H -1.7186806079 -4.1898941839 -4.6742727952
- O -0.7465981534 -0.1725624096 -6.7115923626
- C -0.2932572804 -0.6370320473 -7.9760715983
- Н 0.5947234572 -1.2667491891 -7.8542784269

H -0.0435927397 0.2524824385 -8.5515305608

- H -1.0842704037 -1.2017049895 -8.481168111
- S -1.6732903573 -0.5864221231 -4.2987735572
- C -5.0014258875 -3.2274745673 0.7414982353
- C -5.0326951566 -5.0199264419 -1.5435732453
- N -5.7424151531 -5.9322427055 -1.5692879111
- N -5.737782508 -3.3772776253 1.6201458259
- $DPZ(T_1)$
- E = -1781.397531
- ZPE = 0.2325040
- C 1.0291569767 0.1695055152 -0.5750685312
- C 0.0086928572 0.8203536923 0.0928199077
- C -1.2339129426 0.1968781524 -0.1366502204
- C -1.1622076763 -0.8939072939 -0.9869373517
- H 0.1502319501 1.6942063317 0.7144068116
- H-2.1759116842 0.5153071929 0.293302842
- S 0.4851645562 -1.1889718722 -1.4969270844
- O 2.335714328 0.43222257 -0.6137060292
- C 2.7712190169 1.5416602227 0.1650013418
- H 3.8480987269 1.6021728597 0.0213111649
- H 2.5405845989 1.3778357777 1.2225748649
- H 2.2929967784 2.4619201426 -0.1860196752
- C -2.2432948038 -1.7794731746 -1.3247045956
- C -3.7980932187 -3.964317938 -1.5076058353
- C -2.2261679924 -2.6503723319 -2.5097144201
- N -3.2530629182 -1.8410410778 -0.4768287517
- C -4.1322129728 -2.8466389931 -0.6349485677
- N -2.8642757865 -3.8948970426 -2.3970362961
- C -1.792198931 -2.3080875492 -3.7785956763

C -1.1481748647 -1.2133516192 -5.9256294582
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H -1.3622094832 -3.0654986293 -7.058315081
H -2.0664932255 -4.2272112104 -4.807870033
O -0.7444872549 -0.2913852306 -6.7953599358
C -0.568734132 -0.7282404541 -8.1402818014
Н 0.1965776454 -1.509449582 -8.18686777
H -0.2454854168 0.147771509 -8.6985676181
H -1.5151171684 -1.1029310395 -8.5428510177
S -1.3228947037 -0.6767921634 -4.2674146023
C -5.2953940926 -2.879308882 0.1857379911
C -4.5097567237 -5.2190427045 -1.3621642997
N -5.0893634991 -6.2117524538 -1.2384157689
N -6.2580512505 -2.9094765996 0.8343254908

Fluorescein anion (S₀)

- E = -1145.066646
- ZPE = 0.2622870
- H -2.5273271319 2.0440562848 -0.1397370488
- C -2.4384877212 0.9627213791 -0.1578932018
- C -2.2032738666 -1.821829127 -0.1457534229
- C -1.1572119453 0.3908096627 -0.1499500289
- C -3.5754848288 0.177104486 -0.1608523945
- C -3.4515691565 -1.220073703 -0.1513531211
- C -1.0720622688 -1.0080542501 -0.1407114412
- C 0.0678364461 1.1670515148 -0.1299773165
- O 0.122119884 -1.6502550237 -0.1327229554
- $C \quad 1.2668519421 \quad 0.4947031141 \ \textbf{-} 0.1346455758$
- C 1.2953478961 -0.9516814232 -0.1308981504

C 2.4464998805 -1.673768933 -0.1540172323
C 3.7416417726 -1.0136451016 -0.1621351723
C 3.7122230329 0.4596243946 -0.1211960008
C 2.5541059879 1.1546509897 -0.1092118949
Н 2.55668276 2.2395004942 -0.0625295638
C 0.0108662462 2.6406434142 -0.3522468137
C -0.1012183563 5.3703183978 -0.9487092851
C 0.3802545946 3.1071237552 -1.6216776276
C -0.4172425988 3.5474608681 0.6262335696
C -0.4717368677 4.906491991 0.3105967188
C 0.3272900765 4.46540384 -1.9210253156
H 0.7118426803 2.3931585184 -2.3720568003
H -0.8112111676 5.5739623515 1.0975902465
Н 0.6224582174 4.812732867 -2.9071833073
H -0.1408303535 6.4328784336 -1.1744700365
O -4.5453241272 -2.0379491377 -0.152080658
H -5.3429864244 -1.5007546881 -0.0940426214
C -0.8110207593 3.0639153581 2.0435330091
O -1.297470996 3.9319397721 2.7924510381
O -0.5922668083 1.8462167557 2.2450956448
O 4.809520953 -1.6365960714 -0.1994595541
H -2.1025317527 -2.9004814722 -0.1322318801
Н 2.4160608417 -2.7572432989 -0.166405612
Н 4.6761256947 0.9582498885 -0.0905970408
H -4.5604184457 0.6366970186 -0.152246023

Fluorescein anion (T₁)

E = -1144.992658

ZPE = 0.2590980

H -2.6616731909 1.975849167 -0.6253282506

C -2.5400931036 0.9011050702 -0.5423035216					
C -2.2373838398 -1.8581247808 -0.2540067161					
C -1.233212332 0.3871077627 -0.3477354422					
C -3.6513797109 0.0743513949 -0.6079092651					
C -3.5013651014 -1.3070502434 -0.4663514184					
C -1.1359380518 -1.0237910434 -0.18935074					
C -0.0725953182 1.188675856 -0.2480187299					
O 0.0585929128 -1.6374198979 0.0444717625					
C 1.2060549576 0.5093471724 -0.0242075541					
C 1.2178596812 -0.898786953 0.1008199309					
C 2.3810343049 -1.6093038684 0.2684591032					
C 3.6579747559 -0.9339782095 0.310331028					
C 3.6346566674 0.5031062144 0.1763245603					
C 2.4516083463 1.1800225573 0.0121307017					
Н 2.4469586944 2.2620202763 -0.0714735528					
C -0.1249052025 2.6546642097 -0.4326934284					
C -0.191458317 5.4257999752 -0.8611407359					
C -0.4121873422 3.1773855648 -1.7035684382					
C 0.112499893 3.5371907724 0.633397956					
C 0.0819398743 4.9140587643 0.4042219981					
C -0.4463768788 4.5523308787 -1.9199708913					
H -0.6001503387 2.488470246 -2.524496708					
H 0.2677398642 5.5567099283 1.2600307077					
H -0.6652612418					
H -0.2132427928 6.5003044691 -1.0234675961					
O -4.5629591952 -2.1727735632 -0.5218853872					
H -5.3765374684 -1.6662920359 -0.6182026386					
C 0.3714724312 3.0165041032 2.0623064065					
O 0.5087438726 3.8799780501 2.9457295905					

- H 4.5859847217 1.0230192933 0.2189011956
- H -4.6405765346 0.5026559194 -0.7542830016

14. Characterization data of products

Red solid, Mp 121.7–123.5 °C, 63% yield; ¹H NMR (300 MHz, CDCl₃)
$$\delta$$

7.59–7.54 (m, 2H), 7.10 (t, J = 7.6, 1H), 6.88 (d, J = 7.6, 1H), 3.23 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 183.3, 158.2, 151.4, 138.4, 125.1, 123.8, 117.3, 109.9,

26.1; HRMS (ESI) m/z 162.0554 (M+H⁺), calc. for C₉H₈NO₂ 162.0555.

Red solid, Mp 121.1–122.3 °C; 65% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.53 (d, J = 7.4, 1H), 7.40 (t, J = 7.9, 1H), 7.26 (s, 5H), 7.01 (t, J = 7.4, 1H), 6.70 (d, J = 7.9, 1H), 4.85 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 183.2, 158.2,

150.7, 138.30, 134.4, 129.0, 128.1, 127.4, 125.4, 123.8, 117.6, 111.0, 44.0; HRMS (ESI) m/z 260.0690 (M+Na⁺), calc. for C₁₅H₁₁NO₂Na 260.0687.

Red solid, Mp 127.3–128.7 °C; 67% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.69 (d, J = 7.4, 1H), 7.56–7.54 (m, 3H), 7.43–7.41 (m, 3H), 7.17 (t, J = 7.4, 1H), 6.90 (d, J = 7.9, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 182.9, 157.3, 151.6, 138.3,

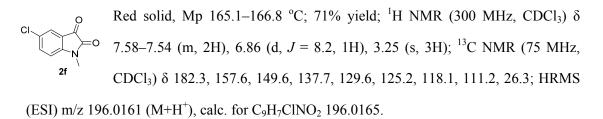
132.8, 129.9, 128.8, 126.0, 125.6, 124.3, 117.4, 111.3; HRMS (ESI) m/z 224.0715 (M+H⁺), calc. for C₁₄H₁₀NO₂ 224.0712.

Red solid, Mp 151.6–153.2 °C; 60% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.53 (t, J = 8.2, 1H), 6.62 (t, J = 8.2, 1H), 6.45 (t, J = 8.2, 1H), 3.97 (s, 3H), 3.22 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 179.9, 158.7, 158.6, 152.3, 140.3, 107.6, 105.7, 102.1, 56.3, 26.4; HRMS (ESI) m/z 192.0659 (M+H⁺), calc. for C₁₀H₁₂NO₃ 192.0661.



Red solid, Mp 150.7–152.5 °C; 65% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.30–7.21 (m, 2H), 6.82 (d, J = 7.0, 1H), 3.19 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 182.7, 160.9, 157.9, 157.7, 147.5, 124.8, 124.5, 117.9, 117.8, 112.4,

112.1, 111.1, 111.0, 26.3; HRMS (ESI) m/z 180.0466 (M+H⁺), calc. for C₉H₇FNO₂ 180.0461.



