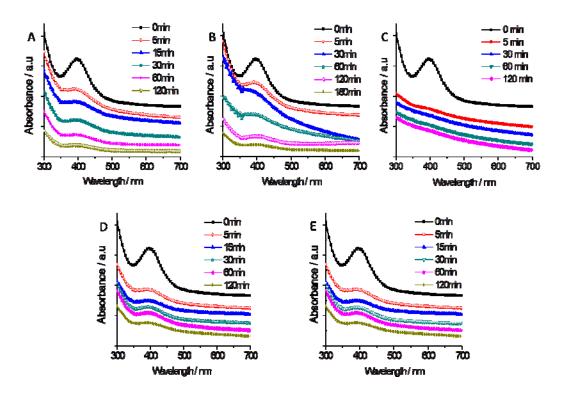
## **Supporting Information (SI)**

# Selective partial hydrogenation of methyl linoleate using highly active palladium nanoparticles in polyethylene glycol

Wei Liu<sup>1</sup>\*, Lige Xu<sup>1</sup>, Guanghui Lu<sup>1</sup>, Hua Zhang<sup>2</sup>\*

<sup>1</sup>College of Food Science and Technology, Henan University of Technology, Lianhua Street, Zhengzhou 450001, P. R. China
<sup>2</sup>College of Chemistry, Nanchang University, 999 Xuefu Avenue, Nanchang 330031, P. R. China
\*Corresponding author: Wei Liu and Hua Zhang
E-mail address: <u>liuwei307@hotmail.com; huazhang@ncu.edu.cn</u>

Total number of pages: 7 Total number of figures: 11 Total number of tables: 2



**Fig S1.** UV-vis spectra of Pd(OAc)<sub>2</sub>/PEG400 (A), Pd(OAc)<sub>2</sub>/PEG600 (B), Pd(OAc)<sub>2</sub>/PEG1000 (C), Pd(OAc)<sub>2</sub>/PEG2000 (D) and Pd(OAc)<sub>2</sub>/PEG4000 (E).

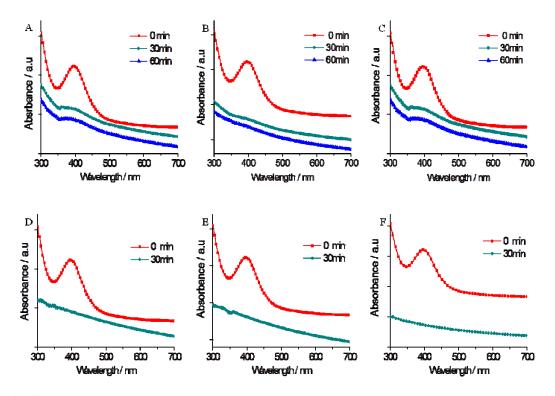
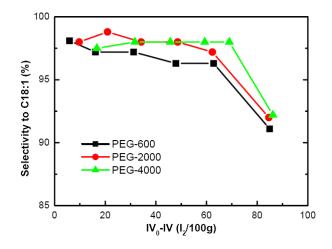


Fig S2. UV-vis spectra of Pd(OAc)<sub>2</sub>/PEG2000 and Pd(OAc)<sub>2</sub>/PEG4000 with different

## Pd concentration: (A) Pd/PEG2000=0.02mmol/g; (B) Pd/PEG2000=0.01mmol/g; (C) Pd/PEG2000=0.005mmol/g; (D) Pd/PEG4000=0.02mmol/g; (E) Pd/PEG4000=0.01mmol/g; (F) Pd/PEG4000=0.005mmol/g.



**Fig S3**. The methyl oleate selectivity with different PEG. (The selectivity= methyl oleate (mol)/converted methyl linoleate (mol);  $IV_0$  corresponded to the iodine value of methyl linoleate (172), IV corresponded to the iodine value of hydrogenated product.)

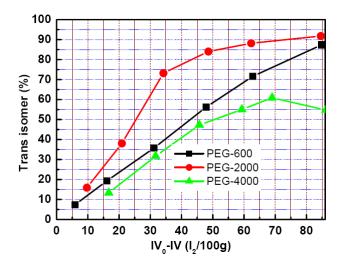


Fig S4. The *trans* isomer selectivity with different PEG. ( $IV_0$  corresponded to the iodine value of methyl linoleate (172), IV corresponded to the iodine value of hydrogenated product.)

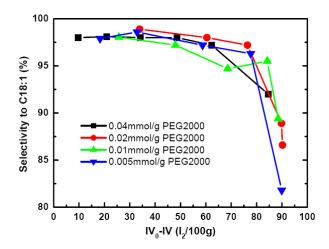


Fig S5. The methyl oleate selectivity at different Pd concentration in PEG2000.