^{Br} ^{2g} ^{Red solid, Mp 150.0–151.7 °C; 60% yield; ¹H NMR (300 MHz, CDCl₃) $\delta =$ 7.72–7.60 (m, 2H), 6.81 (d, J = 7.9, 1H), 3.24 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 182.12, 157.4, 150.1, 140.6, 128.00), 118.5, 116.6, 111.61, 26.32;}

HRMS (ESI) m/z 239.9663 (M+H⁺), calc. for $C_9H_6BrNO_2$ 239.9660.

Red solid, Mp 150.5–151.2 °C; 64% yield; ¹H NMR (300 MHz, CDCl₃) $\delta =$ 7.37–7.30 (m, 2H), 6.75 (d, J = 7.9, 1H), 3.17 (s, 3H), 2.28 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 183.5, 158.2, 149.1, 138.7, 133.5, 125.3, 117.2, 109.7, 26.0, 20.5; HRMS (ESI) m/z 176.0708 (M+H⁺), calc. for C₁₀H₁₀NO₂ 176.0712.

Orange solid, Mp 161.9–163.1 °C; 66% yield; ¹H NMR (300 MHz, CDCl₃) δ Br i (75 MHz, CDCl₃) δ 182.0, 158.0, 152.1, 133.6, 127.0, 126.3, 116.0, 113.7,

26.4; HRMS (ESI) m/z 239.9653 (M+H⁺), calc. for $C_9H_7BrNO_2$ 239.9660.

Red solid, Mp 162.2–163.5 °C; 56% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.57 (d, J = 8.4, 1H), 7.42 (s, 5H), 6.62 (d, J = 8.4, 1H), 6.45 (s, 1H), 5.18 (s, 2H), 3.20 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 180.7, 167.4, 159.6, 154.0

135.2, 128.8, 128.6, 127.9, 127.6, 111.3, 108.5, 98.0, 70.9, 26.1; HRMS (ESI) m/z 268.0972 (M+H⁺), calc. for C₁₆H₁₄NO₃ 268.0974.



Yellow solid, Mp 133.5–134.8 °C; 66% yield; ¹H NMR (300 MHz, CDCl₃) δ 8.47 (d, J = 5.1, 1H), 7.83 (d, J = 7.3, 1H), 7.11–7.07 (m, 1H), 3.34 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 181.9, 163.9, 158.4, 155.9, 132.8, 119.6, 112.1, 25.1;

HRMS (ESI) m/z 163.0512 (M+H⁺), calc. for $C_8H_7N_2O_2$ 163.0508.

Pale yellow solid , Mp 69.0–70.1 °C; 78% yield (86% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 10.03 and 9.92 (s, 1H, CHO rotameric), **3a** O H 8.32 and 8.18 (s, 1H, CHO rotameric), 7.91 (t, *J* = 7.3 Hz, 1H), 7.64 (t, *J* = 7.3 Hz, 1H), 7.47 (t, *J* = 7.3 Hz, 1H), 7.23 (dd, *J* = 16.5, 11.7 Hz, 1H), 3.36 and 3.28 (s, 3H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 189.2, 189.0, 162.9, 162.7, 143.4, 141.2, 135.3, 135.1, 131.7, 130.9, 130.5, 128.4, 127.7, 127.4, 37.8, 34.5; HRMS (ESI) m/z 164.0708 (M+H⁺), calc. for C₉H₁₀NO₂ 164.0712. Yellow oil, 76% yield (96% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, $H_{\text{N},\text{Bn}}$ CDCl₃) δ 9.77 (s, 1H), 8.64 and 8.27 (s, 1H, CHO rotameric), 7.92–7.87 (m, $3b_{\text{O},\text{H}}$ 1H), 7.64–7.55 (m, 1H), 7.43 – 7.51 (m, *J* = 11.3, 7.5, 1H), 7.31–7.28 (m, 1H), 7.25–7.03 (m, 5H), 4.98 and 4.83 (s , 2H, CH₂ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 189.2, 189.0, 162.8, 162.3, 142.0, 139.9, 135.3, 135.2, 134.9, 132.8, 132.3, 130.4, 129.2, 129.1, 129.0, 128.8, 128.6, 128.6, 128.5, 128.2, 55.2, 50.8; HRMS (ESI) m/z 240.1022 (M+H⁺), calc. for C₁₅H₁₄NO₂ 240.1025.

Yellow oil, 71% yield (85% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, H_{Ph} CDCl₃) δ 10.14 and 10.07 (s, 1H, CHO rotameric), 8.84 and 8.54 (s, 1H, CHO $3c \circ H_{H}$ rotameric), 8.01 (dd, J = 7.7, 1.6, 1H), 7.73–7.65 (m, 1H), 7.60–7.48 (m, 1H), 7.43–7.27 (m, 4H), 7.25–7.13 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 189.0, 188.9, 162.3, 161.6, 141.6, 140.4, 135.4, 135.1, 132.2, 130.6, 130.2, 129.9, 129.7, 129.3, 128.9, 128.7, 126.9, 126.6, 124.5,123.4; HRMS (ESI) m/z 226.0870 (M+H⁺), calc. for C₁₄H₁₂NO₂ 226.0868.



Yellow solid, MP 74.0–75.0 °C; 58% yield (89% yield, 10 mol% TEMPO used); ^H ¹H NMR (300 MHz, CDCl₃) δ 10.45 and 10.42 (s, 1H, CHO rotameric), 8.17 and 8.12 (s, 1H, CHO rotameric), 7.57 (t, J = 8.2, 1H), 7.02 (d, J = 8.5, 1H),

6.82 (d, J = 7.8, 1H), 3.96 and 3.93 (s, 1H, CH₃ rotameric), 3.29 and 3.20 (s, 1 H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 189.2, 188.7, 163.2, 162.6, 142.6, 135.6, 135.5, 120.5, 120.2, 111.7, 111.1, 56.3, 56.1, 37.5, 33.7; HRMS (ESI) m/z 194.0822 (M+H⁺), calc. for C₁₀H₁₂NO₃ 194.0817.

Pale yellow solid, Mp 72.1–73.8 °C; 75% yield (87% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 10.04 and 9.91 (s, 1H , CHO rotameric), 8.39 and 8.22 (s, 1H, CHO rotameric), 7.67–7.64(m, 1H), 7.43–7.35 (m, 1H),

7.33-7.28 (m, 1H), 3.41 and 3.33 (s, 3H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 187.8, 187.7, 163.6, 163.1, 162.6, 160.3, 139.9, 133.8, 133.7, 130.2, 130.1, 129.5, 129.4, 122.7, 122.4, 122.1, 116.6, 116.3, 38.0, 35.0; HRMS (ESI) m/z 204.0438 (M+Na⁺), calc. for C₉H₈FNO₂Na 204.0437.

Yellow oil, 86 % yield (92% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 10.07 and 9.94 (s, 1H, CHO rotameric), 8.42 and 8.26 (s, 1H, CHO rotameric), 7.97(dd, J = 2.5, 1H), 7.69 (dd, J = 8.5, 2.5, 1H), 7.35–7.27 (m, 1H), 3.46 and 3.38 (s, 3H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ

187.7, 187.6, 162.9, 162.43, 141.9, 139.6, 135.2, 134.9, 134.7, 134.6, 132.9, 132.8, 130.4, 130.0, 129.1, 128.6, 37.7, 34.7; HRMS (ESI) m/z 198.0321 (M+H⁺), calc. for $C_9H_8CINO_2$ 198.0322.

Yellow oil, 72% yield (86% yield, 10 mol% TEMPO used); 1H NMR (300 H_{N} WHz, CDCl₃) δ 10.03 and 9.91 (s, 1H, CHO rotameric), 8.36 and 8.24 (s, 1H, 3g OH CHO rotameric), 7.83 (dd, J = 8.3, 5.4, 1H), 7.68–7.62 (m, 1H), 7.49–7.44 (m, 1.7, 1H), 3.41 and 3.34 (s, 3H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 188.1, 187.9, 162.9, 162.4, 144.4, 132.0, 131.8, 131.6, 130.7, 130.6, 130.4, 129.9, 129.5, 37.7, 34.5; HRMS (ESI) m/z 263.9635 (M+Na⁺), calc. for C₉H₈BrNO₂Na 263.9636.

Yellow oil, 66% yield (84% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 10.06 and 9.96 (s, 1H, CHO rotameric), 8.37 and 8.21 (s, 1H, Bh OH CHO rotameric), 7.76(d, J = 1.5, 1H), 7.49 (dd, J = 8.0, 1.5, 1H), 7.18 (d, J

= 8.0, 1H), 3.38 and 3.31 (s, 3H, CH₃ rotameric), 2.45 and 2.43 (s, 1H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 189.5, 189.2, 163.0, 162.8, 141.2, 138.8, 136.0, 135.9, 131.6, 131.0, 127.8, 127.5, 38.0, 34.7, 20.9; HRMS (ESI) m/z 178.0866 (M+H⁺), calc. for C₁₀H₁₂NO₂ 178.0868.

Yellow oil, 78% yield (88% yield, 10 mol% TEMPO used); ¹H NMR (300 H MHz, CDCl₃) δ 10.01 and 9.88 (s, 1H, CHO rotameric), 8.37 and 8.22 (s, 1H, **3**i OH CHO rotameric), 8.07 (dd, J = 6.6, 2.4, 1H), 7.80 (dd, J = 8.4, 2.4, 1H), 7.19 (d, J = 8.4, 1H), 3.41 and 3.33 (s, 3H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 187.6, 187.5, 163.0, 162.4, 142.4, 140.1, 138.2, 137.9, 133.5, 133.1, 132.9, 129.3, 128.8, 122.4, 37.7, 34.6; HRMS (ESI) m/z 263.9634(M+Na⁺), calc. for C₉H₈BrNO₂Na 263.9636.

CI

3f

Brown oil, 53% yield (85% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 9.94 and 9.87 (s, 1H , CHO rotameric), 8.36 and 8.25 (s, 1H, CHO rotameric), 7.99–7.87 (m, 1H), 7.43–7.38 (m, 5H), 7.07 (dt, *J* = 5.8, 3.0, 1H), 6.83 (dd, *J* = 5.8, 2.4, 1H), 5.17 and 5.14 (s, 2H, CH₂ rotameric), 3.36 and 3.32 (s, 3H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 188.0, 187.7, 164.0, 162.6, 145.5, 135.2, 133.3, 128.8, 128.8 128.7, 128.6, 127.6, 127.5, 125.3, 114.5, 113.9, 70.7, 37.9, 34.5; HRMS (ESI) m/z 270.1133 (M+H⁺), calc. for C₁₆H₁₆NO₃ 270.1130.

Pale yellow solid, Mp 133.4–134.5 °C; 62% yield (86% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 10.03 and 9.84 (s, 1H, CHO rotameric), 8.70 (dd, J = 4.7, 1.9, 1H), 8.50 and 8.44 (s, 1H,CHO rotameric), 8.29 (dd, J = 7.8, 1.9, 1H), 7.40 (dd, J = 7.8, 4.7, 1H), 3.60 and 3.50 (s, 3H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 187.9, 187.4, 163.5, 163.4, 155.3, 153.7, 152.7, 139.5, 137.0, 124.1, 122.6, 122.0, 34.9, 32.0; HRMS (ESI) m/z 165.0668 (M+H⁺), calc. for C₈H₉N₂O₂ 165.0664.

Red solid, Mp 199.2–200.1 °C; 45% yield; ¹H NMR (300 MHz, DMSO) δ 11.01 (s, 1H), 7.54 (t, J = 7.6, 1H), 7.44 (d, J = 7.2, 1H), 7.02 (t, J = 7.2, 1H), 6.87 (d, J = 7.6, 1H); ¹³C NMR (75 MHz, DMSO) δ 184.4, 159.4, 150.8, 138.4, 124.7, 122.8, 117.8, 112.3; HRMS (ESI) m/z 148.0395 (M+H⁺), calc. for C₈H₆NO₂ 148.0399.

Yellow solid, Mp 258.7–260.1 °C; 56% yield; ¹H NMR (300 MHz, CDCl₃) δ 8.63 (d, J = 8.0, 1H), 8.44 (d, J = 7.8, 1H), 8.03 (d, J = 8.0, 1H), 7.93–7.76 (m, 3H), 7.68 (t, J = 7.6, 1H), 7.43 (t, J = 7.6, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 182.6, 158.1, 146.6, 146.3, 144.3, 138.3, 135.1, 130.7, 130.3, 127.5, 127.2, 125.4, 123.7, 121.9, 118.0; HRMS (ESI) m/z 249.0663 (M+H⁺), calc. For C₁₅H₉N₂O₂ 249.0664.

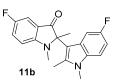
Brown oil, 55% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.15 (t, J = 7.5, 1H), 6.74 (t, J = 7.5, 1H), 6.52 (d, J = 7.5, 1H), 3.79 (s, 1H), 2.20 (s, 1H), 1.69 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 203.5, 171.7, 143.8, 129.4, 118.3, 114.1, 69.5,

53.5, 24.7, 18.6; HRMS (ESI) m/z 222.1135 (M+H⁺), calc. for $C_{12}H_{16}NO_3$ 222.1130.

Colorless solid, Mp 114.8–116.3 °C; 74% yield; ¹H NMR (300 MHz, CDCl₃) δ 9 7.49 (d, J = 7.8, 1H), 7.32 (t, J = 7.8, 1H), 7.11 (t, J = 7.2, 1H), 2.19 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 168.4, 137.8, 129.0, 124.3, 120.0, 24.6; HRMS (ESI)

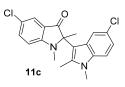
m/z 136.0766 (M+H⁺), calc. for C₈H₁₀NO 136.0762.

Yellow solid, Mp 153.4–155.2 °C; 82% yield; ¹H NMR (300 MHz, DMSO) δ 7.60 (t, J = 7.7, 1H), 7.53 (d, J = 7.5, 1H), 7.40 (d, J = 8.2, 1H), 7.21 (d, J = 7.8, 1H), 7.09–6.99 (m, 2H), 6.87 (t, J = 7.3, 1H), 6.76 (t, J = 7.3, 1H), 3.66 (s, 3H), 2.82 (s, 3H), 2.30 (s, 3H), 1.77 (s, 3H); ¹³C NMR (75 MHz, DMSO) δ 202.7, 158.7, 138.0, 136.2, 135.9, 126.5, 124.5, 120.3, 119.1, 118.8, 117.5, 116.6, 109.5, 108.8, 106.0, 70.4, 29.3, 27.6, 21.6, 11.4; HRMS (ESI) m/z 305.1650 (M+H⁺), calc. for C₂₀H₂₅N₂O 305.1654.



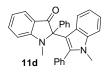
Yellow solid, Mp 121.2–122.6 °C; 82% yield; ¹H NMR (300 MHz, DMSO) δ 7.52 (t, J = 9.2, 1H), 7.42 (dd, J = 9.2, 3.8, 1H), 7.31 (d, J = 7.5, 1H), 7.06 (dd, J = 8.9, 3.8, 1H), 6.90 (d, J = 8.9, 1H), 3.65 (s, 3H),

2.81 (s, 3H), 2.25 (s, 3H), 1.74 (s, 3H); ¹³C NMR (75 MHz, DMSO) δ 202.3, 158.4, 156.2, 155.9, 155.4, 153.1, 138.0, 133.0, 126.5, 126.4, 126.2, 125.9, 117.2, 117.1, 110.6, 110.5, 10.4, 110.3, 109.3, 109.0, 108.3, 107.9, 106.1, 106.0, 71.1, 29.6, 27.8, 21.4, 11.6; HRMS (ESI) m/z 341.1466 (M+H⁺), calc. for C₂₀H₁₉F₂N₂O 341.1465.



Yellow soild, Mp 134.7–136.2 °C; 88% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.62 (s, 1H), 7.47 (d, J = 8.8, 2H), 7.15 (d, J = 8.8, 1H), 7.08 (d, J = 8.7, 1H), 6.74 (d, J = 8.7, 1H), 3.60 (s, 3H), 2.85 (s, 3H), 2.19 (s,

3H), 1.88 (s, 3H);¹³C NMR (75 MHz, CDCl₃) δ 202.2, 157.2, 137.7, 137.1, 135.2, 128.1, 125.4, 124.7, 122.1, 120.9, 119.1, 118.8, 110.0, 109.4, 106.2, 29.7, 27.9, 21.9, 11.9; HRMS (ESI) m/z 373.0877 (M+H⁺), calc. for C₂₀H₁₉Cl₂N₂O 373.0874



Yellow solid, Mp 230.8–232.5 °C; 56% yield; ¹H NMR (300 MHz, DMSO) δ 7.49 (d, J = 7.9, 2H), 7.27–7.21 (m, 7H), 7.14 (d, J = 7.8, 2H), 6.98–6.79 (m, 5H), 6.56 (t, J = 7.4, 1H), 6.40 (d, J = 8.1, 1H), 3.43

(s, 3H), 2.80 (s, 3H); ¹³C NMR (75 MHz, DMSO) δ 199.1, 159.0, 139.9, 138.2, 137.6, 136.4,

130.9, 130.8, 130.6, 128.2, 127.7, 127.4, 126.9, 124.1, 121.2, 120.6, 119.2, 118.5, 116.4, 110.3, 108.9, 108.5, 75.5, 30.5, 29.4; HRMS (ESI) m/z 429.1970 (M+H⁺), calc. for $C_{30}H_{25}N_2O$ 429.1967.

Yellow solid, Mp 156.3–157.7 °C; 91% yield; ¹H NMR (300 MHz, DMSO) δ 7.60 (d, J = 7.6, 1H), 7.51 (t, J = 7.6, 1H), 7.39 (d, J = 8.2, 1H), 7.25–7.17 (m, 7H), 7.07–6.90 (m, 6H), 6.80 (t, J = 7.4, 1H), 6.74 (d, J =

8.2, 1H), 5.43 (s, 2H), 4.58 (d, J = 16.5, 1H), 4.33 (d, J = 16.5, 1H), 2.18 (s, 3H), 1.76 (s, 3H); ¹³C NMR (75 MHz, DMSO) δ 202.6, 158.5, 138.4, 138.1, 137.9, 136.2, 128.6, 128.3, 27.1, 126.9, 125.9, 124.6, 120.7, 119.6, 118.9, 117.9, 117.3, 110.0, 109.5, 107.2, 71.1, 46.5, 45.4, 22.6, 11.8; HRMS (ESI) m/z 457.2278 (M+H⁺), calc. for C₃₂H₂₉N₂O 457.2280.

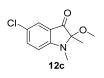


Yellow oil, 66% yield; ¹H NMR (300 MHz, MeOD) δ 7.56 (t, J = 7.7, 1H), 7.48 (d, J = 7.7, 1H), 6.88 (d, J = 7.9, 1H), 6.75 (t, J = 7.9, 1H), 2.98 (s, 3H), 2.93 (s, 3H), 1.31 (s, 3H); ¹³C NMR (75 MHz, MeOD) δ 202.8, 162.1, 140.4,

125.2, 119.4, 118.8, 109.8, 93.7, 51.6, 26.6, 19.9; HRMS (ESI) m/z 192.1026 (M+H⁺), calc. for $C_{11}H_{14}NO_2$ 192.1025.