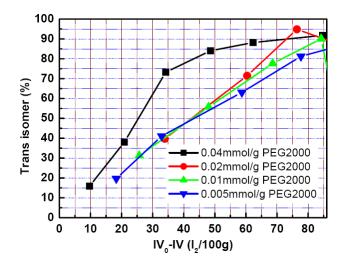


Fig S6. The trans isomer selectivity at different Pd concentration in PEG2000.

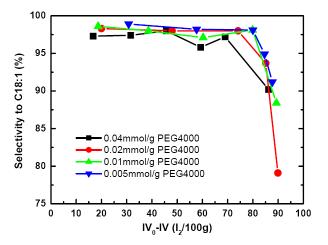


Fig S7. The methyl oleate selectivity at different Pd concentration in PEG4000.

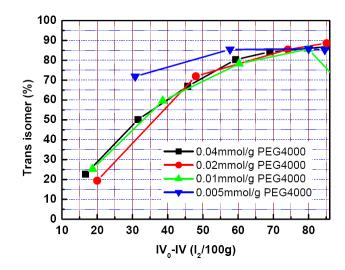


Fig S8. The trans isomer selectivity at different Pd concentration in PEG4000.

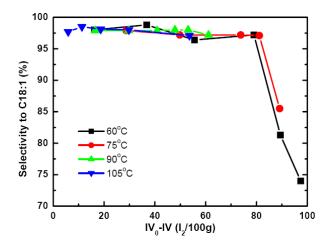


Fig S9. The methyl oleate selectivity at different reduction temperatures.

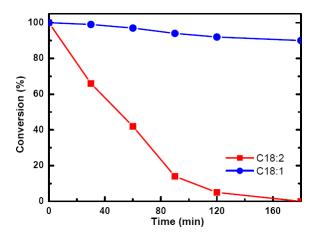


Fig S10. Hydrogenation rate of methyl oleate in comparison with methyl linoleate in the presence of Pd-PEG4000 catalyst.

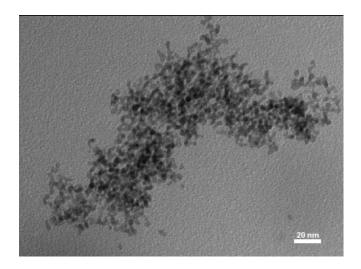


Fig S11. Pd-PEG4000 has be characterized by TEM after the recycling.

Table S1. Com	paration of presented	work and wetness	impregnation	for Pd catalysts
	paration of presented	in one and nothers	mpregnation	IOI I a catalysts

Nano-Pd catalyst	Reducing agent	Temp.& Time	Atmosphere	Lab instrument	Experimental skill	
Pd-PEGn	PEG	75°C, 0.5-1.0 h	Air	-	Middle	
Pd/SiO <sub>2</sub>	TT	Calcined at 300-400 °C for	$O_2$ for			
	H <sub>2</sub> ( <i>Handling</i>	2-3 h, and reduced at	calcination,	Tube	High	
	(Itanating carefully!)	250-300 °C for 2-4 h	$H_2$ for	furnace	nigii	
		under a H <sub>2</sub> flow.	reducing			

### Table S2. Comparison of *trans*-isomers produced in hydrogenation of FAMEs by

### different Pd catalysts.

Pd catalyst	Reaction conditions	Biodiesel	Trans isomers/C18:0 (%)	IV of hydrogenated product (I <sub>2</sub> /100g)	Reference
Pd/SiO <sub>2</sub>	80°C, <b>30</b> atm of H <sub>2</sub>	Rapeseed oil FAME	29.9%/8.5%	75	J Am Oil Chem Soc (2013) 90:1431-1438
Pd/ImS3-12@Al <sub>2</sub> O <sub>3</sub>	80°C, <b>75</b> atm of H <sub>2</sub>	Soybean oil FAME	33%/18%	65	Applied Catalysis A: General (2012) 433-434:

					109-114
				31	Applied
					Catalysis A:
Pd/C 5%	$80^{\circ}$ C, <b>75</b> atm of H <sub>2</sub>	Soybean oil FAME	24%/50%		General (2012)
					433-434:
					109-114
				81	
Pd(0)/BMI BF4	80°C, <b>10</b> atm of $H_2$		32%/5%		Catal. Sci.
		Soybean oil FAME			Technol.,
Pd(0)/BMI BF4	80°C, <b>75</b> atm of H <sub>2</sub>			75	(2011) 1:
			24%/11%		480-488
		Soybean oil FAME			
	75°C, <b>10</b> atm of $H_2$	Sunflower oil	24%/6%	77	
Pd-PEGn		FAME			This work