Yellow oil, 67% yield; ¹H NMR (300 MHz, DMSO) δ 7.50 (t, J = 9.1, 1H), 7.26 (d, J = 7.4, 1H), 7.00 (d, J = 5.6, 1H), 2.88 (d, J = 1.8, 6H), 1.24 (s, 3H); ^{12b} ¹³C NMR (75 MHz, DMSO) δ 199.2, 156.6, 156.0, 152.8, 126.0, 125.7,

117.2, 117.1, 110.0, 109.9, 108.6, 108.3, 92.0, 50.3, 26.0, 18.8; HRMS (ESI) m/z 210.0928 (M+H⁺), calc. For C₁₁H₁₃FNO₂ 210.0930.



Yellow oil, 76% yield; ¹H NMR (300 MHz, DMSO) δ 7.59 (d, J = 8.8, 1H), 7.45 (s, 1H), 7.01 (d, J = 8.8, 1H), 2.88 (s, 6H), 1.25 (s, 3H); ¹³C NMR (75 MHz, DMSO) δ 198.5, 158.1, 137.7, 122.5, 120.8, 118.2, 110.3, 91.8, 50.3,

25.8, 18.7; HRMS (ESI) m/z 226.0639 (M+H⁺), calc. for C₁₁H₁₃ClNO₂ 226.0635.



Yellow solid, Mp 73.4–75.1 °C; 85% yield; ¹H NMR (300 MHz, MeOD) δ 7.62 (t, J = 7.7, 1H), 7.47 (d, J = 7.7, 1H), 7.32 (s, 5H), 6.96 (d, J = 7.8, 1H), 6.78 (t, J = 7.8, 1H), 3.21 (s, 3H), 2.81 (s, 3H); ¹³C NMR (75 MHz, MeOD) δ

201.4, 163.5, 140.7, 136.8, 130.0, 129.9, 127.4, 126.0, 119.5, 119.2, 109.2, 96.9, 51.9, 27.6; HRMS (ESI) m/z 254.1184 (M+H⁺), calc. for C₁₆H₁₆NO₂ 254.1181.

Yellow solid, Mp 79.2–80.8 °C; 64% yield. ¹H NMR (300 MHz, MeOD) δ 7.53 (d, J = 7.8, 1H), 7.47 (t, J = 7.8, 1H), 7.35–7.27 (m, 5H), 6.77 (t, J = 7.8, 1H), 6.70 (d, J = 7.8, 1H), 4.59 (q, J = 16.7, 2H), 2.99 (s, 3H), 1.34 (s, 3H); ¹³C NMR (75 MHz, MeOD) δ 202.5, 162.2, 140.3, 139.6, 129.9, 128.4, 128.2, 125.5, 119.8, 119.4, 110.7, 94.2, 52.1, 46.0, 21.4; HRMS (ESI) m/z 268.1340 (M+H⁺), calc. for C₁₇H₁₈NO₂ 268.1338.

Pale yellow solid, Mp 52.6–54.3 °C; 50% yield (71% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 10.15 and 9.99 (s, 1H, CHO rotameric), 7.99 (d, J = 7.7, 1H), 7.71 (t, J = 7.7, 1H), 7.55 (t, J = 7.7, 1H), 7.30 (d, J = 7.7, 1H), 3.40 and 3.30 (s, 3H, CH₃ rotameric), 2.32 and 1.81 (s, 3H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 188.4, 169.4, 145.2, 134.8, 131.4, 129.2, 128.1, 128.0, 36.9, 21.5;

HRMS (ESI) m/z 178.0873 (M+ H^+), calc. for C₁₀H₁₂NO₂ 178.0868.

Pale yellow oil, 66% yield (74% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 10.08 and 9.91 (s, 1H, CHO rotameric), 7.66–7.57 (m, 1H), **13b** of 7.40 (t, *J* = 8.1, 1H), 7.39–7.28 (m, 1H), 3.41 and 3.29 (s, 1H, CH₃ rotameric), 2.31 and 1.81 (s, 1H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 188.0, 170.4, 163.7, 160.4, 142.5, 142.4, 134.2, 134.1, 131.2, 131.1, 123.0, 122.7, 116.2, 115.9, 38.1, 22.5; HRMS (ESI) m/z 196.0773 (M+H⁺), calc. for C₁₀H₁₁FNO₂ 196.0774.

Pale yellow oil, 73% yield (80% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 10.02 and 9.84 (s, 1H, CHO rotameric), 7.88 and 7.80 (s, 1H, CH rotameric), 7.62–7.52 (m, 1H), 7.21–7.12 (m, 1H), 3.34 and 3.22 (s, 3H, CH₃ rotameric), 2.25 and 1.75 (s, 3H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 188.3, 188.0, 170.1, 144.6, 135.7, 135.3, 134.9, 133.6, 130.6, 130.1, 129.9, 129.2, 39.7, 38.0, 22.4, 22.0; HRMS (ESI) m/z 212.0477 (M+H⁺), calc. for C₁₀H₁₁CINO₂ 212.0478.

Pale yellow Solid, Mp 106.5–107.5 °C; 60% yield (80% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 10.05 (s, 1H), 7.69 (d, J = 6.1, 1H), 7.47 (t, 1H), 7.29 (t, 1H), 7.29–7.05 (m, 6H), 3.45 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 189.2, 170.9, 146.4, 135.0, 131.8, 130.6, 129.9, 129.3, 128.4, 127.9, 127.9, 39.2; HRMS (ESI) m/z 240.1023 (M+H⁺), calc. for C₁₅H₁₄NO₂ 240.1025.

Pale yellow oil, 52% yield (67% yield, 10 mol% TEMPO used); ¹H NMR (300 MHz, CDCl₃) δ 9.63 (s, 1H), 7.84 (d, *J* = 7.8, 1H), 7.55 (t, *J* = 7.4, 1H), 7.43 (t, *J* = 7.4, 1H), 7.18 (s, 3H), 7.08 (s, 2H), 7.00 (d, *J* = 7.8, 1H), 4.85 (s, 2H), 1.74 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 189.1, 169.9, 144.3, 135.9, 135.4, 133.0, 123.0, 129.6, 129.5, 128.9, 128.6, 128.0, 53.5, 22.7; HRMS (ESI) m/z 254.1186 (M+H⁺), calc. for C₁₆H₁₆NO₂ 254.1181.

Yellow solid, Mp 194.8–196.5 °C; 87% yield; ¹H NMR (300 MHz, DMSO) δ 10.90 (s, 1H), 7.74 (s, 1H), 7.52–7.44 (m, 2H), 7.25–7.21(m, 2H), 6.96–6.86 (m, 2H), 6.81–6.69 (m, 2H), 2.40 (s, 3H), 1.74 (s, 3H); ¹³C NMR (75 MHz, DMSO) δ 203.9, 159.9, 137.5, 134.7, 133.0, 127.2, 124.4, 120.0, 119.4, 118.4, 117.7, 117.0, 111.8, 110.5, 108.47, 66.2, 24.4, 14.0; HRMS (ESI) m/z 227.1345 (M+H⁺), calc. for C₉H₁₀NO₂ 227.1341.

O H Ph Ph Ph 15b Yellow solid, Mp 227.2–228.8 °C; 95% yield; ¹H NMR (300 MHz, DMSO) δ 11.36 (s, 1H), 8.36 (s, 1H), 7.52 (t, *J* = 7.6, 1H), 7.38–7.34 (m, 3H), 7.26 (d, *J* = 7.6, 1H), 7.15 (t, *J* = 8.3, 3H), 7.06–6.98 (m, 7H), 6.79–6.70 (m, 2H), 6.62 (d, *J* = 8.3, 1H); ¹³C NMR (75 MHz, DMSO) δ 200.5, 160.1, 139.8,

138.0, 137.5, 135.8, 133.2, 129.5, 127.6, 127.4, 127.3, 27.1, 127.0, 124.4, 121.2, 120.3, 118.7, 118.5, 117.5, 111.9, 111.3, 111.0, 71.2; HRMS (ESI) m/z 401.1652 (M+H⁺), calc. for $C_{28}H_{21}N_2O$ 401.1654.

Pale yellow solid, Mp 71.3–72.5 °C; 58% yield; ¹H NMR (300 MHz, CDCl₃) δ 8.19 (d, J = 7.8, 1H), 7.79 (t, J = 7.8, 1H), 7.55–7.50 (m, 2H), 2.47 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 160.2, 159.7, 146.4, 136.5, 128.4, 128.2, 126.4, 116.6, 21.4; HRMS (ESI) m/z 162.0559 (M+H⁺), calc. for C₉H₈NO₂ 162.0555. Red solid, Mp 108.9–110.1 °C; 50% yield; ¹H NMR (300 MHz, CDCl₃) δ 8.32 (d, J = 7.7, 2H), 8.25 (d, J = 8.0, 1H), 7.83 (t, J = 7.7, 1H), 7.70 (d, J = 8.0, 1H), 7.58–7.50 (m, 4H); ¹³C NMR (75 MHz, CDCl₃) δ 159.6, 157.1, 147.0, 136.6, 132.6, 130.2, 128.7, 128.6, 128.3, 128.3, 127.2, 117.0; HRMS (ESI) m/z 224.0714 (M+H⁺), calc. for C₁₄H₁₀NO₂ 224.0712.



Yellow oil, 81% yield; ¹H NMR (300 MHz, CDCl₃) δ 9.82 (s, 1H), 8.26 (s, 1H), 7.44 (dd, J = 17.8, 8.1, 2H), 6.69 (t, J = 8.1, 2H), 2.94 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 193.9, 151.6, 136.6 135.9, 118.4, 114.7, 110.4, 29.1; HRMS

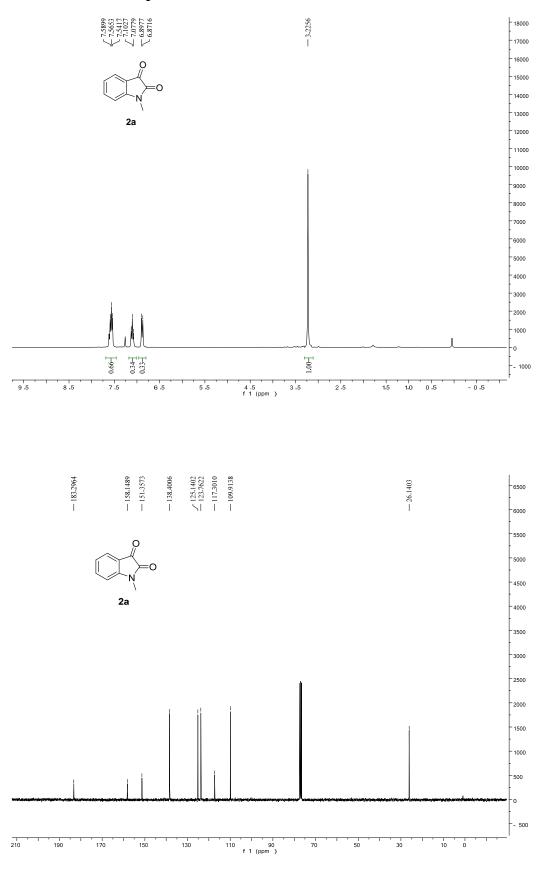
(ESI) m/z 136.0765 (M+H⁺), calc. for $C_8H_{10}NO$ 136.0762.

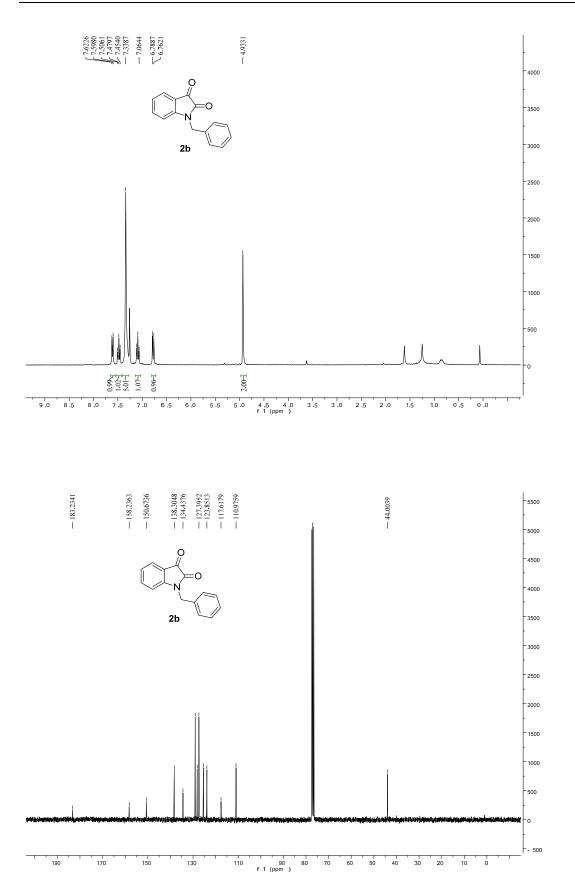
Red solid, Mp 105.8–106.1 °C; 71% yield; ¹H NMR (300 MHz, CDCl₃) δ 8.38 (s, 1H), 7.94 (d, J = 7.6, 1H), 7.86 (d, J = 7.6, 1H), 7.64 (q, J = 8.1, 1H), ¹⁸ 7.46 (t, J = 7.3, 1H), 7.34 (t, J = 7.3, 1H), 7.17–7.12 (m, 1H), 4.23 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 156.11, 155.3, 136.8, 130.3, 129.9, 129.2, 128.0, 127.9, 123.9, 121.8, 124.0, 120.7, 119.8, 117.6, 114.1, 33.0; HRMS (ESI) m/z 233.1075 (M+H⁺), calc. for C₁₂H₁₆NO₃ 233.1079.

Yellow solid, Mp 135.7–136.1 °C; 26% yield (Conditions A), 84% (Conditions B); ¹H NMR (300 MHz, CDCl₃) δ 8.19 and 8.17 (d, J = 7.4 Hz, 1H, CHO rotameric), 7.74 (dd, J = 17.1, 7.6 Hz, 1H), 7.67–7.56 (m, 1H), 7.55–7.41 (m, 1H), 7.31 (d, J = 8.3 Hz, 1H), 3.30 and 3.42 (d, J = 35.9 Hz, 3H, CH₃ rotameric), 2.60 and 2.59 (d, J = 13.5 Hz, 3H, CH₃ rotameric); ¹³C NMR (75 MHz, CDCl₃) δ 200.1, 199.6, 162.6, 162.4, 139.9, 137.4, 137.2, 136.6, 132.7, 132.3, 129.8, 128.4, 128.0, 127.7, 127.4, 37.8, 33.8, 29.5, 28.6; HRMS (ESI) m/z 178.0870 (M+H⁺), calc. for C₁₀H₁₂NO₂ 178.0868.

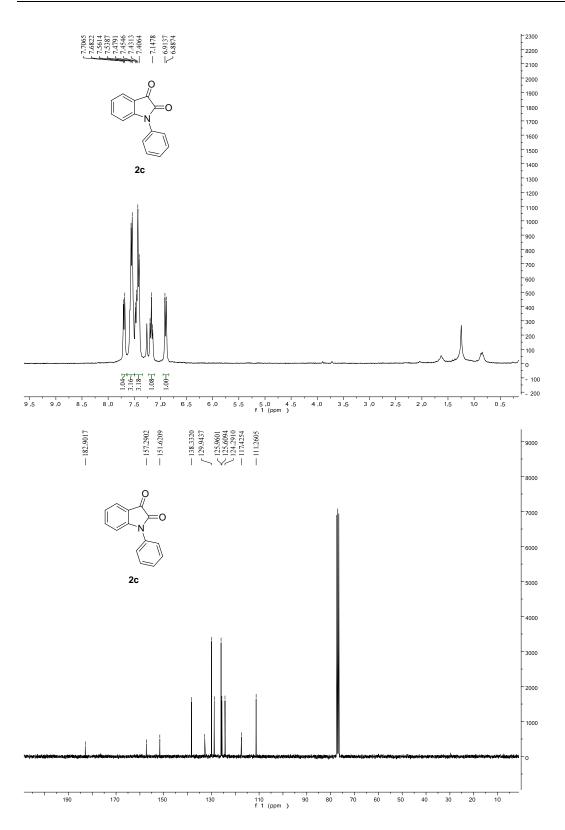
 $\begin{array}{c} \begin{array}{c} \circ & \text{Ph} \\ \mathsf{N} & \text{Yellow solid, Mp 187.3-190.1 °C; 44\% yield; ^{1}H NMR (300 MHz, CDCl_3) \delta} \\ \hline & \mathsf{N} & \mathsf{Ph} \\ \mathbf{22} \end{array} & 7.65-7.04 (m, 14H), 4.08 (d, J = 16.0 Hz, 1H), 3.72 (d, J = 16.1 Hz, 1H), 3.34 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) \delta 198.4, 170.0, 143.6, 135.6, 133.6, 132.6, 130.2, 129.6, 129.6, 128.7, 128.6, 127.6, 127.3, 127.1, 126.6, 47.4, 38.4. HRMS (ESI) m/z 314.1549 (M+H^+), calc. for C_{22}H_{20}NO 314.1545. \end{array}$

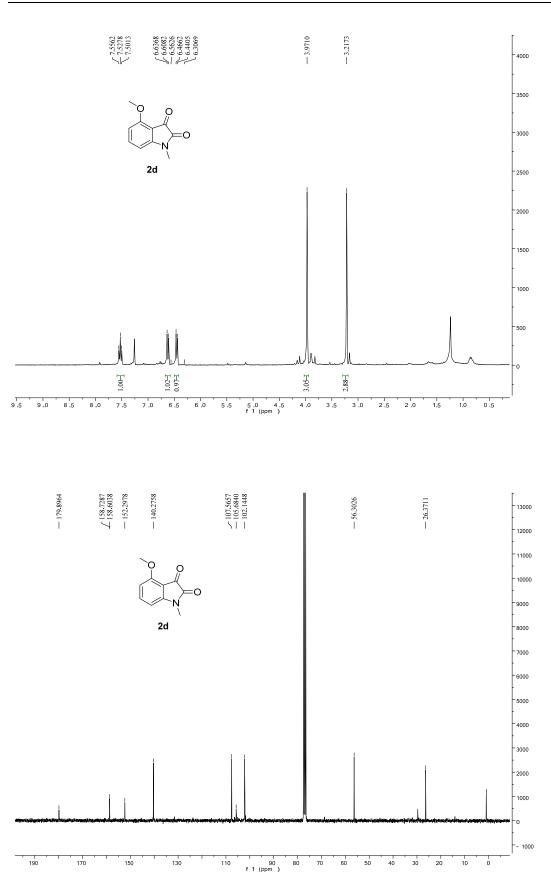
Ph Vellow solid, Mp 119.8–120.5 °C; 38% yield; ¹H NMR (300 MHz, CDCl₃) δ 8.04 (d, J = 7.7 Hz, 1H), 7.54 (d, J = 7.6 Hz, 2H), 7.50–7.21 (m, 9H), 7.14 (t, J = 7.3 Hz, 2H), 3.71 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 192.9, 146.4, 140.1, 137.2, 130.9, 130.9, 130.7, 129.2, 128.7, 128.0, 127.6, 127.5, 114.7, 109.8, 31.2. HRMS (ESI) m/z 314.1387 (M+H⁺), calc. for C₂₂H₁₈NO 314.1388.

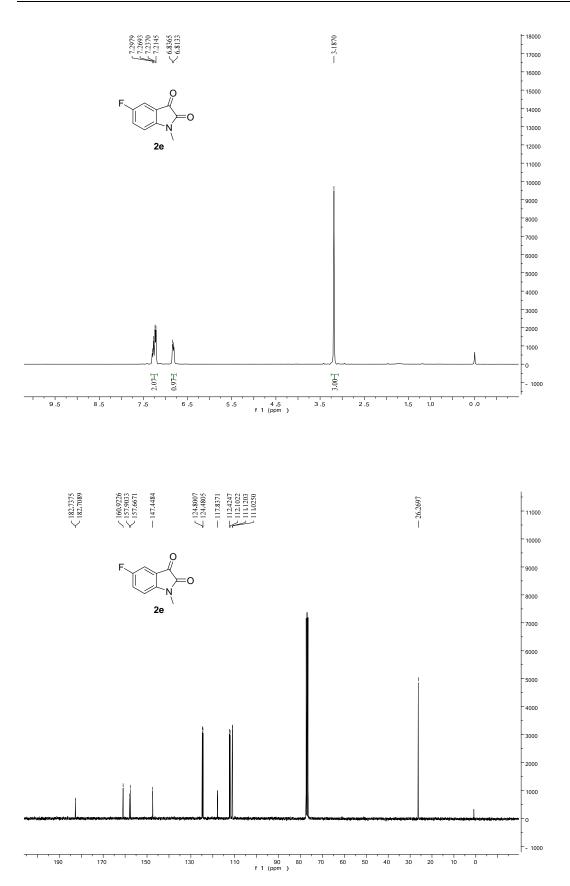


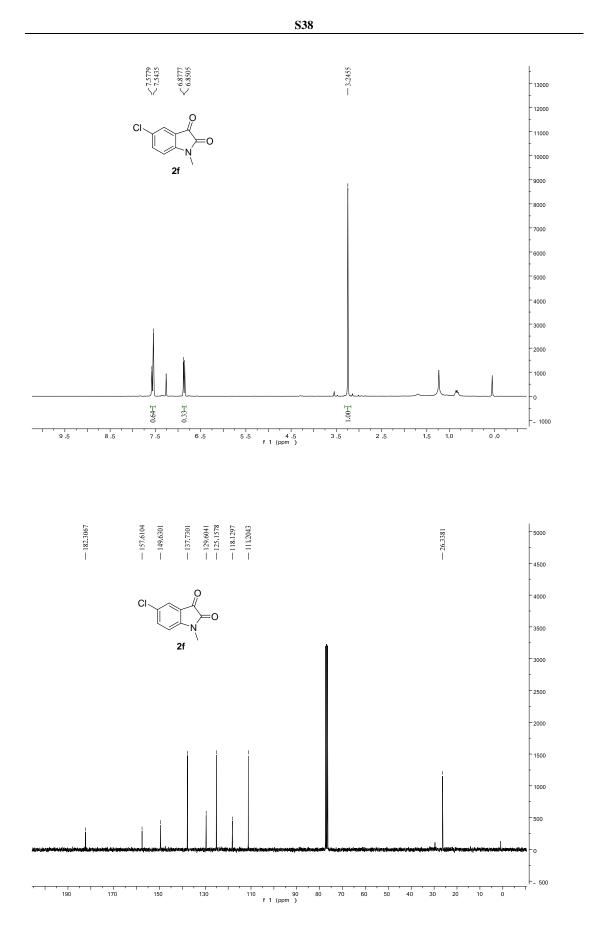


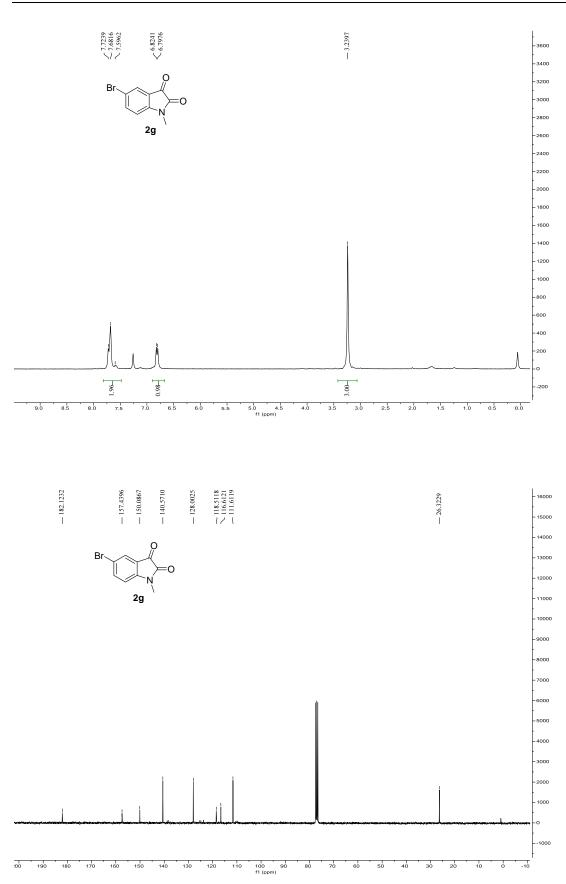
S34

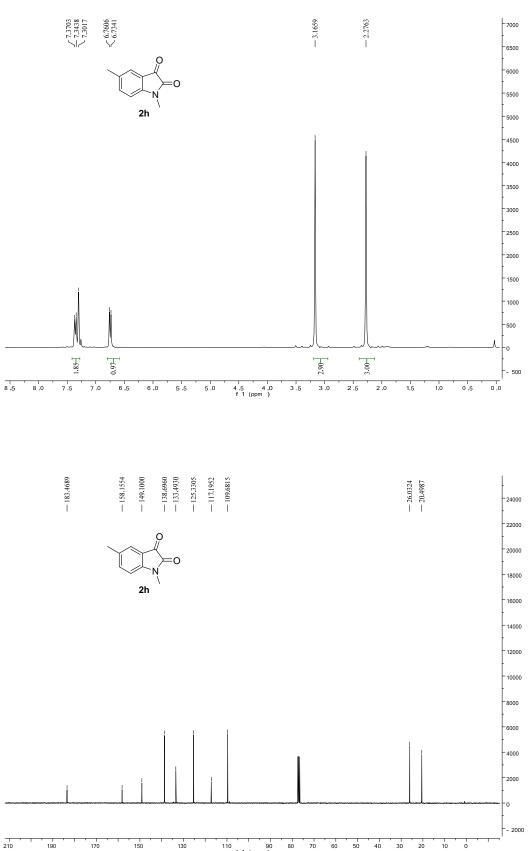






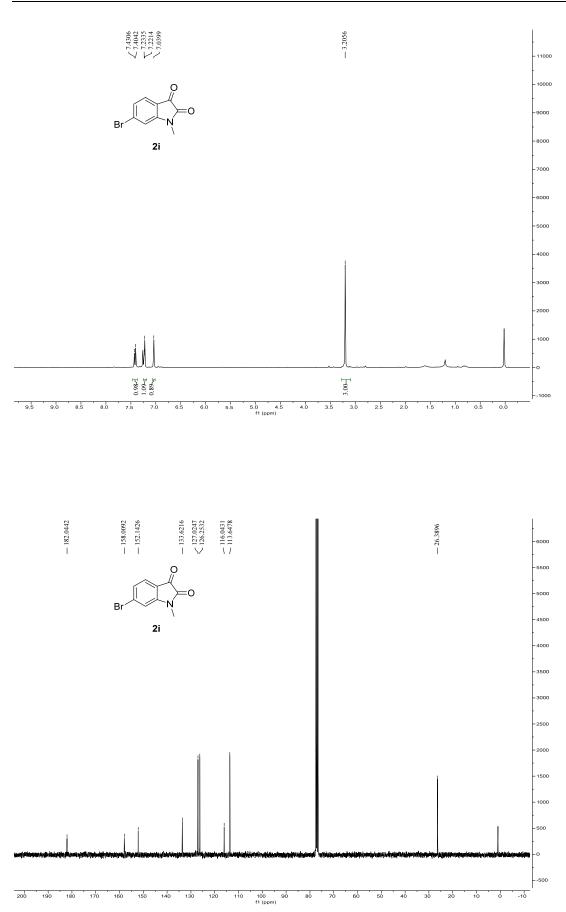


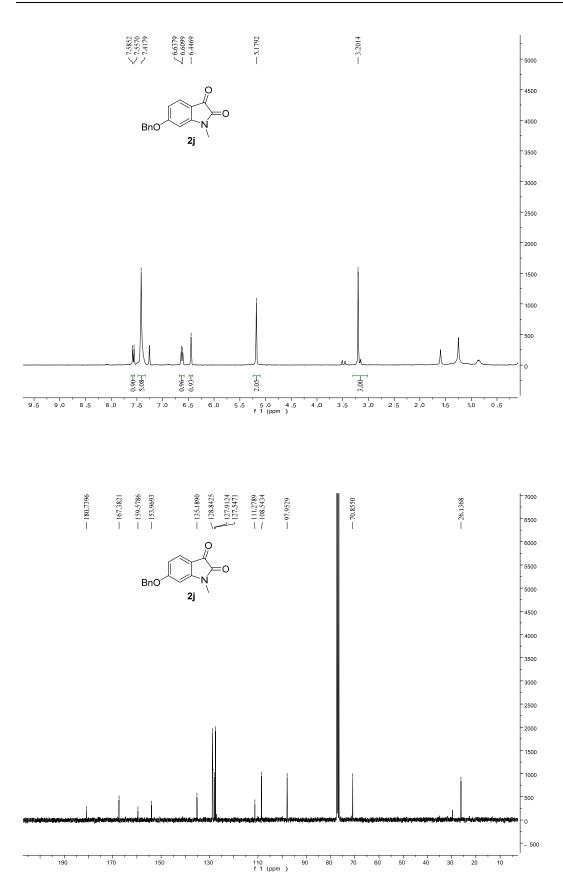


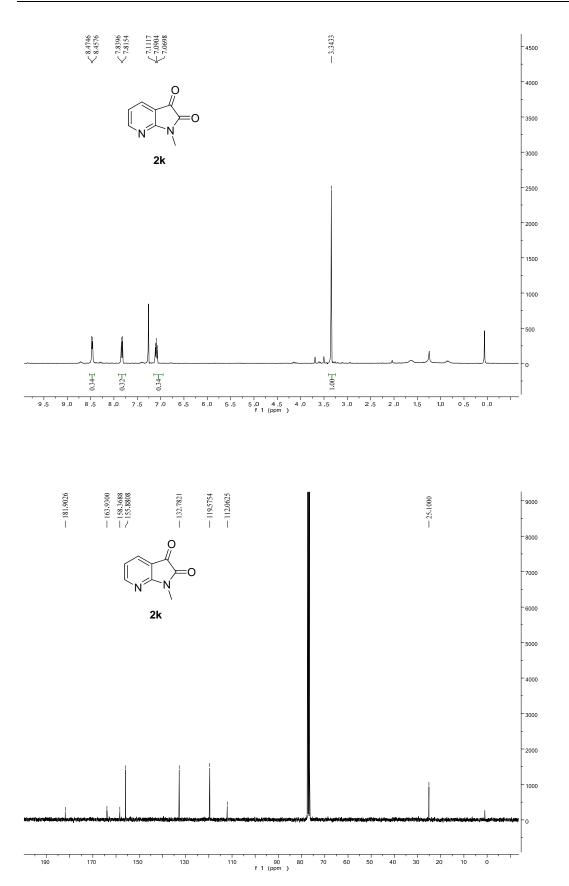


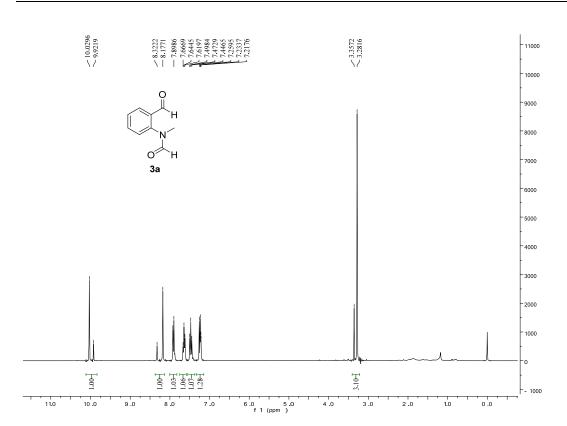
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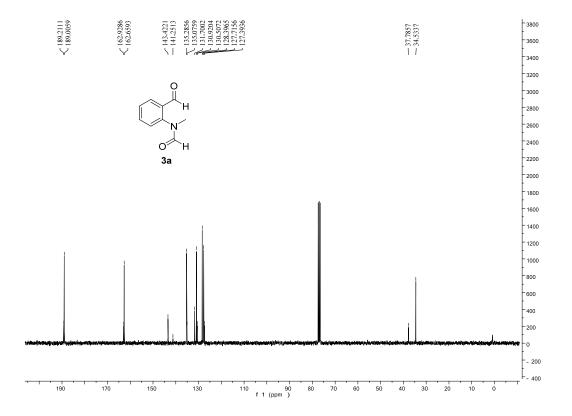
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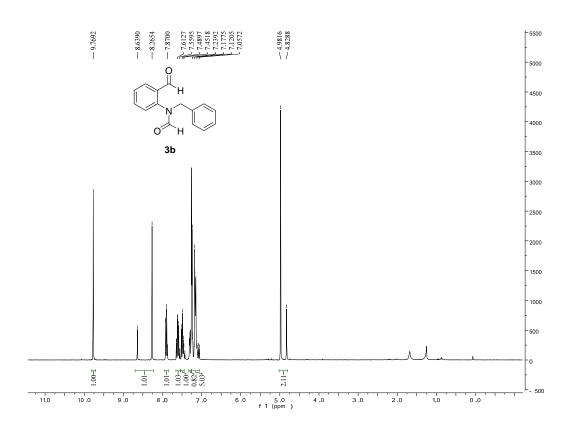


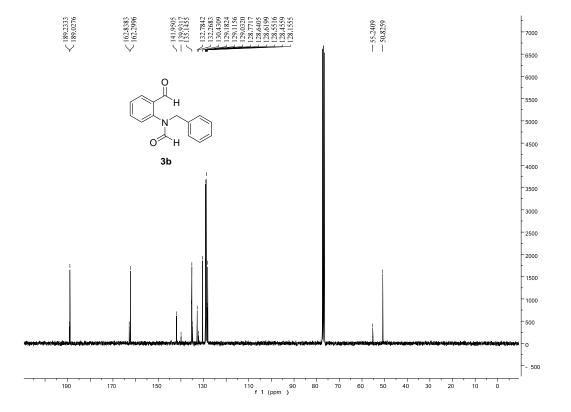


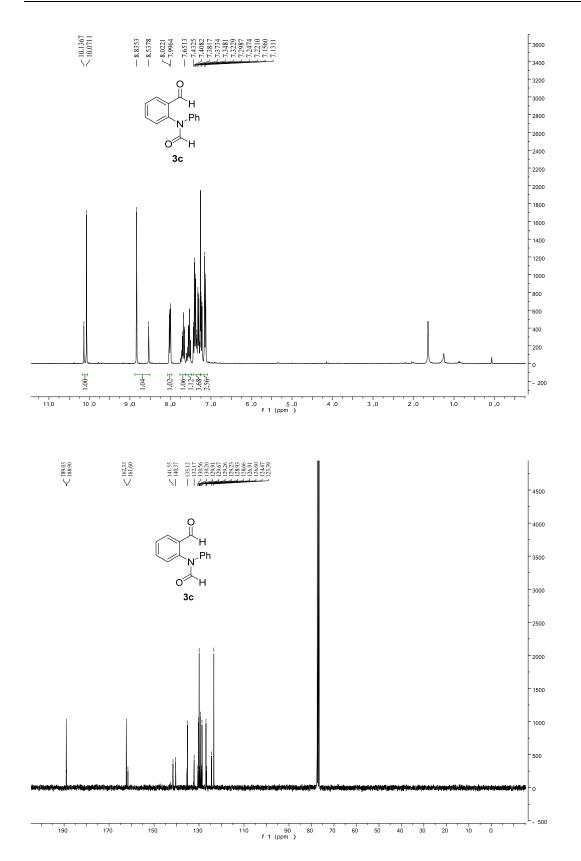


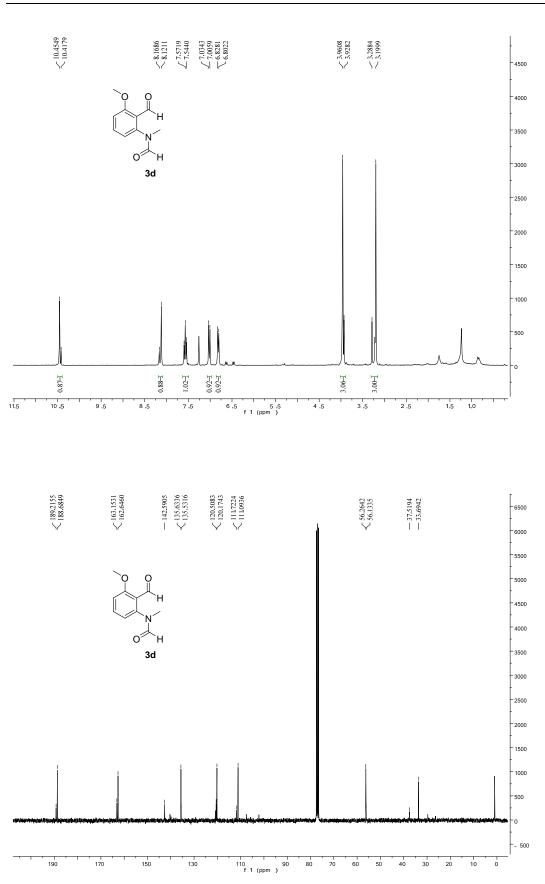


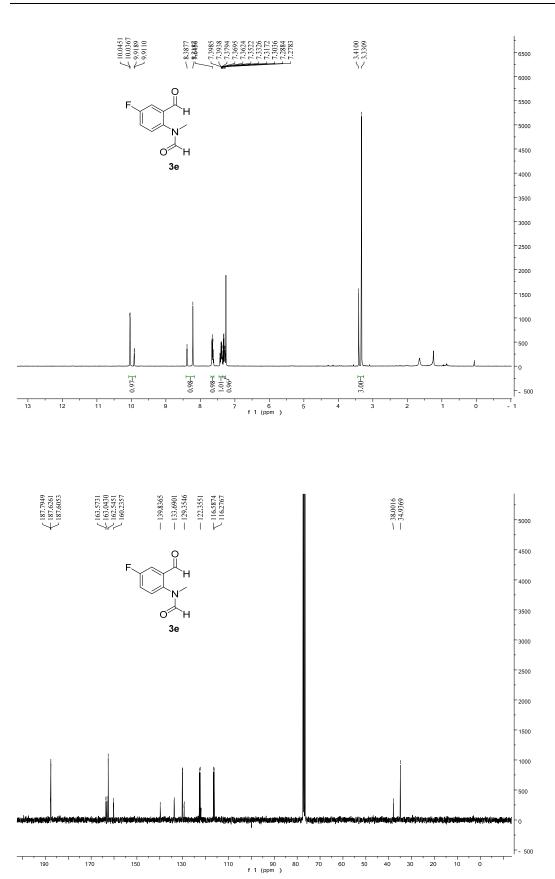


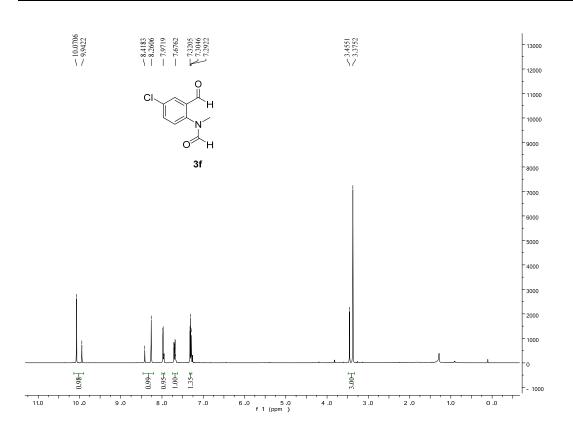


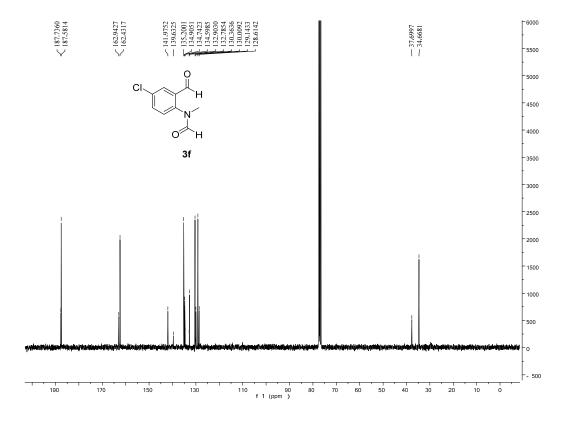


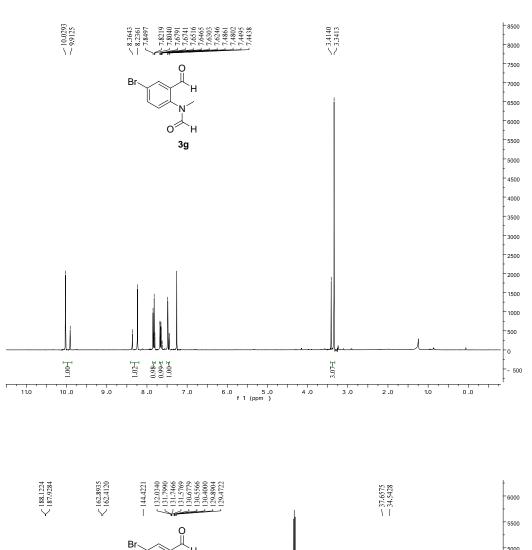


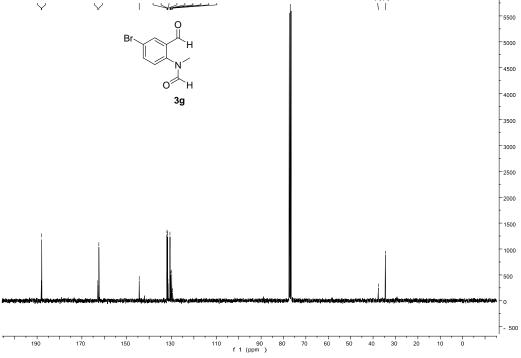


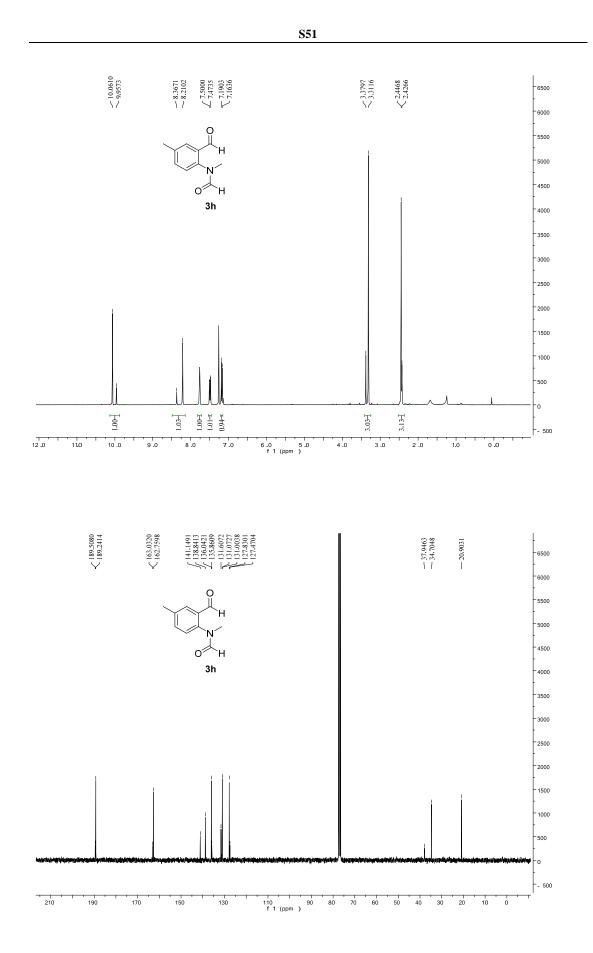


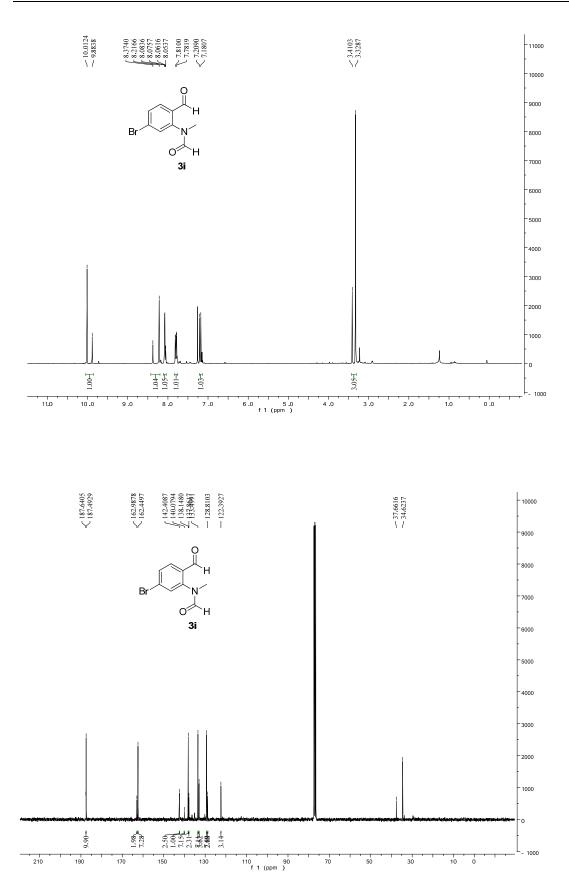


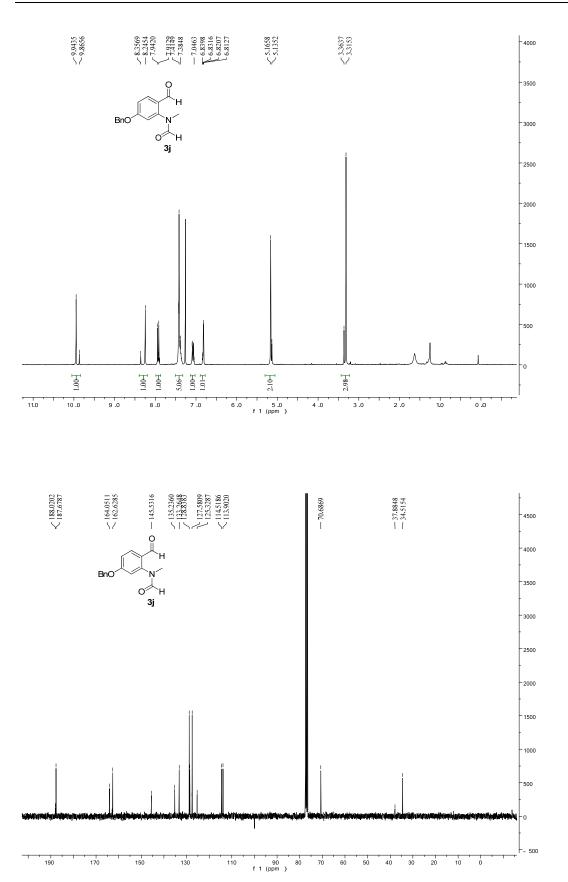


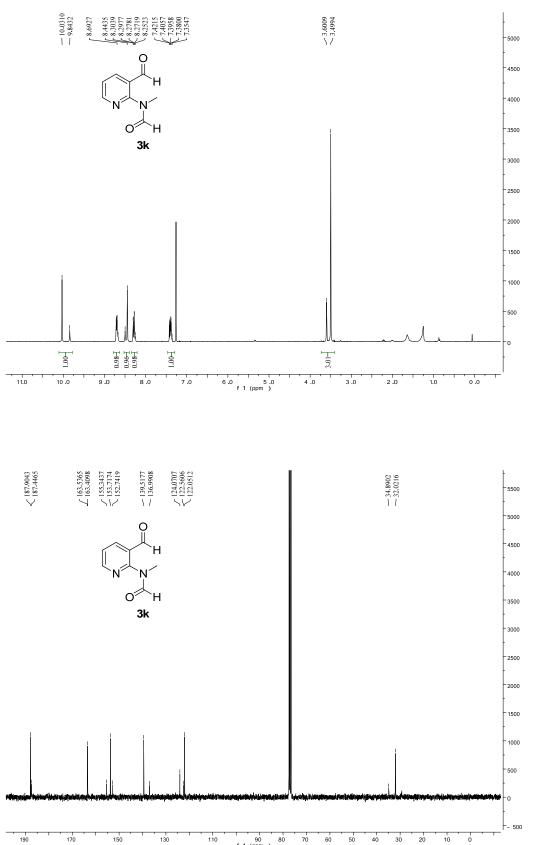












90 80 70 60 50 f 1 (ppm)

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